Electronic Supplementary Information (ESI)

Construction of Magnetic visible-light-driven Fe₃O₄@SiO₂@AgCl:Ag nanophotocatalysts

Changhua An*, Xijuan Ming, Jizhuang Wang and Shutao Wang

State Key Laboratory of Heavy Oil Processing and Department of Materials Physics and Chemistry, Key Laboratory of New Energy Physics & Materials Science in Universities of Shandong, College of Science, China University of Petroleum, Qingdao, Shandong 266555, P. R. China
*To whom correspondence should be addressed.
Fax: 86-532-86981787
Email: anchh@upc.edu.cn

Synthesis of Fe(acac)₃ particles

Saturated aqueous solution of FeCl₃ was added gradually to a mixture of ethanol and acetylacetone with the volume ratio of 1:1 to form a homogeneous solution. CH₃COONa saturated solution was then dropwisely added to the above solution until membrane-shaped product emerged. After being filtered and dried in a vacuum, a crude blood-red product was collected. Finally, the product was recrystallized in methanol.



Figure S1. XRD pattern of Fe(acac)₃ particles



Figure S2. Size distribution diagrams of (A) Fe_3O_4 nanoparticles, (B) $Fe_3O_4@SiO_2$ nanoparticles, and (C) $Fe_3O_4@SiO_2@AgCl:Ag$ nanocomposites.



Figure S3. SEM-EDS elemental mapping of $Fe_3O_4@SiO_2@AgCl:Ag$ nanocomposite. Upper left is an SEM image of the nanocomposite



Figure S4. SEM EDS line profile of the as-synthesized sample. The image shows that the target structure is composed of $Fe_3O_4@SiO_2@AgCl$.



Figure S5. (A) TEM image of big cubic shaped $Fe_3O_4@SiO_2@AgCl:Ag$ nanoparticles, (B) Higher magnification image of labeled B in (A), (C, D) HRTEM images of C, D portion in (B). These images clearly show that $Fe_3O_4@SiO_2@AgCl$ nanoparticles are successfully constructed via the present facile solution-based synthetic route.



Figure S6 HRTEM image of small spherical nanoparticles. The image reveals that small nanoparticles also possess the target $Fe_3O_4@SiO_2@AgCl$ structures.

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Figure S7. The normalized concentration of the MO (A) and MB (B) molecules as a function of reaction time. The inset is dye color changes with the reaction proceeds.



Figure S8. The dye concentration during photodegradation as a function of reaction time with the addition of different scavengers.