

Supplementary Material (ESI) for Journal of Materials Chemistry

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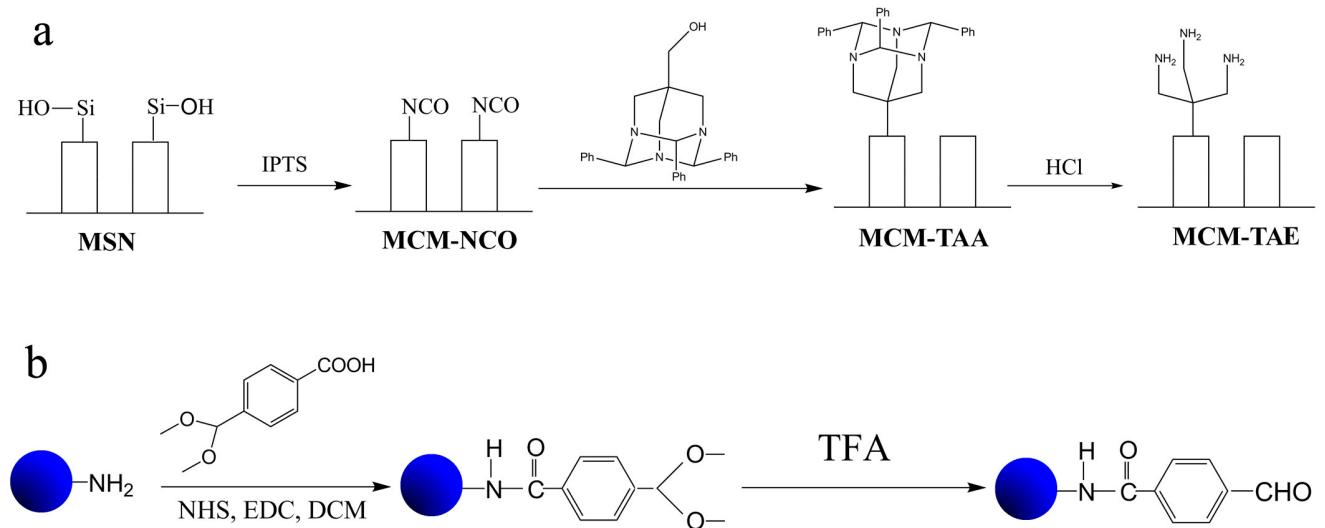
**Endosomal pH-activatable magnetic nanoparticles-capped mesoporous  
silica for intracellular controlled release**

