

## **Supporting Information**

“Versatile Approach to Synthesis of 2-D Hexagonal Ultra-Large-Pore Periodic Mesoporous Organosilicas”

**Manik Mandal and Michal Kruk**

Center for Engineered Polymeric Materials, Department of Chemistry, College of Staten Island, City University of New York, 2800 Victory Boulevard, Staten Island, New York 10314, and Graduate Center, City University of New York, 365 Fifth Avenue, New York, New York 10016

**Supporting Table S1.** The interplanar spacing and other physicochemical properties of PMOs.

| Bridging group, initial temperature, amount of precursor | $d_{100}$<br>(nm) | $S_{BET}$<br>( $m^2/g$ ) | $V_t$<br>( $cm^3/g$ ) | $V_{mic}$<br>( $cm^3/g$ ) | $w_{BJH}$<br>(nm) |
|--|-------------------|--------------------------|-----------------------|---------------------------|-------------------|
| Methylene, 15°C, x=0.2                                   | 17.8              | 1197                     | 1.28                  | 0.26                      | 12.0              |
| Methylene, 15°C, x=0.3, H 100°C <sup>a</sup>             | 20.8              | 1230                     | 1.61                  | 0.28                      | 22.2              |
| Methylene, 15°C, x=0.3, H 130°C                          | 22.7              | 1007                     | 1.75                  | 0.11                      | 25.1              |
| Methylene, 15°C, x=0.3, H 150°C                          | 21.8              | 837                      | 1.66                  | (-0.01) <sup>f</sup>      | 23.6              |
| Methylene, 15°C, x=0.4                                   | 17.2              | 1204                     | 1.10                  | 0.35                      | 12.6              |
| Ethylene, 15°C, x=0.4, BTME                              | 16.1              | 429                      | 0.65                  | 0.08                      | 13.7              |
| Ethylene, 15°C, x=0.4, BTEE                              | 15.8              | 542                      | 0.67                  | 0.10                      | 11.6              |
| Ethylene, 12°C, x=0.4, BTEE                              | 18.2              | 258                      | 0.40                  | 0.05                      | 15.8              |
| Ethylene, 11°C, x=0.4, BTEE                              | 20.8              | 82                       | 0.15                  | 0.02                      | 18.8              |
| Ethylene, 15°C, x=0.7, BTME                              | 16.8              | 534                      | 0.56                  | 0.16                      | 13.8              |
| Ethylene, 15°C, x=1.0, BTME <sup>b</sup>                 | 16.7              | 690                      | 0.51                  | 0.25                      | 13.4              |
| Ethylene, 15°C, x=1.5, BTME                              | 18.0              | 581                      | 0.41                  | 0.22                      | 14.6              |
| Ethenylene, 15°C, x=0.7                                  | 14.0              | 139                      | 0.27                  | 0.00                      | 11.1              |
| Phenylene, 17°C <sup>c,d</sup>                           | 15.9              | 642                      | 0.56                  | 0.22                      | 12.7              |
| Phenylene, 16°C <sup>b,d</sup>                           | 18.2              | 514                      | 0.49                  | 0.16                      | 14.8              |
| Phenylene, 15.5°C <sup>c,d</sup>                         | 19.6              | 620                      | 0.50                  | 0.23                      | 16.4              |
| Phenylene, 15°C <sup>b,d</sup>                           | 22.7 <sup>e</sup> | 553                      | 0.52                  | 0.19                      | 21.1              |

Notation:  $d_{100}$ , (100) interplanar spacing for either calcined or extracted sample.  $S_{BET}$ , BET specific surface area;  $w_{BJH}$ , BJH pore diameter;  $V_t$ , total pore volume,  $V_{mic}$ , micropore volume,  $V_{mes}$ , mesopore volume.

<sup>a</sup> Repeated synthesis. Data for other sample listed in Table 1.

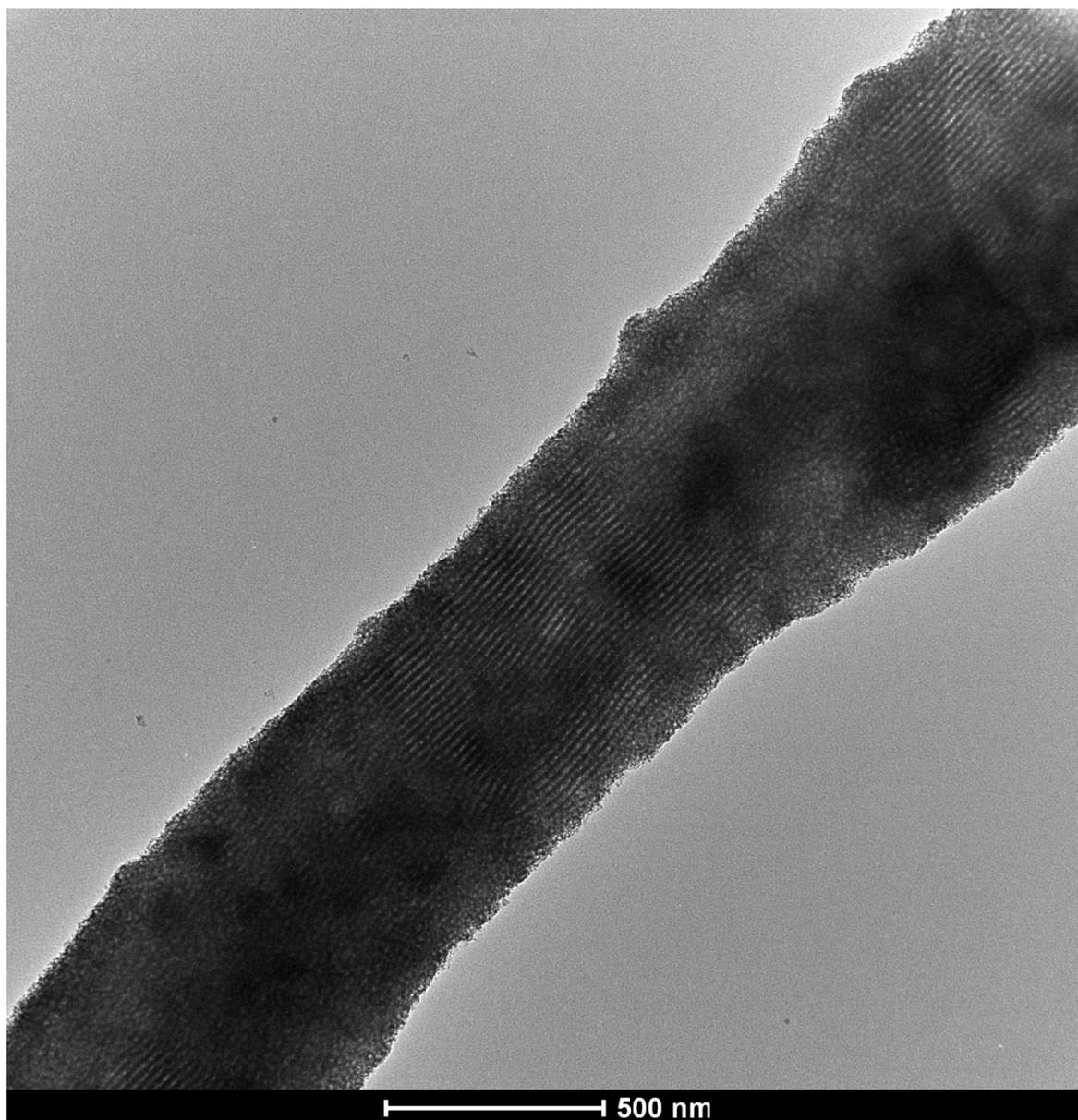
<sup>b</sup> Sample was calcined at 300 °C under nitrogen.

