

Supporting Information
for
Magnetic Fe₃O₄@C@Cu₂O Composites with Bean-like
Core/Shell Nanostructures: Synthesis, Properties and
Application in Recyclable Photocatalytic Degradation of Dye
Pollutants

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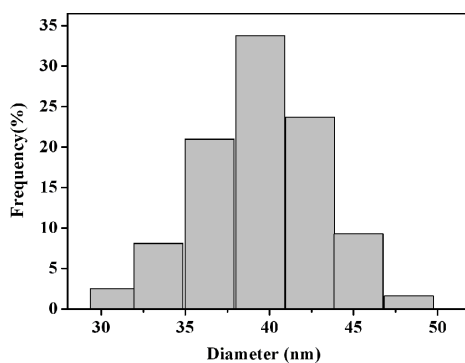


Figure S1. Histogram of size distribution of the as-prepared Fe₃O₄ nanoparticles.

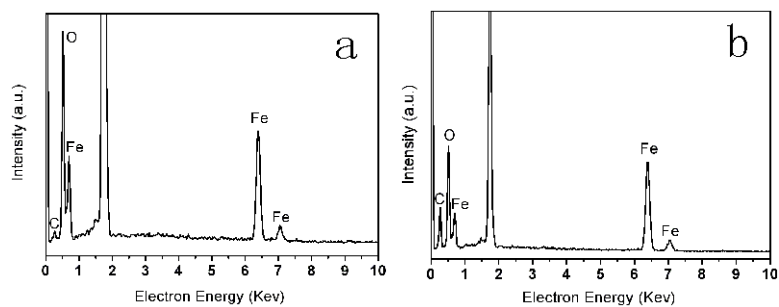


Figure S2. EDX spectra of the as-prepared Fe₃O₄ nanoparticles (a), and Fe₃O₄@C microspheres (b), respectively.

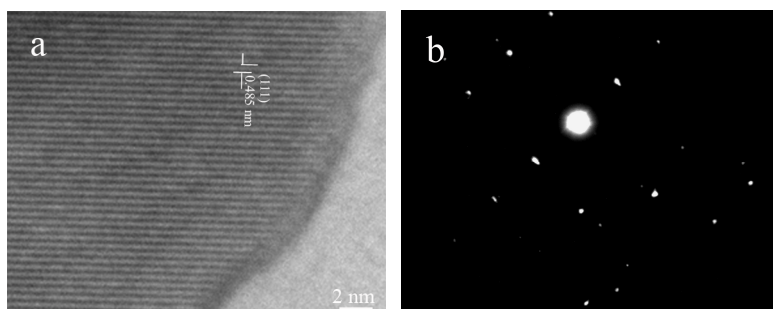


Figure S3. HRTEM image (a) and SAED pattern (b) of the as-prepared Fe₃O₄ nanoparticles.

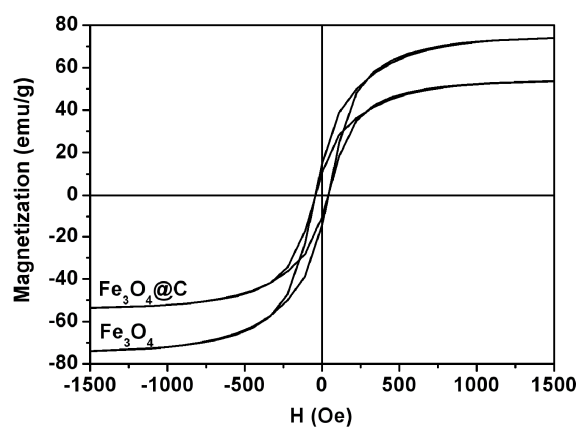


Figure S4. Hysteresis loops measured at 300 K for the as-prepared Fe₃O₄ nanoparticles and Fe₃O₄@C microspheres.

