

Supporting Information

Shape Controllable Synthesis and Upconversion Properties of NaYbF₄/NaYbF₄:Er³⁺ and YbF₃/YbF₃: Er³⁺ Microstructures

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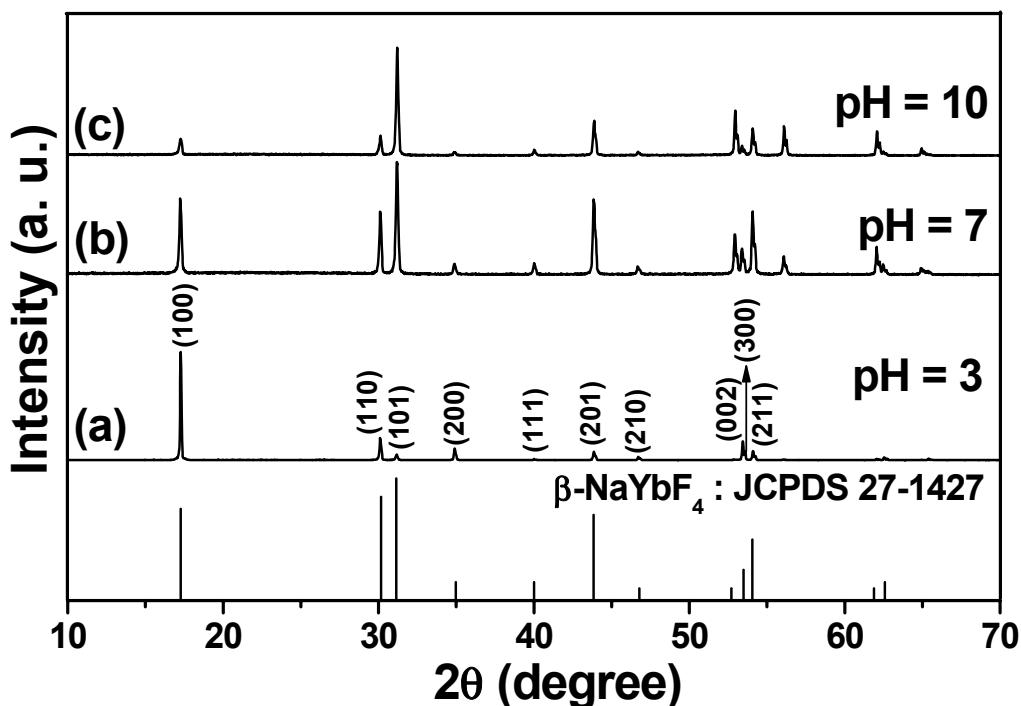


Fig. S1 The XRD patterns of the as-prepared β -NaYbF₄ products using NH₄F as fluoride source at 180 °C for 24 h at different pH values of (a) 3; (b) 7; (c) 10 and the standard data of β -NaYbF₄ (JCPDS card no. 27-1427).

