Electronic Supplementary Information

Kinetically Controlled Way to Create Highly Uniform Mono-dispersed ZnO Sub-microrods for Electronics

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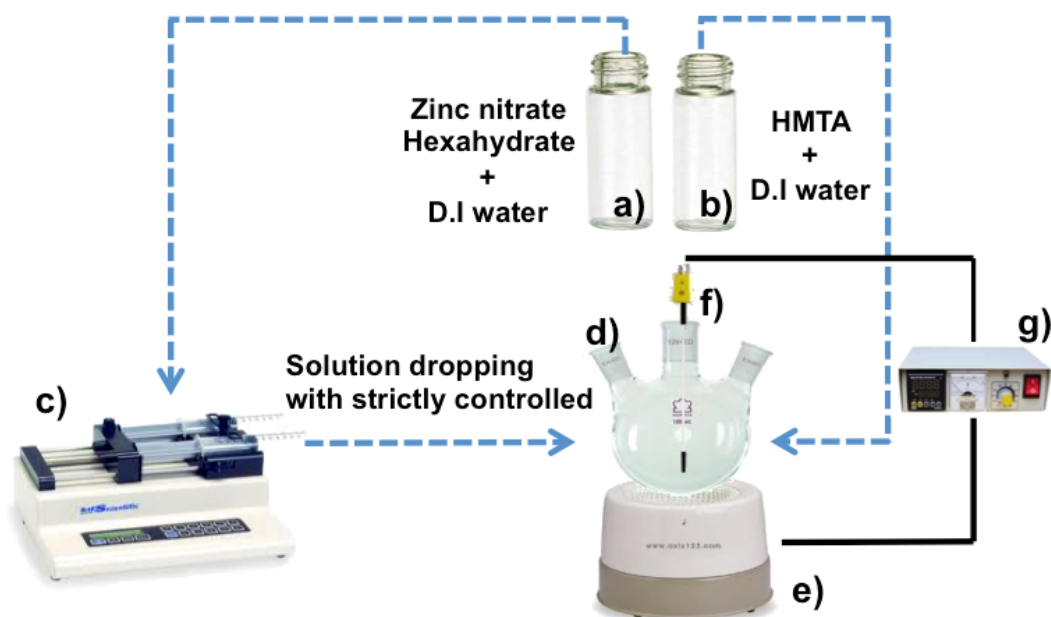
Figure S1.

Figure S1. Schematic diagram for the kinetically controlled ZnO sub-microrods synthesis system. a) zinc nitrate solution, b) hexamethylenetetramine (HMTA) solution, c) dual-syringe pump, d) three-necks ground flask, e) heating-mantle, f) thermocouple, g) thermal controller.
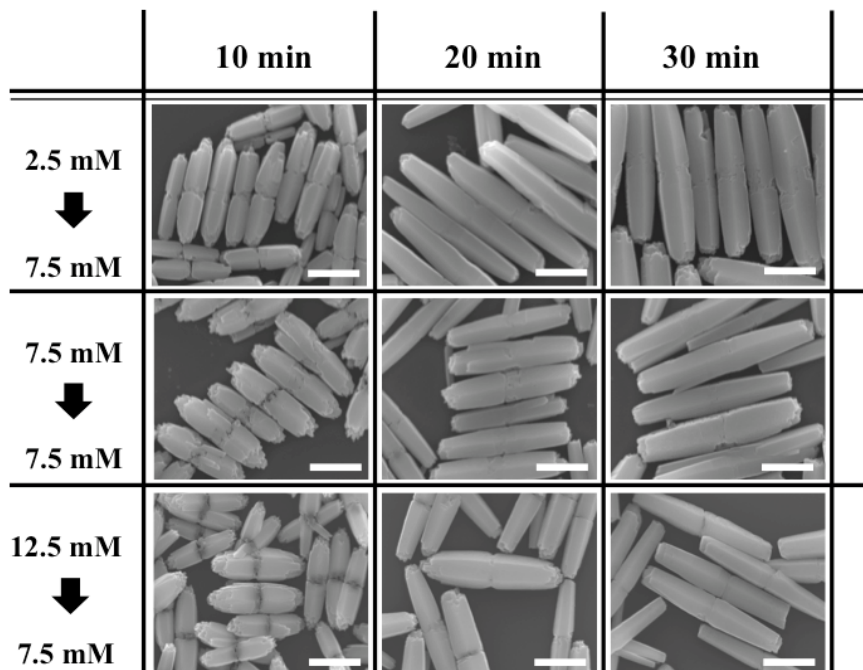
Figure S2.

