

# Supporting Information

## A New Family of Metal Borohydride Ammonia Borane Complexes: Synthesis, Structures, and Hydrogen Storage Properties

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Table S1. Calculated structural parameters of  $\text{Li}_2(\text{BH}_4)_2(\text{NH}_2\text{BH}_3)$  ( $\text{LiBH}\cdot\text{AB}$ , Space group  $Pnma$ , No. 62, refined lattice parameters  $a=8.3118(8)$  Å,  $b=12.428(1)$ Å,  $c=6.5944(7)$ Å)

Atom	Site	x	y	z
Li1	8d	0.3318	0.9228	0.9681
B2	8d	0.0995	0.5736	0.7190
H3	8d	0.1709	0.6520	0.7861
H4	8d	0.1636	0.4892	0.7734
H5	8d	-0.03842	0.5740	0.7845
H6	8d	0.1020	0.5809	0.5327
H7	8d	0.2996	0.1705	0.6705
H8	8d	0.0767	0.1837	0.9487
B9	8c	0.7848	0.75	0.3091
H10	8c	0.8982	0.75	0.4280
H11	8c	0.7616	0.75	0.9778
N12	8c	0.8533	0.75	0.0830

Table S2. Calculated structural parameters of  $\text{Ca}(\text{BH}_4)_2(\text{NH}_2\text{BH}_3)_2$  ( $\text{CaBH}\cdot\text{AB}$ , Space group  $Aba2$ , No. 41, refined lattice parameters  $a=8.265(1)$  Å,  $b=13.478(2)$ Å,  $c=8.136(1)$ Å)

Atom	Site	x	y	z
B1	$8b$	0.0544	0.2092	0.7237
B2	$8b$	0.7273	0.0082	0.5747
N3	$8b$	0.0007	0.1874	0.2995
H4	$8b$	0.0674	0.3381	0.1879
H5	$8b$	0.1361	0.1655	0.8259
H6	$8b$	0.3630	0.2255	0.1029
H7	$8b$	0.0880	0.1530	0.2291
H8	$8b$	0.0940	0.3622	0.8057
H9	$8b$	0.0470	0.1941	0.4173
H10	$8b$	0.1314	0.0067	0.5362
H11	$8b$	0.3520	0.0650	0.5352
H12	$8b$	0.2875	0.4806	0.2240
H13	$8b$	0.1788	0.4165	0.5045
Ca14	$8a$	0.5	0	0.3034

Table S3. Average calculated bond lengths of  $\text{Li}_2(\text{BH}_4)_2\text{NH}_3\text{BH}_3$  and  $\text{Ca}(\text{BH}_4)_2(\text{NH}_3\text{BH}_3)_2$

Bond	Length (Å)		
	$\text{Li}_2(\text{BH}_4)_2\text{NH}_3\text{BH}_3$	$\text{Ca}(\text{BH}_4)_2(\text{NH}_3\text{BH}_3)_2$	$\text{NH}_3\text{BH}_3$
N-B	1.596	1.591	1.592
N-H(AB)	1.030	1.032	1.030
B-H(AB)	1.222	1.222	1.222
B-H( $\text{BH}_4$ )	1.228	1.229	
$\text{H}_\text{N}-\text{H}_\text{B}$ (in AB layer)	2.439	1.735	2.02 <sup>1</sup>
$\text{H}_\text{N}-\text{H}_{\text{BH}_4}$	2.248	2.036	

<sup>1</sup> Klooster, W.T.; Koetzle, T.F.; Siegbahn, P. E. M.; Richardson, T. B.; Crabtree, R. H. *J. Am. Chem. Soc.* **1999**, *121*, 6337.

