

Supporting Information

Surfactant-Free Nonaqueous Synthesis of Lithium Titanium Oxide (LTO) Nanostructures for Lithium Ion Battery Applications

Seung-Ho Yu¹, Andrea Pucci², Tobias Hertrich², Marc-Georg Willinger², Seung-Hwan Baek³, Yung-Eun Sung^{1,3} and Nicola Pinna^{2,3,}*

1 - School of Chemical & Biological Engineering and Research Center for Energy Conversion & Storage, Seoul National University (SNU), Seoul 151-744, Korea

2 - Department of Chemistry, CICECO, University of Aveiro, 3810-193 Aveiro, Portugal.

3 - World Class University (WCU) program of Chemical Convergence for Energy & Environment (C2E2), School of Chemical and Biological Engineering, College of Engineering, Seoul National University (SNU), Seoul 151-744, Korea.

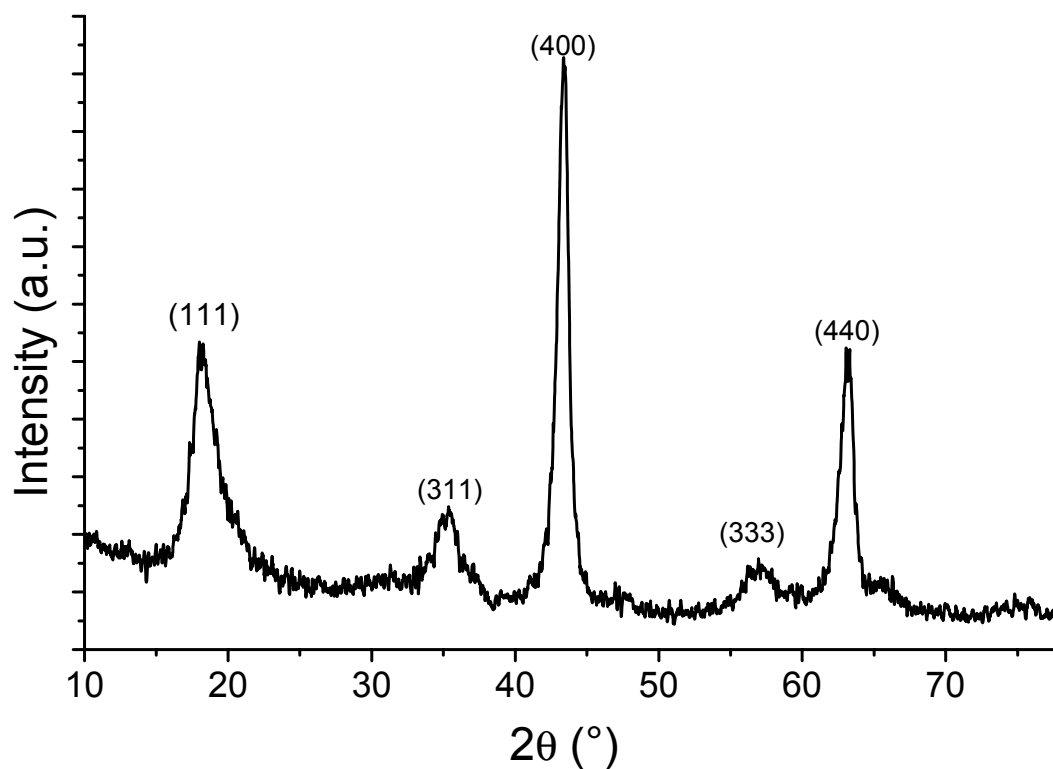


Figure S1. XRD pattern of the as-prepared LTO nanostructures with a Li:Ti concentration ratio of 8.5:8 mmol. The excess of lithium permits to eliminate the TiO₂ anatase phase impurity.