Figure S2. The influence of the concentration of initially supplied Zn^{2+} ions on the synthesis of ZnO sub-microrods. (scale bar =1μm)
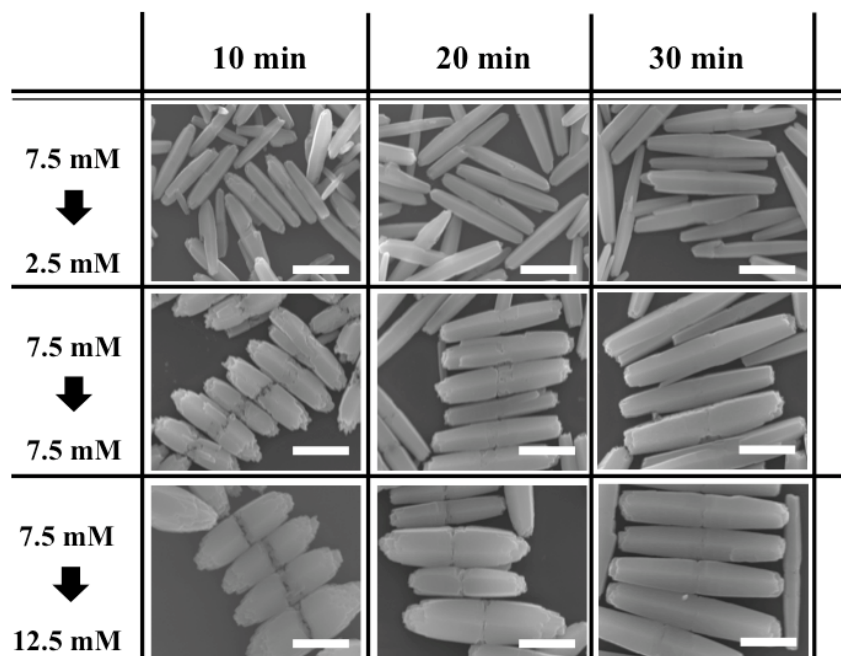
Figure S3.

Figure S3. The influence of the concentration of additional solution supplied Zn^{2+} ions on the synthesis of ZnO sub-microrods. (scale bar =1μm)
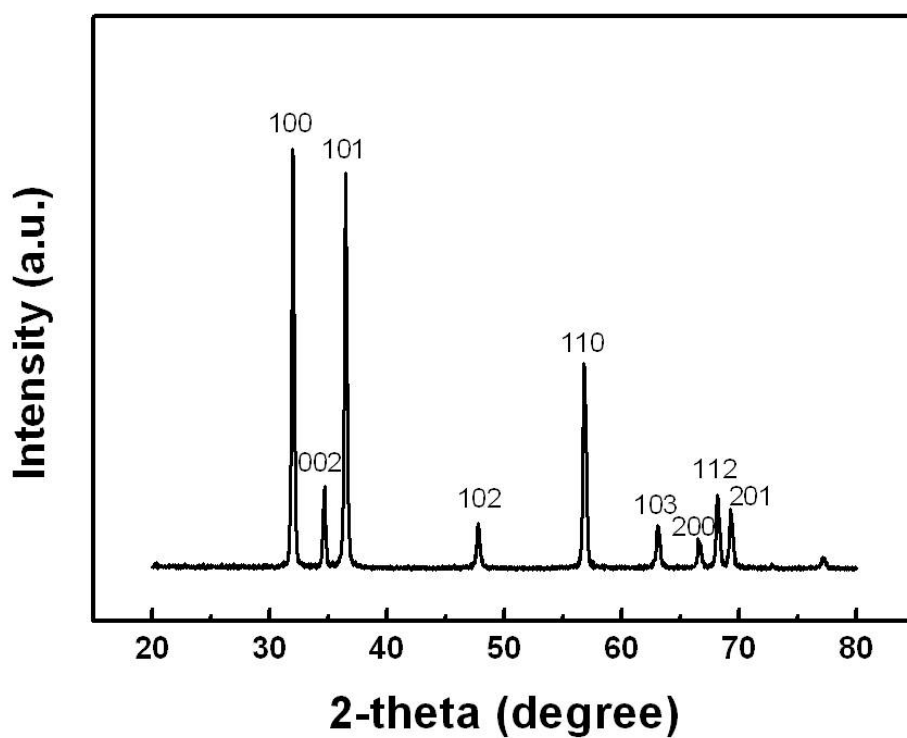
Figure S4.

Figure S4. XRD pattern of ZnO sub-microrods. (growth time: 10s, seed: 7.5mM and growth 7.5mM conditions)

Table S1. The summarization of the average length and diameter of ZnO sub-microrods shown in Figure S3.

Parameter	Unit (μm)	10 min	20 min	30 min
7.5 →2.5 mM	Length	1.33	1.81	2.37
	Diameter	0.29	0.31	0.37
7.5 → 7.5 mM	Length	1.86	2.35	3.05
	Diameter	0.46	0.49	0.52
7.5 → 12.5 mM	Length	1.76	2.34	3.09
	Diameter	0.53	0.57	0.61