<sup>c</sup> Sample was extracted with ethanol and calcined at 250 °C under air.

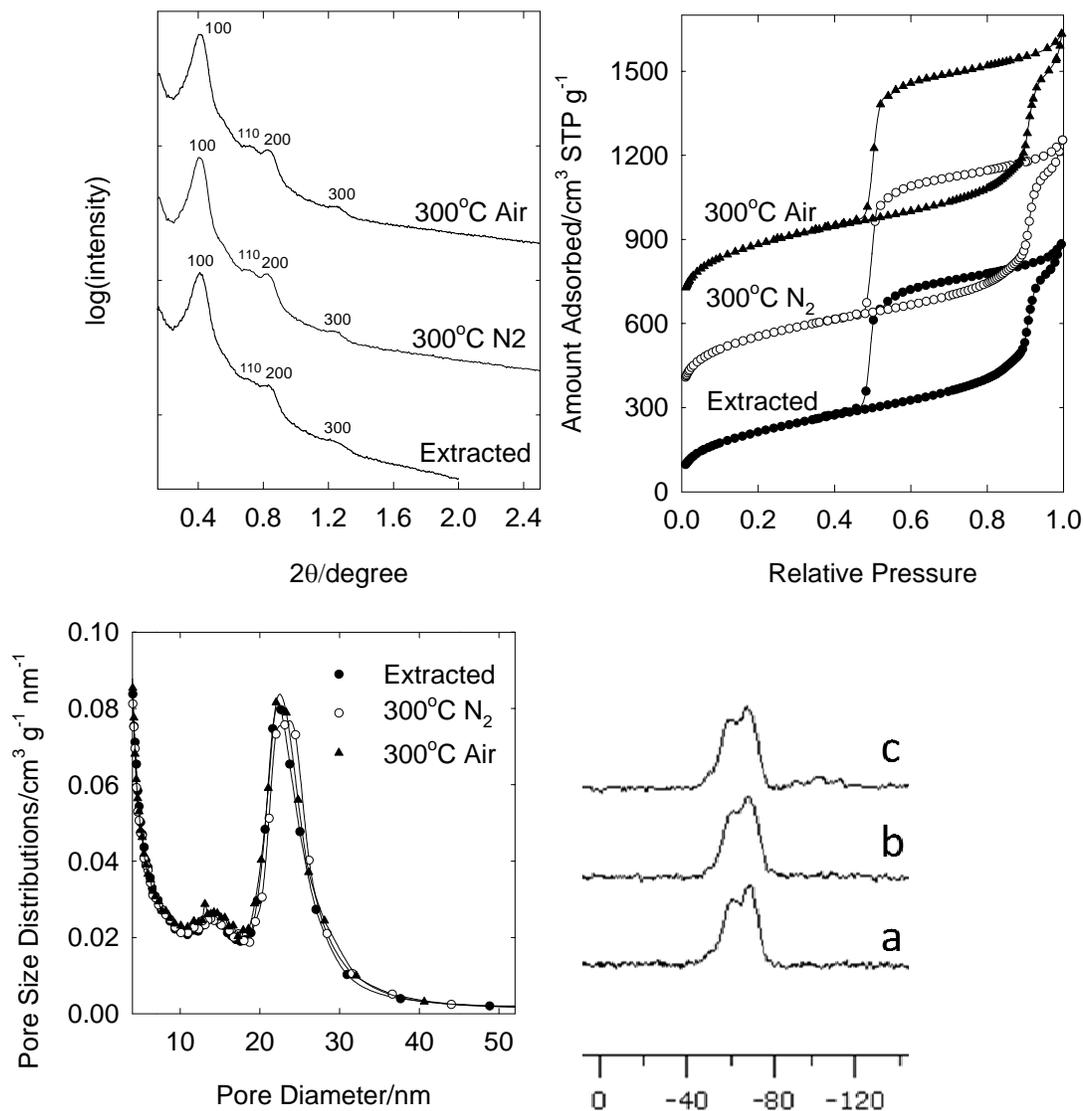
<sup>d</sup> Synthesized using 0.6 mL TIPB per 0.6 g of P123.

<sup>e</sup> Calculated assuming that the first peak is (100) reflection of 2-D hexagonal structure, which is consistent with TEM observation.