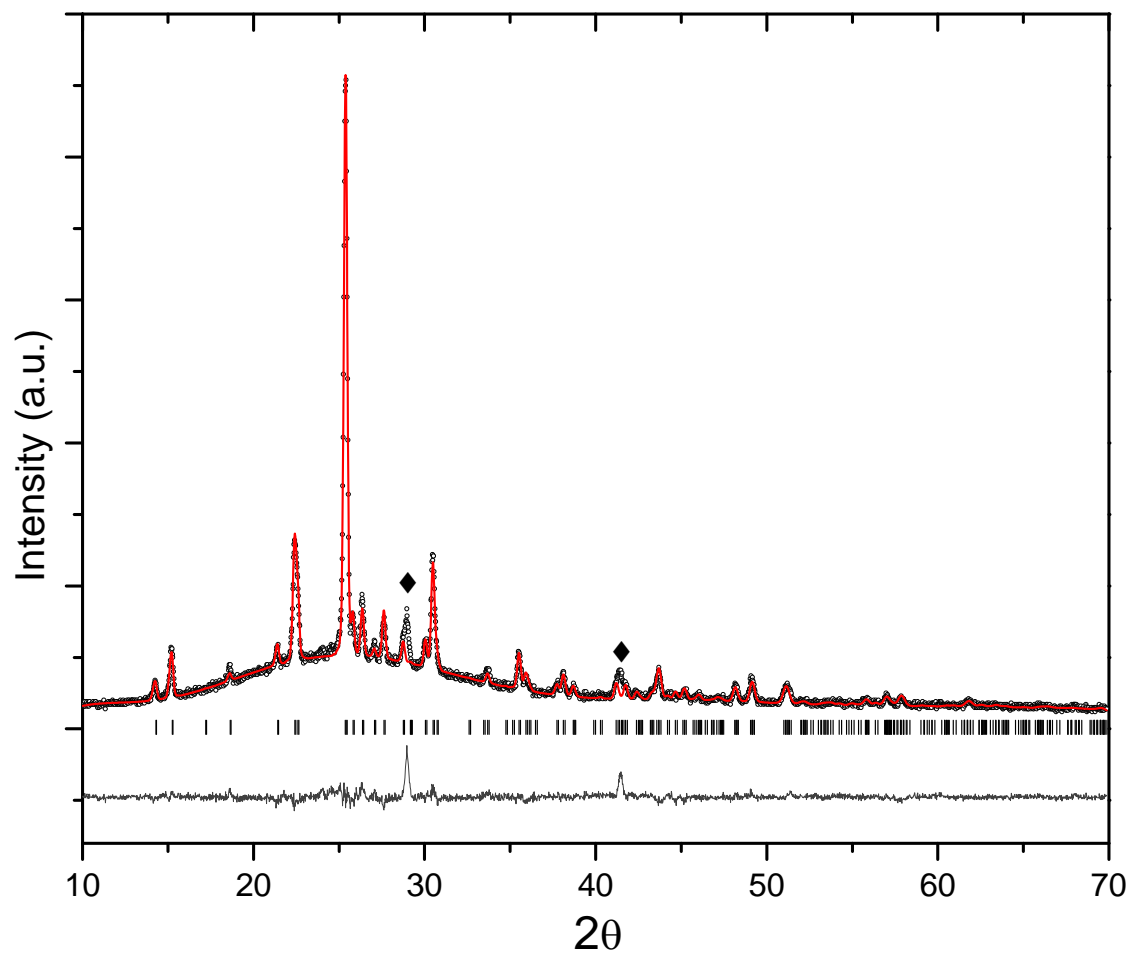


Figure S1. Experimental (circles), fitted (line), and difference (line below observed and calculated patterns) XRD profiles for  $\text{Li}_2(\text{BH}_4)_2\text{NH}_3\text{BH}_3$  at 298 K. Vertical bars indicate the calculated positions of Bragg peaks for  $\text{Li}_2(\text{BH}_4)_2\text{NH}_3\text{BH}_3$ . The patterns also contain peaks of impurity phase from the  $\text{NH}_3\text{BH}_3$  precursor (90% Aldrich, indicated by solid diamond) as identified in the previous study<sup>2</sup>.

<sup>2</sup> Wu, H.; Zhou, W.; Yildirim, T. *J. Am. Chem. Soc.* **2008**, *130*, 14834–14839.