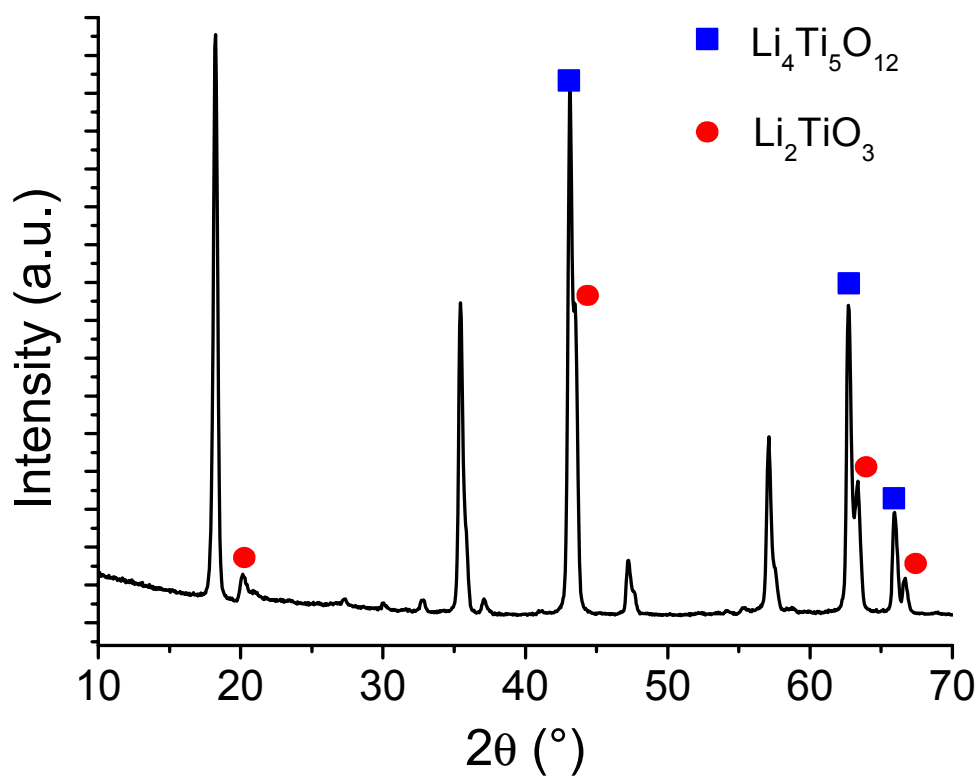


Figure S2. XRD pattern of LTO nanostructures for a Li:Ti concentration ratio of 8.5:8 mmol. The excess of lithium permits to eliminate the TiO₂ anatase phase impurity, however it promote the formation on Li₂TiO₃.