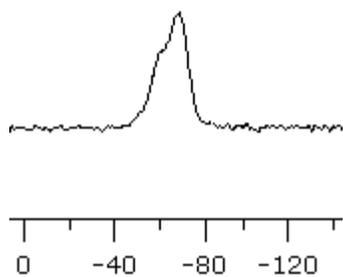
<sup>f</sup> Indicates that the micropore volume is negligible.



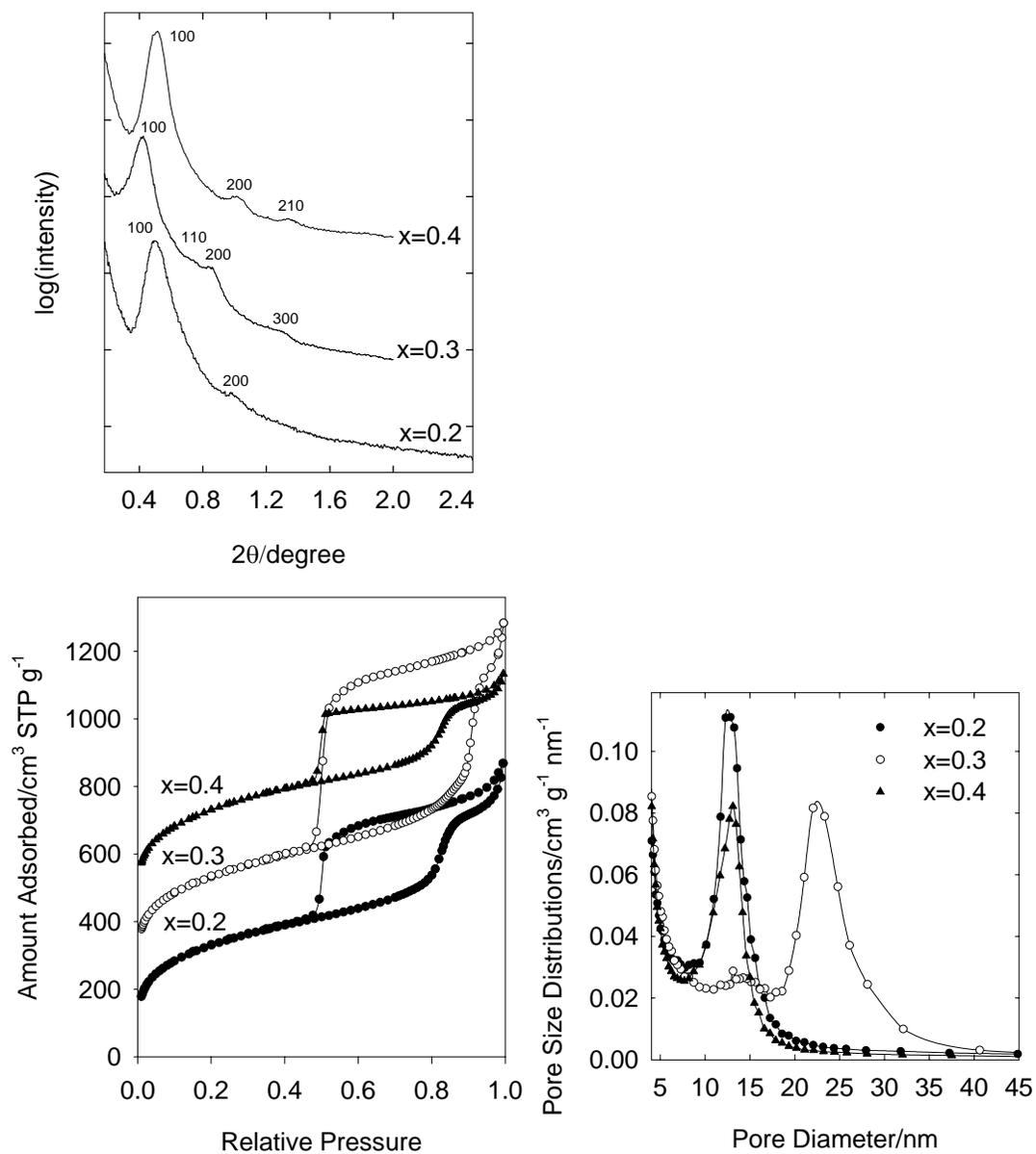
**Supporting Figure S1.** TEM image of calcined methylene-bridged PMO synthesized using a hydrothermal treatment at 150 °C for two days.



