Supporting information

Synthesis and evaluation of high-birefringence polymethacrylate having diphenyl-diacetylene LC moiety in side chain

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Instruments

The ¹H NMR and ¹³C NMR spectra were measured in CDCl₃ on a JEOL LNM-EX 400 at room temperature using tetramethylsilane (TMS) as an internal standard. The FT-IR spectra were recorded on a JASCO FT-IR 460 plus spectrometer. Gel permeation chromatography (GPC) was carried out by a JASCO UV-2070 detector and a JASCO RI-2031 detector (TOSOH TSKgel G4000H_{XL} column) using tetrahydrofuran (THF) as the eluent after calibration with polystyrene standards. The transition behavior was investigated by polarizing optical microscopy (POM) (Leica DM2500P microscopy with a Mettler FP90 hot stage) and differential scanning calorimetry (Perkin Elmer DSC7) with heating and cooling scans performed at 10 °Cmin⁻¹. The transmittance of light was observed by a microscope spectroscopic method using a Nikon LV100 Pol optical microscope equipped with a USB4000 (Ocean photonics) spectrometer.

Materials

Trimethylsilylacetylene, Pd(PPh₃)₄, 4-(hexyloxy)benzaldehyde, 6-bromo-1-hexanol, 4-iodophenol, tetrabutylammoniumfluoride (TBAF), *tert*-butyldimethylchlorosilane and methacryloylchloride were purchased from TCI, and PPh₃, CuI, ZnCl₂, 2-bromo-6-naphthol, triethylamine, N-ethyldiisopropylamine, chloromethylmethylether and tetrabromomethane were purchased from Wako, and *n*-butyllithium was purchased from Kanto chemical. Unless otherwise noted, all chemical were commercially available and use as received.

Synthesis

4-(6-Hydroxyhexyloxy)iodobenzene (1)

A mixture of 4-iodophenol (5.5 g, 25.1 mmol), 6-bromo-1-hexanol (5.0 g, 27.6 mmol),

potassium carbonate (4.2 g, 30.1 mmol) and DMF (50 mL) was refluxed for 18 h. The insoluble solids were removed by filtration with diethylether, and the filtrate was extracted with diethylether, washed with brine, and dried over MgSO₄. After removing the solvents, the residue was purified by silica gel column chromatography (EtOAc/hexane = 1/3) to afford the title compound as a colorless solid. **Yield**: >99%; ¹**H NMR** (400 MHz, CDCl₃) δ 7.54 (d, 2H, *J* = 8.5 Hz), 6.67 (d, 2H, *J* = 8.5 Hz), 3.92 (t, 2H, *J* = 6.5 Hz), 3.66 (dt, 2H, *J* = 4.0, 7.0 Hz,), 1.79 (tt, 2H, *J* = 6.5, 7.4 Hz), 1.64-1.36 (m, 6H), 1.26 (t, 1H, *J* = 7.0 Hz) ppm.

4-(6-Hydroxyhexyloxy)-1-(2-trimethylsilylethynyl)benzene (2)

To a mixture of 4-(6-hydroxyhexyloxy)iodobenzene (8.0 g, 25 mmol), Pd(PPh₃)₄ (0.58 g, 0.050 mmol), CuI (95 mg, 0.050 mmol), PPh₃ (0.13 g, 0.050 mmol), and triethylamine (20 mL) in THF (20 mL) was added trimethylsilylacetylene (5.5 mL, 40 mmol), and it was stirred at 45 °C for 2 days. A portion of diethylether was added, and insoluble solids were removed by filtration. The filtrate was extracted with diethylether, consequently washed with aqueous 2 M HCl and brine, and dried by MgSO₄. After removal of solvents, the residue was purified by silica gel column chromatography (EtOAc/hexane = 1/2) to afford the title compound as a pale yellow liquid. **Yield**: >99%; ¹**H NMR** (400 MHz, CDCl₃): δ 7.34 (d, 2H, *J* = 8.3 Hz), 6.80 (d, 2H, *J* = 8.3 Hz), 3.95 (t, 2H, *J* = 6.6 Hz), 3.69-3.63 (brm, 2H), 1.79 (tt, 2H, *J* = 6.6, 6.6 Hz), 1.64-1.1.39 (m, 6H), 1.26 (t, 1H, *J* = 7.2 Hz) ppm.