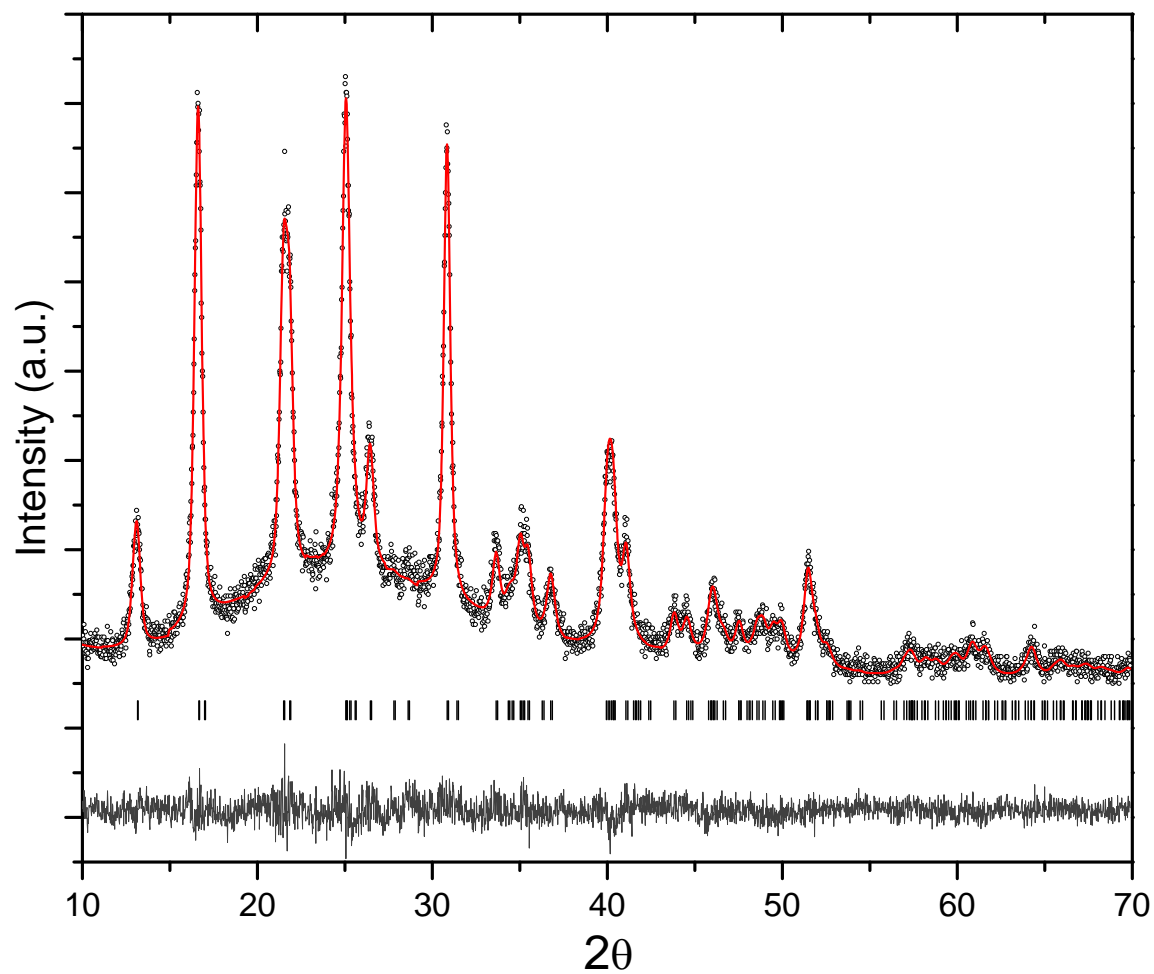


Figure S2. Experimental (circles), fitted (line), and difference (line below observed and calculated patterns) XRD profiles for  $\text{Ca}(\text{BH}_4)_2(\text{NH}_3\text{BH}_3)_2$  at 298 K. Vertical bars indicate the calculated positions of Bragg peaks for  $\text{Ca}(\text{BH}_4)_2(\text{NH}_3\text{BH}_3)_2$ .