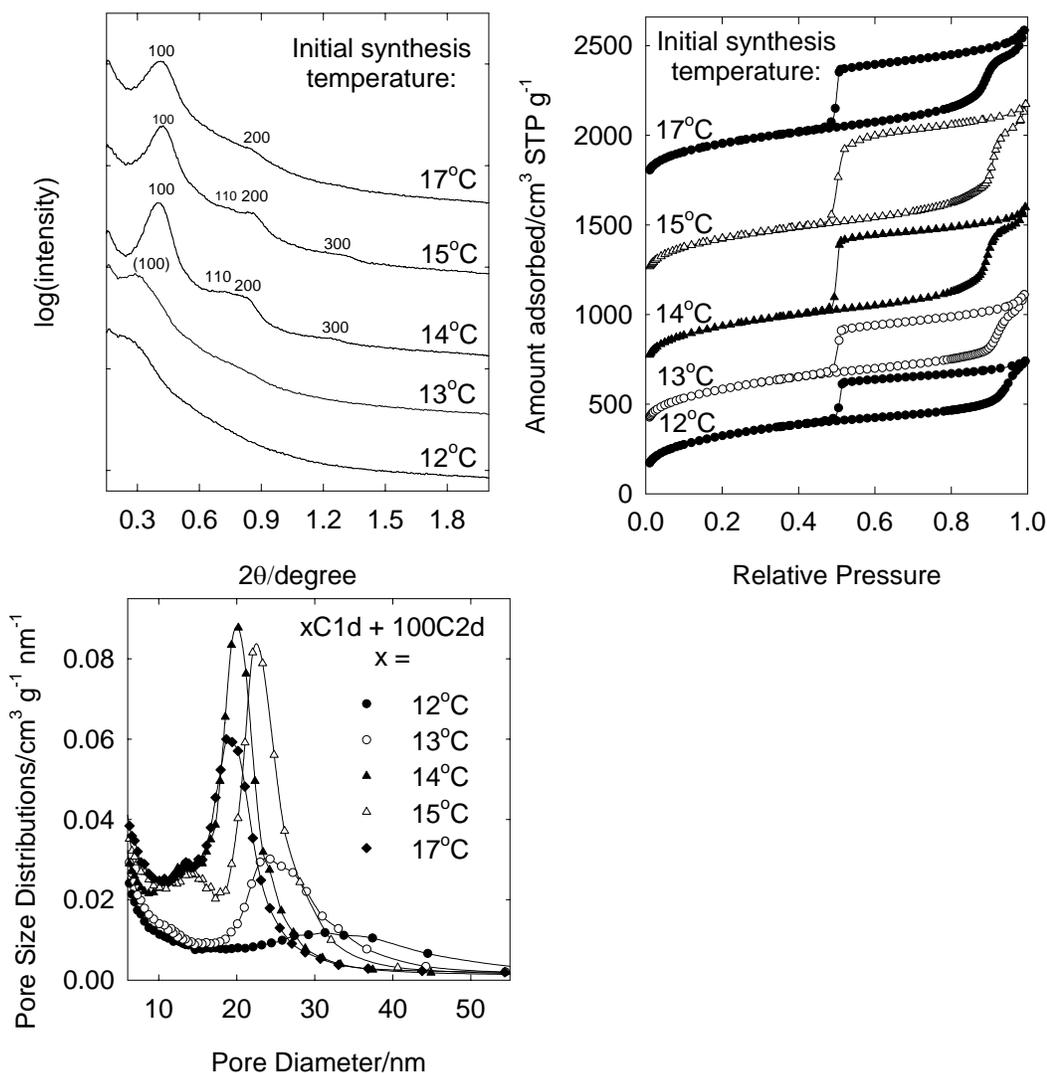
**Supporting Figure S2.** SAXS patterns (top left), nitrogen adsorption isotherms (top right), pore size distributions (bottom left) for methylene-bridged PMOs. Isotherms for samples calcined at 300 °C under N<sub>2</sub> and under air were offset vertically by 250 and 550 cm<sup>3</sup> STP g<sup>-1</sup> for clarity. <sup>29</sup>Si CP MAS NMR (bottom right) for methylene-bridged PMOs a) as synthesized, b) calcined at 300 °C under nitrogen, c) calcined at 300 °C under air (scale in ppm).



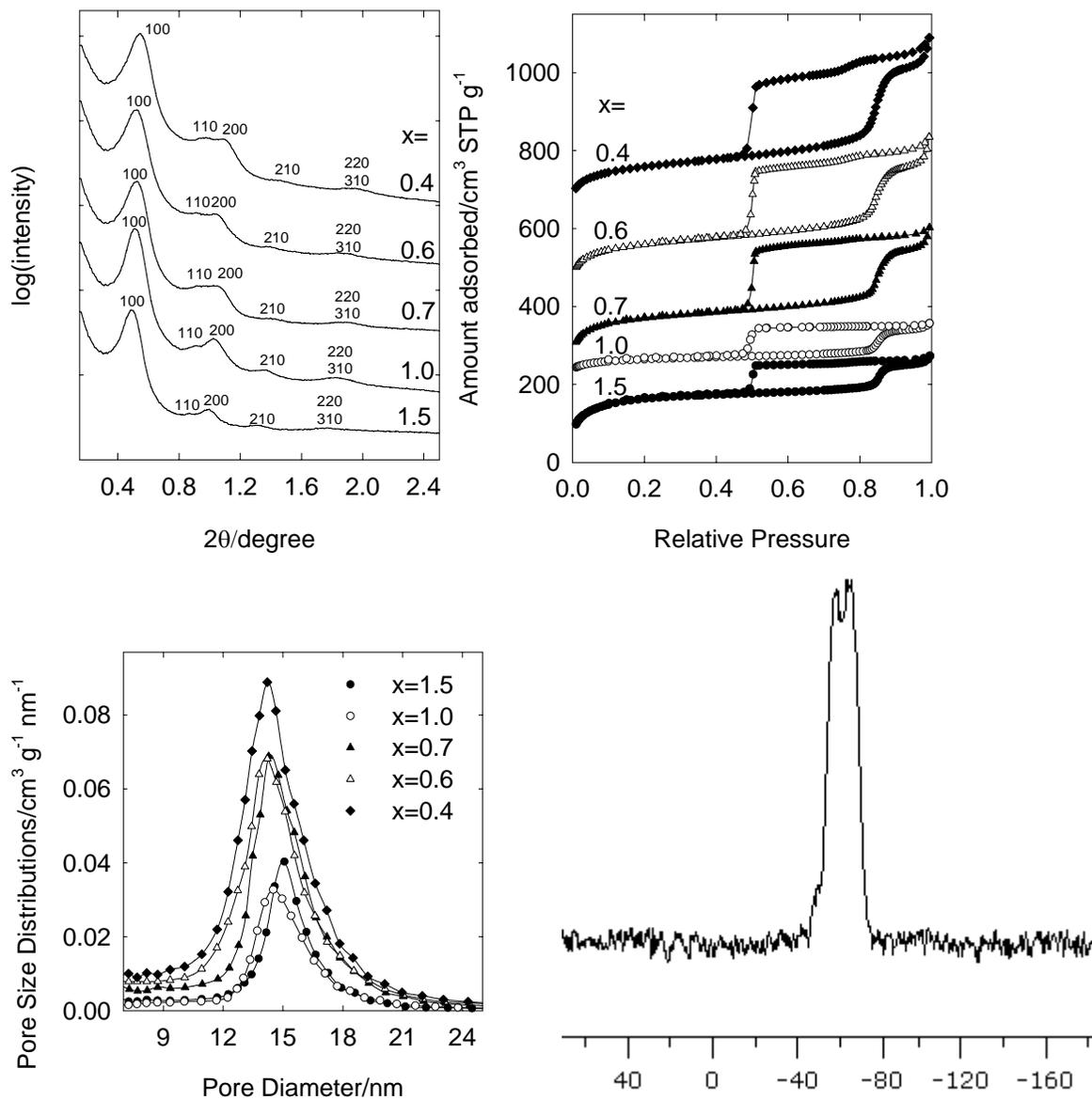
**Supporting Figure S3.**  $^{29}\text{Si}$  CP MAS NMR for a methylene-bridged PMO hydrothermally treated at 150 °C for two days (scale in ppm).



