Supporting information

Liquid crystalline and charge transport properties of double-decker cerium phthalocyanine complexes

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Additional Figures and Tables:





Table S1.	Elemental	analyses	data	for a	ll homologues.

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Molecule	Molecular formula		Elemental analysis; Found % (calculated %)		
Wolecule		Molecular weight (g.mol)	С	Н	N
(1a) [(C ₈ H ₁₇) ₈ Pc] ₂ Ce	C192H288CeN16S16	3473.69	66.28 (66.39)	8.24 (8.36)	6.21 (6.45)
(1b) [(C ₁₀ H ₂₁) ₈ Pc] ₂ Ce	C ₂₂₄ H ₃₅₂ CeN ₁₆ S ₁₆	3922.55	68.72 (68.59)	9.05 (9.05)	5.67 (5.71)
(1c) [(C ₁₂ H ₂₅) ₈ Pc] ₂ Ce	C256H416CeN16S16	4371.42	70.19 (70.34)	9.60 (9.59)	5.10 (5.13)
(1d) [(C ₁₄ H ₂₉) ₈ Pc] ₂ Ce	C ₂₈₈ H ₄₈₀ CeN ₁₆ S ₁₆	4820.29	71.48 (71.76)	9.98 (10.04)	4.64 (4.65)
(1e) [(C ₁₆ H ₃₃) ₈ Pc] ₂ Ce	C320H544CeN16S16	5269.15	72.64 (72.94)	10.44 (10.41)	4.17 (4.25)
(1f) [(C ₁₈ H ₃₇) ₈ Pc] ₂ Ce	C352H608CeN16S16	5718.02	73.64 (73.94)	10.59 (10.72)	3.62 (3.92)

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 Table S2. XRD data of cerium phthalocyanine compounds 3a-3f.

Compound	Temperatures/ °C	Lattice constant/ Å	Spacing / Å		Miller indices (h,k,l)
			Observed	Calculated for Hex	Hex
[(C ₈ H₁ ₇ S) ₈ Pc]₂Ce (3a)			25.7	25.7	(100)
			14.4	14.8	(110)
	400	a = 29.6	12.4	12.8	(200)
	100		9.3	9.7	(210)
			8.2	8.8	(300)
			6.8	7.1	(310)
			6.1 4.2	6.4	(400) Halo of molten alkylthio chains
			27.4	27.4	(100)
	120	a = 31.7	15.4	15.8	(110)
			13.3	13.7	(200)
			10.0	10.4	(210)
			8.8	9.1	(300)
			7.3	7.6	(310)
			6.5	6.9	(400)
			4.3	00.0	Halo of molten alkylthio chains
[(C ₁₀ H ₂₁ S) ₈ Pc] ₂ Ce (3b)			29.0	29.0	(100)
			16.4	16.8	(110)
		22 F	14.1	14.5	(200)
		a = 33.5 b = 7.0	10.0	0.7	(210)
	23	c = 3.3	8.0	9.7	(300)
		0 0.0	7.7	7.3	(400)
			7.0		h
			4.3		Halo of molten alkylthio chains
			3.3		c
	160	a = 31.5	27.2	27.2	(100)
			15.7	15.7	(110)
			13.5	13.6	(200)
			10.3	10.3	(210)
			4.6		Halo of molten alkylthio chains
[(C ₁₂ H ₂₅ S) ₈ Pc] ₂ Ce (3c)			28.7	28.7	(100)
			16.6	16.5	(110)
	22	a = 33.1	14.3	14.3	(200)
	22	n = 7.0	10.9	10.8	(210)
		c - 3.4	7.0	5.0	(300) h
			4.4		Halo of molten alkylthio chains
			3.4		с
			31.2	30.3	(100)
	80	a = 35.0	17.3	17.5	(110)
			15.2	15.2	(200)
			10.0	11.5	(210)
			8.4	8.4	(310)
			7.5	7.6	(400)
			6.0	6.1	(500)
			4.7	21.4	Halo of molten alkylthio chains
[(C ₁₄ H ₂₉ S) ₈ Pc]₂Ce (3d)	23	a = 36.3 c = 3.3	32.1	31.4	(100)
			18.4	18.1	(110)
			10.7 11.8	15.7	(∠00) (210)
			10.5	10.5	(300)
			8.7	8.7	(310)
			7.9	7.9	(400)
			6.8	6.9	(410)
			0.3 4.4	0.3	(500) Halo of molten alkylthic chains
			4.3		Halo of molten alkylthio chains
			3.3	1	c

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(Continued) 33.4 32.4 (100) 18.5 18.7 (110) (200) (210) 16.2 16.2 80 a = 37.4 12.2 12.4 10.7 (300) 10.8 7.9 (400) 8.1 7.0 7.1 (410) Halo of molten alkylthio chains 4.6 33.6 33.0 (100) [(C₁₆H₃₃S)₈Pc]₂Ce (3e) 19.1 19.1 (110) 16.5 (200) 16.5 10.9 11.0 (300) a = 38.1 9.3 9.2 (310) 23 c = 3.3 8.2 8.3 (400) 7.2 6.6 7.3 (410) (500) Halo of molten alkylthio chains Halo of molten alkylthio chains 6.8 4.4 4.1 3.3 С 33.4 33.4 (100) 19.9 19.3 (110) (200) (210) 16.4 16.7 80 a = 38.6 12.6 12.6 11.0 11.1 (300) 8.4 (400) 8.1 4.6 Halo of molten alkylthio chains 33.7 33.7 (100) [(C18H37S)8Pc]2Ce (3f) 19.9 19.5 16.8 (110) (200) 16.7 12.7 12.7 (210) a = 38.9 11.1 11.2 (300) 25 c = 3.3 9.2 9.3 (310) 8.3 8.4 (400) 7.2 7.4 6.7 (410) (500) Halo of molten alkylthio chains 6.6 4.2 3.3 с

²⁰ h: Stacking periodicity between double-decker complexes.

C: Stacking periodicity between single-decker complexes.



-40 -30 -20 -10 0 10 20 30 40 50 60 70 80 90 100 110 120 130 140 150 160 170 180 190 200 Temperature (°C)

Fig. S3: DSC thermogram of 3c at 1 °C/min.



Fig. S4: DSC thermogram of 3d at 1 °C/min.



Fig. S5: DSC thermogram of 3e at 1 °C/min.

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Fig. S6: DSC thermogram of 3f at 1 °C/min.