

Electronic Supplementary Information (ESI)

CuO nanoclusters coated with mesoporous SiO₂ as highly active and stable catalyst for olefin epoxidation

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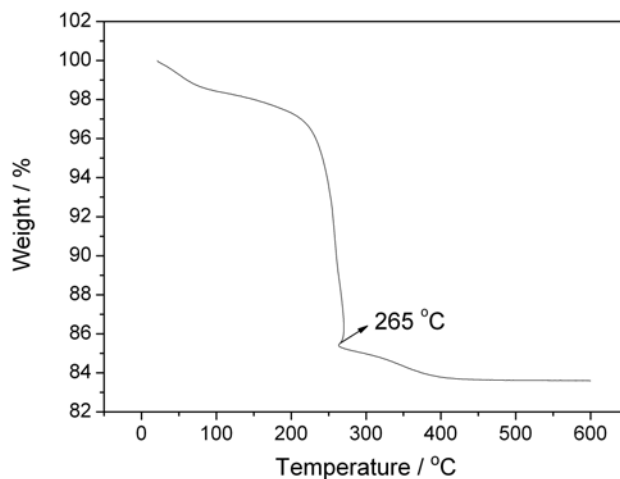


Fig. S1 TG profile of CuO CNCs. The TG plot shows three major weight loss regimes, the two areas before 250 °C can be mainly assigned to the loss of adsorbed and hydrated water in the CuO CNCs. The final one starting around 265 °C is corresponded to the degradation of PVP, resulting in 1.8% weight loss.¹

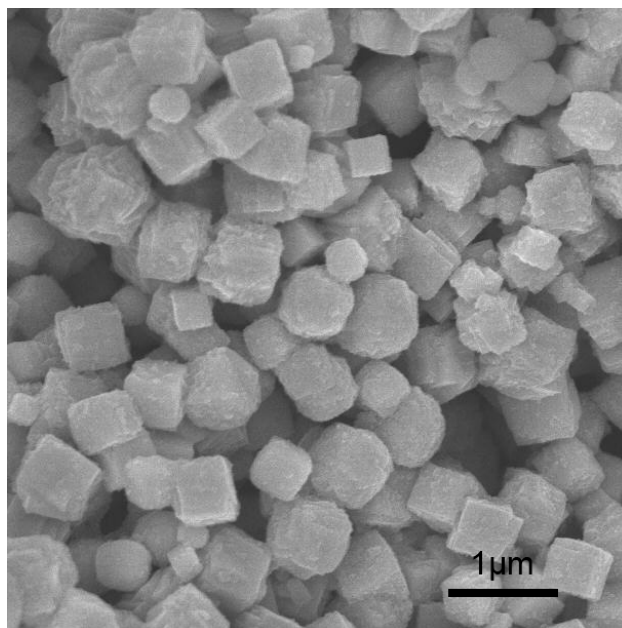


Fig. S2 SEM image of sample obtained at the same reaction conditions without the addition of surfactant PVP.

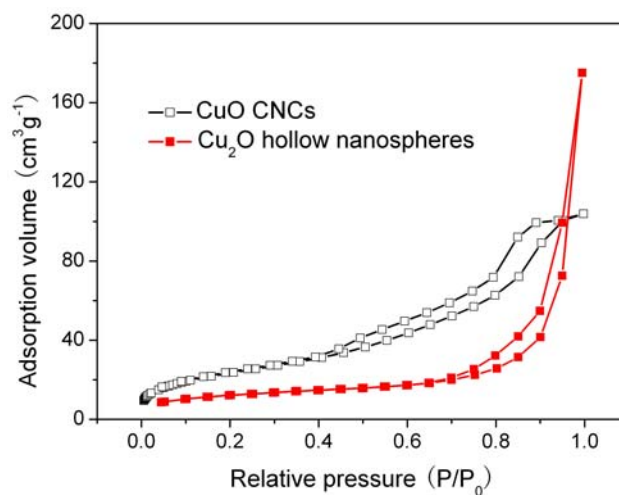


Fig. S3 N_2 adsorption-desorption isotherms of the CuO CNCs and Cu_2O hollow nanospheres.

