

ESI

Microwave-assisted, surfactant-free synthesis of air-stable copper nanoparticles and their SERS study

M. Ibrahim Dar^{a,b}, S. Sampath^c and S.A. Shivashankar^{*a,b}

^a Materials Research Centre, ^c Department of Inorganic and Physical Chemistry, ^b Centre for Nano Science and Engineering, Indian Institute of Science, Bangalore 560012, India

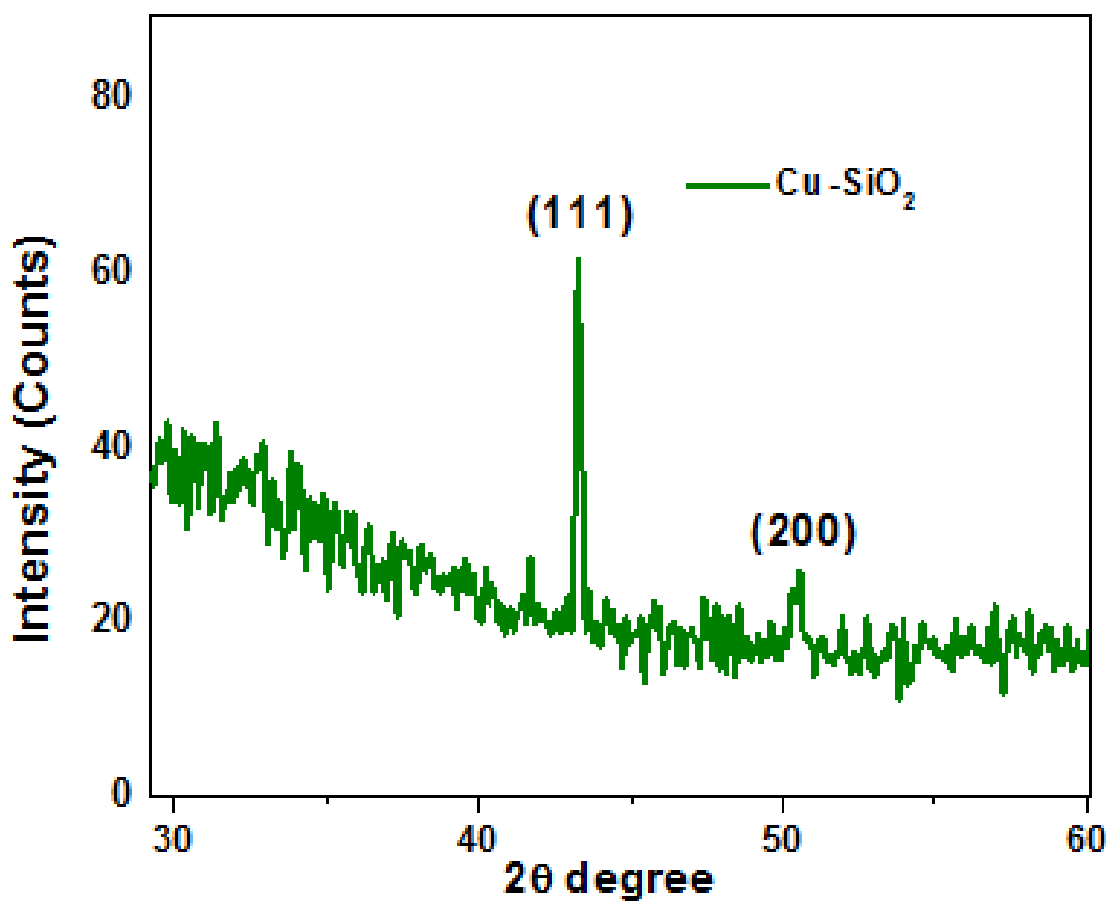


Fig. S1: Powder XRD pattern of Cu-SiO₂

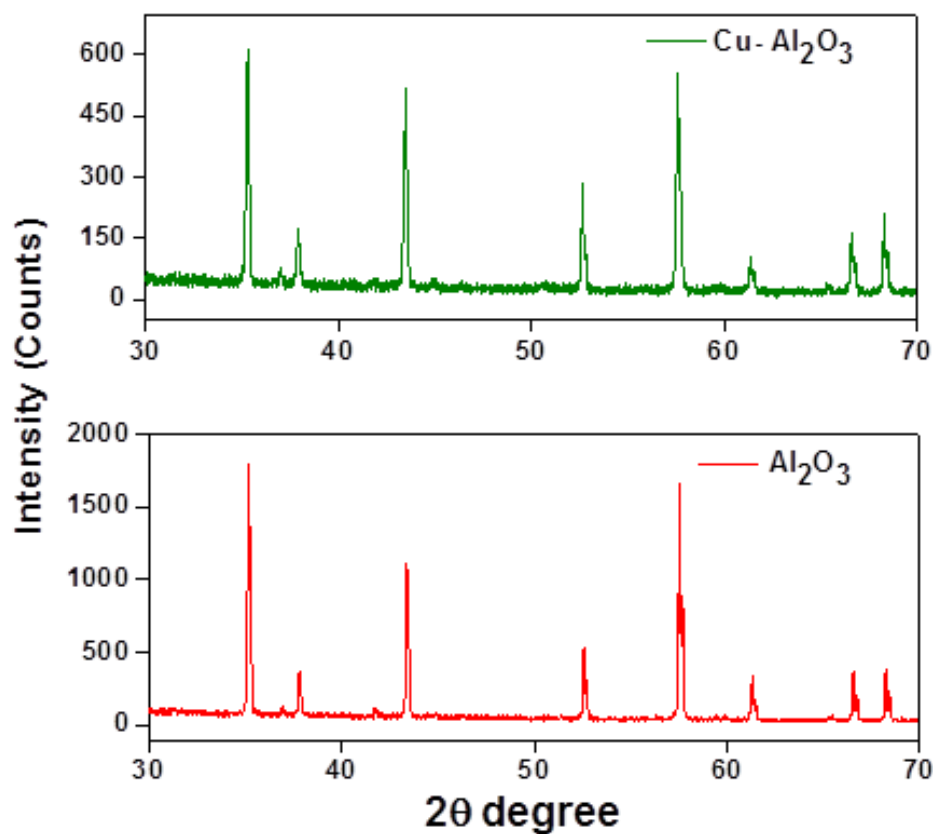


Fig. S2: Powder XRD pattern of Al₂O₃ and Cu-Al₂O₃

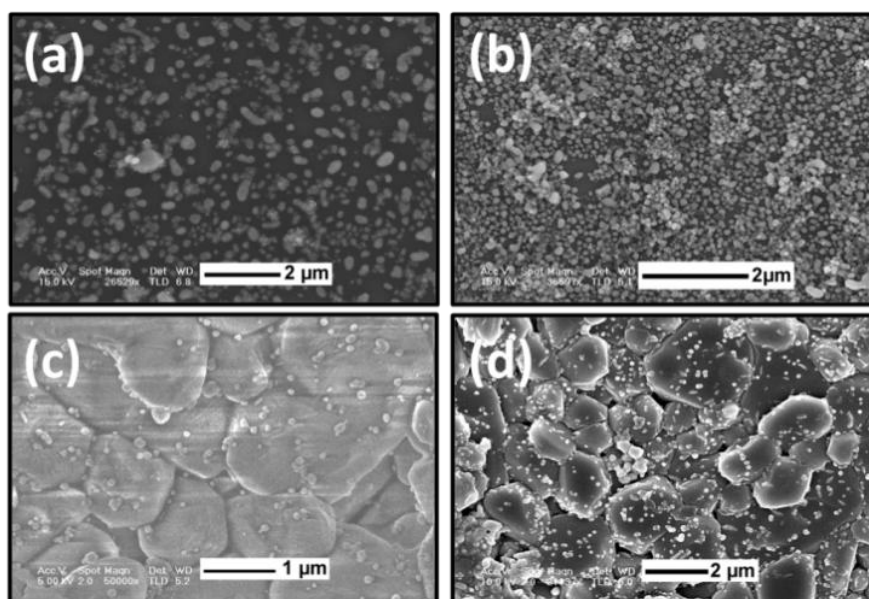


