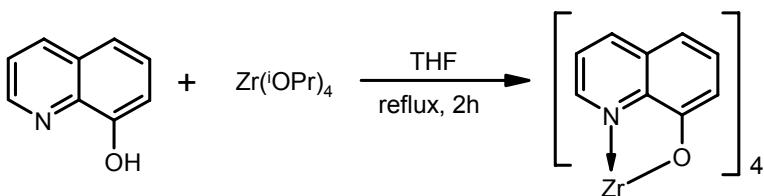
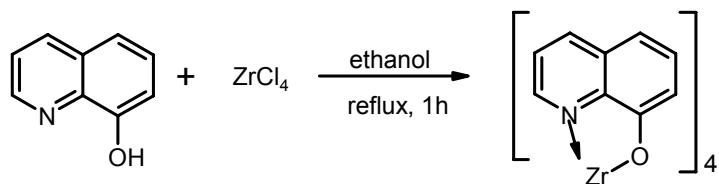


**Preparation of Zirconium tetrakis(8-hydroxyquinolate) [ $\alpha$ -ZrQ<sub>4</sub>]**



8-Hydroxyquinoline, 99+% (Aldrich) (100 g; 0.69 mole) was dissolved in tetrahydrofuran, 99+% (ACS reagent) (400 ml). To the magnetically stirred solution was added zirconium isopropoxide (66.8 g; 0.17 mole) in tetrahydrofuran (175 ml), all at once. A yellow precipitate formed immediately. The reaction mixture was magnetically stirred and refluxed under nitrogen for 2 hours. After cooling the reaction mixture in the refrigerator the product was filtered off under suction, washed with fresh tetrahydrofuran and dried under vacuum at 80 °C for 10 hours. Yield 101 g (88 %), mp 388 °C (DSC, onset). The product was sublimed at 340 °C ( $2 \times 10^{-6}$  Torr) to give a sublimation yield of 91 %. Overall yield was 81%. Found: C; 64.49, H; 3.57, N; 8.33, Zr; 13.79, Cl < 0.10.  $C_{36}H_{24}N_4O_4Zr$  requires, C; 64.75, H; 3.62, N; 8.39, Zr; 13.66 %. The trace elements analysis: Fe <1 ppm, Cr<1 ppm, Co<1 ppm, Co <1 ppm, Ni< 1 ppm, Al<1 ppm, Cu <1 ppm, Mo < 1 ppm, Ta < 25 ppm , Pt < 1 ppm, Ti < 2 ppm, Hf 5 ppm and W < 1ppm.

**Preparation of Zirconium tetrakis(8-hydroxyquinolate) [ $\beta$ -ZrQ<sub>4</sub>]**



Ethanol was dried over potassium hydroxide and distilled before use. To a solution of 8-hydroxyquinoline (100.0 g; 0.69 mol) in ethanol, 95% (700 ml) was carefully added zirconium (IV) chloride (40.2 g; 0.17 mol) in ethanol (200 ml) at 10 °C (The solution was cooled in an ice-water bath). After 10 minutes, the solution was warmed to 50 °C and piperidine (75 ml; 0.76 mole) was slowly added, during which time a yellow precipitate separated out. The suspension was refluxed for 1h, and allowed to cool to room temperature. The precipitate was collected by suction filtration on a Buchner funnel, washed thoroughly with ethanol, tetrahydrofuran and finally with diethyl ether. The product was dried under vacuum at 80 °C. Crude yield 106.5 g (93 %). Further purification of the product (65 g) was performed by Soxhlet extraction (65 g) with 1,4-dioxane for 24 h. Concentration of the extract yielded a yellow precipitate which was collected on a Buchner funnel and washed with ethanol, (95%; 100 ml). The product was dried under vacuum at 80 °C for 8h. Yield, 50 g (77 %), mp., 388 °C (DSC, onset). The product was further purified by sublimation to give a sublimed yield of 92 %. Overall yield was 66 %. Found C 64.42, H 3.58, N 8.16 ,Zr 13.54 , Cl < 0.10.  $C_{36}H_{24}N_4O_4Zr$  requires, C 64.75, H 3.62, N 8.39 and Zr13.66 %. The trace elements analysis: Fe 5ppm, Cr<1ppm, Co<1ppm, Ni <1ppm, Al<1ppm, Cu < 1ppm, Mo<1 ppm, Ta 23 ppm, Pt < 1ppm, Ti<1 ppm, Hf 6ppm, W< 1ppm.

