

Supporting Information

Hierarchical $\text{SnO}_2/\text{Fe}_2\text{O}_3$ Heterostructures as Lithium-Ion Battery Anodes

*Yanli Wang, Jingjie Xu, Hao Wu, Ming Xu, Zheng Peng, and Gengfeng Zheng**

Laboratory of Advanced Materials, Department of Chemistry, Fudan University, Shanghai,
200433, China

* Corresponding author: gfzheng@fudan.edu.cn (G.Z.)

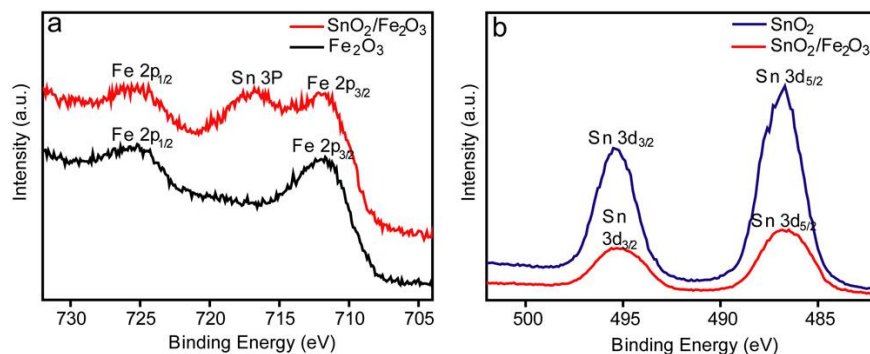


Figure S1. (a) Fe 2p XPS spectra collected for the SnO₂/Fe₂O₃ heterostructures (upper) and the pure Fe₂O₃ nanorods (lower). (b) Sn 3d xps spectra collected for the SnO₂/Fe₂O₃ heterostructures (lower) and the pure SnO₂ nanosheets (upper). The Sn 3p spectrum of SnO₂/Fe₂O₃ heterostructures was also shown in (a).

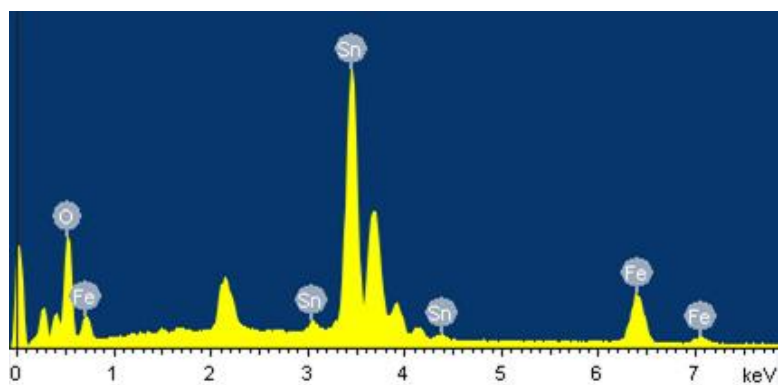


Figure S2. The EDX spectrum of SnO₂/Fe₂O₃ heterostructures showing a Sn/Fe ratio as 2.3 : 1.

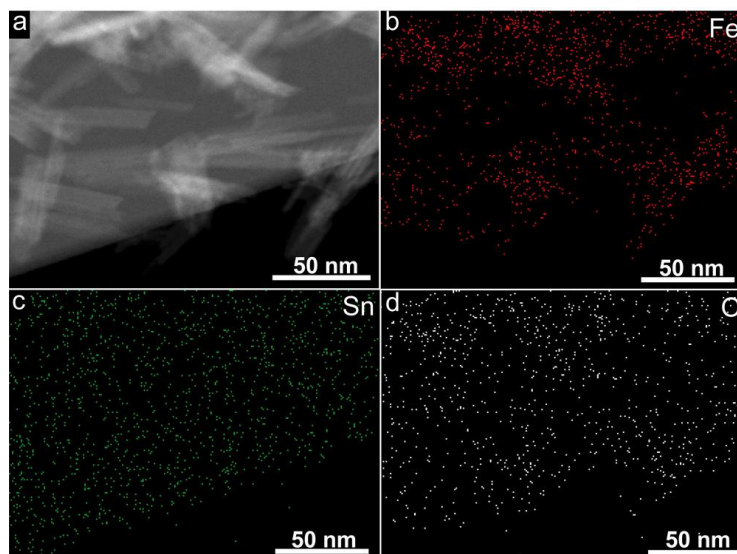


Figure S3. TEM images of (a) single branched nanosheet, and (b-d) the corresponding elemental mapping of Fe, Sn and O.

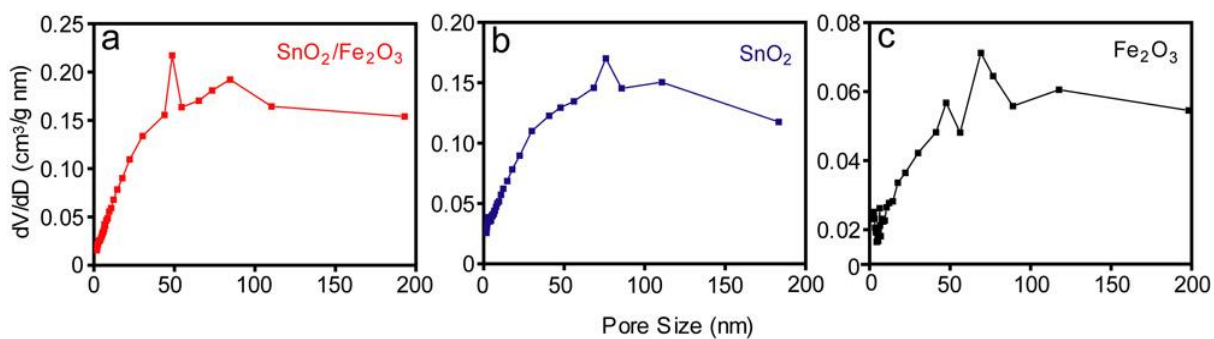


Figure S4. Pore size distributions calculated by the BJH method from the adsorption branch of in Figure 3: (a) SnO₂/Fe₂O₃ heterostructures; (b) SnO₂ nanosheets; (c) Fe₂O₃ nanorods.

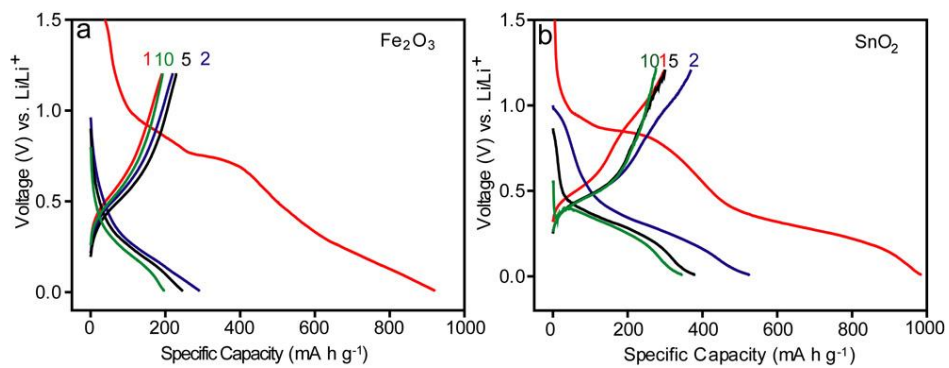


Figure S5. Charge-discharge profiles at a current rate of 400 mA g⁻¹ for the first, second, fifth and tenth cycles: (a) Fe₂O₃ nanorods; (b) SnO₂ nanosheets.