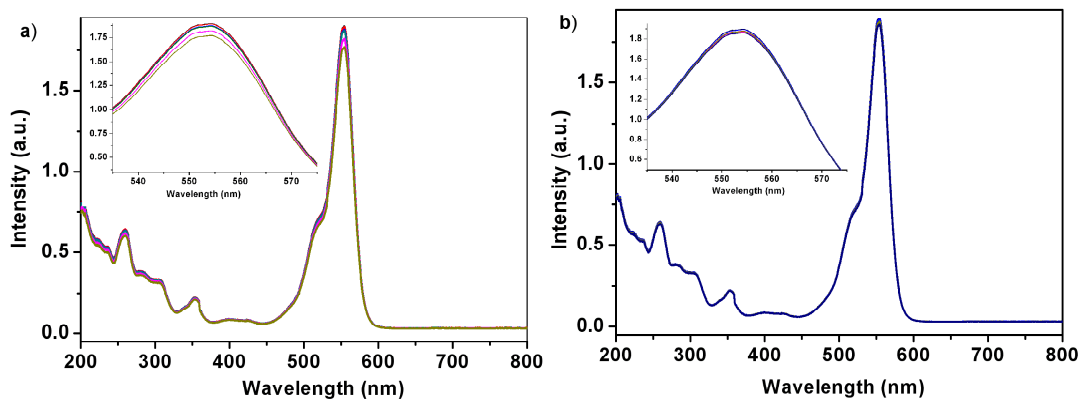


Figure S5. a) Time-dependent absorption of 50 mL 1.0×10^{-4} M RhB solution under visible-light irradiation in the absence of the as-prepared magnetic $\text{Fe}_3\text{O}_4@\text{C}@\text{Cu}_2\text{O}$ core/shell composites; b) Similar solution with (a) was performed in dark in the presence of $\text{Fe}_3\text{O}_4@\text{C}@\text{Cu}_2\text{O}$ composites (50 mg).

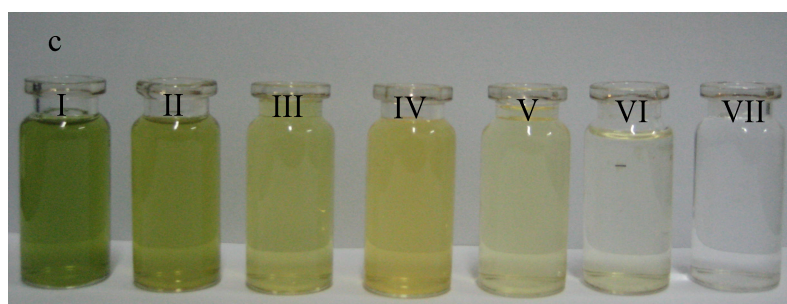
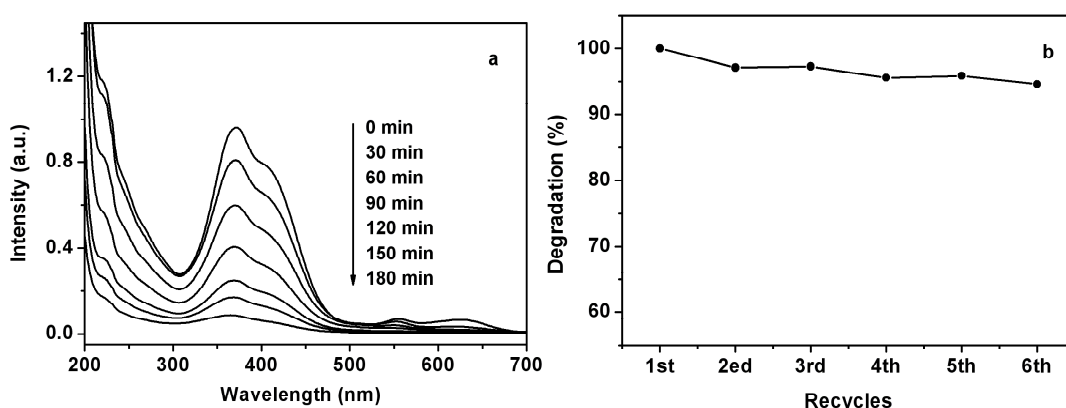


Figure S6. a) Time-dependent absorption of the practical polluted water under visible-light irradiation in the presence of the as-prepared magnetic $\text{Fe}_3\text{O}_4@\text{C}@\text{Cu}_2\text{O}$ core/shell composites (50 mg); b) 6 cycles of the degradation of the practical polluted water on the magnetic $\text{Fe}_3\text{O}_4@\text{C}@\text{Cu}_2\text{O}$ composites; c) Photographs of the corresponding degraded practical polluted water.