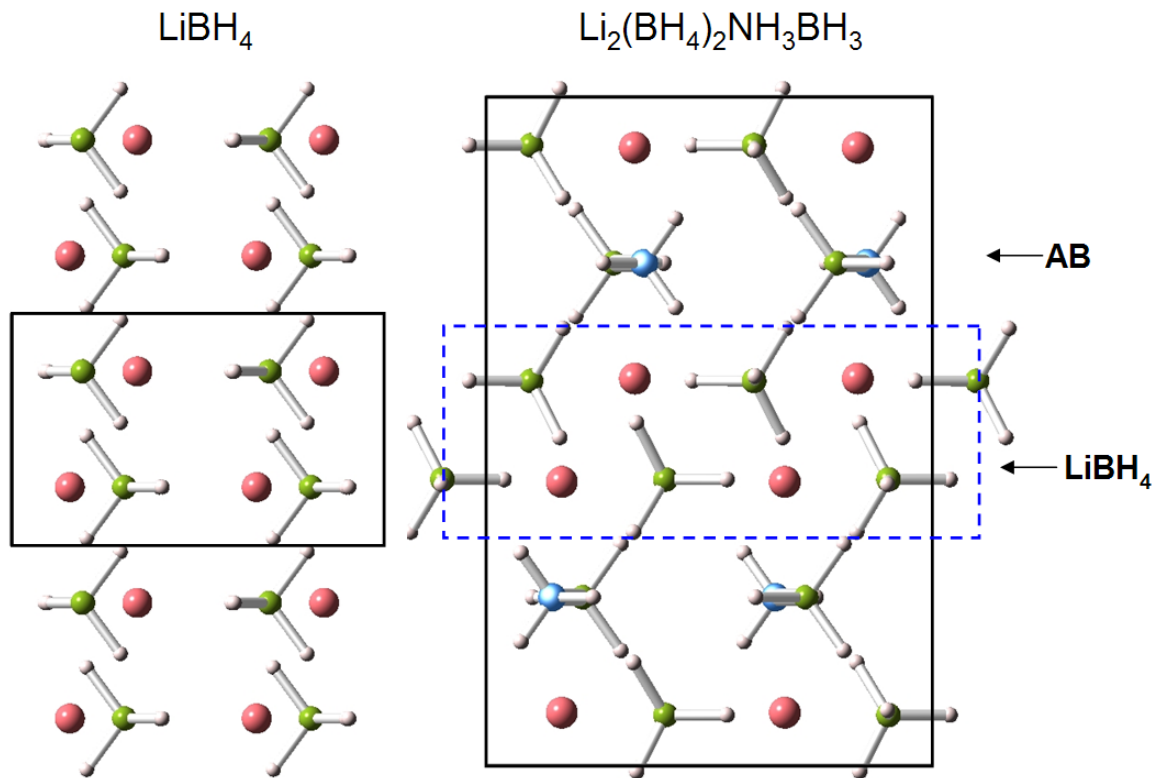


Figure S3 Comparison of the structures ([001] view) of  $\text{LiBH}_4$  and  $\text{Li}_2(\text{BH}_4)_2\text{NH}_3\text{BH}_3$ . Note:  $\text{Li}_2(\text{BH}_4)_2\text{NH}_3\text{BH}_3$  contains alternating layers of  $\text{LiBH}_4$  and  $\text{NH}_3\text{BH}_3$ . The unit cells of  $\text{LiBH}_4$  and  $\text{Li}_2(\text{BH}_4)_2\text{NH}_3\text{BH}_3$  are plotted in grey line. The  $\text{LiBH}_4$  block in  $\text{Li}_2(\text{BH}_4)_2\text{NH}_3\text{BH}_3$  cell is highlighted in blue.