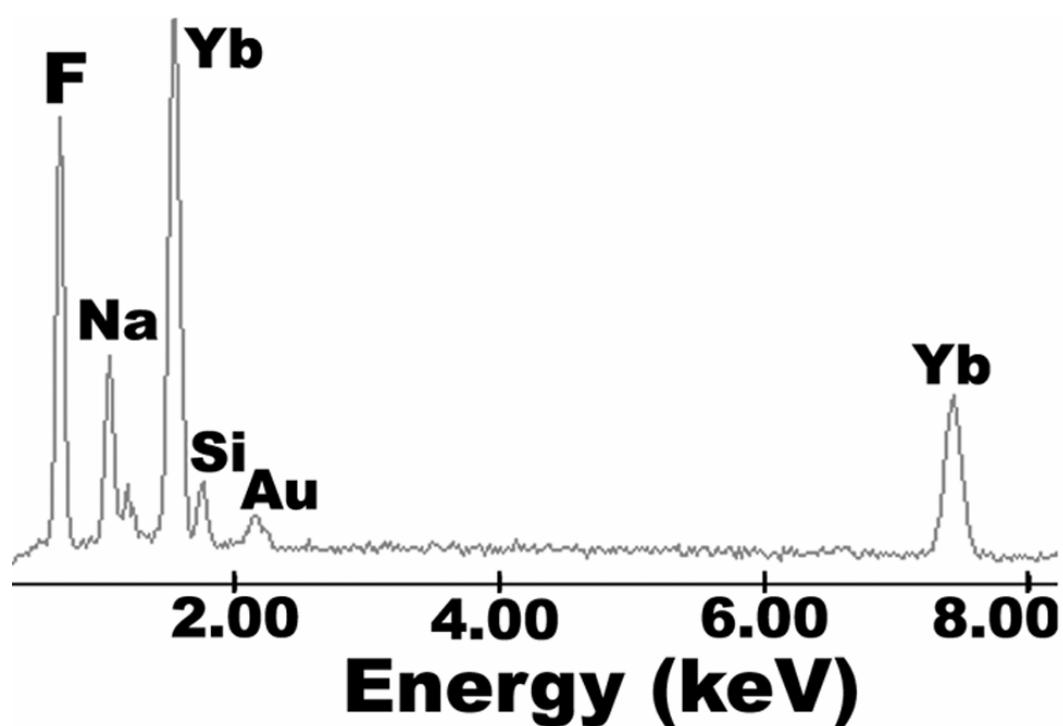


Fig. S2 EDX spectrum of β -NaYbF₄ irregular microprisms (**P1**), revealing the presence of Na, Yb and F.

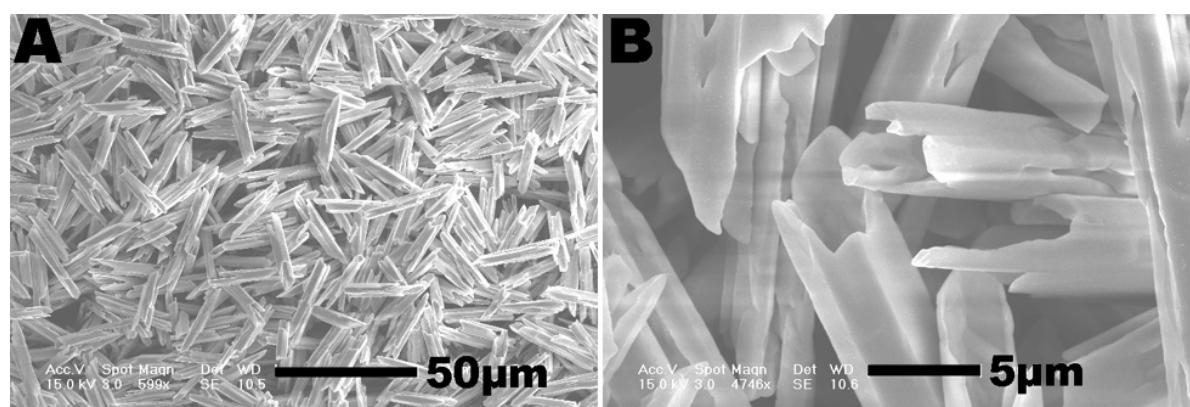


Fig. S3 SEM images of β -NaYbF₄ microtubes using NaBF₄ as fluoride source (pH = 10).