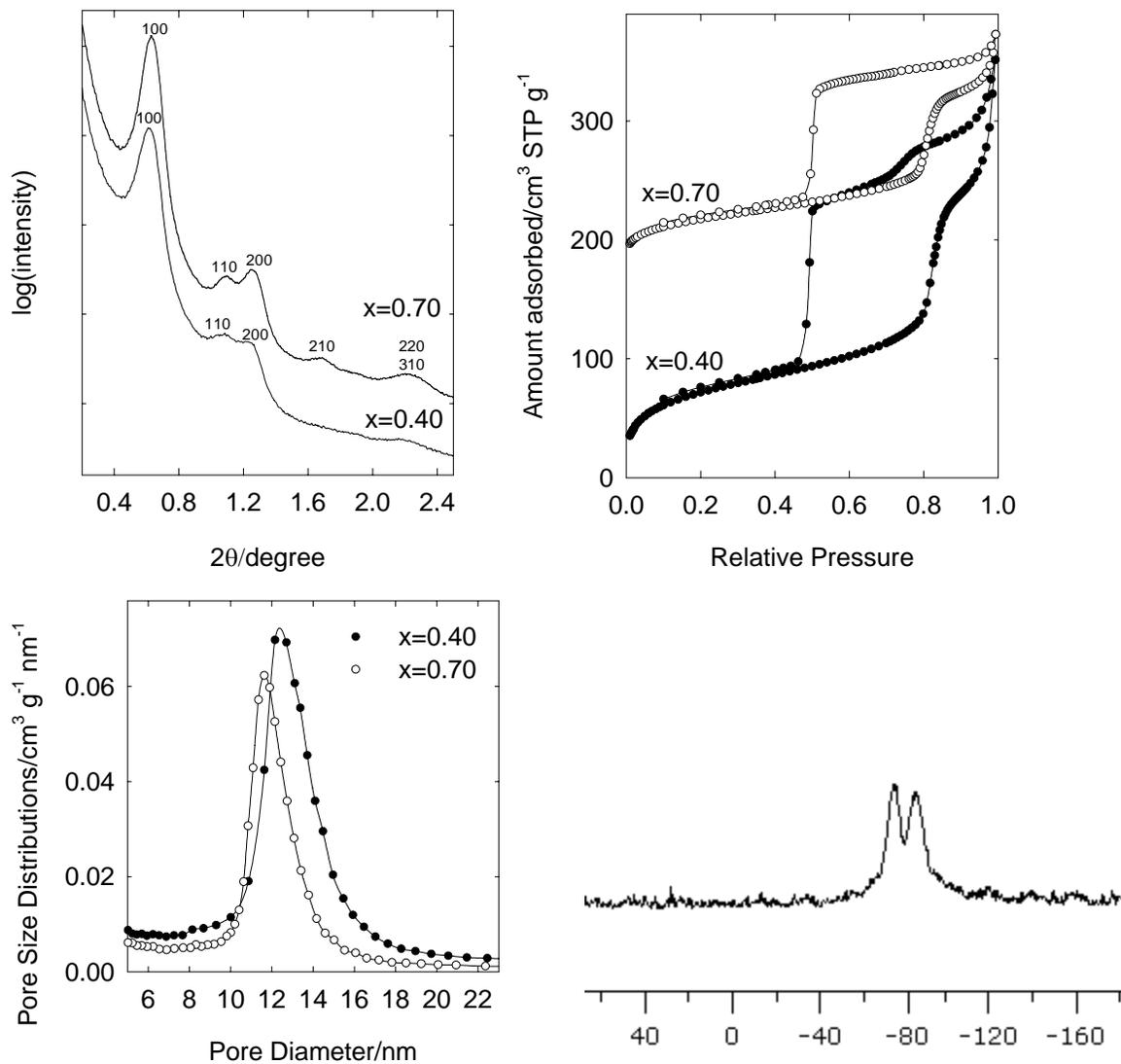
**Supporting Figure S4.** (top) SAXS patterns, (bottom left) nitrogen adsorption isotherms and (bottom right) pore size distributions of calcined methylene-bridged PMO synthesized with different amount of precursor in the presence of cyclohexane as a micelle expander. The isotherms are offset vertically by 200, and 400 cm<sup>3</sup> STP g<sup>-1</sup> for x=0.3, and x=0.4 respectively.