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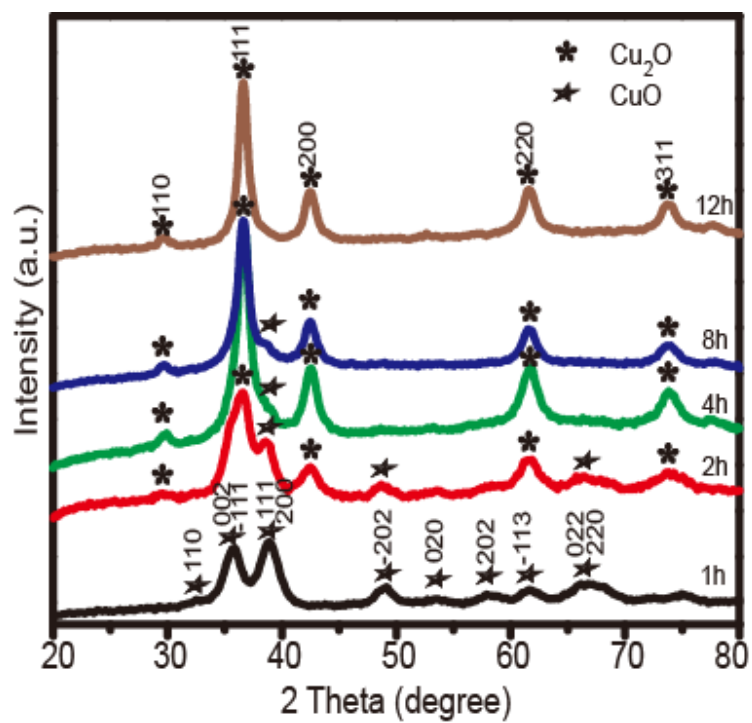


Fig. S4 X-ray diffraction patterns of samples obtained with different reaction times.

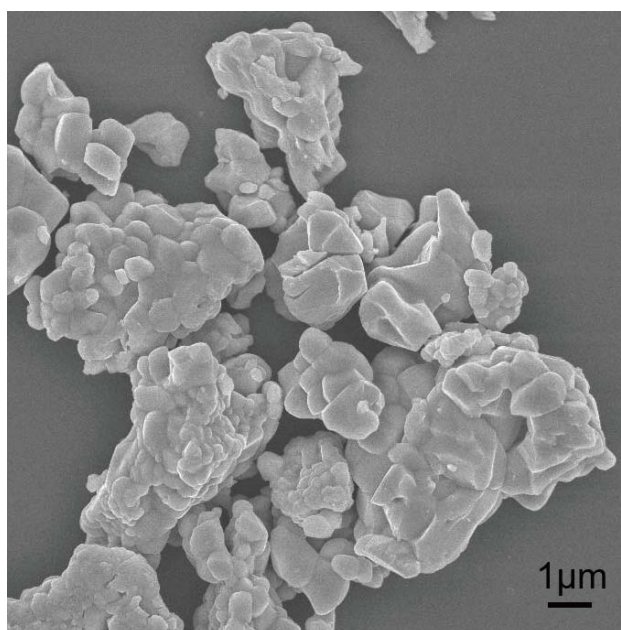


Fig. S5 SEM image of commercial available CuO.

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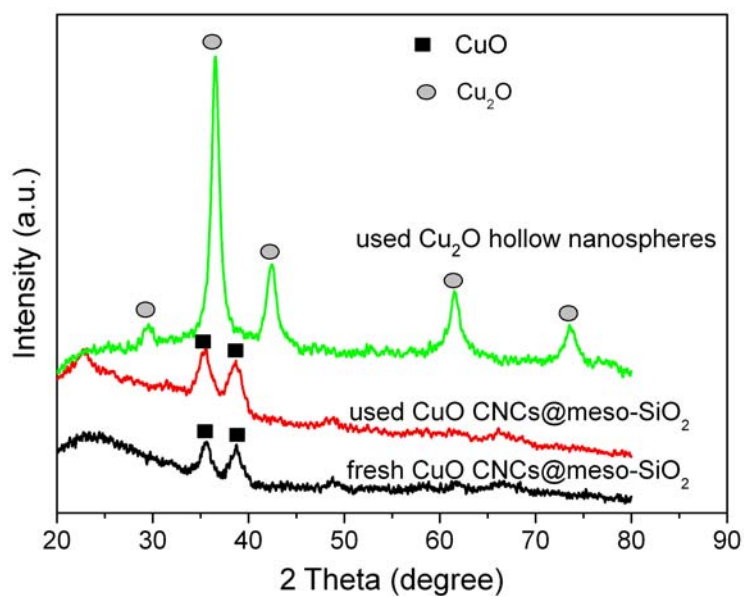


Fig. S6 X-ray diffraction patterns of fresh, used CuO CNCs@meso-SiO₂ nanocomposite and used Cu₂O hollow nanospheres.