4-(6-Hydroxyhexyloxy)-1-ethynylbenzene (3)

To a solution of 4-(6-hydroxyhexyloxy)-(2-trimethylsilylethynyl)benzene (3.07 g, 14.1 mmol) in THF (50 mL) was added TBAF (1.0 M in THF solution, 16 mL, 16 mmol), and stirred for 15 min at room temperature. A portion of water was added, and the

mixture was extracted with diethylether, washed with brine and dried over MgSO₄. After removal of solvents, the residue was purified by silica gel column chromatography (EtOAc/hexane = 1/3) to afford the title compound as a colorless solid. **Yield**: 68%; ¹**H NMR** (400 MHz, CDCl₃) δ 7.41 (d, 2H, *J* = 8.8 Hz), 6.82 (d, 2H, *J* = 8.8 Hz), 3.96 (t, 2H, *J* = 6.5 Hz), 3.68-3.64 (brm, 2H), 2.99 (s, 1H), 1.80 (tt, 2H, *J* = 6.5, 7.4 Hz), 1.64-1.41 (m, 6H), 1.22 (brm, 1H) ppm.

4-[6-(tert-Butyldimethylsiloxy)hexyloxy]-1-ethynylbenzene (4)

To a solution of **3** (1.10 g, 5.00 mmol) and imidazole (0.45 g, 6.50 mmol) in CH₂Cl₂ (30 mL) was added tert-butyl-dimethyl-silylchloride (0.98 g, 6.50 mmol). The reaction mixture was stirred at room temperature for 3 h. The reaction mixture was extracted with CH₂Cl₂ and washed with water. The organic layer was dried over MgSO₄ and concentrated in vacuo. The crude product was purified by flash column chromatography on silica gel (eluent: ethyl acetate/ hexane = 1/10) to afford the title compound as a colorless liquid. **Yield**: >99%; ¹**H** NMR (400 MHz, CDCl₃) δ 7.41, (d, 2H, *J* = 8.6 Hz), 6.82 (d, 2H, *J* = 8.8 Hz), 3.95 (t, 2H, *J* = 6.5 Hz), 3.61 (t, 2H, *J* = 6.5 Hz), 2.99 (s, 1H), 1.79 (tt, 2H, *J* = 6.5, 7.3 Hz), 1.56-1.34 (m, 6H), 0.89 (s, 9H), 0.04 (s, 6H) ppm.

4-(2,2-Dibromoethenyl)-1-hexyloxybenzene (5)

A solution of PPh₃ (10.5 g, 40 mmol) in CH₂Cl₂ was added tetrabromomethane (6.6 g, 20 mmol) at 0 °C. After stirring for 15 min, 4-hexyloxybenzaldehyde (2.10 g, 10 mmol) was added to the mixture, and it was stirred for further 3 h. A portion of methanol and water were sequentially added, and the mixture was extracted with chloroform, sequentially washed with aqueous NaS₂O₃ and brine, and dried over MgSO4. After removal of solvents, the residue was purified by silica gel column chromatography (EtOAc/hexane = 1/10) to afford the title compound as a colorless liquid. **Yield**: >99%;

¹**H NMR** (400 MHz, CDCl₃) δ 7.50, (d, 2H, *J* = 7.8 Hz), 7.40 (s, 1H), 6.88 (d, 2H, *J* = 7.8 Hz), 3.96 (t, 2H, *J* = 6.6 Hz), 1.78 (tt, 2H, *J* = 6.6, 7.4 Hz), 1.49-1.29 (m, 6H), 0.91 (t, 3H, *J* = 7.1 Hz) ppm.