Sample Number LIB\_000312\_NMR\_OLED  
Project E246-QSV

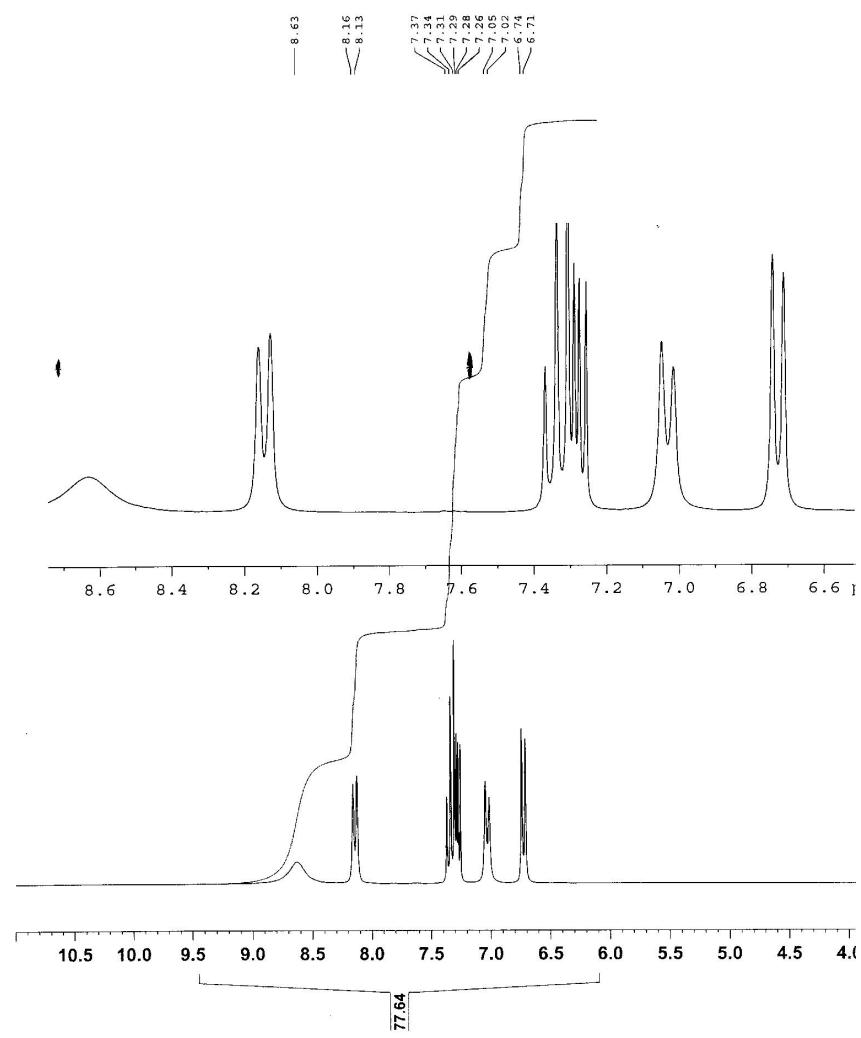


Fig. 16: <sup>1</sup>H NMR of  $\alpha$ -Zr<sub>4</sub> in DMSO-d<sup>6</sup>

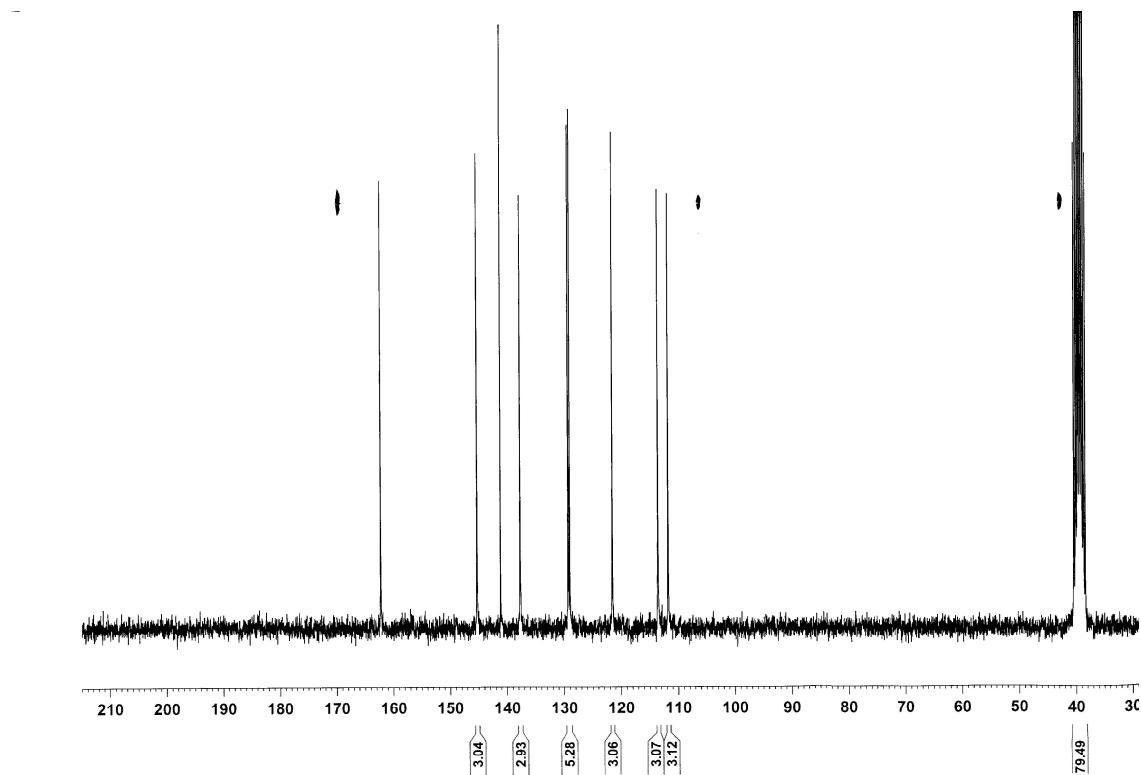