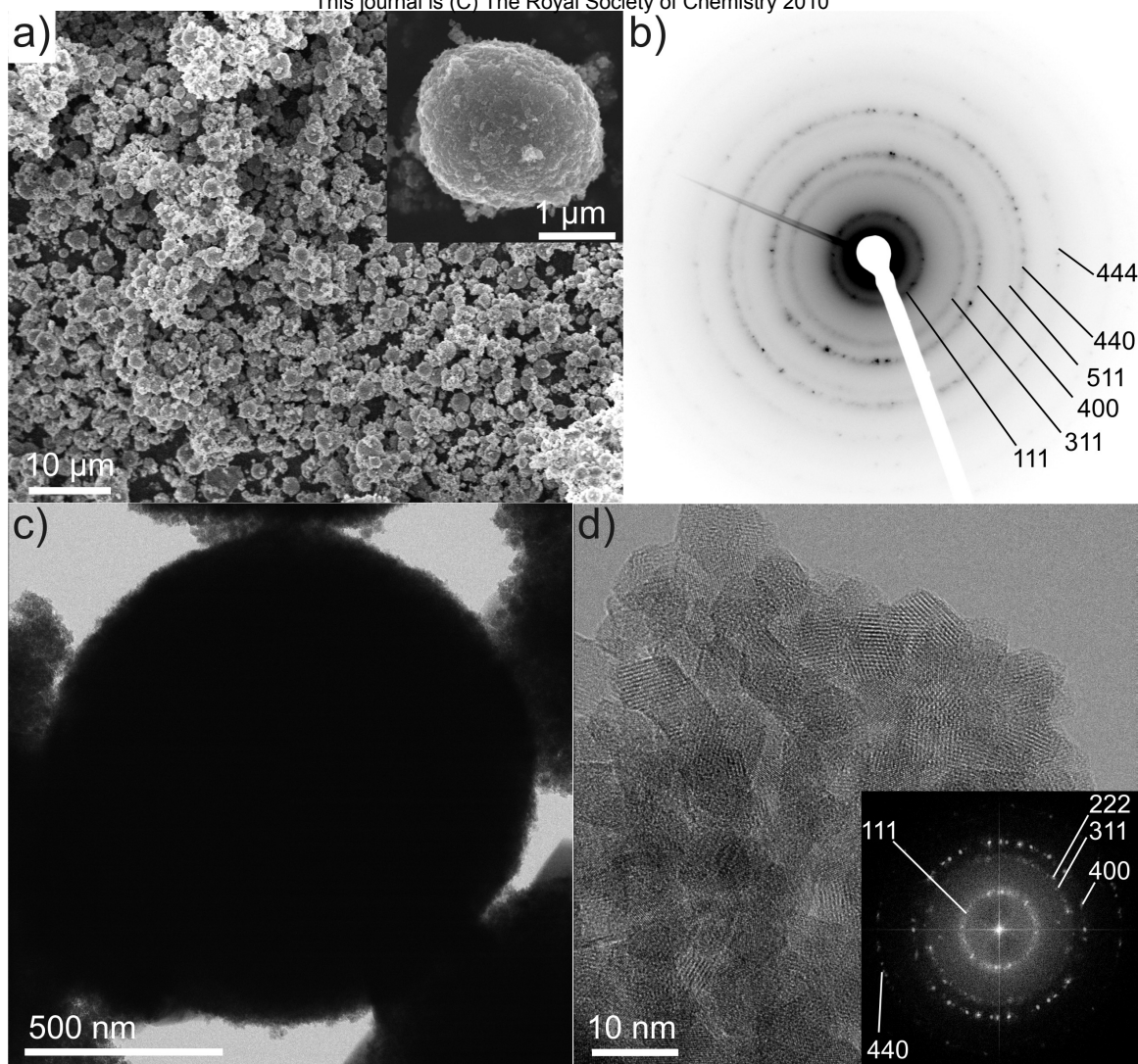


Figure S3. High quality image of Figure 2

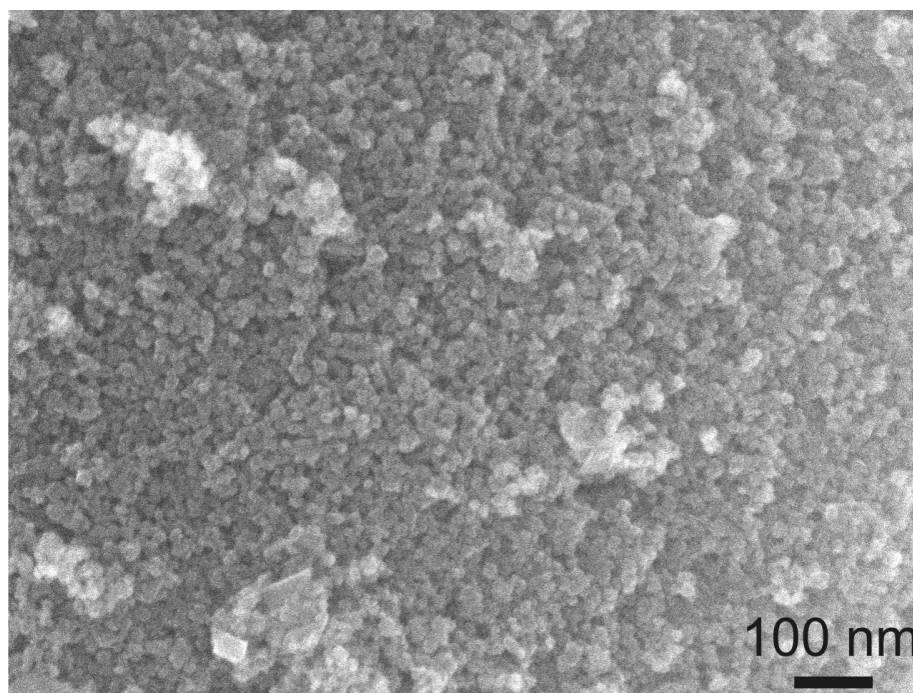


Figure S4. HRSEM image of the surface of as-prepared nanostructured LTO micrometric sized particle.