2-Bromo-1-(4-hexyloxyphenyl)-4-[4-(6-*tert*-butyldimethylsiloxyhexyloxy)phenyl]-1-(Z)butene-3yne (6)

To a solution of acetylene **4** (1.9 g, 5.7 mmol) in THF (30 mL) was added *n*-BuLi (2.6 M in hexane, 2.2 mL, 5.8 mmol) at -78 °C. After stirring 30 min, a solution of dry ZnCl₂ (0.79 g, 5.8 mmol) in THF (15 mL) was added to the mixture, and it was stirred for further 15 min. The mixture was allowed to warm to room temperature over 30 min. This mixture was added to another flask at 0 °C, which contained a mixture of dibromostyrene **5** (2.0 g, 5.7 mmol) and Pd(PPh₃)₄ (0.33 g, 0.28 mmol) in THF (5 mL). After stirring at 0 °C for 2 days, a portion of aqueous NH₄Cl was added, and it was extracted with diethylether, sequentially washed with aqueous NaHCO₃ and brine, and dried over MgSO₄. After removal of solvents, the residue was purified by silica gel column chromatography (EtOAc/hexane = 1/10) to afford the title compound as a colorless solid. **Yield**: 47%; ¹**H NMR** (400 MHz, CDCl₃) δ 7.69 (d, 2H, *J* = 8.8 Hz), 7.41 (d, 2H, *J* = 8.8 Hz), 7.21 (s, 1H), 6.89 (d, 2H, *J* = 8.8 Hz), 6.85 (d, 2H, *J* = 8.8 Hz), 3.97 (dt, 4H, *J* = 6.0, 6.3 Hz), 3.62 (t, 2H, *J* = 6.5 Hz), 1.86-1.73 (m, 4H), 1.53-1.31(m, 12H), 0.93-0.88 (m, 12H), 0.05 (s, 6H) ppm.

1-(4-Hexyloxyphenyl)-4-{4-[6-(*tert*-butyldimethylsiloxy)hexyloxy]phenyl}-1,3-butad iyne (7)

To a solution of compound **6** (1.60 g, 2.60 mmol) in THF (20 ml) was added tetra-*n*-butylammonium fluoride (1.0 M in THF, 4.80 mL, 4.80 mmol), and the mixture was stirred for 1 day at room temperature. A portion of water was added, and the

mixture was extracted with chloroform, washed with brine, and dried over MgSO₄. After removal of solvents, the residue was purified by silica gel column chromatography (CHCl₃/EtOAc/hexane = 1/1/1) to afford the title compound as a colorless solid. **Yield**: 96%; ¹**H NMR** (400 MHz, CDCl₃) δ 7.44, (dd, 2H, *J* = 4.4 Hz), 7.44, (dd, 2H, *J* = 4.4 Hz), 6.83 (d, 2H, *J* = 4.4 Hz), 6.83 (d, 2H, *J* = 4.4 Hz), 3.98-3.94 (m, 4H), 3.66 (t, 2H, *J* = 6.6 Hz), 1.83-1.74 (m, 4H), 1.64-1.33 (m, 12H), 1.27-1.20 (brs, 1H) 0.90 (t, 3H, *J* = 6.8 Hz) ppm; **HRMS** (FAB) M⁺ Calcd for C₂₈H₃₄O₃ 418.2503; Found, 418.2500.

1-(4-Hexyloxyphenyl)-4-[4-(6-hydroxyhexyloxy)phenyl]-1,3-butadiyne (8)

