

CuGaO₂ a promising alternative for NiO in p-type dye solar cells

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Experimental details

¹⁵ Electrochemical impedance spectroscopy (EIS) measurements.

Pellets for electrochemical impedance spectroscopy (EIS) measurements were formed from CuGaO₂ nanoparticles pressed under 100 bars and sintered at 450°C for 3 hours in nitrogen atmosphere. The back side was painted with a silver glue and contacted with copper coil before being mounted in a polypropylene tube using chemically resistant epoxy. EIS measurements were carried out with a potentiostat/galvanostat model VSP ²⁰ from Biologic Sciences Instruments on polished samples (used SiC-Paper, grit 1200 and 4000 (Struers)) in an electrolyte composed of 1 M LiClO₄ in water (pH ~ 6.3) with a platinum counter-electrode and a saturated calomel reference electrode (SCE).

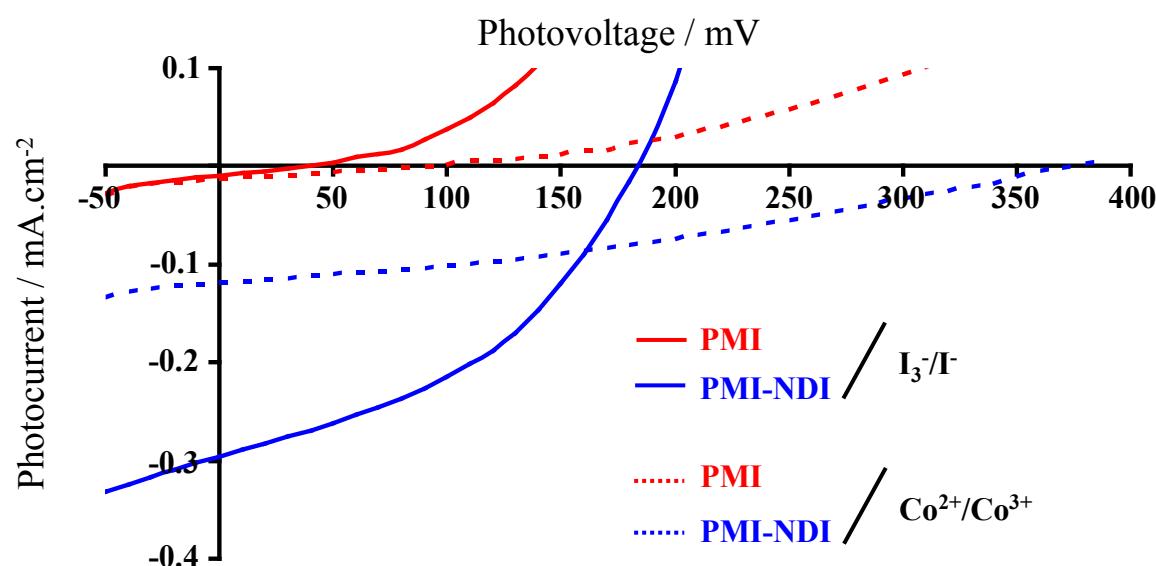
Preparation of the CuGaO_2 photocathode.

The doctor Blade paste is composed of CuGaO_2 nanoparticles ball milled in an adapted mixture of organic ingredients to have a suitable rheology to homogeneously spread CuGaO_2 on FTO (Solems 10 Ohm TEC7) (see table S1).

5 **Table S1.** Detailed composition of the paste for doctor blading.

	Component	Quantity
Semiconductor	CuGaO_2	100 mg
Binder	PEG 300 (Aldrich)	20.5 mg
Rheological agent	Ethylcellulose (Aldrich)	4 mg
Acidifying agent	4-hydrobenzoic acid (Merck)	2 mg
Solvant	2-ethyl-1-hexanol (Aldrich)	750 μL

The as-obtained electrodes were then dipped at room temperature for 48 hours into an acetonitrile solution, containing either perylene monoimide (PMI) or perylene monoimide sensitizer connected to a naphthalene diimine electron acceptor (dyad PMI-NDI). The dyed electrodes were thoroughly rinsed with acetonitrile.



10 **Fig. S1.** Photoresponse under AM1.5 illumination (1000 W/m^2) of solar cells constructed from CuGaO_2 p-SC with dyad PMI-NDI or PMI as sensitizer, and I/I_3^- or $\text{Co}^{2+}/\text{Co}^{3+}$ in PC as redox mediator.