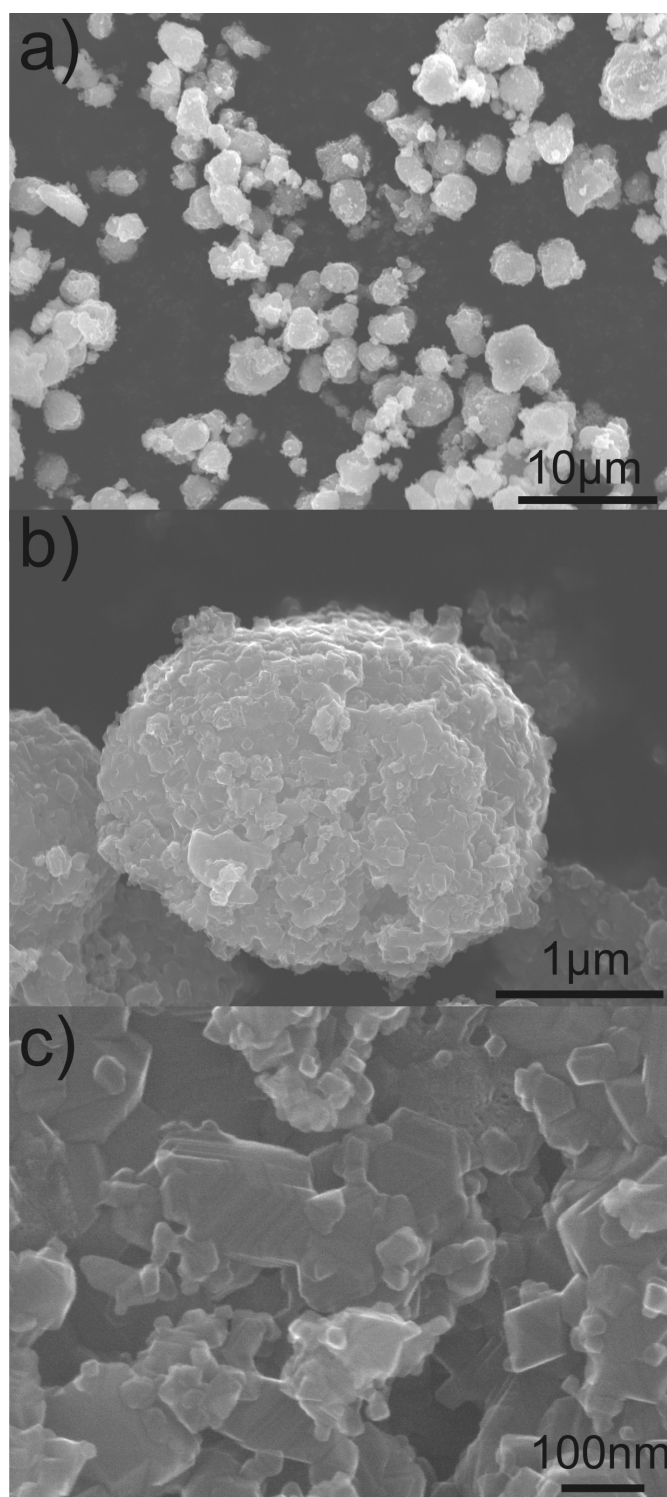


Figure S5. HRSEM images of the LTO micrometric sized particles annealed at 750 °C. a) Overview, b) single particle and c) detail of the nanostructured surface.

