Electronic Supplementary Information (ESI)

Metal-Free Organic Dyes Derived from Triphenylethylene for Dye-Sensitized Solar Cells: Tuning of the Performance by Phenothiazine and Carbazole

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Fig. S1. UV-vis absorption spectra of C3, P2, C2 and P3 dyes in different solvents.



Fig. S2.UV-vis absorption spectra of 10^{-5} M C3, P2, C2 and P3 dyes in THF/H₂O (1:1) mixture with different acidity and alkalinity.



Fig. S3.UV-vis absorption spectra of P3 on TiO₂ film.

Table S1 P3 amount vs time						
Dye	P3 (8 h)	P3 (16 h)	P3 :CDCA (1:1, 16 h)			
Amount / 10^{-7} mol cm $^{-2}$	4.1	4.1	3.7			



Fig. S4. EIS Nyquist and Bode plots of dye P3 with coadsorbent (CDCA, 1:1) in the dark.

Lfetime: With CDCA: 381.3 * 0.001072=408.8 ms Without CDCA: 481.1 ms



Fig. S5 Photocurrent density-photovoltage curves of P3 in the dark.

Table 52 Terrormance of D55C under unterent conditions							
Dye	$J_{\rm sc}$ / mA cm ⁻²	$V_{\rm oc}/~{ m mV}$	$\eta(\%)$	Fill factor (ff)			
8 h without CDCA	12.09	822	6.36	0.64			
16 h without CDCA	12.18	826	6.55	0.65			
16h with CDCA	10.94	806	6.07	0.69			

Table S2 Performance of DSSC under different conditions



MS and NMR spectra

















Fig. S10 ¹H NMR spectrum of P2Br



Fig. S11 ¹³C NMR spectrum of **P2Br**.







Fig. S13 ¹H NMR spectrum of P2B.















Fig. S17 ¹³C NMR spectrum of 4a.







Fig. S19 ¹H NMR spectrum of 4b.











Fig. S22 ¹H NMR spectrum of 5a.



Fig. S23 ¹³C NMR spectrum of 5a.







Fig. S25 ¹H NMR spectrum of 5b.











Fig. S29 ¹³C NMR spectrum of **P2**.



















Fig. S35 13 C NMR spectrum of C3.







Fig. S37 ¹H NMR spectrum of C2.