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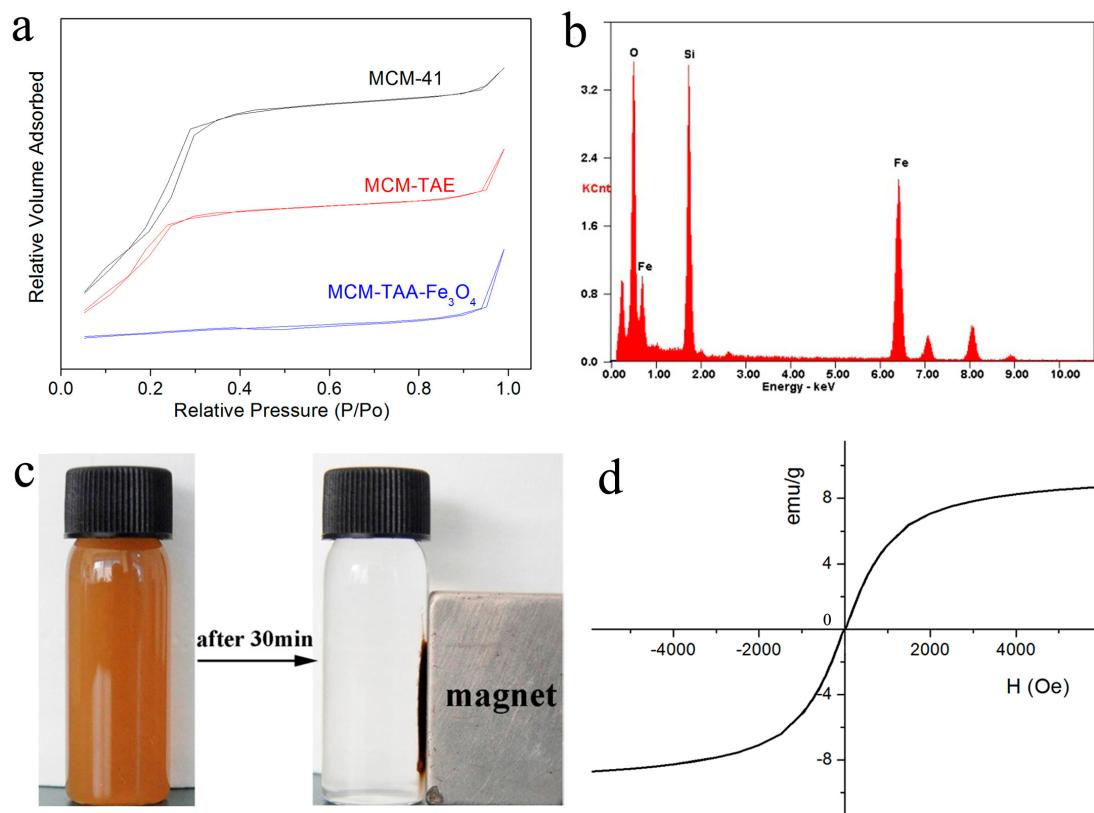
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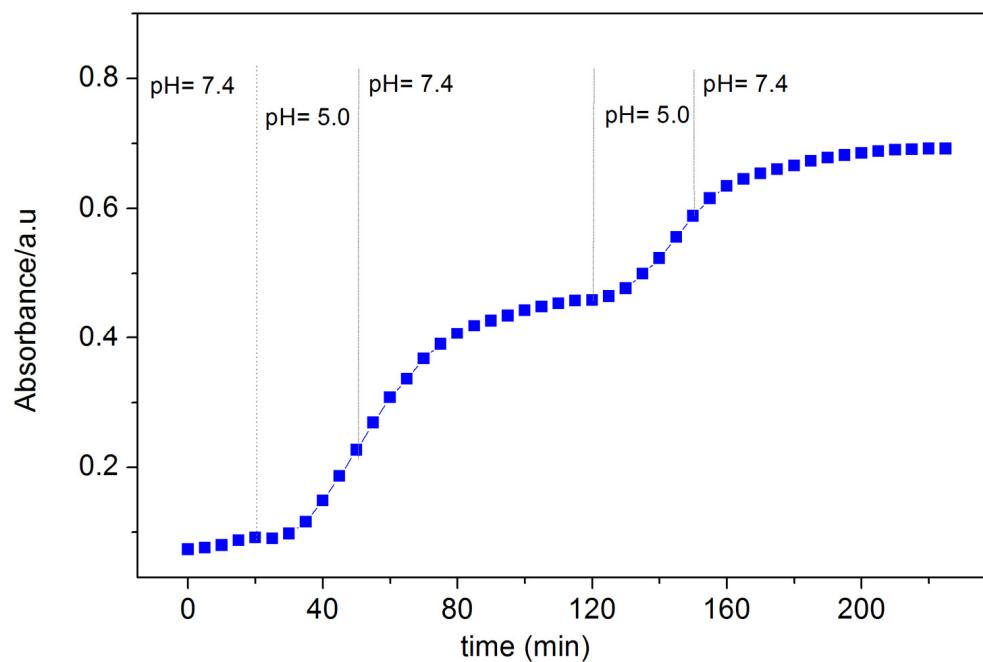
**SUPPORTING INFORMATION**