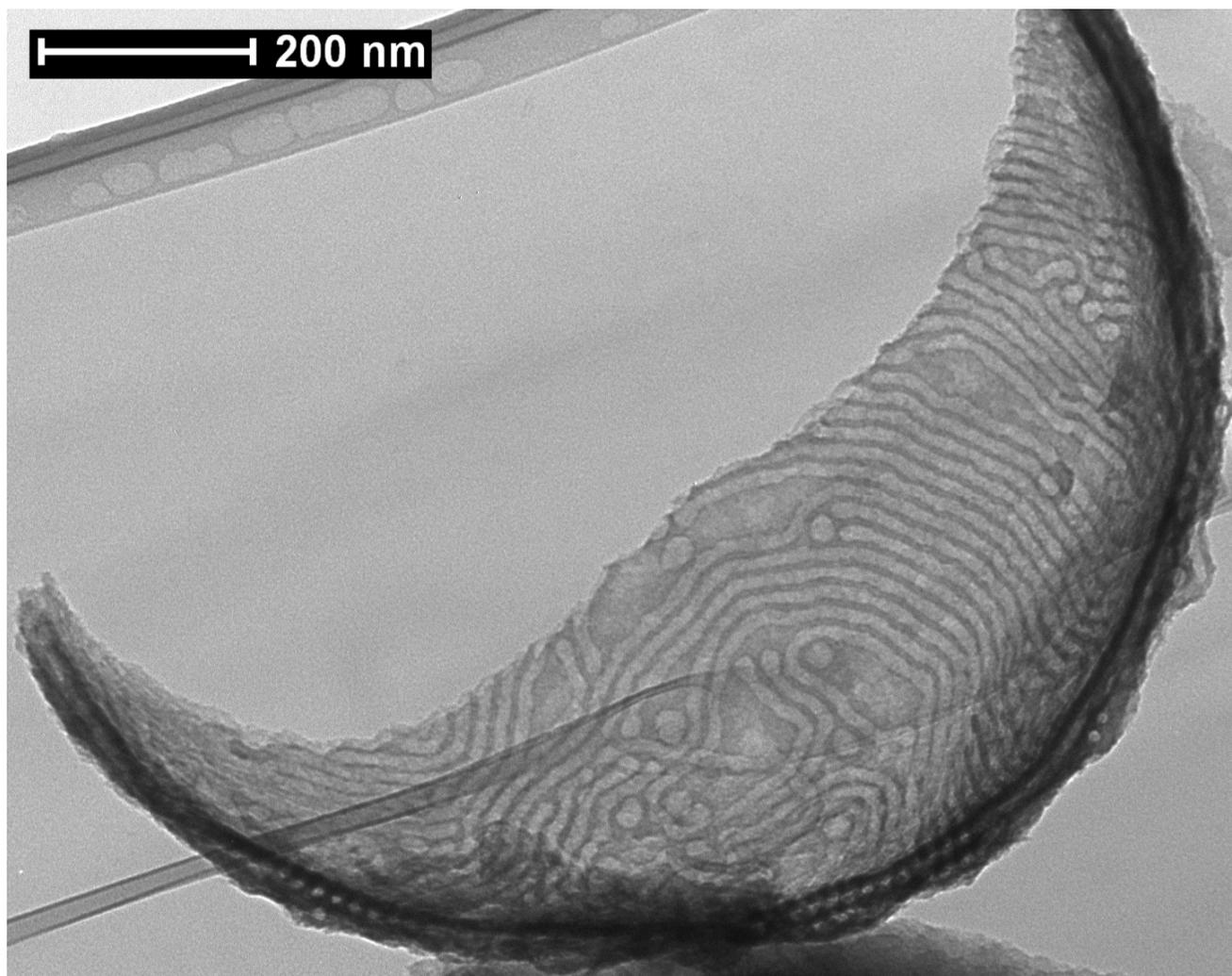
**Supporting Figure S5.** (top left) SAXS patterns, (top right) nitrogen adsorption isotherms and (bottom left) pore size distributions for calcined methylene-bridged PMOs synthesized at different initial temperature using cyclohexane as micelle expander. The isotherms were offset vertically by 260, 600, 1090 and 1640 cm<sup>3</sup> STP g<sup>-1</sup> for samples synthesized at initial synthesis temperature of 13, 14, 15 and 17 °C respectively.



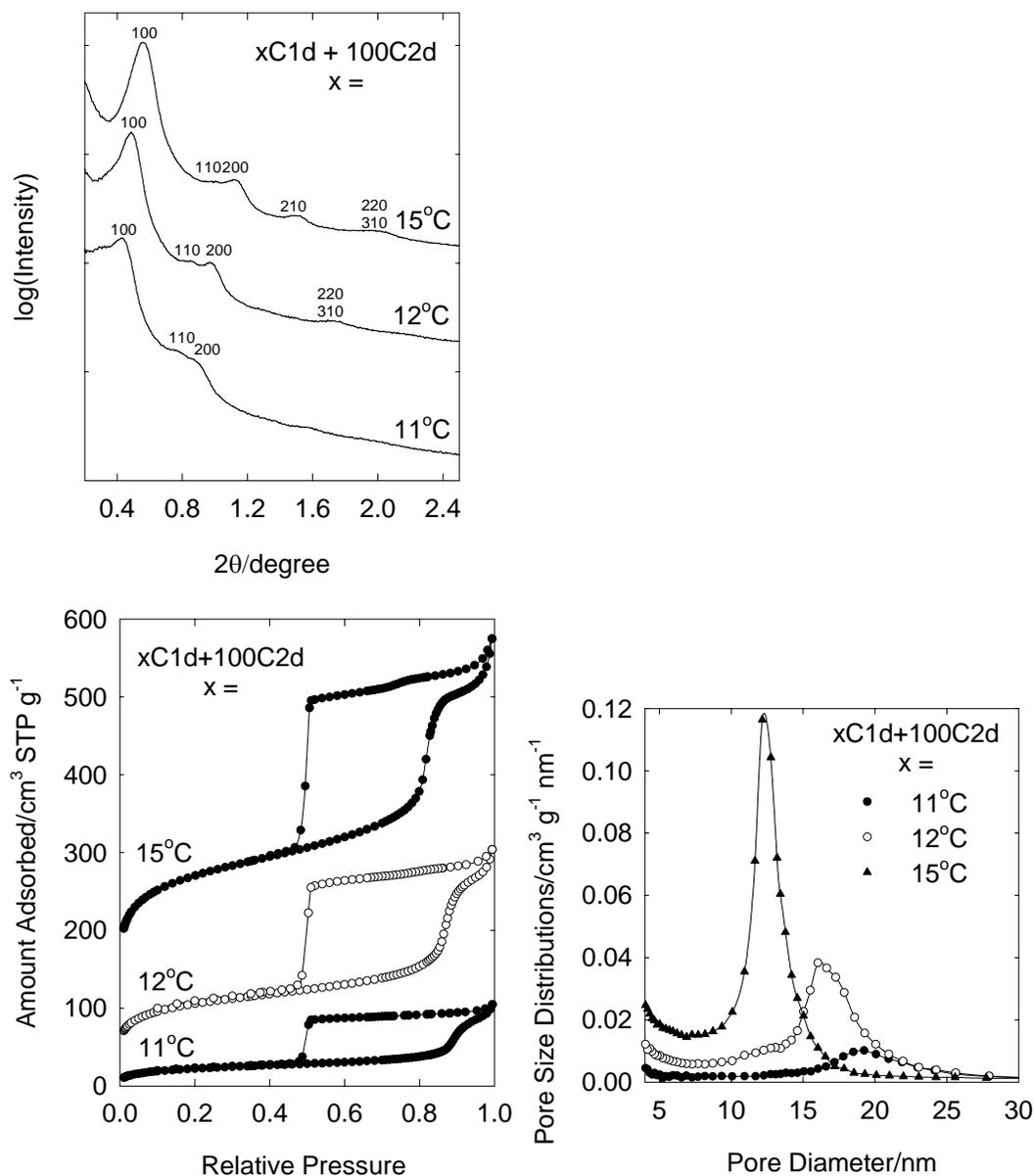
**Supporting Figure S6.** (top left) SAXS patterns, (top right) nitrogen adsorption isotherms and (bottom left) pore size distributions of extracted ethylene-bridged PMOs with different amounts of precursors using cyclohexane as a micelle expander at initial temperature of 15 °C. Isotherms were offset vertically by 220, 220, 420, and 640 cm<sup>3</sup> STP g<sup>-1</sup> for samples synthesized with x= 1.0, 0.7, 0.6, and 0.4 respectively. (bottom right) <sup>29</sup>Si CP MAS NMR for extracted ethylene-bridged PMOs.