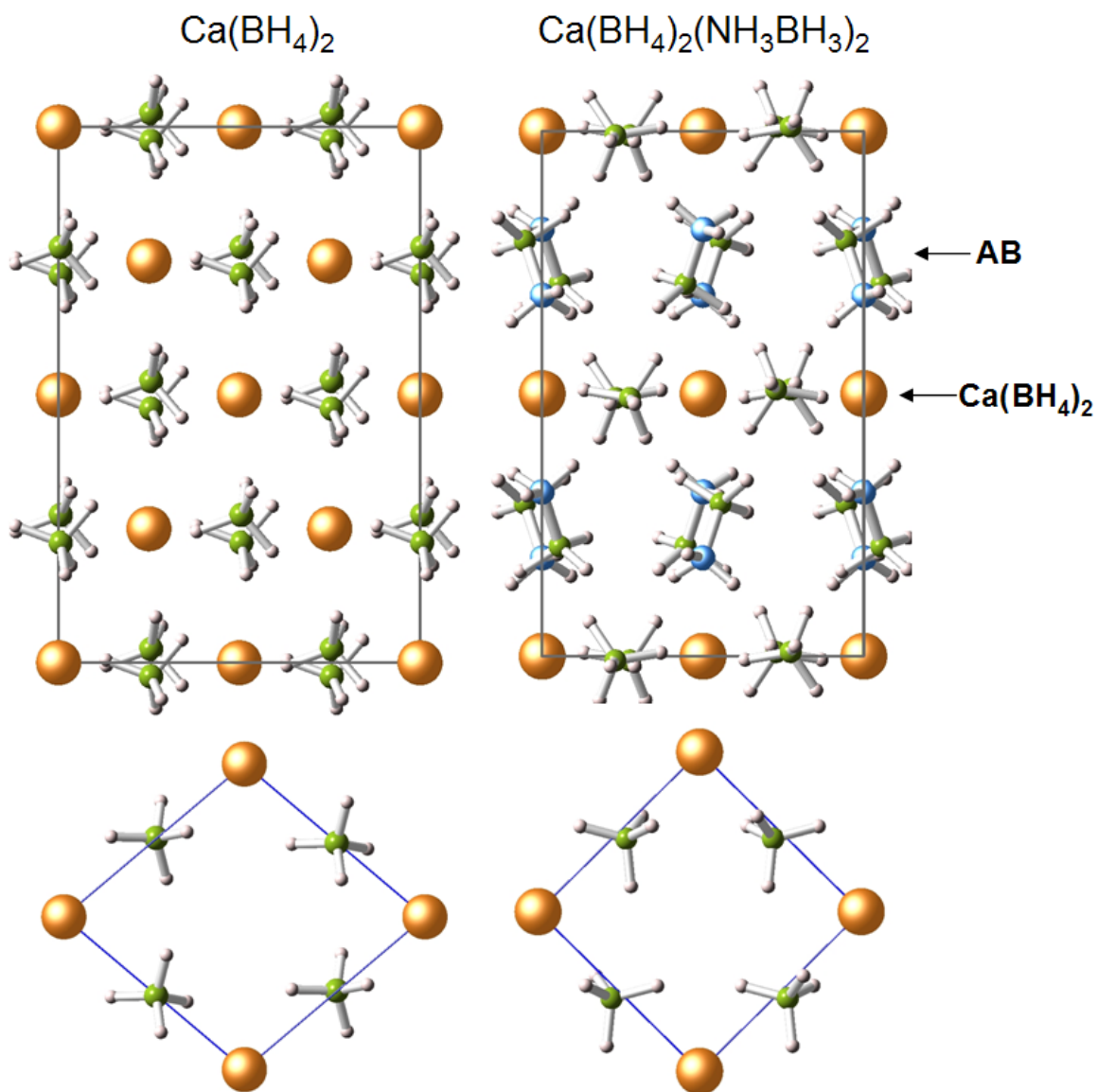


Figure S4 Top: Comparison of the structures ([001] view) of  $\text{Ca}(\text{BH}_4)_2$  and  $\text{Ca}(\text{BH}_4)_2(\text{NH}_3\text{BH}_3)_2$ . Note:  $\text{Ca}(\text{BH}_4)_2(\text{NH}_3\text{BH}_3)_2$  contains alternating layers of  $\text{Ca}(\text{BH}_4)_2$  and  $\text{NH}_3\text{BH}_3$ . Bottom: comparison of lattice arrangement of  $\text{Ca}(\text{BH}_4)_2$  and the  $\text{Ca}(\text{BH}_4)_2$  layer of  $\text{Ca}(\text{BH}_4)_2(\text{NH}_3\text{BH}_3)_2$ .

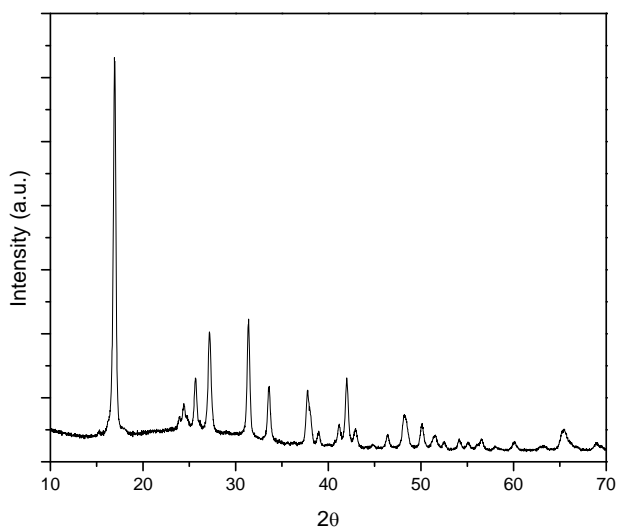
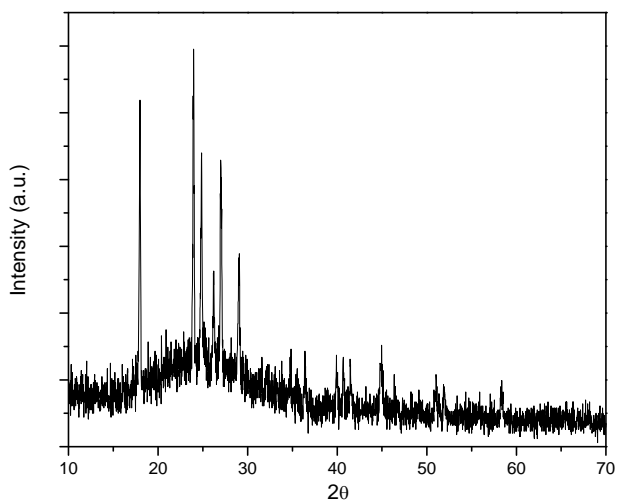