To a solution of compound **7** (0.97 g, 2.30 mmol) and triethylamine (3.00 ml, 3.00 mmol) in CH₂Cl₂ (35 mL) was added methacryloyl chloride (0.29 mL, 3.00 mmol), and the mixture was stirred at room temperature for 7 h. A portion of 2 M HCl was added, and the mixture was extracted with chloroform, washed with water, and dried over MgSO₄. After removal of solvents, the residue was purified by silica gel column chromatography (EtOAc/hexane = 1/2) to afford the title compound as a colorless solid. **Yield**: 67%; ¹**H NMR** (400 MHz, CDCl₃) δ 7.44 (d, 2H, *J* = 4.3 Hz), 7.44 (d, 2H, *J* = 4.3 Hz), 6.83 (d, 2H, *J* = 4.4 Hz), 6.83 (d, 2H, *J* = 4.4 Hz), 6.10 (d, 1H, *J* = 1.7 Hz), 5.55 (d, 1H, *J* = 1.7 Hz), 4.16 (t, 2H, *J* = 6.6 Hz), 3.98-3.94 (m, 4H), 1.94 (s, 3H), 1.82-1.67 (m, 6H), 1.53-1.31 (m, 10H), 0.91 (t, 3H, *J* = 6.7 Hz) ppm; ¹³**C NMR** (400 MHz, CDCl₃) δ 167.5, 159.8, 159.7, 136.4, 134.0, 134.0, 125.2, 114.6, 114.6, 113.7, 113.6, 81.3, 81.2, 72.9, 72.8, 68.1, 67.8, 64.6, 31.5, 29.1, 29.0, 28.5, 25.7, 25.7, 25.6, 22.6, 18.3, 14.0 ppm; **HRMS** (FAB) M⁺ Calcd for C₃₂H₃₈O₄ 486.2770; Found, 486.2774.

Polymerization

To a Schlenk tube, which contained **8** (0.10 g, 0.21 mmol) and THF (0.41 ml) was added dropwise BuLi (2.6 M in hexane, 3.9 μ L, 0.010 mmol) at 0 °C, and the mixture was stirred for 24 h at the same temperature. The mixture was poured dropwise into excess methanol, and it was filtrated. The solids were purified by reprecipitation with MeOH to afford the title polymer as a colorless solid. The polymerization result explained following: **Yield** = 40%, Mn = 7990, Mw / Mn = 1.69. ¹H NMR (400 MHz, CDCl₃) δ 7.46-7.35 (brm, 4H), 6.85-6.74 (brm, 4H), 4.02-3.81 (br, 6H), 1.82-1.08 (brm, 23H), 1.94 (s, 3H), 0.93-0.81 (m, 3H) ppm.

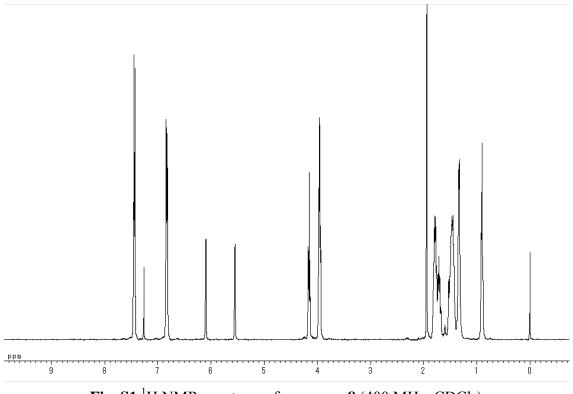


Fig. S1 ¹H NMR spectrum of monomer 8 (400 MHz, CDCl₃).

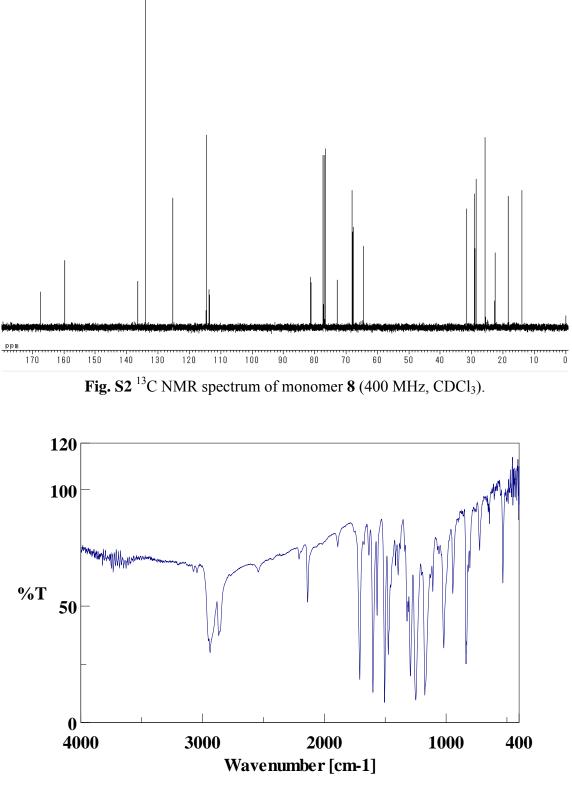


Fig. S3 FT-IR spectrum of monomer 8 (KBr).

