Nonlinear Optical Activity in Dipolar Organic-

Lanthanide Complexes

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Supporting Information for publication

Synthesis of Complexes and Characterization by FT-IR and ESI-MS.

Figure S1 Selected SHG of lanthanide complexes (Ln = Dy, Er, Gd and Tb, $\lambda_{ex} = 1.15 \,\mu$ m). **Figure S2.** Powder XRD of the series of lanthanide complexes (Type I, Ln = Y, Ce, Pr, Nd, Eu and Gd;

Type II, Ln = Lu, Yb, Tm and Er) in the solid state. (Full powder pattern of Figure 9).

Figure S3. UV-vis absorption spectrum of cinnamic acid and its lanthanide complexes $(1.0 \times 10^{-5} \text{ M}, 10^{-5} \text{ M})$

MeOH).

Figure S4. ESI-MS spectra of Eu, Gd, Tb complexes.

Table S1. Crystallographic data collection, intensity measurements and structure refinement and

geometric parameters for $Ln[C_9H_7O_2)_3]_n$.

Table S2. Selected bond lengths (Å) of Ln = Eu 6 (Type I) and Ln = Er 12 (Type II) complexes.

Table S3. Selected bond angles (°) of Ln = Eu 6 (Type I) and Ln = Er 12 (Type II) complexes.

Synthesis of lanthanide complexes. [(*trans*-cinnamate)₃Ln]_n (Ln = La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Er, Ho, Yb, Tm, Lu and Y): Trans-cinnamic acid (300 mg) was used as the chelating ligand with various lanthanide nitrate salts (Aldrich, 99.99%) in the mole ratio 3.5 to 1. The mixtures were stirred for 12 hours in methanol at room temperature and the pale colored precipitates were filtered out and were re-crystallized with acetonitrile. The purified products were isolated as pale-colored crystals. Ln = La, 1: white powder; yield: 519 mg (44%). MS (ESI): m/z 581 (M⁺ + 2), 580 (M⁺ + 1), 579 (M^+) . Ln = Ce, 2: pale yellow powder; yield: 632 mg (51%). MS (ESI): m/z 582 $(M^+ + 2)$, 581 (M^+) + 1), 580 (M^+). Ln = Pr, **3**: pale green powder; yield: 706 mg (58%). MS (ESI): m/z 583 (M^+ + 2), $582 (M^+ + 1)$, $581 (M^+)$. Ln = Nd, 4: pale purple powder; yield: 711 mg (60%). MS (ESI): m/z 588 $(M^{+} + 3)$, 587 $(M^{+} + 2)$, 586 $(M^{+} + 1)$, 585 (M^{+}) , 584 $(M^{+} - 1)$, 583 $(M^{+} - 2)$, 582 $(M^{+} - 3)$. Ln = Sm, 5: white powder; yield: 603 mg (48%). MS (ESI): m/z 596 (M⁺ + 5), 595 (M⁺ + 4), 594 (M⁺ + 3), 593 $(M^+ + 2)$, 592 $(M^+ + 1)$, 591 (M^+) , 590 $(M^+ - 1)$, 589 $(M^+ - 2)$, 588 $(M^+ - 3)$, 587 $(M^+ - 4)$. Ln = Eu, 6: pale yellow powder; yield: 680 mg (57%). MS (ESI): m/z 594 (M⁺ + 1), 593 (M⁺), 592 (M⁺ -1), 591 (M^+ - 2). IR (KBr): 1533, 1504 cm⁻¹ asym v(COO⁻), 1448, 1402 cm⁻¹ sym v(COO⁻). Ln = Gd, 7: white powder; yield: 358 mg (30%). MS (ESI): $m/z \ 602 \ (M^+ + 3), \ 601 \ (M^+ + 2), \ 600 \ (M^+ + 3), \ 601 \ (M^+$ 1), 599 (M⁺), 598 (M⁺ - 1), 597 (M⁺ - 2), 596 (M⁺ - 3). IR (KBr): 1533, 1504 cm⁻¹ asym v(COO⁻), 1449, 1402 cm⁻¹ sym v(COO⁻). Ln = Tb, 8: white powder; yield: 530 mg (44%). MS (ESI): m/z 600 (M^+) , 599 $(M^+ - 1)$, 598 $(M^+ - 2)$. IR (KBr): 1533, 1504 cm⁻¹ asym v(COO⁻), 1448, 1404 cm⁻¹ sym $v(COO^{-})$. Ln = Dy, 9: white powder; yield: 579 mg (47%). MS (ESI): m/z 605 (M⁺ + 2), 604 (M⁺ + 1), 603 (M^+), 602 (M^+ - 1), 601 (M^+ - 2), 600 (M^+ - 3). Ln = Y, 10: white powder; yield: 501 mg (46%). MS (ESI): m/z 532 (M⁺ + 2), 531 (M⁺ + 1), 530 (M⁺). Ln = Ho, 11: pale pink powder; yield: 599 mg (48%). MS (ESI): m/z 606 (M⁺), 605 (M⁺ - 1), 604 (M⁺ - 2). Ln = Er, 12: pale pink powder; vield: 582 mg (47%). MS (ESI): $m/z 609 (M^+ + 1), 608 (M^+), 607 (M^+ - 1), 606 (M^+ - 2), 605 (M^+ - 2), 605 (M^+ - 1), 606 (M^+ - 2), 605 (M^+ - 2), 605$ 3). Ln = Tm, 13: white powder; yield: 609 mg (49%). MS (ESI): $m/z \ 611 \ (M^+ + 2), \ 610 \ (M^+ + 1),$ 609 (M⁺). Ln = Yb, 14: white powder; yield: 689 mg (55%). MS (ESI): m/z 617 (M⁺ + 3), 616 (M⁺ + 2), 615 $(M^+ + 1)$, 614 (M^+) , 613 $(M^+ - 1)$, 612 $(M^+ - 2)$, 611 $(M^+ - 3)$, 610 $(M^+ - 4)$; IR (KBr): 1533, 1504 cm⁻¹ asymv(COO⁻), 1448, 1402 cm⁻¹ sym v(COO⁻). Ln = Lu, 15: white powder; yield: 689 mg (55%). MS (ESI): m/z 616 (M^+ + 2), 615 (M^+ + 1) IR (KBr): 1534, 1504 cm⁻¹ asym v(COO⁻¹)), 1448, 1402 cm⁻¹ sym v(COO⁻). The crystal data of the 14 lanthanide complexes are listed in Table S1.