Figure S5 Top: XRD pattern of  $\text{Li}_2(\text{BH}_4)_2\text{NH}_3\text{BH}_3$  after the 1<sup>st</sup> step desorption, indicating the product is  $\text{LiBH}_4$ . Bottom: XRD pattern of  $\text{Ca}(\text{BH}_4)_2(\text{NH}_3\text{BH}_3)_2$  after the 1<sup>st</sup> step desorption, the main product is  $\text{Ca}(\text{BH}_4)_2$ .



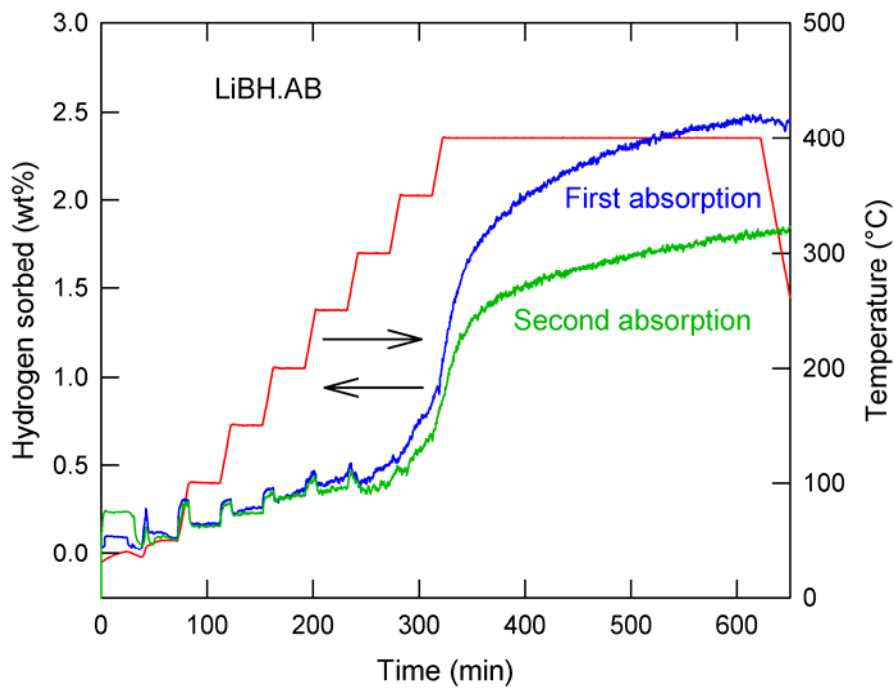


Figure S6(a) First and second rehydrogenation for LiBH·AB up to 400°C.

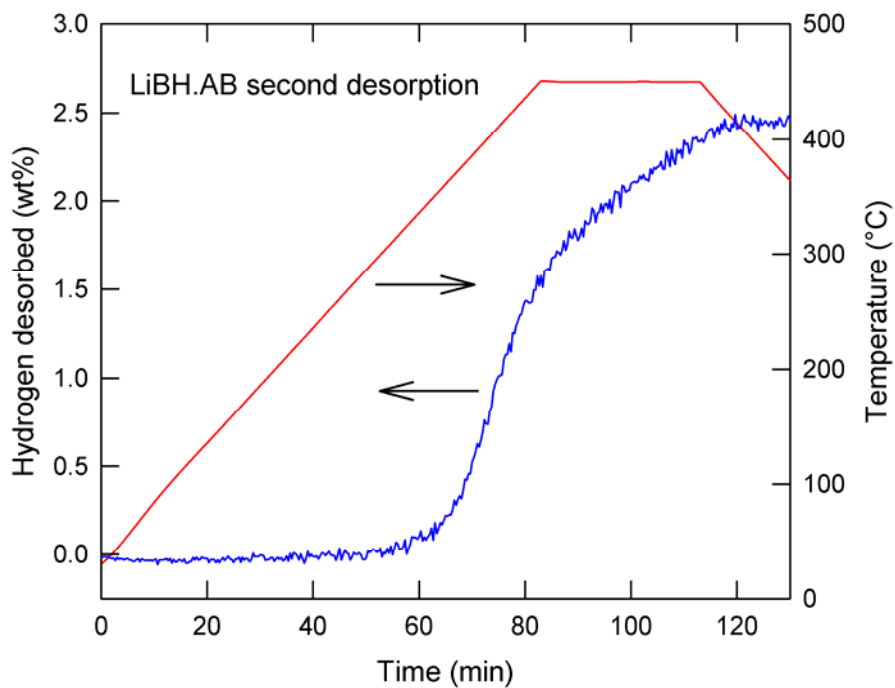


Figure S6(b) Second desorption (between 1<sup>st</sup> and 2<sup>nd</sup> hydrogenation) for LiBH·AB. RGA indicates that the second desorption is comprised only of H<sub>2</sub>.

