## **Supporting Information**

Self-Assembly of Well-ordered Whisker-like Manganese Oxide Arrays on Carbon Fiber Paper and Its Application as Electrode Materials for Supercapacitors

Yongsong Luo,<sup>a,b,g</sup> Jian Jiang,<sup>a</sup> Weiwei Zhou,<sup>a</sup> Huanping Yang,<sup>a</sup> Jingshan Luo,<sup>a</sup> Xiaoying Qi,<sup>c</sup> Hua

Zhang,<sup>c</sup> Denis Y. W. Yu,<sup>b</sup> Chang Ming Li<sup>d,e</sup> and Ting Yu<sup>\*a,b,f</sup>

<sup>a</sup> Division of Physics and Applied Physics, School of Physical and Mathematical Sciences, Nanyang Technological University, 637371, Singapore

<sup>b</sup> Energy Research Institute at Nanyang Technological University, 639789 Singapore

- <sup>c</sup> School of Materials Science and Engineering, Nanyang Technological University, 639798, Singapore
- <sup>d</sup> Institute for Clean Energy and Advanced Materials, Southwest University, Chongqing 400700, P. R China
- <sup>e</sup> Division of Bioengineering, School of Chemical and Biomedical Engineering, Nanyang Technological University, 637371, Singapore
- <sup>f</sup>Department of Physics, Faculty of Science, National University of Singapore, 117542 Singapore
- <sup>g</sup> School of Physics and Electronic Engineering, Xinyang Normal University, Xinyang 464000, P. R. China

<sup>\*</sup> To whom correspondence should be addressed: E-mail: <u>yuting@ntu.edu.sg</u>(T. Yu.).

Representive	Specific	Current density		
nanostructures	capacitance	/Scan rate	Remarks	Ref.
MnO <sub>2</sub> nanowhisker arrays			High specific capacitance,	
on CFP	274.1 F g <sup>-1</sup>	$0.1 \mathrm{A  g^{-1}}$	excellent cycling performance	Present
			Good electrochemical	
Graphene oxide/MnO <sub>2</sub>	197.2 F g <sup>-1</sup>	$200 \text{ mA g}^{-1}$	behaviors	22
			Excellent electrochemical	
Graphene/ MnO <sub>2</sub>	310 F g <sup>-1</sup>	2 mV s <sup>-1</sup>	stablity	26
			Good rate capability, excellent	
Graphene/ MnO <sub>2</sub> -textile	$315 \text{ F g}^{-1}$	2 mV s <sup>-1</sup> cycling performance		27
			Good cycle performance,	
Co <sub>3</sub> O <sub>4</sub> @MnO <sub>2</sub> arrays	480 F g <sup>-1</sup>	$2.67 \text{ Ag}^{-1}$	remarkable rate capability	46
	$660.7 \text{ F g}^{-1}$	10 mV s <sup>-1</sup>	Good cycling stability,	
MnO2 nanorod arrays	485.2 F g <sup>-1</sup>	$3 \text{ Ag}^{-1}$	excellent specific capacitance	51
			Best electrochemical capacitive	
MnO <sub>2</sub> /carbon nanotubes	162.2 F g <sup>-1</sup>	$200 \text{ mA g}^{-1}$	performance	52
Carbon nanotube				
sheet/MnO <sub>x</sub>	1250 F g <sup>-1</sup>	1 A g <sup>-1</sup>	High specific capacitance	53

Table S1	Summarization of	of the supercap	pacitor performan	ce of different MnO <sub>2</sub>	nanocomposites
----------	------------------	-----------------	-------------------	----------------------------------	----------------



Fig. S1 XRD pattern of the well-ordered MOWAs and the pristine CFP.



**Fig. S2** (a) Optical images of two pieces of samples before and after hydrothermal reaction (Enlarged image of the area marked by a rectangle); (b) Low-magnification SEM image of MOWAs; (c-e) SEM images of pristine CFP at various magnifications; (f) High-magnification SEM image of CFP after hydrothermal reaction.



**Fig. S3** (a) SEM image of MOCSs; (b) SEM image of MOCSs with few plumules; (c) SEM image of MOWAs; (d) SEM image of I-MOCSs.



**Fig. S4** Nitrogen adsorption-desorption and pore-size distribution isotherm for the obtained MOWAs products.