Figure S1 Selected SHG of lanthanide complexes (Ln = Dy, Er, Gd and Tb, $\lambda_{ex} = 1.15 \mu m$).



Figure S2. Powder XRD of the series of lanthanide complexes (Type I, Ln = Y, Ce, Pr, Nd, Eu and Gd; Type II, Ln = Lu, Yb, Tm and Er) in the solid state. (Full powder pattern of Figure 10).



Figure S3. UV-vis absorption spectrum of cinnamic acid and its lanthanide complexes $(1.0 \times 10^{-5} \text{ M}, \text{MeOH})$.



| | P | Ce | PN | Sm | Eu | Gd | Tb | > | DV | 우 | Ъ | Ē | ٩X | Lu |
|--|------------|-------------|---------------|---------------|------------|------------|------------|------------|----------------------|------------|----------------------|----------------------|----------------------|------------|
| F. W. | 580.36 | 581.58 | 585.7 | 591.86 | 593.42 | 598.71 | 600.38 | 530.37 | 603.96 | 606.39 | 608.72 | 610.39 | 614.50 | 616.43 |
| T/°C | 301 | 301 | 301 | 301 | 303 | 301 | 303 | 301 | 301 | 301 | 301 | 301 | 301 | 301 |
| Crystal system | Trigonal | Trigonal | Trigonal | Trigonal | Trigonal | Trigonal | Trigonal | Monoclinic | Monoclinic | Monoclinic | Monoclinic | Monoclinic | Monoclinic | Monoclinic |
| Space Group | R3c (#161) | R3c (#161) | R3c (#161) | R3c (#161) | R3c (#161) | R3c (#161) | R3c (#161) | P21 (#4) | P2 ₁ (#4) | P21 (#4) | P2 ₁ (#4) | P2 ₁ (#4) | P2 ₁ (#4) | P21 (#4) |
| a /Å | 22.636(2) | 22.646(2) | 22.657(3) | 22.646(2) | 22.649(4) | 22.654(6) | 22.647(4) | 11.543(10) | 11.520(11) | 11.544(1) | 11.548(1) | 11.564(1) | 11.574(2) | 11.598(2) |
| ¢/q | 22.636(2) | 22.646(2) | 22.657(3) | 22.646(2) | 22.649(4) | 22.654(6) | 22.647(4) | 7.918(7) | 7.936(7) | 7.917(9) | 7.886(1) | 7.870 (9) | 7.848(2) | 7.817(2) |
| c /Å | 7.7554(10) | 7.9212(7) | 7.830(1) | 7.7670(11) | 7.744(1) | 7.722(2) | 7.703(1) | 13.062(11) | 13.078(12) | 13.071(2) | 13.048(1) | 13.042(2) | 13.043(3) | 13.021(3) |
| α/° | 06 | 06 | 06 | 06 | 06 | 06 | 06 | 06 | 06 | 06 | 06 | 06 | 06 | 06 |
