Supplementary Material (ESI) for Journal of Materials Chemistry This journal is (c) The Royal Society of Chemistry 2010

Supporting information

Toward High Surface Area TiO₂ Brookite with Morphology Control

D. Dambournet, I. Belharouak, J. Ma, and K. Amine

Chemical Sciences and Engineering Division, Argonne National Laboratory, 9700 South Cass Avenue, Argonne, Illinois 60439 USA

Fig SI-1. Scanning electron micrographs of the sample obtained at the early stage of precipitation. a. typical large agglomerate observed in a very small fraction. Energy-dispersive X-ray spectroscopy performed on those large agglomerates indicates a titanium dioxide composition. b. For comparison, EDS analysis of rounded-type particles corresponding to the titanium oxalate hydrate phase.



Supplementary Material (ESI) for Journal of Materials Chemistry This journal is (c) The Royal Society of Chemistry 2010

Fig SI-2. X-ray diffraction powder patterns of the samples prepared during 3 and 4 hours of reaction using sodium oxalate.



Fig SI-3. Scanning electron micrographs of the titanium oxalate hydrate prepared using sodium oxalate.