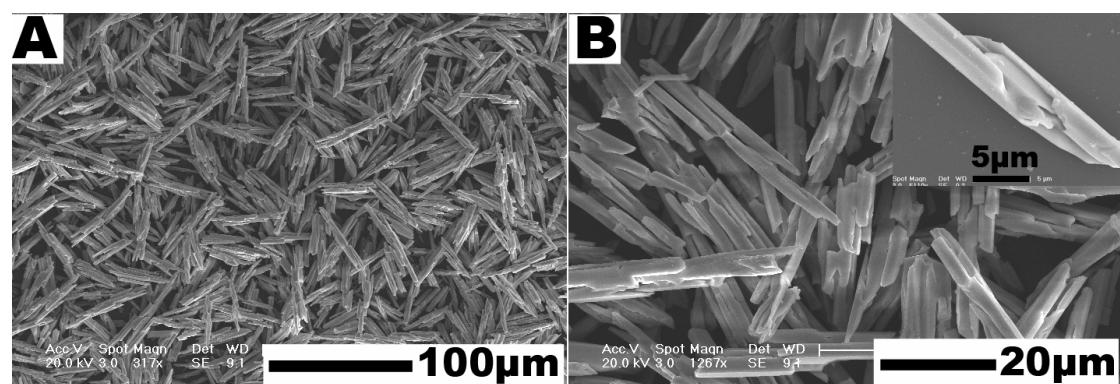


Fig. S4 SEM images of β -NaYbF₄ in the absence of Cit³⁺ and other conditions are similar to those preparing **P2**.

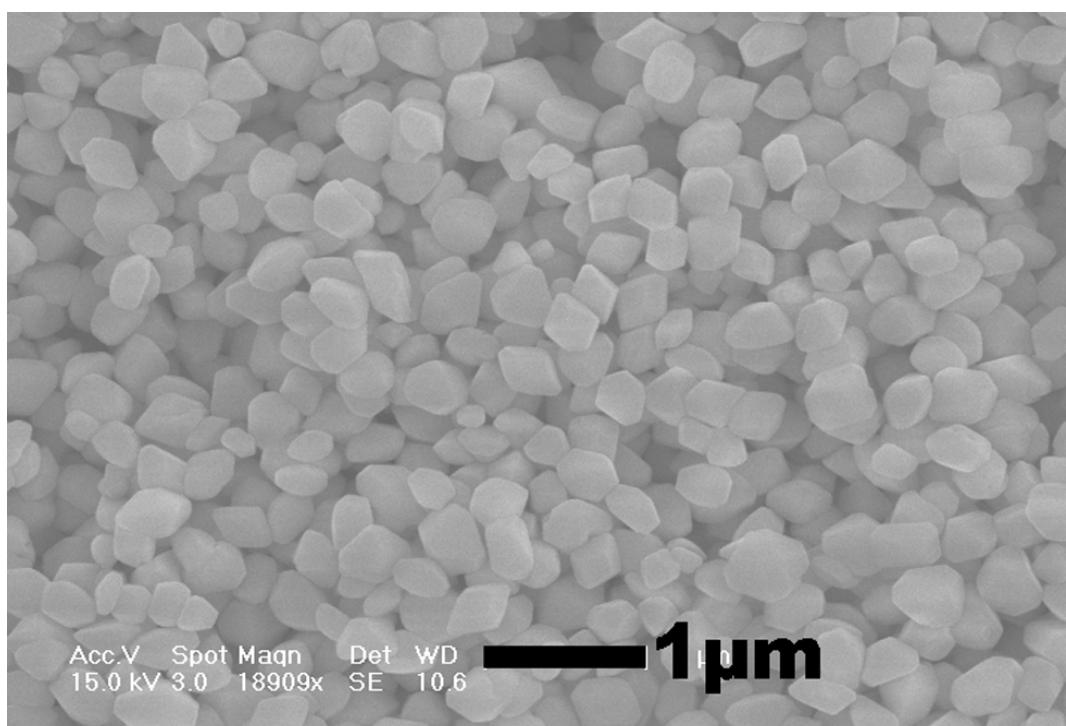


Fig. S5 SEM images of YbF₃ in the absence of Cit³⁺ and other conditions are similar to those preparing **P7**.