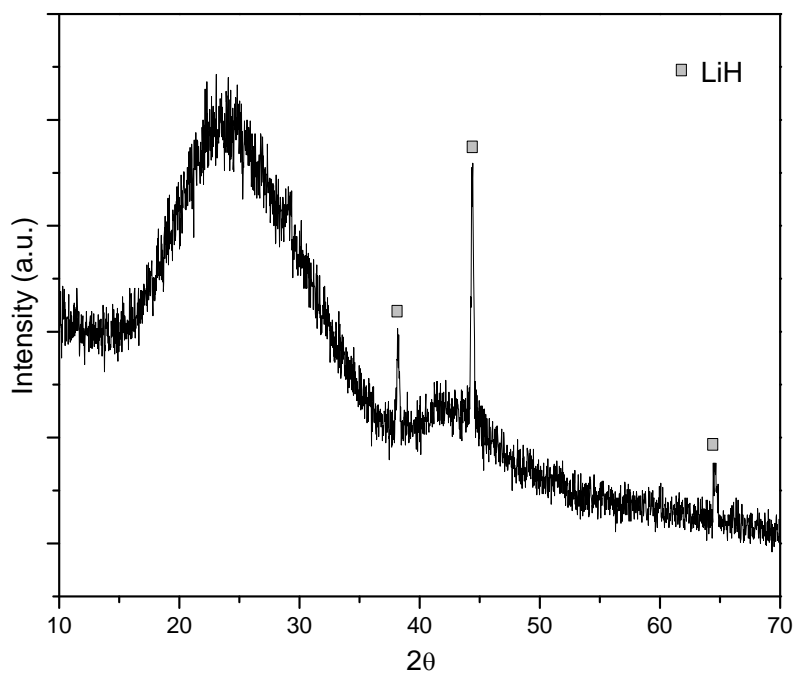


Figure S7 XRD pattern of LiBH·AB after rehydrogenation

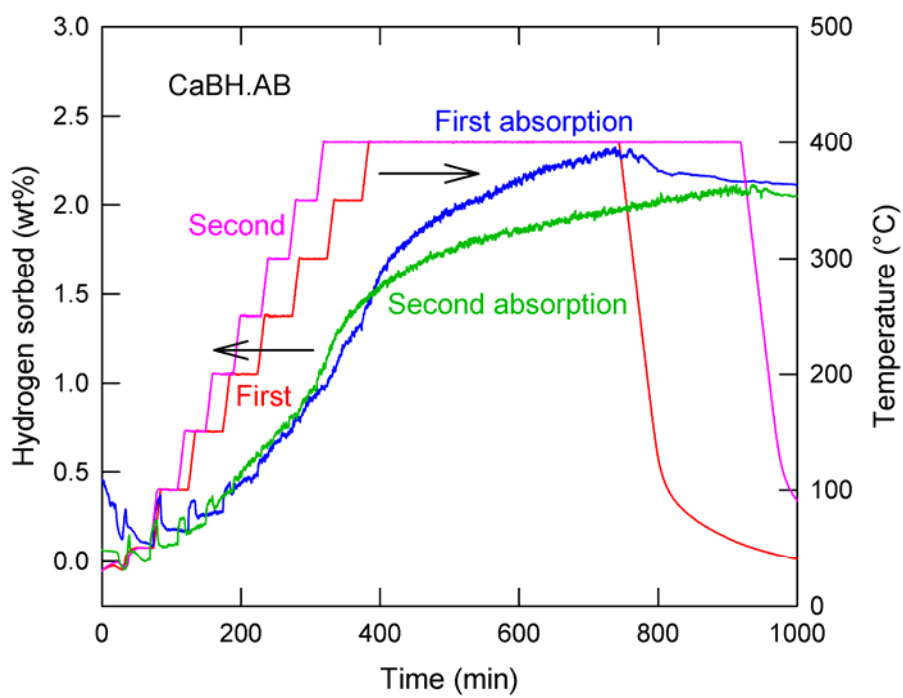


Figure S8(a) First and second rehydrogenation for CaBH·AB up to 400°C.

