

Microwave-assisted Hydrothermal Synthesis of Perovskite NaTaO₃ Nanocrystals and its Photocatalytic Properties

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Supporting Information

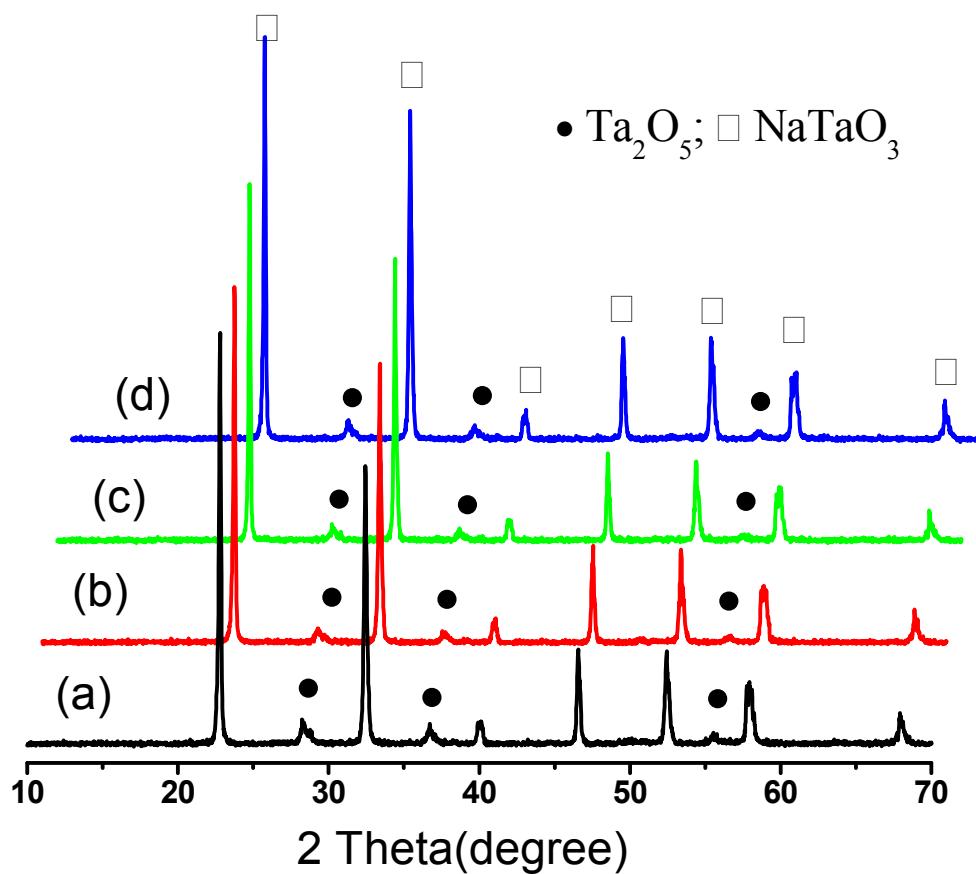


Fig. S1 XRD patterns of the as-prepared samples derived at 160 °C in 1.75 M NaOH solution for (a) 6 h by MHT technique, (b) 12 h, (c) 18 h, and (d) 24 h by CHT method.

