Supporting Information

Solution-Processable Ambipolar Host Oligomers with High Triplet Energies for Phosphorescent Green Emitters

Shashi U. Pandya,^a Hameed A. Al Attar,^b Vygintas Jankus,^b Yonghao Zheng,^a

Martin R. Bryce^{*,a} and Andrew P. Monkman,^{*,b}

^a Department of Chemistry, Durham University, South Road, Durham, DH1 3LE, UK

^b Department of Physics, Durham University, South Road, Durham, DH1 3LE, UK

Emails: m.r.bryce@durham.ac.uk (M.R.B.); a.p.monkman@durham.ac.uk (A.P.M.)

Synthesis of 9. Following the reported procedure, ¹ **9** was obtained from its dibromo precursor² and isolated as a white solid in 68% yield.

1*H***-Tetrazole precursors.** 2-Phenyltetrazole (Aldrich), 2-(4-pyridyl)tetrazole (Aldrich) and 2-(4-*t*-butylphenyl)tetrazole (Apollo Scientific) were used as supplied. 2-(4-*n*-Octyloxyphenyl)tetrazole was synthesized following the reported procedure.³

General synthesis of oxadiazole derivatives 10-13. A mixture of 3,5-dibromobenzoic acid (1.73 g, 6.18 mmol) and thionyl chloride (10 mL) was stirred and heating at 85 °C for 4 h. Excess thionyl chloride was removed by evaporation under reduced pressure, and then to the residue was added the corresponding tetrazole derivative (6.8 mmol, 1.1. equiv.) in dry pyridine (20 mL). The mixture was stirred at 120 °C for 18 h. The cooled solution was poured into water, and the crude product was extracted with dichloromethane and washed with water three times. The organic layer was separated, dried (MgSO₄) and evaporated to give a white solid which was recrystallized from DCM/hexane to give the desired oxadiazole product.

2-(3,5-Dibromophenyl)-5-phenyl-1,3,4-oxadiazole **10.**¹ Yield: 85%.

2-(*3*,5-*Dibromophenyl*)-5-(4-*pyridyl*)-1,3,4-oxadiazole **11.** Yield: 40%. Mp: 196-197 °C. ¹H NMR (400 MHz, CDCl₃): 7.88 (1H, s), 7.99 (2H, d, *J* 6.0), 8.24 (2H, s), 8.86 (2H, d, *J* 6.0). ¹³C NMR (100 MHz, CDCl₃): 163.4, 163.0, 151.0, 137.6, 130.5, 128.6, 126.4, 123.9, 120.4. Anal. Calcd. for C₁₃H₇Br₂N₃O: C, 40.98; H, 1.85; N, 11.03; Found: C, 40.88; H, 1.82; N, 10.78. MS (EI, 70 eV): *m*/*z* 357 (M⁺).

2-(*3*,5-*Dibromophenyl*)-*5*-(*4*-*t*-*butylphenyl*)-*1*,*3*,*4*-*oxadiazole* **12.** Yield: 85%. Mp: 145-146 °C. ¹H NMR (400 MHz, CDCl₃): 1.38 (9H, s), 7.56 (2H, d, *J* 8.5), 7.84 (1H, t, *J* 1.8), 8.06 (2H, d, *J* 8.4), 8.22 (2H, d, *J* 1.8). ¹³C NMR (100 MHz, CDCl₃): 165.3, 161.9, 155.9, 136.9 128.4,

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127.2, 127.0, 126.2, 123.7, 120.6, 35.1, 31.1 Anal. Calcd. for C₁₈H₁₆Br₂N₂O: C, 49.57; H, 3.70; N, 6.42; Found: C, 49.61; H, 3.67; N, 6.35. MS (EI, 70 eV): *m/z* 421 (M⁺).

2-(3,5-Dibromophenyl)-5-(4-n-octyloxyphenyl)-1,3,4-oxadiazole **13.** Yield: 60%. Mp: 81-82 °C. ¹H NMR (200 MHz, CDCl₃) δ 8.20 (2H, d, *J* 1.7), 8.06 (2H, d, *J* 8.9), 7.83 (1H, t, *J* 1.7), 7.02 (2H, d, *J* 9.0), 4.04 (2H, t, *J* 6.5), 1.81 (2H, m), 1.38 (10H, m), 0.88 (3H, t, *J* 6.3). ¹³C NMR (100 MHz, CDCl₃) 165.2, 162.6, 161.6, 136.8, 128.9, 128.3, 127.2, 123.6, 115.5, 115.1, 70.9, 39.4, 30.5, 29.1, 23.9, 23.0, 14.0, 11.1. Anal. Calcd. for C₂₂H₂₄Br₂N₂O₂: C, 51.99; H, 4.76; N, 5.51; Found: C, 51.87; H, 4.69; N, 5.39. MS (EI, 70 eV): *m*/*z* 508(M⁺).