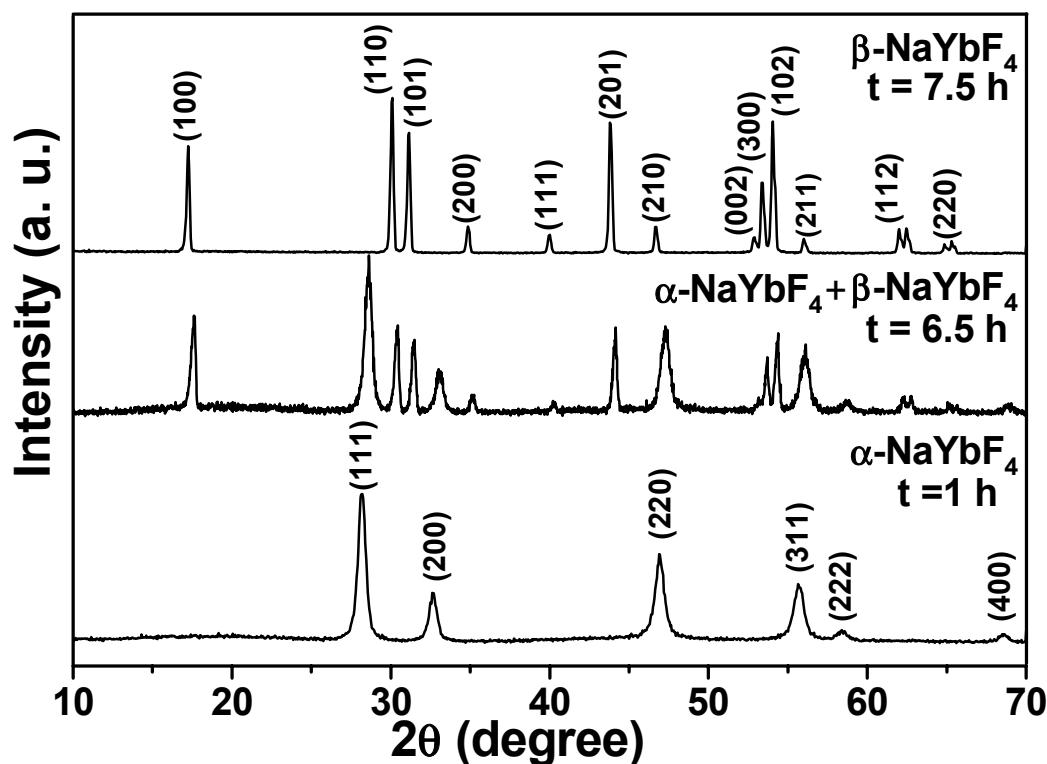


Fig. S6 XRD patterns of the as-prepared NaYbF_4 samples at 180 $^{\circ}\text{C}$ for different periods of the reaction time of (a) 1 h; (b) 6.5 h; (c) 7.5 h using NaBF_4 as fluoride source. These samples were prepared under the similar conditions for synthesizing P9.