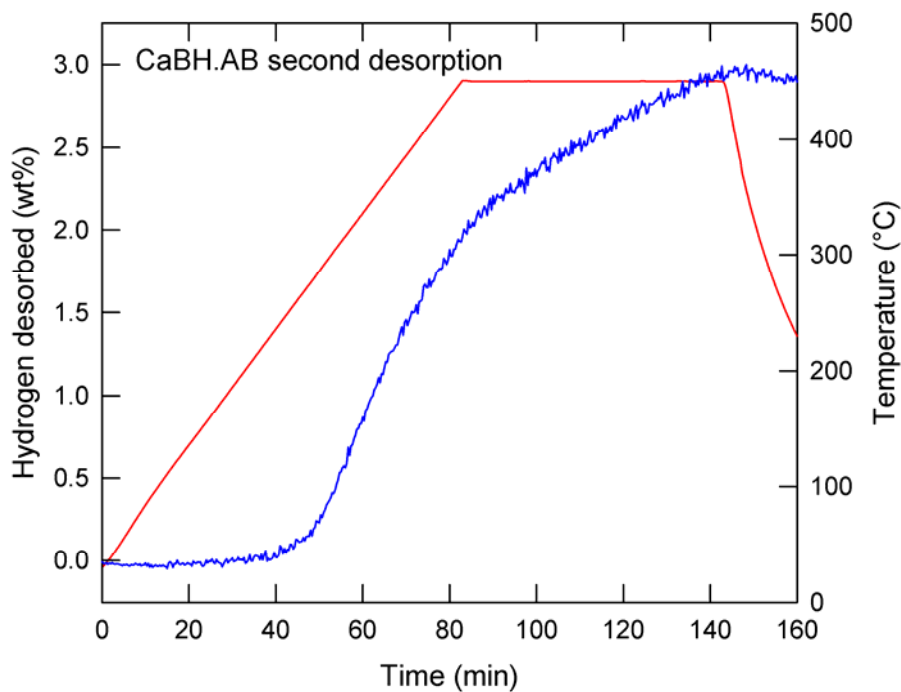


Figure S8(b) Second desorption (between 1<sup>st</sup> and 2<sup>nd</sup> hydrogenation) for CaBH·AB. RGA indicates that the second desorption is comprised only of H<sub>2</sub>.

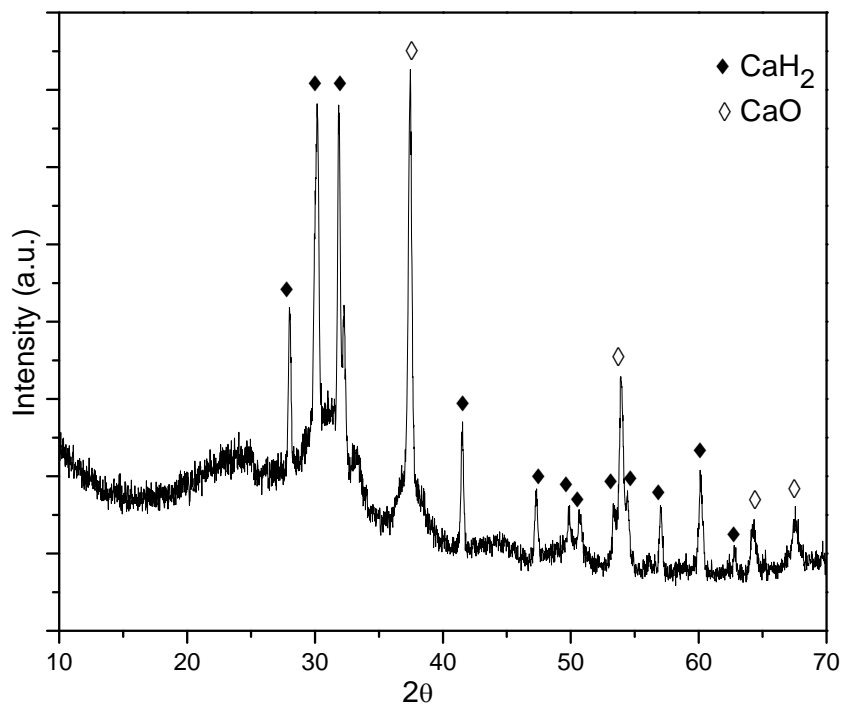


Figure S9 XRD pattern of CaBH·AB after rehydrogenation