| β/° | 06 | 06 | 06 | 06 | 06 | 06 | 06 | 94.139(2) | 94.049(2) | 94.136(2) | 94.15(1) | 94.280(2) | 94.384(3) | 94.354(3) |
| ۰/٨ | 120 | 120 | 120 | 120 | 120 | 120 | 120 | 06 | 6 | 06 | 06 | 06 | 06 | 06 |
| V / Å ³ | 3441.4(6) | 3518.2(5) | 3481.3(9) | 3449.5(7) | 3440.3(10) | 3432.0(16) | 3421.5(10) | 1190.8(2) | 1192.7(2) | 1191.5(3) | 1185.1(2) | 1183.6(2) | 1181.3(4) | 1176.1(4) |
| Z | 9 | 9 | 9 | 9 | 9 | 9 | 9 | 2 | 2 | 2 | 2 | 2 | 2 | 2 |
| d _{calc} / g cm ⁻³ | 1.680 | 1.647 | 1.676 | 1.709 | 1.718 | 1.738 | 1.748 | 1.479 | 1.682 | 1.690 | 1.706 | 1.713 | 1.727 | 1.740 |
| µ / mm-1 | 1.896 | 1.981 | 2.277 | 2.601 | 2.770 | 2.948 | 3.136 | 2.492 | 3.117 | 3.364 | 3.572 | 3.780 | 3.999 | 4.2378 |
| F(000) | 1728 | 1734 | 1746 | 1758 | 1764 | 1770 | 1776 | 540 | 594 | 596 | 598 | 009 | 602 | 604 |
| | 0.01X0.02X | 0.06X0.07X0 | 0.02X0.04X0.4 | 0.05X0.07X0.3 | 0.08X0.18X | 0.05X0.06X | 0.04X0.05X | 0.08X0.17X | 0.15X0.21X | 0.13X0.14X | 0.12X0.16X | 0.06X0.17X | 0.08X0.17X | 0.02x0.03x |
| urystal size / mm | 0.48 | .41 | - | 9 | 0.42 | 0.48 | 0.48 | 0.38 | 0.41 | 0.30 | 0.44 | 0.35 | 0.38 | 0.48 |
| No. of unique refins. (Total) | 6925 | 7227 | 7574 | 6304 | 6841 | 6917 | 2000 | 7456 | 6882 | 7328 | 6869 | 7338 | 7370 | 7162 |
| No. of obsd refins. | 3011 | 2011 | 1036 | | 1CDE | 1704 | CU21 | ONED | 0140 | 0020 | 1030 | 0000 | r r J G | 1764 |
| (>2ơ(l)) | 0.041 | 1430 | 0621 | 1001 | 0001 | 10/1 | CD / I | 74.00 | 6112 | 0017 | 1007 | 0707 | 4402 | 10/6 |
| R₁ (I>2ơ(I)) | 0.031 | 0.018 | 0.015 | 0.015 | 0.019 | 0.018 | 0.035 | 0.029 | 0.019 | 0.21 | 0.022 | 0.020 | 0.030 | 0.033 |
| wR (all data) | 0.034 | 0.021 | 0.017 | 0.018 | 0.024 | 0.022 | 0.038 | 0.028 | 0.023 | 0.26 | 0.027 | 0.021 | 0.037 | 0.041 |
| Goodness of fit/S | 1.034 | 1.006 | 1.067 | 1.054 | 1.054 | 1.043 | 1.081 | 1.024 | 1.014 | 1.038 | 1.024 | 1.046 | 1.029 | 1.001 |
| Max. shift/error | 0.043 | <0.001 | 0.013 | 0.028 | 0.031 | 0.021 | 0.003 | <0.0001 | 0.015 | 0.028 | 0.034 | 0.00 | 0.023 | 0.012 |