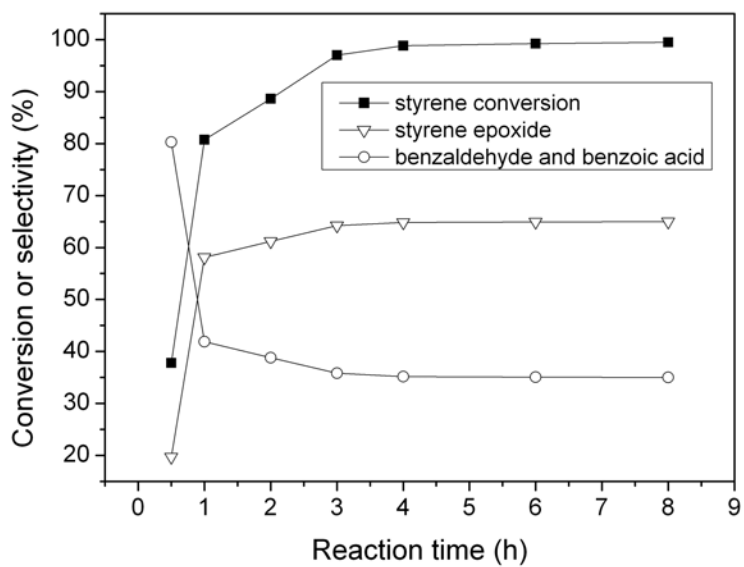


Fig. S7 The styrene epoxidation over CuO CNCs@meso-SiO₂ nanocomposite as a function of reaction time. Conditions: 1 mmol styrene, 5 mmol TBHP, 10 mg catalyst, stirred in 10 mL of acetonitrile under reflux (70°C).

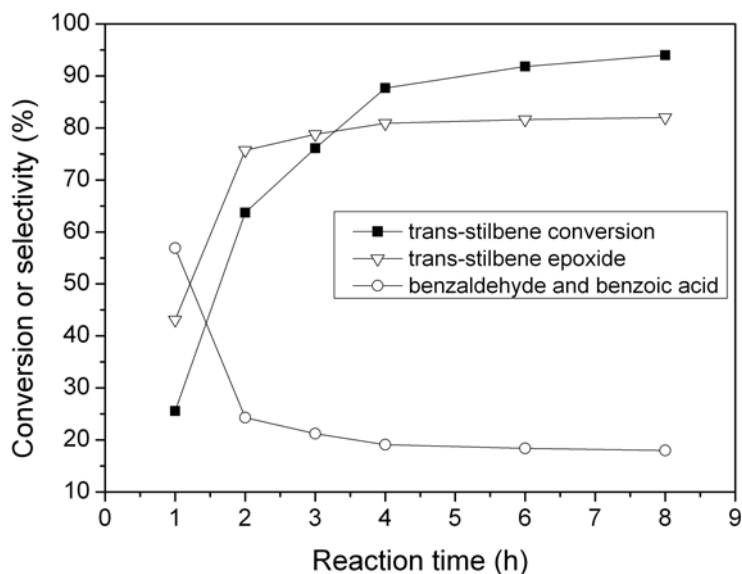
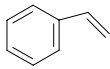
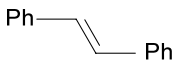


Fig. S8 The trans-stilbene epoxidation over CuO CNCs@meso-SiO₂ nanocomposite as a function of reaction time. Conditions: 1 mmol trans-stilbene, 5 mmol TBHP, 10 mg catalyst, stirred in 10 mL of acetonitrile under reflux (70°C).

Table S1 Influence of reaction temperature on alkene epoxide selectivities over CuO CNCs@meso-SiO₂ nanocomposite*

Temperature, °C						
	Time, h	Conversion, %	Selectivity of alkene epoxide, %	Time, h	Conversion, %	Selectivity of alkene epoxide, %
60	2	51.1	28.9	4	45.2	58.6
70	2	88.6	61.2	4	87.7	80.9
80	2	98.2	56.1	4	93.4	78.3

*1 mmol alkene, 5 mmol TBHP, and 10 mg catalyst, stirred in 10 mL of acetonitrile under given reaction temperature

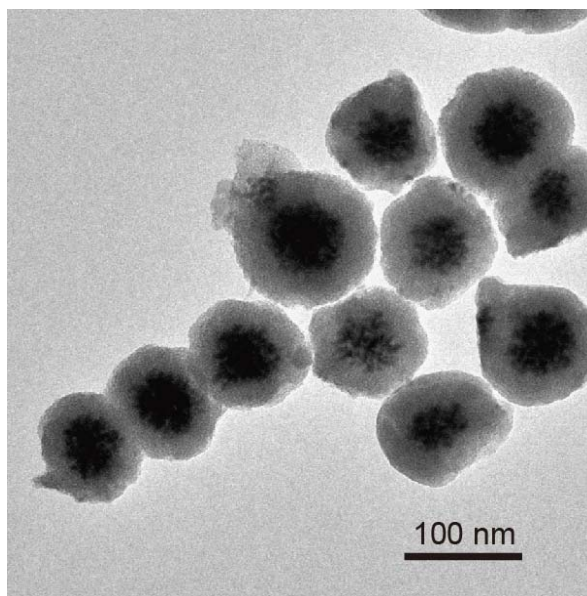


Fig. S9 TEM image of CuO CNCs@meso-SiO₂ nanocomposite stirred uninterruptedly in the same reaction mixture for 720 h.

References:

1. Azhari, S.; Diab, M. *Polymer Degradation and Stability* **1998**, *60*, 253-256.