## **Supporting Information**

## Uncoordinated carbonyl groups of MOFs as anchoring sites for the preparation of highly active Pd nano-catalysts

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Empirical formula	$C_9H_{13}N_3O_{11}Tm$		
Formula weight	508.15	$D_{\rm c}({\rm mg/cm}^3)$	1.905
Crystal system	Triclinic	F (000)	490
Space group	<i>P</i> -1	$\mu (\text{mm}^{-1})$	5.064
<i>a</i> / Å	6.0200(5)	Refections collected	4623
<i>b</i> / Å	12.7722(11)	Independent reflections	3147 ( $R_{int} = 0.0279$ )
<i>c</i> / Å	13.1612(11)	Data/restraints/parameters	3147/5/212
$\alpha/^{\circ}$	118.8290(10)	GOF	1.047
$eta/^{\circ}$	90.6090(10)	$\theta$ range for data collection (°	)1.77 to 25.20
γ/°	91.6870(10)	Final <i>R</i> indices $[I > 2\sigma(I)]$	$R_1 = 0.0402$
			$\omega R_2 = 0.0987$
$V(\text{\AA}^3)$	885.79(13)	R indices (all data)	$R_1 = 0.0466$
			$\omega R_2 = 0.1036$
Ζ	2	Largest diff. peak an	d 1.98 and 0.99
		hole/eÅ <sup>-3</sup>	

Table S1. Crystal data and structure refinement for Tm-MOF.

 $R = \sum (F_{o} \mid - \mid F_{c}) / \sum \mid F_{o} \mid$ .

 $wR = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o)^2]^{1/2}.$ 

The Palladium contents of the samples were determined quantitatively by atomic absorption spectroscopy (AAS) on a HITACHI Z-2300 instrument. The calibration curve were obtained with a series of aqueous standard solutions of  $Pd(NO_3)_2$ . Approximately 0.0040 g of sample was weighed precisely and dissolved in 5 mL of agua regia (HCl:HNO<sub>3</sub> = 3:1), and the solution was diluted with deionized water to reach a concentration of Pd approximately in the range from 0.5 to 1.5 mg L<sup>-1</sup>.



Figure S1. The AAS calibration curve for Pd using aqueous standard solution.



**Figure S2.** Powder X-ray diffraction patterns of ZIF-8: (a) stimulated, (b) as-synthesized, and (c) 1.0 wt% Pd/ZIF-8.