Table S1. Crystallographic data collection intensity measurements and structure refinement and geometric parameters for I $n[C_0H_2O_2]^2$

Table S1. (cont)

| | La | Ce | Ŋ | Sm | Eu | Gd | Тb | ٢ | Dy | Ю | Er | Tm | ٩ | Lu |
|-------------------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|------------------------|--------------------------------|--------------------------------|--------------------------------|-----------------------------------|--------------------------------|---------------------------------|---------------------------------|
| Coordination | 8 | 6 | 6 | 6 | 6 | 6 | 6 | 7 | 7 | 7 | 7 | 7 | 7 | 7 |
| À / MM | 3.878(4) | 3.961 | 3.915(1) | 3.884(1) | 3.872(2) | 3.861(1) | 3.852(3) | 4.124(1) | 4.135(1) | 4.122(2) | 4.107(2) | 4.096(1) | 4.083(3) | 4.067(2) |
| M0/Å | 2.397(5) - 2.625(4) | 2.466(2) - 2.678(2) | 2.429(3) – 2.644(3) | 2.397(7) – 2.633(3) | 2.389(3) - 2.633(3) | 2.377(4) - 2.624(3) | 2.359(8) – 2.628(6) | 2.236(3) - 2.478(3) | 2.254(3) - 2.489(3) | 2.247(4)- 2.474(4) | 2.228(4)- 2.465(4) | 2.221(4)- 2.453(4) | 2.209(7)- 2.443(7) | 2.190(6)- 2.445(7) |
| Dihedral angle / ° (COOM) vs C=C | 9.3(7) | 10.2(3) | 10.4(3) | 10.4(4) | 10.0(5) | 9.8(4) | 10.0(9) | 2.2(6) 11.7(2), 13.0(4) | 1.6(8), 11.5(3), 12.6(5) | 2.7(7), 12.3(3), 13.1(6) | 24.9(8), 105.2(5), 148.1(3) | 4.0(8), 12.0(2), 13.7(5) | 2.8(18), 13.6(4), 14.5(9) | 4.2(16), 11.6(5), 31.6(5) |
| Dihedral angle / ° (COOM) vs—Ph | 25.4(2) | 24.9(1) | 25.4(1) | 25.8(1) | 25.8(2) | 26.1(2) | 26.3(4) | 8.7(1), 15.6(1), 30.3(2) | 7.9(2), 9.3(2), 28.9(2) | 8.2(2), 14.9(2), 30.2(2) | 30.6(2), 94.5(5),145 .9(2) | 8.1(2), 16.1(2), 30.4(2) | 8.6(3), 17.6(3), 31.0(4) | 8.5(3), 16.0(3), 34.2(3) |