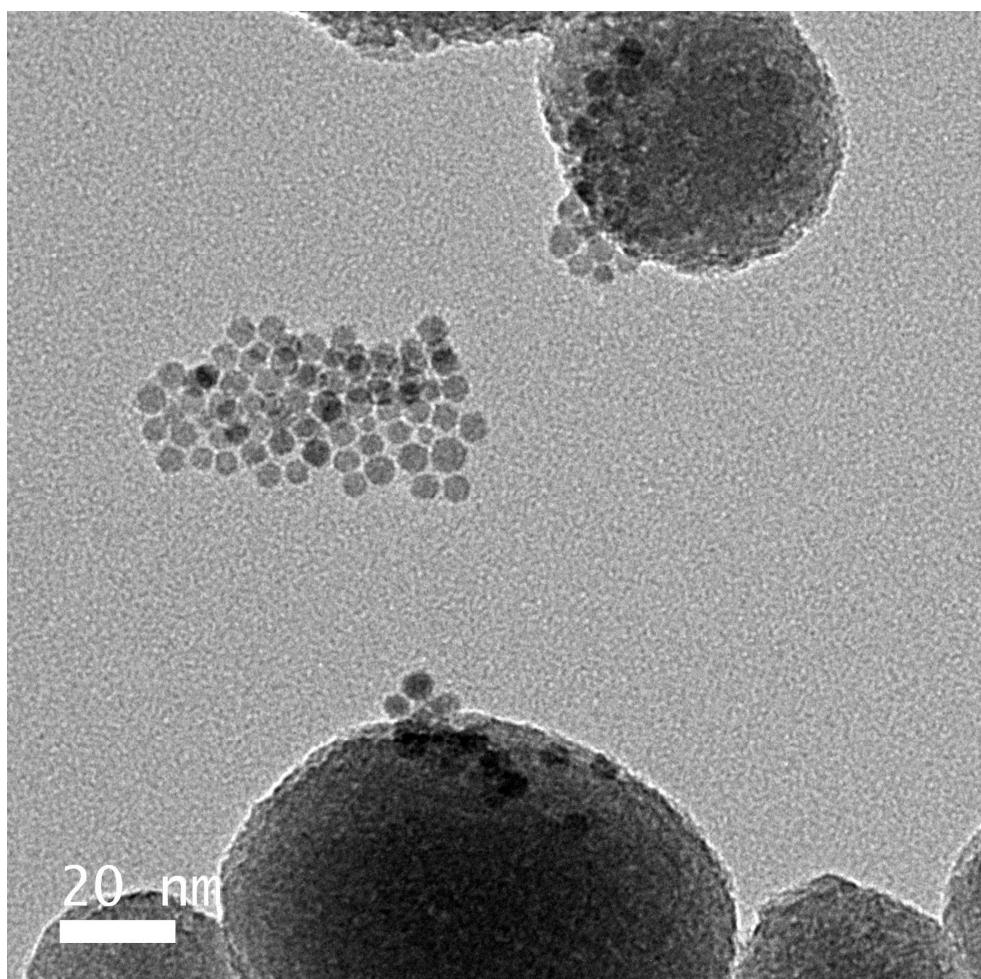
**Scheme S1.** Schematic synthesis outline of MCM-TAE (a) and Fe<sub>3</sub>O<sub>4</sub>-CBA (b).



**Fig. S1** (a) N<sub>2</sub> adsorption/desorption isotherms of MCM-41 as-synthesized, tri(aminomethyl)ethane-functionalized MCM-41 (MCM-TAE), and the solid MCM-TAA-Fe<sub>3</sub>O<sub>4</sub>, (b) EDS spectra of the solid MCM-TAA-Fe<sub>3</sub>O<sub>4</sub>, (c) photographs show the homogeneous aqueous solution, and the MCM-TAA-Fe<sub>3</sub>O<sub>4</sub> after magnetic separation from the solution under an external magnetic field, and (d) field-dependent magnetization at 300K of MCM-TAA-Fe<sub>3</sub>O<sub>4</sub>.



**Fig. S2** The dexamethasone released profile from MCM-TAA- $\text{Fe}_3\text{O}_4$  with two pH stimuli.



**Fig. S3** TEM image of MCM-TAA- $\text{Fe}_3\text{O}_4$  nanoparticles at pH 5.0.

**Table S1.** Zeta-potentials of Fe<sub>3</sub>O<sub>4</sub>-CBA, MCM-TAE and MCM-TAA-Fe<sub>3</sub>O<sub>4</sub> nanopaticles at pH 5.0 and 7.4.

Zeta-potentials (mV)	Fe <sub>3</sub> O <sub>4</sub> -CBA	MCM-TAE	MCM-TAA-Fe <sub>3</sub> O <sub>4</sub>
pH 5.0	+19.5	+19.1	-
pH 7.4	-14.3	-33.9	-22.8