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**Supramolecular Electroactive Organogel and Conducting Nanofibers with
*C*₃-Symmetrical Architectures**

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SUPPORTING INFORMATION

Electrochemical studies: Cyclic voltammetry measurements were performed using a three-electrode cell equipped with a platinum millielectrode of 0.126 cm^2 area, an Ag/Ag^+ pseudo-reference and a platinum wire counter-electrode. The potential values were then re-adjusted with respect to the saturated calomel electrode (SCE), using the ferrocene as internal reference. The electrolytic media involved a 0.1 mol.L^{-1} solution of $(n\text{-Bu}_4\text{N})\text{PF}_6$ in CH_2Cl_2 . All experiments have been performed at room temperature at 0.1 V.s^{-1} . Experiments have been carried out with an EGG PAR 273A potentiostat with positive feedback compensation.

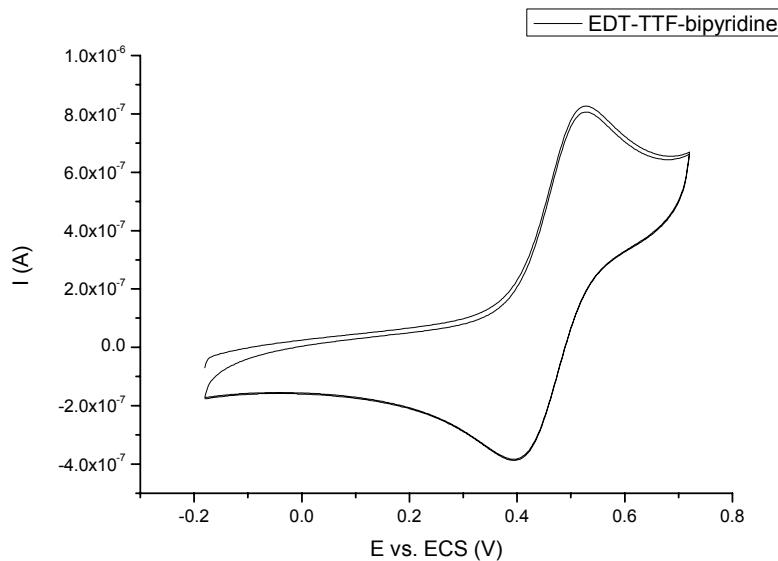


Fig. S1 Cyclic voltammetry of **2a**. First oxidation wave. $E = 0.48\text{ V}$.

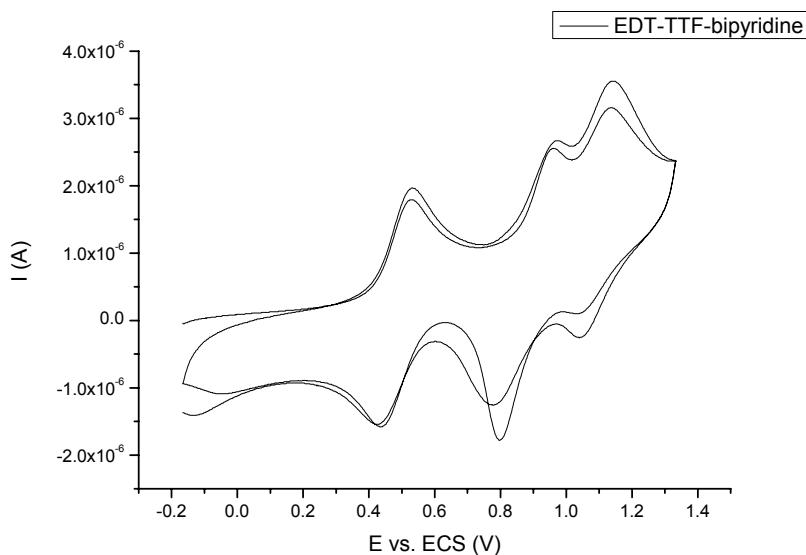


Fig. S2 Cyclic voltammetry of **2a**. $E_1=0.48\text{ V}$, $E_2=0.88\text{ V}$, $E_3=1.09\text{ V}$.

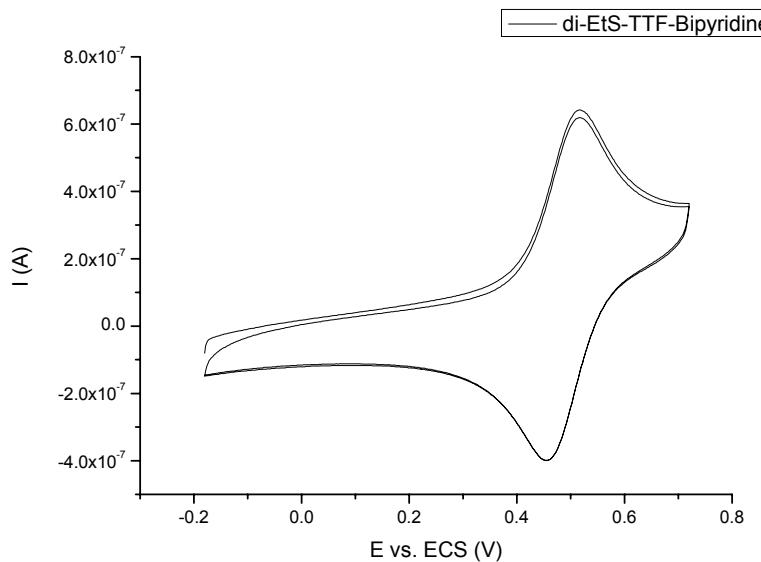


Fig. S3 Cyclic voltammetry of **2b**. First oxidation wave. E = 0.49 V.

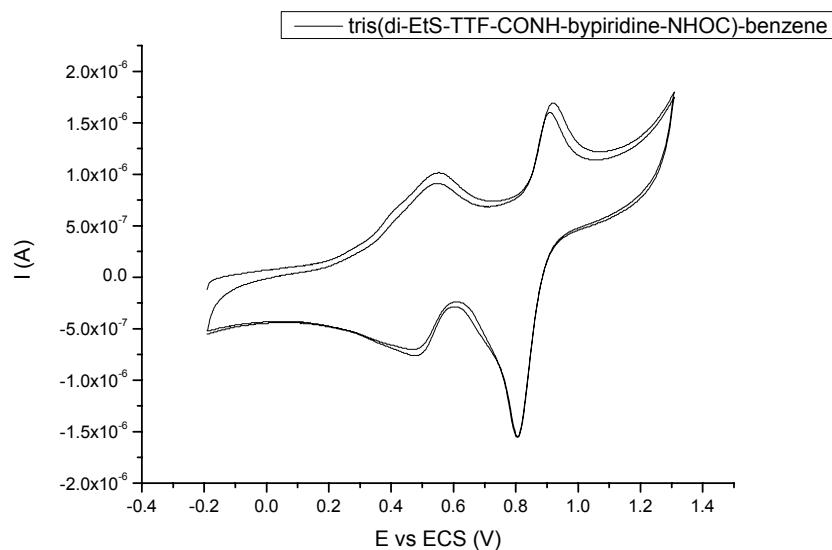


Fig. S4 Cyclic voltammetry of **4b**. E₁ = 0.515 V, E₂ = 0.86 V.