Supplementary Material (ESI) for Journal of Materials Chemistry This journal is (c) The Royal Society of Chemistry 2010 **Table S2.** Selected bond lengths (Å) of Ln = Eu 6 (Type I) and Ln = Er 12 (Type II) complexes.

| Eu | | E | r |
|-------------------------|----------|--|----------|
| Eu(1)-O(1) | 2.633(3) | Er(1)-O(1) | 2.265(5) |
| $Eu(1)-O(1)^{a}$ | 2.633(3) | Er(1)-O(2) | 2.242(5) |
| $Eu(1)-O(1)^{b}$ | 2.633(3) | Er(1)-O(3) | 2.302(4) |
| Eu(1)-O(1) ^c | 2.443(3) | $Er(1)-O(3)^{a}$ | 2.465(4) |
| $Eu(1)-O(1)^{e}$ | 2.443(3) | Er(1)-O(4) | 2.360(5) |
| Eu(1)-O(2) | 2.389(3) | Er(1)-O(5) | 2.228(4) |
| $Eu(1)-O(2)^{a}$ | 2.389(4) | Er(1)-O(6) | 2.250(5) |
| Eu(1)Eu(1) ^d | 3.872(2) | $\frac{\operatorname{Er}(1)}{\operatorname{Er}(1)^{\mathrm{a}}}$ | 4.107(2) |

Symmetry code of **6**: a = -Y,+X-Y,+Z; b = -X+Y,-X,+Z; c = -X+Y,+Y,1/2+Z; d = +X,+X-Y,1/2+Z; e= -Y,-X,1/2+Z. Symmetry code of **12**: a = -X,1/2+Y,-Z.

| Eu | | Er | |
|-------------------------------------|-----------|------------------------|----------|
| $O(1)^{a}$ -Eu(1)-O(1) | 65.26(9) | O(2)-Er(1)-O(1) | 154.1(2) |
| $O(1)^{b}$ -Eu(1)-O(1) | 65.26(8) | O(3)-Er(1)-O(1) | 77.2(2) |
| $O(1)^{c}$ -Eu(1)-O(1) | 124.75(9) | $O(3)^{a}$ -Er(1)-O(1) | 131.7(2) |
| $O(1)^{d}$ -Eu(1)-O(1) | 102.92(9) | O(4)-Er(1)-O(1) | 77.8(2) |
| $O(1)^{e}$ -Eu(1)-O(1) | 161.3(1) | O(5)-Er(1)-O(1) | 92.6(2) |
| $O(1)^{d}$ -Eu(1)-O(1) ^c | 71.1(1) | O(6)-Er(1)-O(1) | 93.1(2) |
| O(2)-Eu(1)-O(1) | 51.4(1) | O(3)-Er(1)-O(2) | 76.8(2) |
| O(2)-Eu(1)-O(1) ^c | 73.9(1) | $O(3)^{a}$ -Er(1)-O(2) | 74.2(2) |
| $O(2)^{a}$ -Eu(1)-O(1) | 116.1(1) | O(4)-Er(1)-O(2) | 128.0(2) |
| $O(2)^{b}$ -Eu(1)-O(1) | 80.7(1) | O(5)-Er(1)-O(2) | 87.5(2) |
| O(2)-Eu(1)-O(1) ^e | 140.3(1) | O(6)-Er(1)-O(2) | 89.0(2) |
| $Eu(1)-O(1)-Eu(1)^{c}$ | 99.34(8) | $Er(1)-O(3)-Er(1)^{a}$ | 118.9(2) |

Table S3. Selected bond angles (°) of Ln = Eu 6 (Type I) and Ln = Er 12 (Type II)complexes.

Symmetry code of **6**: a = -Y,+X-Y,+Z; b = -X+Y,-X,+Z; c = -X+Y,+Y,1/2+Z; d = +X,+X-Y,1/2+Z; e= -Y,-X,1/2+Z. Symmetry code of **12**: a = -X,1/2+Y,-Z.