## Synthesis and Characterization of a Novel Kind of Near-Infrared Electrochromic Polymers Containing Anthraquinone Imide Group and Ionic Moieties

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5 Table S1: Solubility of the polymers

polymers <b>-</b>	solvents <sup>a</sup>						
	DMF	DMSO	NMP	DMAc	СН3ОН	H <sub>2</sub> O	MeEtIm <sup>+</sup> Br <sup>-</sup>
1a	++	++	++	++	-	-	±
1b	++	++	++	++	+	+	+
1c	+ +	+ +	+ +	+ +	+ +	++	+ +

*a* The qualitative solubility was tested with 1 mg of a sample in 1 mL of stirred solvent. ++, soluble at room temperature; +, soluble on heating;  $\pm$ , partially soluble; -, insoluble even on heating.



.a. UV-vis spectra of 3 in DMSO with varying concentration



20 Figure S1: UV-vis calibration curves for determining the AQI content in copolymers



a. UV-vis spectra of poly(ViEtIm \* Br - ) in DMSO with varying concentration



b. Calibration curve of the plots at 220 nm.

Figure S2: UV-vis calibration curves for determining the poly(ionic liquid) content in copolymers











5 Figure S4: GPC traces of the polymers (c=10 mg/mL in DMF)



Figure S6: Cyclic voltammograms of polymer **1b** in DMF and in film, potentials vs Ag/AgCl.



Figure S7: Cyclic voltammograms of polymer **1c** in DMF and in film, potentials vs Ag/AgCl.



25 Figure S9: UV-vis-NIR spectra of polymer 1b in film. ( THF containing 0.1 M TBAP )





Figure S10: UV-vis-NIR spectra of polymer 1c in DMF (5×10<sup>-3</sup> M) containing TBAP in its neutral, anionic, and dianionic states.



Figure S11: UV-vis-NIR spectra of polymer **1c** in film. (THF containing 0.1 M TBAP )

Figure S8: UV-vis-NIR spectra of polymer 1b in DMF (5×10<sup>-3</sup>
M) containing TBAP in its neutral, anionic, and dianionic states.

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Figure S12: Optical attenuation of film of polymer 1a at540 nm
on ITO glass as a function of applied potential with a switching time of 20 seconds and a stepping potential (0.4 V vs silver electrode).