Fig. 17:  $^{13}\text{C}$  NMR of  $\alpha\text{-ZrQ}_4$  in  $\text{DMSO-d}^6$

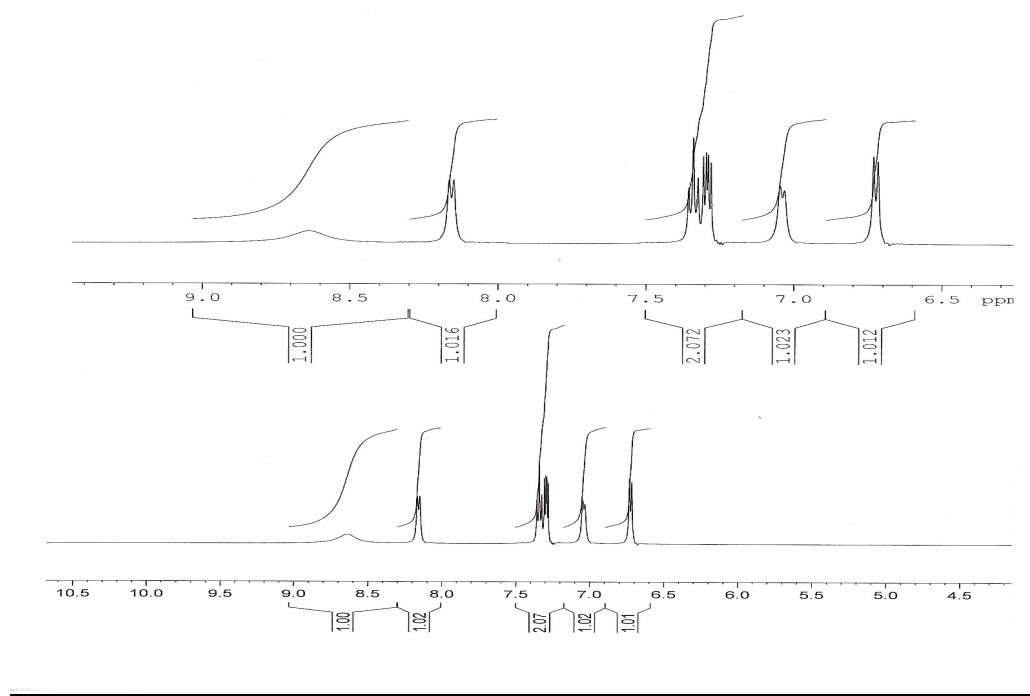


Fig. 18: <sup>1</sup>H NMR of  $\beta$ -ZrQ<sub>4</sub> in DMSO-d<sup>6</sup>

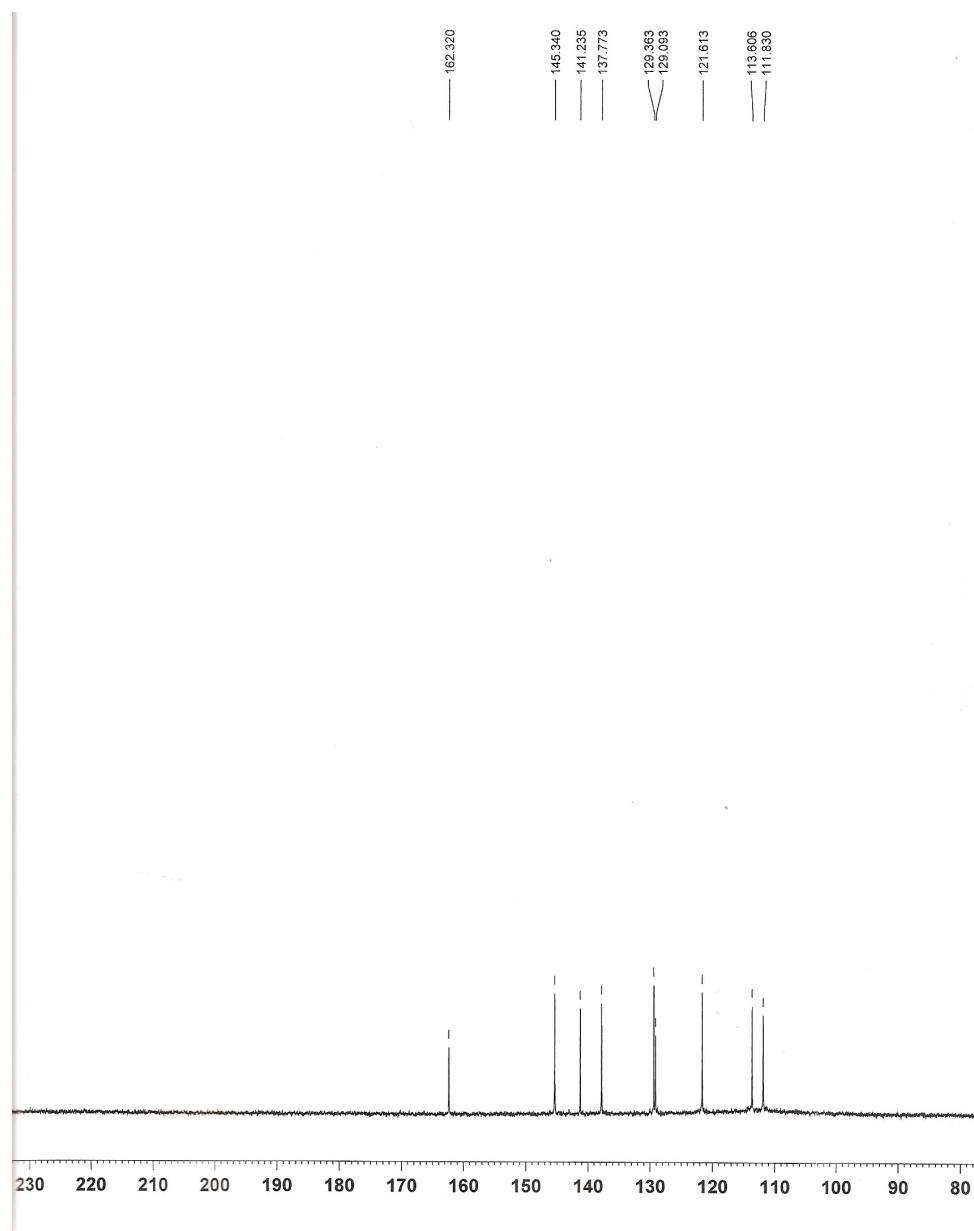


Fig. 19:  $^{13}\text{C}$  NMR of  $\beta\text{-ZrQ}_4$  in  $\text{DMSO-d}^6$

Supplementary Material (ESI) for Journal of Materials Chemistry  
This journal is (c) The Royal Society of Chemistry 2011