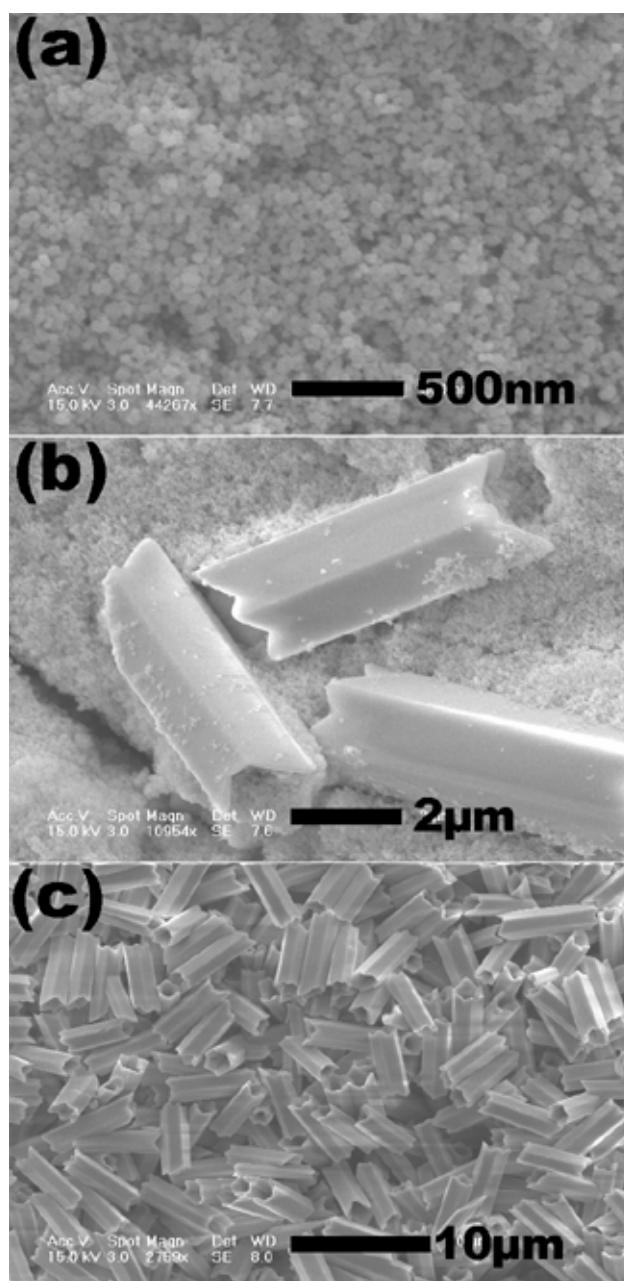


Fig. S7 SEM images of the as-prepared samples using NaBF₄ as fluoride source at 180 °C for different periods of the reaction time of (a) 1 h; (b) 6.5 h; (c) 7.5 h, revealing the morphological evolution process of the β-NaYbF₄ microtubes. These samples were prepared under the similar conditions for synthesizing **P9**.

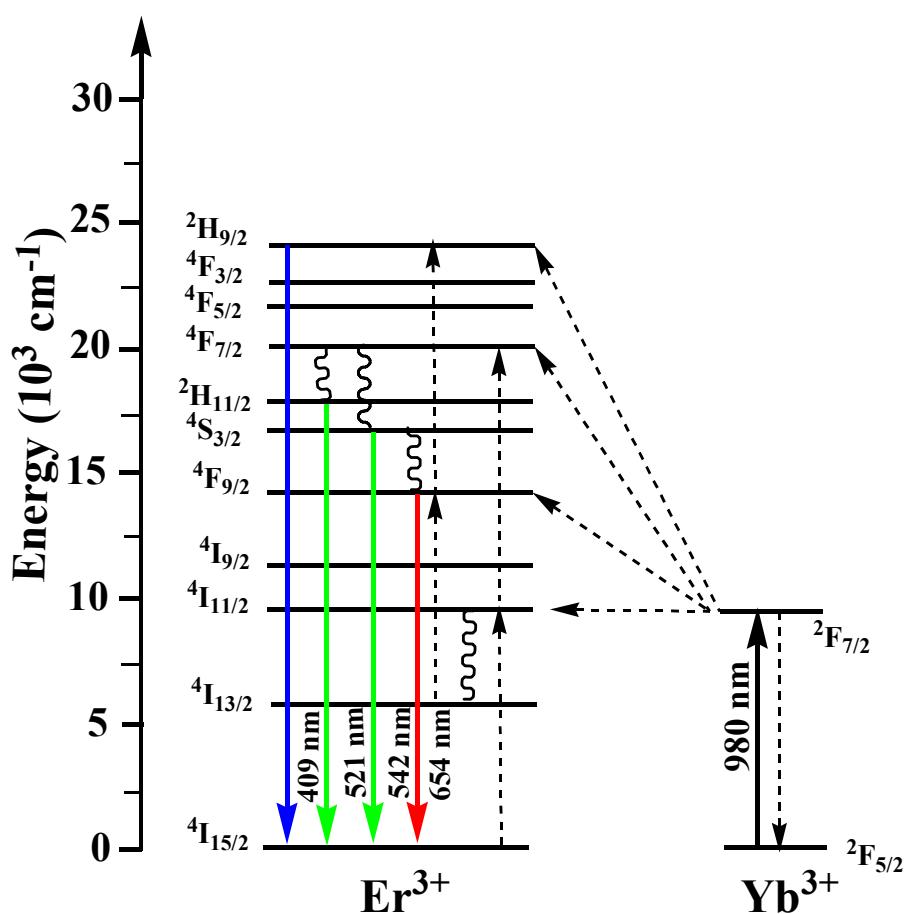


Fig. S8 A scheme showing the energy-level and up-conversion luminescence process for Er^{3+} doped ytterbium fluoride compounds. The full, dotted arrows, and curly lines represent emission, energy transfer, and multiphonon relaxation processes, respectively.