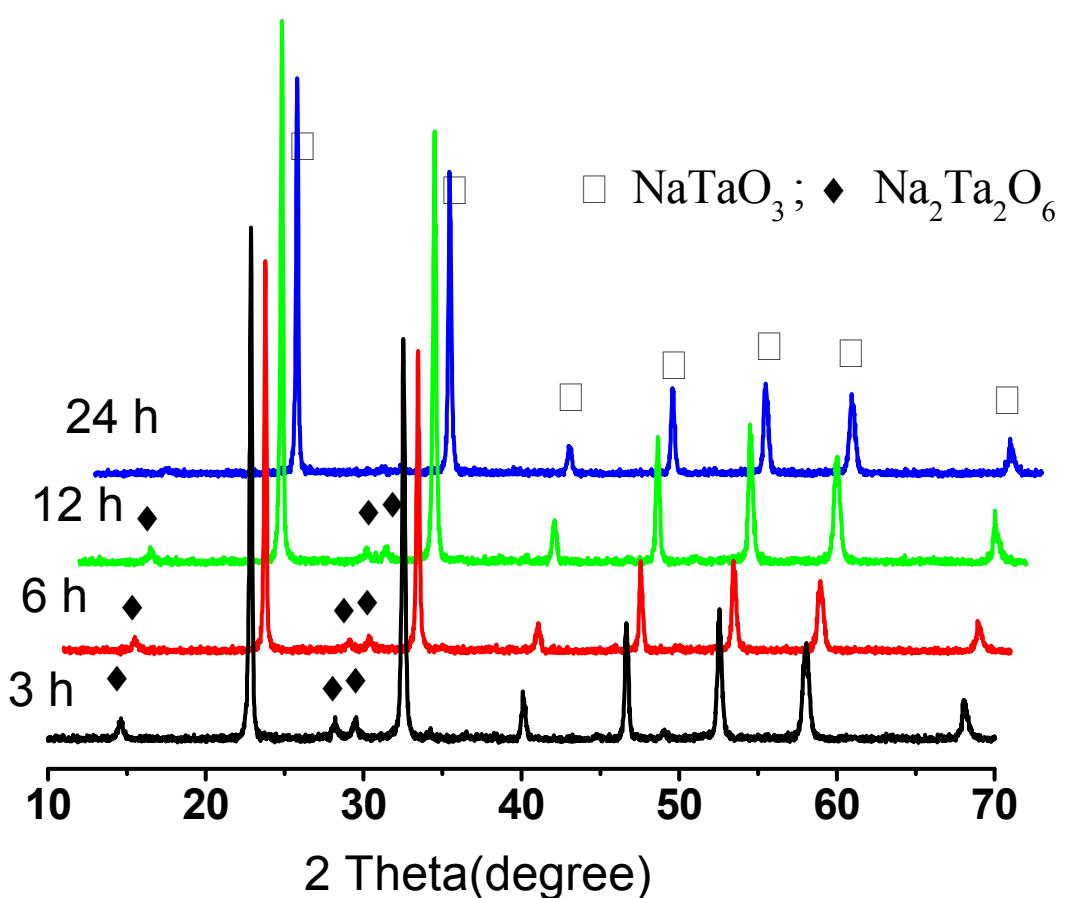


Fig. S2 XRD patterns of the as-prepared samples derived at 160 °C in 1.75 M NaOH solution for different reaction times with 6-hour-ball milled Ta₂O₅ as starting material in the CHT process.

