

Hierarchical Porous Carbonaceous Materials via Ionothermal Carbonization of Carbohydrates

By Zai-Lai Xie¹, Robin J. White², Jens Weber², Andreas Taubert^{1,2,*} and Magdalena M Titirici^{2*}

1. Institute of Chemistry, University of Potsdam, 14476 Potsdam
2. Max-Planck-Institute of Colloids and Interfaces, D-14476 Golm.

Supplementary Materials

Table S1: Macropore information obtained from Hg intrusion measurement.

<i>Samples</i>	<i>Specific area</i> <i>[m² g⁻¹]</i>	<i>Total porosity</i> <i>[%]</i>	<i>$\rho_{density}$</i> <i>[g cm⁻³]</i>
C-Glucose	76.1	92.4	0.11
C-Fructose	27.3	48.9	0.57
C-Xylose	76.6	83.2	0.30
C-Starch	52.3	60.6	0.44

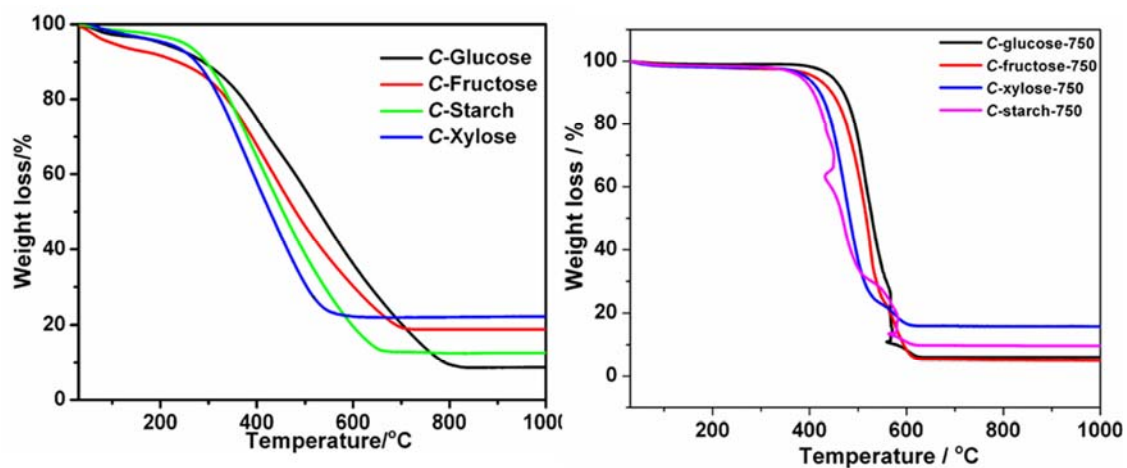


Fig S1. TGA curves of (A): all ionothermal carbons at N₂ atmosphere; (B) all post-carbonized materials at O₂ atmosphere.