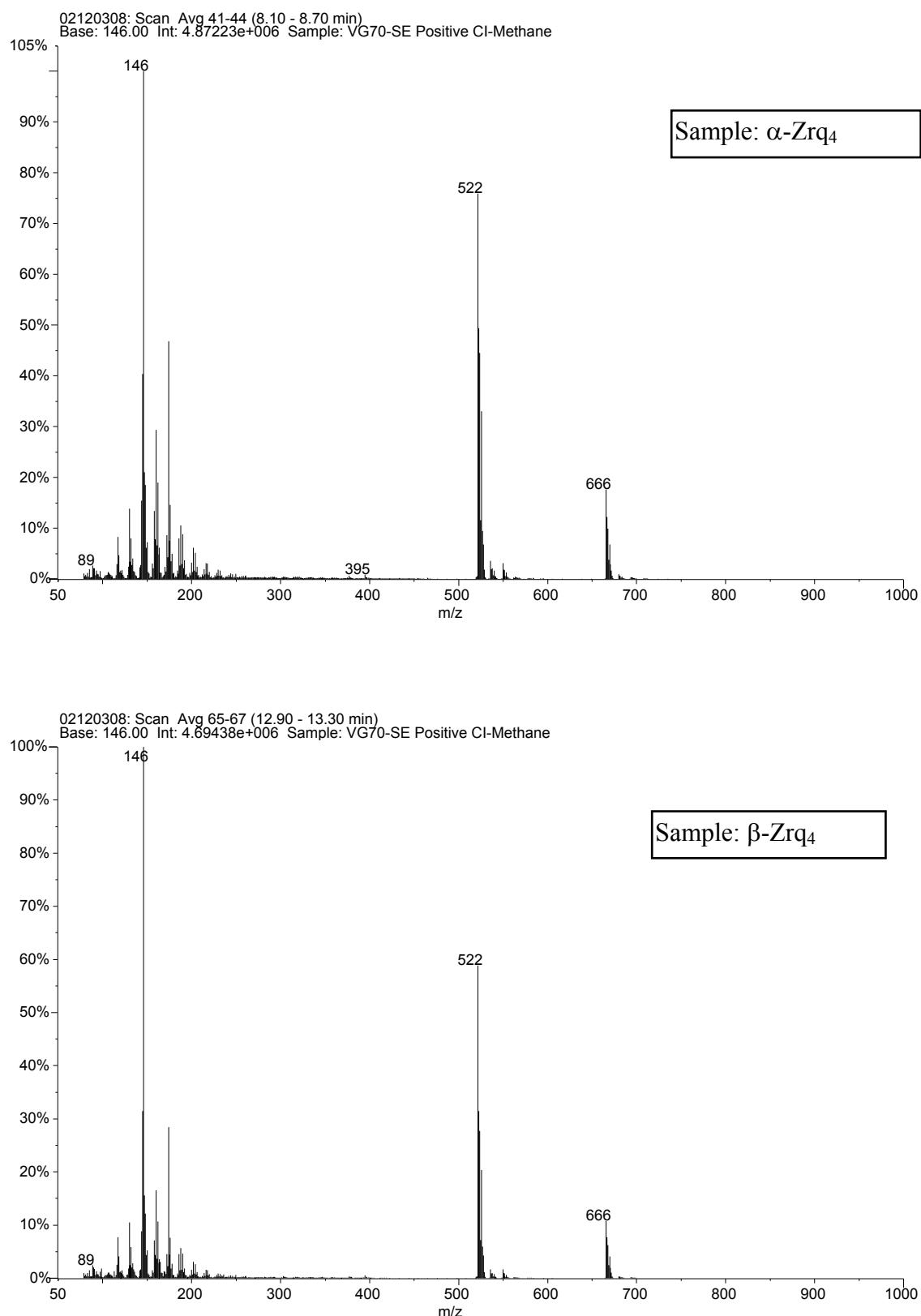


Fig. 20: Chemical ionization mass spectra of  $\alpha\text{-Zr}_4$  and  $\beta\text{- Zr}_4$

**Table S.1.** Selected hydrogen-bond and contact parameters

$\alpha$ -form:

$D\text{--H}\cdots A$	$D\text{--H} (\text{\AA})$	$\text{H}\cdots A (\text{\AA})$	$D\cdots A (\text{\AA})$	$D\text{--H}\cdots A (^{\circ})$
C7—H7···O4 <sup>i</sup>	0.95	2.49	3.430 (3)	171

```

Cg(I) Res(I)   Cg(J)  [ ARU(J) ]      Cg-Cg Transformed J-Plane P, Q, R, S      Alpha   Beta  Gamma   CgI_Perp
CgJ_Perp Slippage
Cg(6) [ 1 ] > Cg(6) [ 3755.01 ] 3.4983(14) -0.2617-0.9035-0.3394 -1.1877 0 2.14 2.14 -3.4958(10) -
3.4985(10) 0.130
[3755] = -2-X,-Y,-Z

```

### Analysis of X-H...Cg(Pi-Ring) Interactions ( $H \ldots Cg < 3.0$ Ang. - Gamma < 30.0 Deg)

$\beta$ -form:

No centroid-centroid distances below 3.95 Å.

### Analysis of X-H...Cg(Pi-Ring) Interactions ( $H \ldots Cg < 3.0$ Ang. - Gamma < 30.0 Deg)