Figure S1. ¹H NMR spectrum of oligomer 16



Figure S2. Cyclic voltammogram of oligomer 14



Figure S3. Cyclic voltammogram of oligomer 15



Figure S4. Cyclic voltammogram of oligomer 16



Figure S5. Cyclic voltammogram of oligomer 17

Computational Data

Model compounds **A** and **B** were computed at the B3LYP/6-31G(d) level. **A** and **B** were chosen as they are the repeat units in oligomers 14 and 15, respectively, with Me replacing the octyl chains to reduce the computation time. HOMO and LUMO distribution maps of **A** and **B** are shown in Figures S6-S7. For both compounds the HOMO is located on the bicarbazolyl

unit, whereas the LUMO is on the diaryloxadiazole fragment. The HOMO-LUMO gaps are 3.50 eV for **A**, and 3.15 eV for **B**, which demonstrates that the electron-withdrawing pyridyl substituent of **B** reduces the HOMO-LUMO gap.



Figure S6. HOMO (left) and LUMO (right) distribution maps for A.



Figure S7. HOMO (left) and LUMO (right) distribution maps for B.

TABLE S1. Computational details for A

opt b3lyp/6-31g(d) geom=connectivity 1/14 = -1, 18 = 20, 19 = 15, 26 = 3, 38 = 1, 57 = 2/1, 3;2/9=110,12=2,17=6,18=5,40=1/2; 3/5=1,6=6,7=1,11=2,16=1,25=1,30=1,71=1,74=-5/1,2,3; 4//1; 5/5=2,38=5/2; 6/7=2,8=2,9=2,10=2,28=1/1; 7//1,2,3,16; 1/14=-1,18=20,19=15/3(2); 2/9=110/2; 99//99; 2/9=110/2; 3/5=1,6=6,7=1,11=2,16=1,25=1,30=1,71=1,74=-5/1,2,3; 4/5=5,16=3/1; 5/5=2,38=5/2; 7//1,2,3,16; 1/14=-1,18=20,19=15/3(-5); 2/9=110/2;6/7=2,8=2,9=2,10=2,19=2,28=1/1; 99/9=1/99; _____

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C	3 88262 -6 60844 -4 13936
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C H C H C H	-3.83124 6.15065 4.96692 -3.12885 4.47558 3.78831 -6.24742 6.48754 4.91842 -7.41139 5.10455 3.72631 -4.94429 6.86484 5.36735 -2.8247 6.43216 5.30626 -7.2182 7.41263 5.40700
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C H C H C H C C C C C	-3.83124 6.15065 4.96692 -3.12885 4.47558 3.78831 -6.24742 6.48754 4.91842 -7.41139 5.10455 3.72631 -4.94429 6.86484 5.36735 -2.8247 6.43216 5.30626 -7.2182 7.41363 5.49709 -6.51151 8.35994 6.30154 -8.59612 7.48398 5.37751 7.19331 0.35812 6.07092
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C H C H C H C C C C C C C H C C H	-3.83124 6.15065 4.96692 -3.12885 4.47558 3.78831 -6.24742 6.48754 4.91842 -7.41139 5.10455 3.72631 -4.94429 6.86484 5.36735 -2.8247 6.43216 5.30626 -7.2182 7.41363 5.49709 -6.51151 8.35994 6.30154 -8.59612 7.48398 5.37751 -7.19331 9.35812 6.97092 -9.27808 8.50041 6.06081 -9.14678 6.7599 4.76082 8.58905 9.42182 6.84278
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TABLE S2. Computational details for B

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- Brunner, K.; van Dijken, A.; Börner, H.; Bastiaansen, J. J. A. M.; Kiggen, N. M. M.; Langeveld, B. M. W. J. Am. Chem. Soc. 2004, 126, 6035-6042.
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