**Supporting Figure S7.** (top left) SAXS patterns of extracted ethylene-bridged PMOs with different amount of precursor using cyclohexane as a micelle expander. (top right) nitrogen adsorption isotherms and (bottom left) pore size distributions of extracted ethylene-bridged PMOs with different amount of precursor using cyclohexane as a micelle expander. The isotherms are offset vertically by 180 cm<sup>3</sup> STP g<sup>-1</sup> for x=0.70. (bottom right) <sup>29</sup>Si CP MAS NMR for extracted ethylene-bridged PMOs.



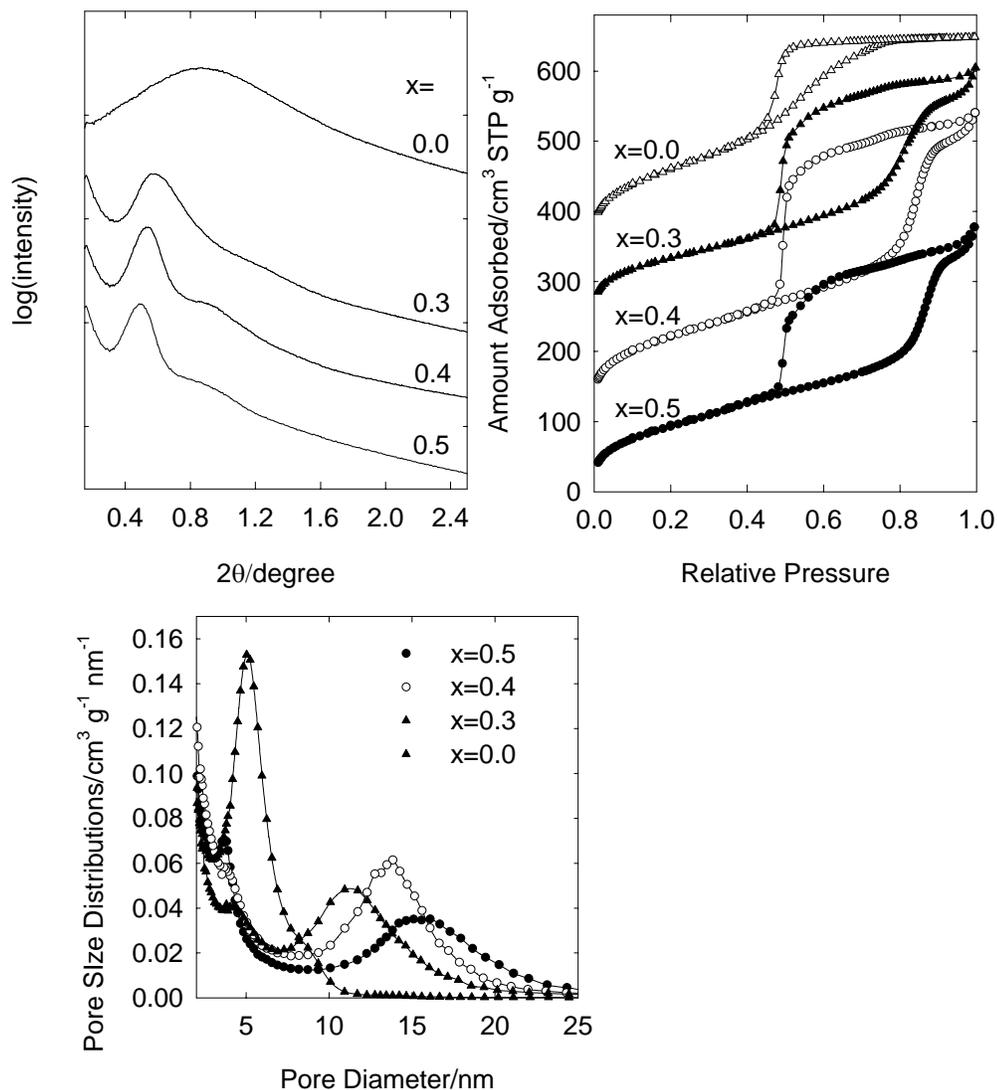
**Supporting Figure S8.** TEM image of as-synthesized ethylene-bridged PMO synthesized at initial temperature of 10.5 °C.