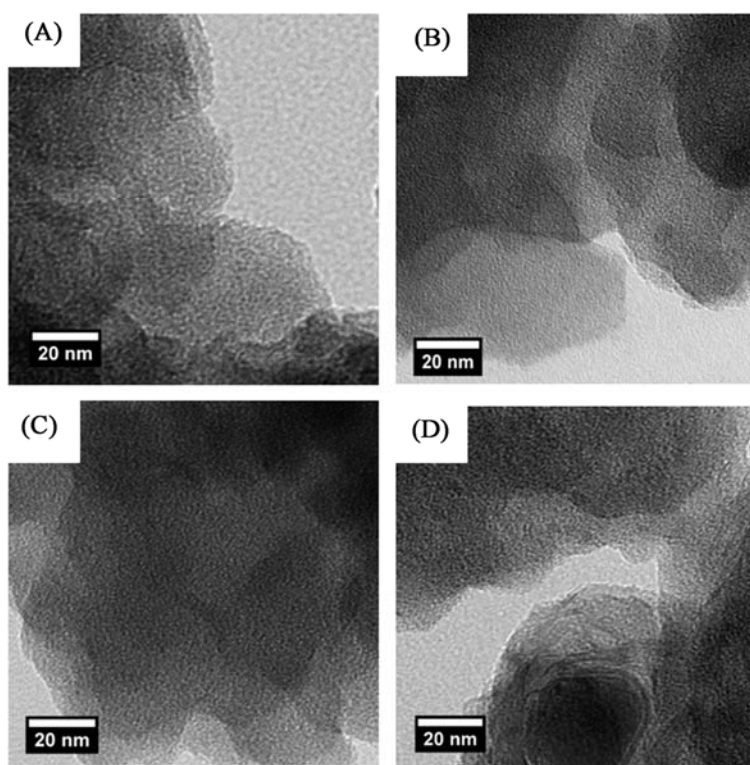


Figure S2: TEM micrographs of carbon materials after carbonization at 750 °C under N₂; (A) C-Glucose, (B) C-Fructose, (C) C-Xylose and (D) C-Starch.

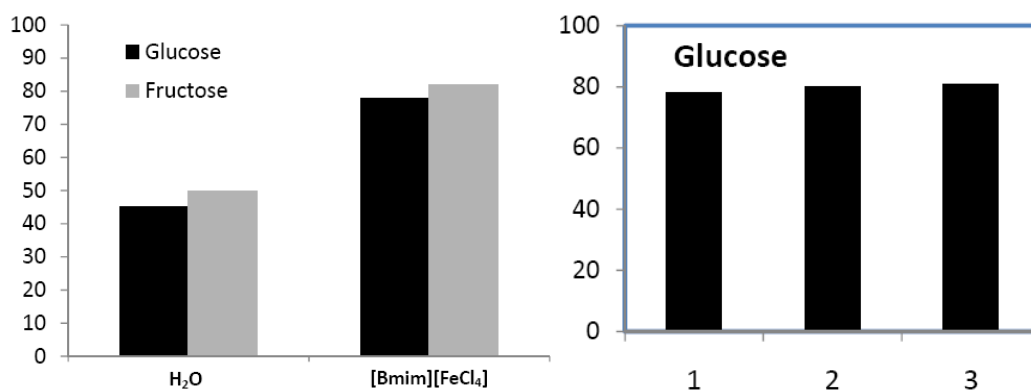


Figure S3. Yield of glucose and fructose to carbon in H₂O and [Bmim][FeCl₄] (Left) and the results of recycling of [Bmim][FeCl₄] for D-Glucose (right). A comparative experiment using the IL without Fe ([Bmim]Cl) was also tried for Glucose. The yield based on carbon is 50%.

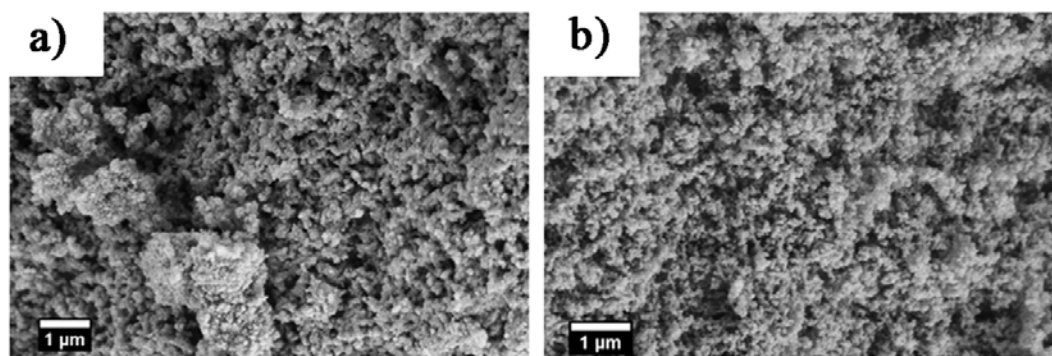


Figure S4. SEM micrographs of carbon materials from fructose. (a) 1st run, (b) 2nd run.