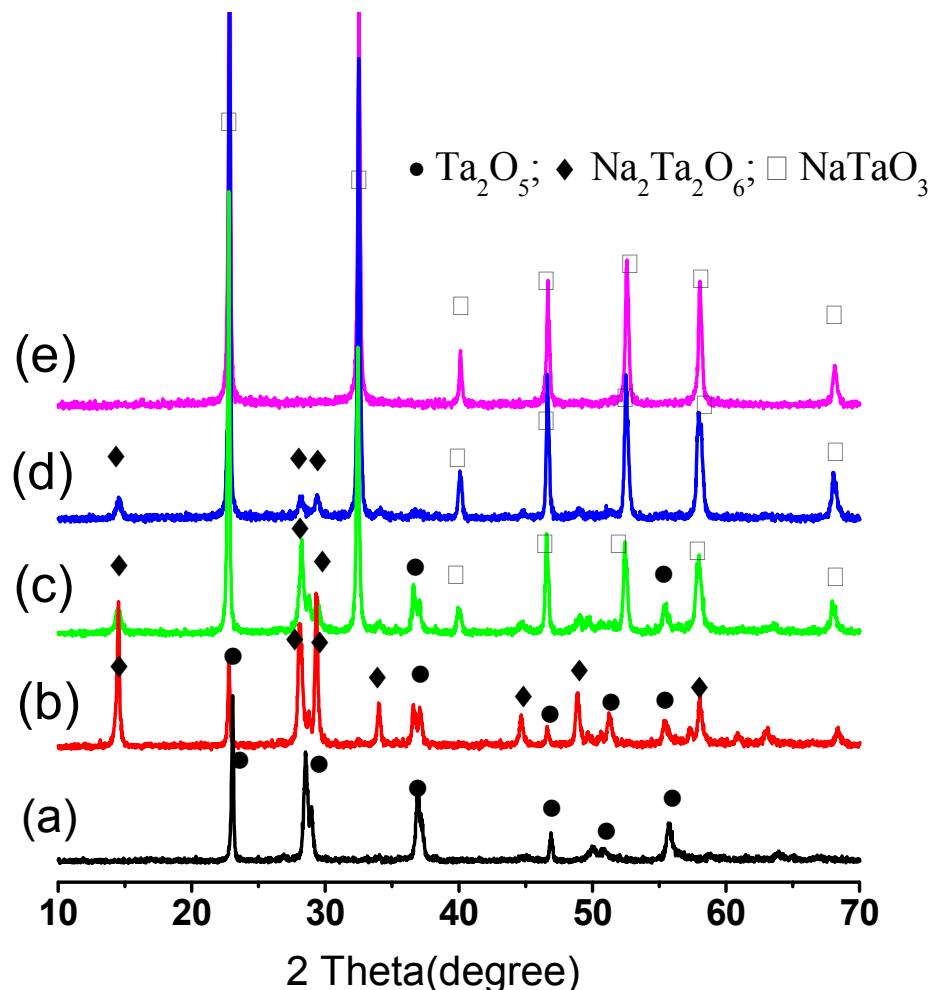


Fig. S3 Phase evolution from (a) to (e) detected by XRD analysis during the MHT reaction: (a) Ta_2O_5 raw material, (b) mixed phases of Ta_2O_5 and $Na_2Ta_2O_6$ obtained in 0.2 M NaOH solution after 3 h, (c) mixed phases of Ta_2O_5 , $Na_2Ta_2O_6$ and $NaTaO_3$ obtained in 1.0 M NaOH solution after 5 minutes with 12-hour-ball milled Ta_2O_5 as raw material, (d) mixed phases of $Na_2Ta_2O_6$ and $NaTaO_3$ obtained in 0.5 M NaOH solution, and (e) pure $NaTaO_3$ phase obtained in 1.75 M NaOH solution after 3 h with 6-hour-ball milled Ta_2O_5 as raw material at 160 °C.

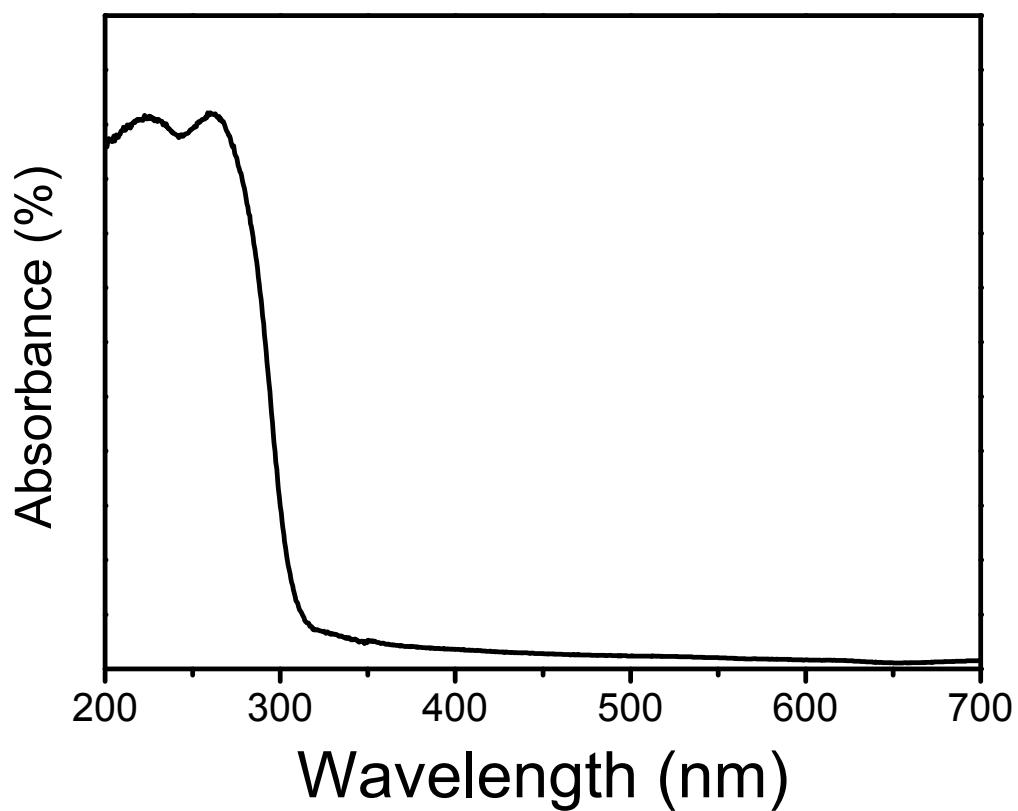


Fig. S4 Representative UV-Vis diffuse reflectance spectroscopy of as-prepared sample with pure NaTaO_3 phase. The sample was obtained in 1.75 M NaOH solution at 160 °C with 6-hour-milled Ta_2O_5 as precursor under MHT conditions for 3 h.