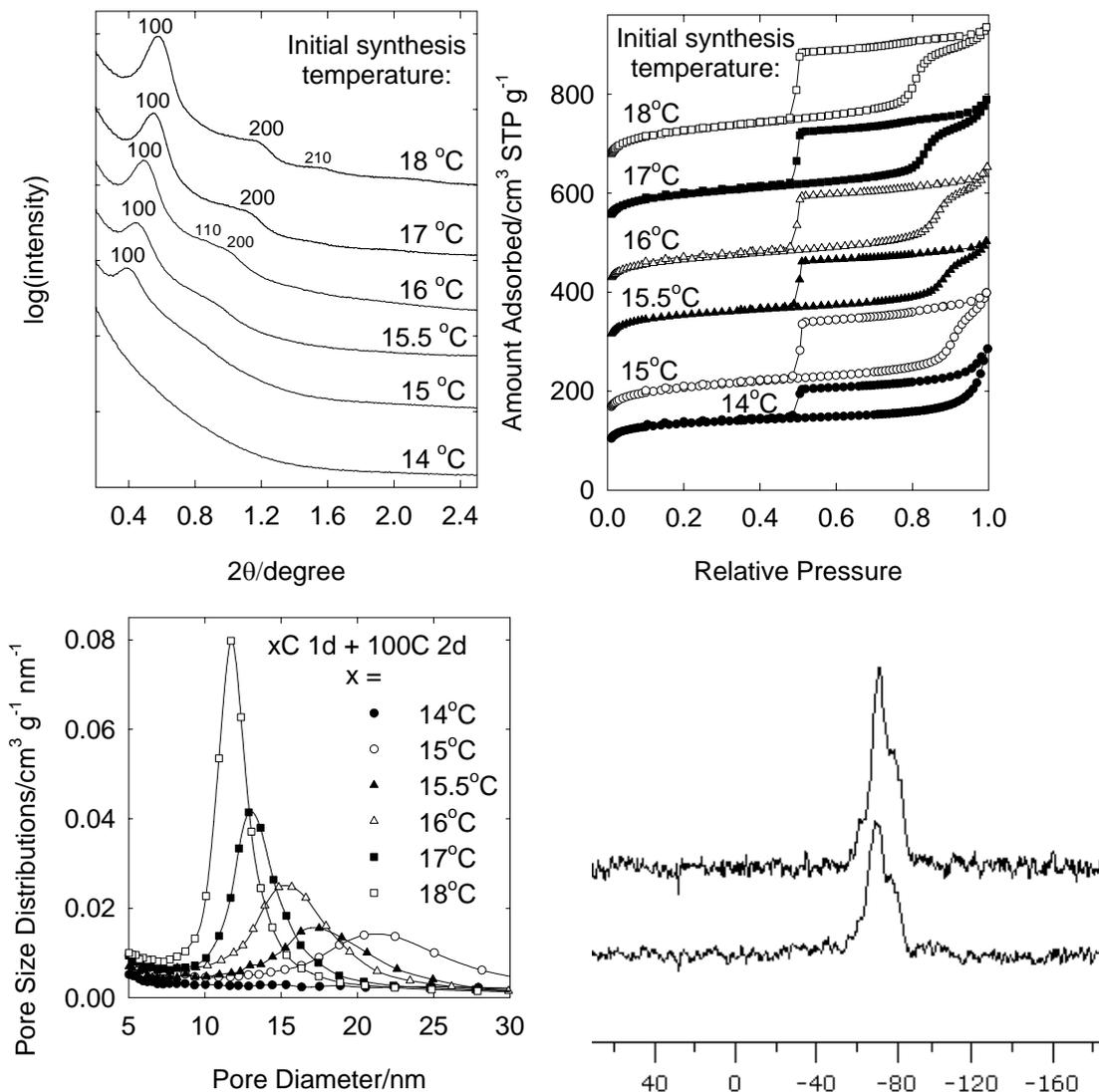
**Supporting Figure S9.** (top) SAXS patterns, (bottom left) nitrogen adsorption isotherms and (bottom right) pore size distributions for extracted ethylene-bridged PMOs synthesized at different initial synthesis temperature using bis(triethoxysilyl)ethane (BTEE) as precursor and cyclohexane as a micelle expander. The isotherms were offset vertically by 35, and 120  $\text{cm}^3 \text{ STP g}^{-1}$  for samples synthesized at initial synthesis temperature of 12, and 15 °C, respectively.



**Supporting Figure S10.** TEM image of extracted ethylene-bridged PMO synthesized at an initial temperature of 11.00 °C.



**Supporting Figure S11.** (top left) SAXS patterns, (top right) nitrogen adsorption isotherms and (bottom) PSDs of extracted PMOs synthesized using  $-\text{CH}_2\text{C}(\text{=CH}_2)\text{CH}_2$ -bridging group as precursor with different amount of precursor at an initial temperature  $15\text{ }^\circ\text{C}$  using cyclohexane as a micelle expander.



**Supporting Figure S12.** (top left) SAXS patterns of calcined phenylene-bridged PMOs synthesized at different initial temperature using 0.6 mL TIPB expander per 0.6 g P123. (top right) nitrogen adsorption isotherms and (bottom left) pore size distributions for the corresponding samples. Isotherms for samples synthesized at initial synthesis temperature of 15, 15.5, 16, 17, and 18 °C are offset vertically by 50, 180, 320, 430, and 530 cm<sup>3</sup> STP g<sup>-1</sup> respectively. Note that the sample prepared at 18 °C was prepared under the same conditions as the sample denoted 18 °C (a) in Figure 8, but it was from a different batch. (bottom right) <sup>29</sup>Si CP MAS NMR for phenylene-bridged samples calcined at 300 °C under nitrogen (top spectrum) and calcined at 250 °C under air after extraction (bottom spectrum).