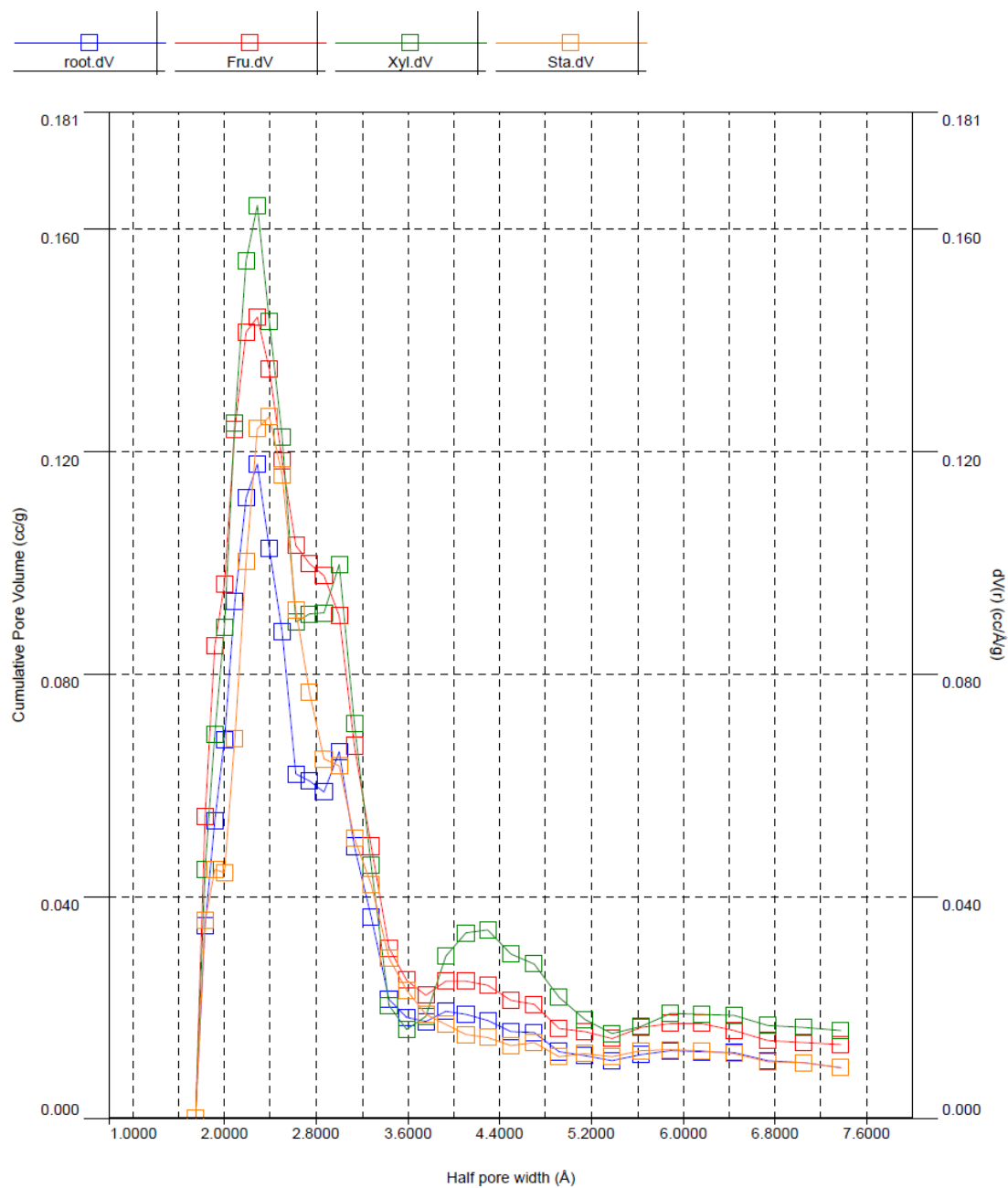


Figure S5. PSD of the post carbonized materials, extracted from CO₂ adsorption isotherms (@273K) by the GCMC model; blue: C-Glucose-750; red: C-Fructose-750; red: C-Xylose-750; orange: C-Starch-750;