

Electronic Supplementary Information

Phase Selective Synthesis of Metastable Orthorhombic Cu₂ZnSnS₄

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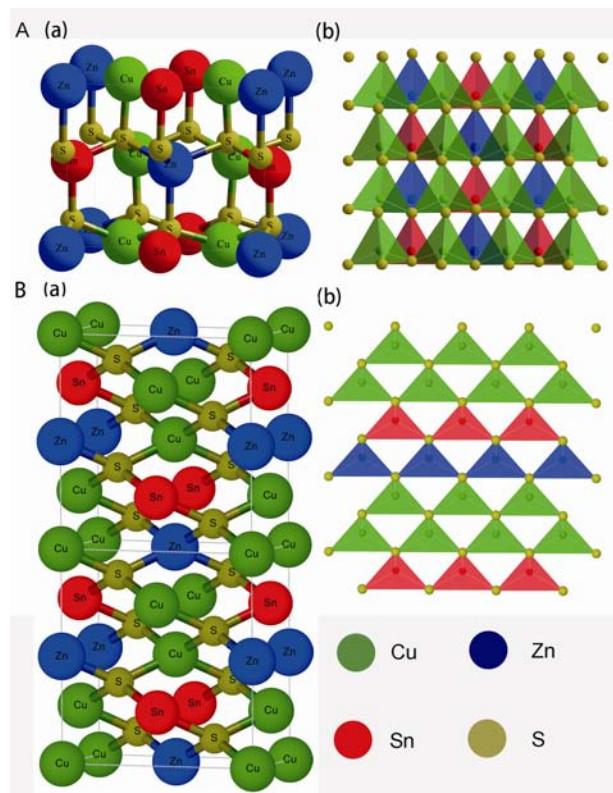


Figure S1 Crystal structures of orthorhombic CZTS (A) and kesterite CZTS (B).

| | |
|--|-----------------------------|
| Molecular | $\text{Cu}_2\text{ZnSnS}_4$ |
| Crystal system | Orthorhombic |
| Space group | $\text{Pmn}2_1$ |
| Number of formula units per unit cell | 2 |
| a (nm) | 0.75385 |
| b (nm) | 0.64304 |
| c (nm) | 0.62038 |
| Cell volume (nm³) | 0.30073 |
| Number of atoms in the cell | 16 |

Table S1 Crystal Data of the orthorhombic CZTS.

| Atom | x/a | y/b | z/c |
|------|--------|--------|--------|
| Cu | 0.2504 | 0.3270 | 0 |
| Zn | 0 | 0.8411 | 0.9971 |
| Sn | 0 | 0.1786 | 0.4982 |
| S1 | 0.2335 | 0.3420 | 0.3688 |
| S2 | 0 | 0.2068 | 0.864 |
| S3 | 0 | 0.8480 | 0.398 |

Table S2 Atomic coordinates for the $\text{Cu}_2\text{ZnSnS}_4$ compound.¹

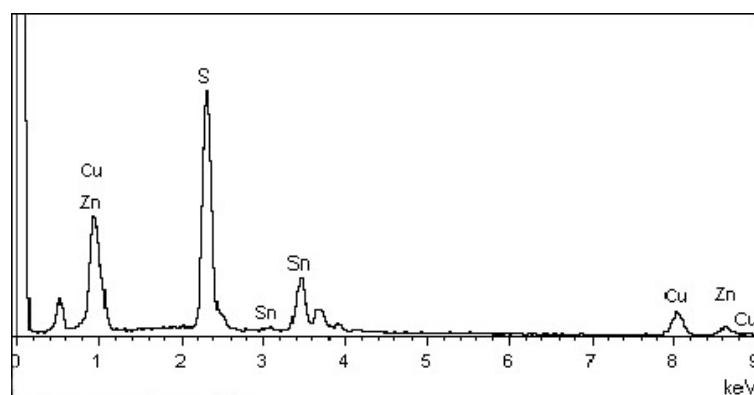


Figure S2. Spectrum of the Energy Dispersive X-ray Spectroscopy (EDS) of wurtzite-stannite CZTS nanocrystals.

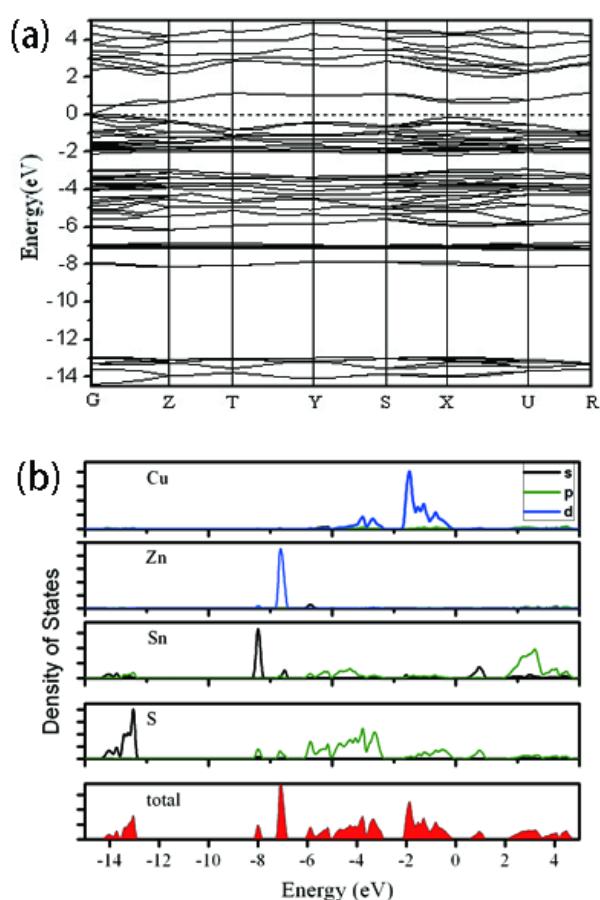


Figure S3. (a) Band structures of as-synthesized orthorhombic CZTS. (b) DOS and PDOS of orthorhombic CZTS.

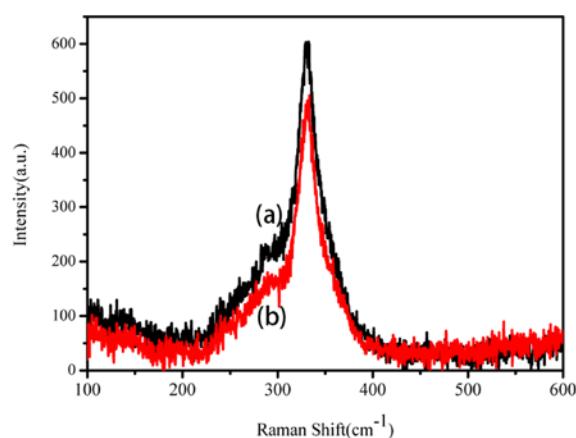


Figure S4. Raman spectra of orthorhombic CZTS synthesized in ultrapure water (a) and the mixed solvent of ultrapure water and ethylenediamine (b).

1. O. V. Parasyuka, Y. E. Romanyuk and I. D. Olekseyuk, *Journal of Crystal Growth*, 2005, 275, 4.