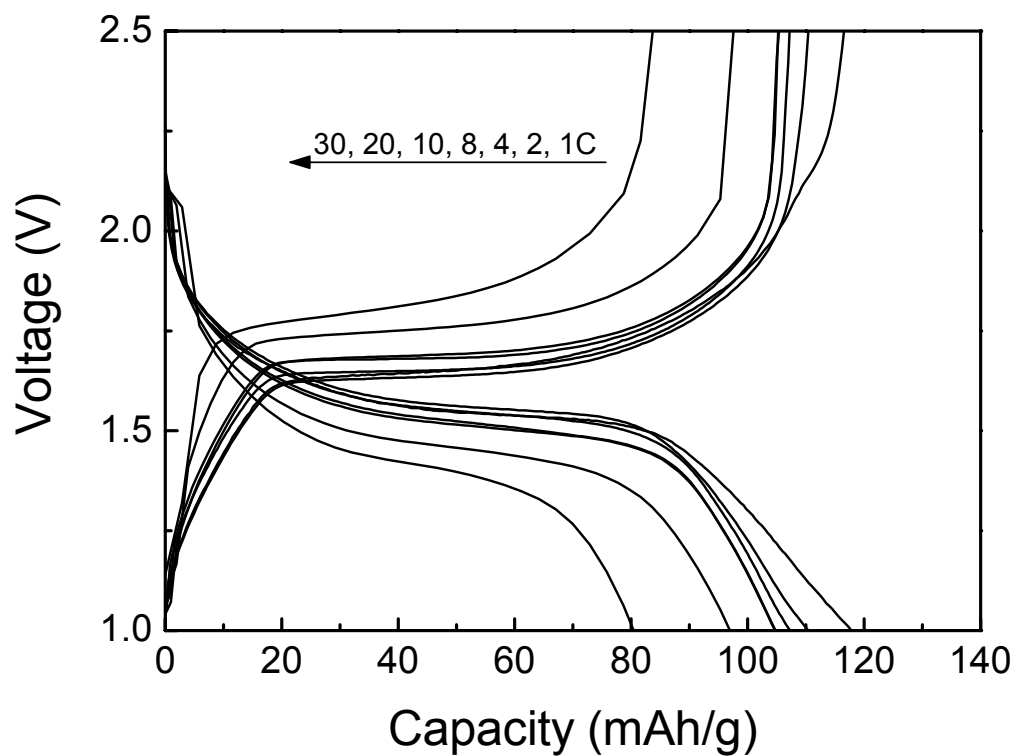


Figure S6. Capacity-voltage profiles for as-synthesized nanocrystalline $\text{Li}_4\text{Ti}_5\text{O}_{12}$

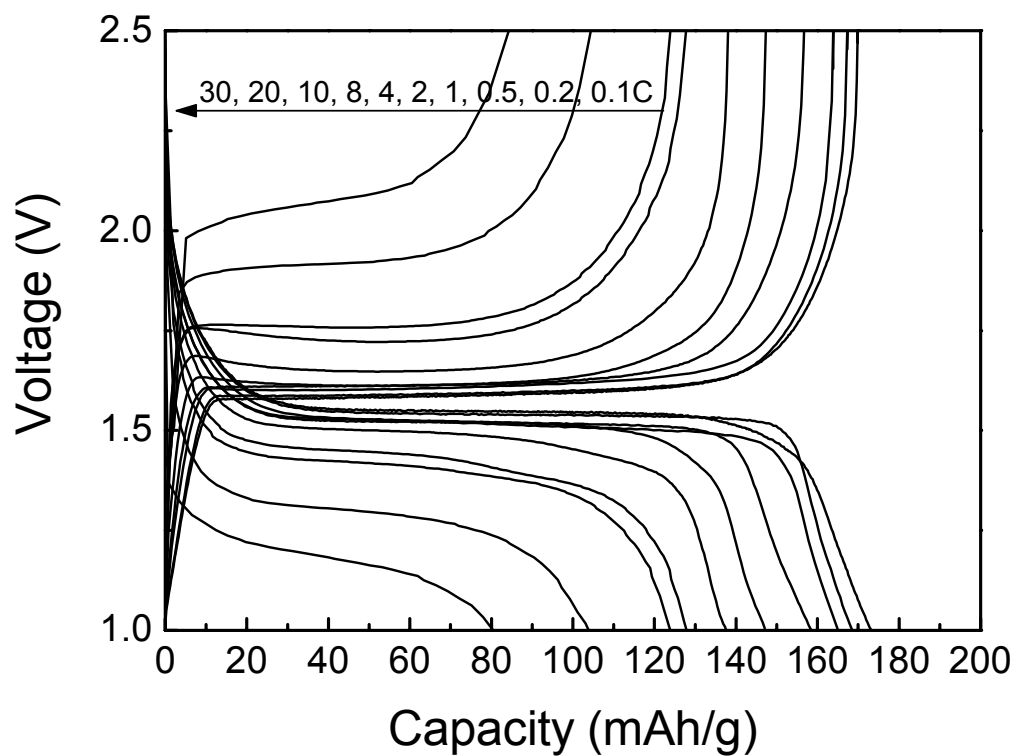


Figure S7. Capacity-voltage profiles for calcined nanocrystalline $\text{Li}_4\text{Ti}_5\text{O}_{12}$