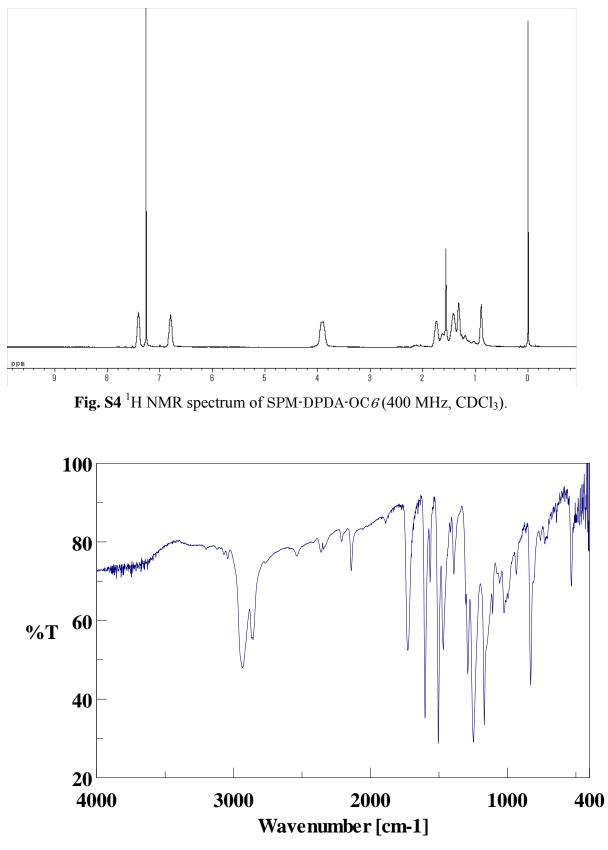


Fig. S5 FT-IR spectrum of SPM-DPDA-OC6(KBr).

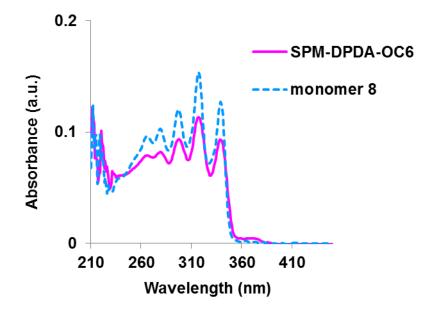


Fig. S6 UV-visible spectra of SPM-DPDA-OC6 and monomer 8.

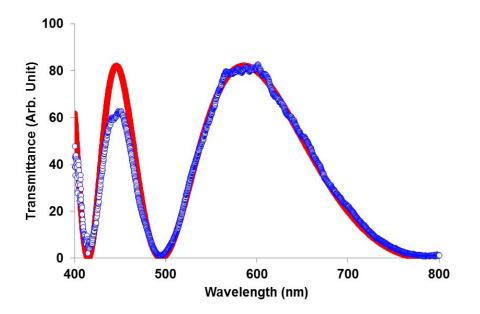


Fig. S7 Wavelength dependence of light intensity (open circles) transmitted of SPM-DPDA-OC6 under cross polarization condition. And the solid curve is the fitting of equation (1).

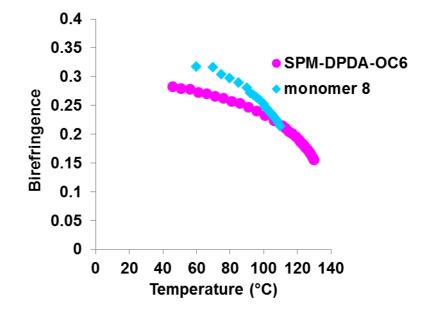


Fig. S8 Temperature dependence of birefringence measured at wavelength of 550 nm for SPM-DPDA-OC6 and monomer 8.