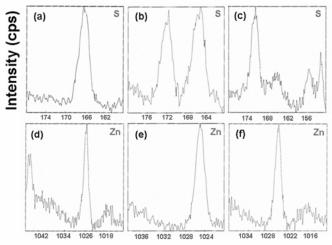
## **Supplementary Information for**

## Supramolecular Nanotubes with High Thermal Stability: A Rigidity Enhanced Structure Transformation Induced by Electron-Beam Irradiation and Heat

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Binding Energy (eV)

**Fig. S1** ESCA spectra acquired at specified binding energy (eV) for Zn and S of precursor zinc hexadecylxanthate (ZnHDX), half-reacted products (HRPs) and as-prepared nanotubes (NTs). (a), (b) and (c) correspond to the S binding-energy region of ZnHDX, HRPs and NTs, respectively; (d), (e) and (f) represent Zn signal of ZnHDX, HRPs and NTs, respectively. Supplementary Material (ESI) for Journal of Materials Chemistry This journal is The Royal Society of Chemistry 2007

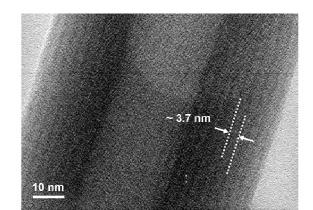
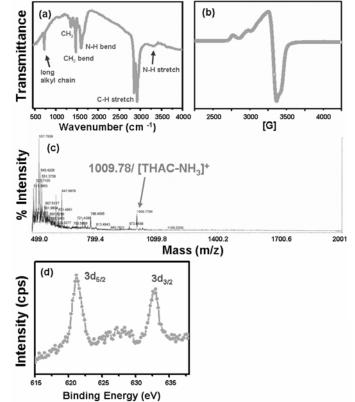


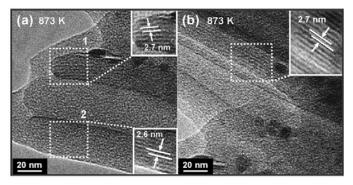
Fig. S2 TEM image of pristine nanotube. The multi-wall structure in the wall region has been characterized to have layer spacing of  $\sim 3.7$  nm.



**Fig. S3** FTIR, ESR, and MALDI-TOF mass spectra of tetrahexadecylamine copper (II) (THAC)-dioxygen complex. (a) FTIR spectra of THAC complex recorded with KBr medium; (b) Regular axial (frozen, 77K) ESR result of powder form THAC complex; (c) Positive-ion MALDI-Mass spectra with detected fragment ion [THAC-NH<sub>3</sub>]<sup>+</sup> at m/z 1009; (d) ESCA result of collected products from the reaction of NaI and THAC-dioxygen complex. Additive shift of iodine  $3d_{5/2}$  and  $3d_{3/2}$  photoelectron lines stands for the suggested oxidation process.

MALDI-TOF mass spectra (Figs. S2, c) shows fragment ion at m/z 1009, which is considered to present in the form of fragment ion [THAC-NH<sub>3</sub>]<sup>+</sup>. The active ESR signal (Fig. S2, b) acquired from powder form of resulted complex with calculated spectroscopic g factor of  $g_{\parallel}$  (2.28) and  $g_{\perp}$  (2.06) implicates an axial environment for the copper (II) ion, i.e.,  $g_{\parallel} > 2.1$  and  $g_{\perp} > 2.0$ .<sup>s1</sup> This can be expected for the ground state symmetry of copper (II)  $d^9$  ion is  $d_{x^2-y^2}^2$  and sustained the square-planar geometry of copper (II) ion with HDA molecules. Characteristic FTIR peaks (3260, 2956, 2919(s) 2851, 1596, 1468, 1352 and 721 cm<sup>-1</sup>) also inform the THAC complex formation (Fig. S2, a).<sup>s2</sup>

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**Fig. S4** Multi-wall nanotube structure formed during *in-situ* TEM examination at 873 K. (a) TEM image of multi-layer structured nanotube. Two insets show the fast Fourier-transform (FFT) images of area 1 and 2, respectively. The insets reveal the layer spacing is close to 2.7 nm; (b) Additional TEM and FFT (inset) images taken at 873 K of a nanotube.

## **References for supplementary information**

- (S1) B. Lucchese, K. J. Humphreys, D.-H. Lee, C. D. Incarvito, R. D. Sommer, A. L. Rheingold, K. D. Karlin, *Inorg. Chem.*, 2004, **43**, 5987.
- (S2) S. Muñoz, G. W. Gokel, Inorganica Chimica Acta., 1996, 250, 59.