Fig. S3: FESEM images (a & b) Cu / SiO₂ and (c & d) Cu / Al₂O₃ nanostructures. (a & c are obtained after 5 minutes) and (b & d are obtained after 10 minutes)

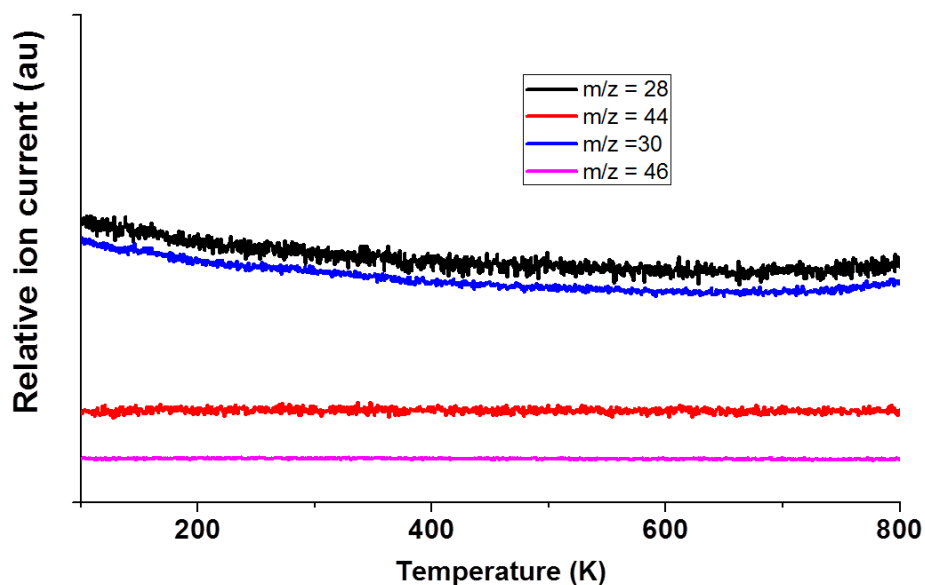


Fig. S4: Mass spectra of copper nanostructures under oxygen ambient

Synthesis of PVP stabilised Copper nanoparticles: In a round bottom flask, 50 ml reaction mixture of 26.2 mg $\text{Cu}(\text{acac})_2$ and 0.05 g of PVP, dissolved in benzyl alcohol was exposed to microwaves at 800 watt (100%) for a duration of two minute under reflux condition. The TEM sample was prepared directly from the colloidal solution of PVP stabilized copper nanoparticles.

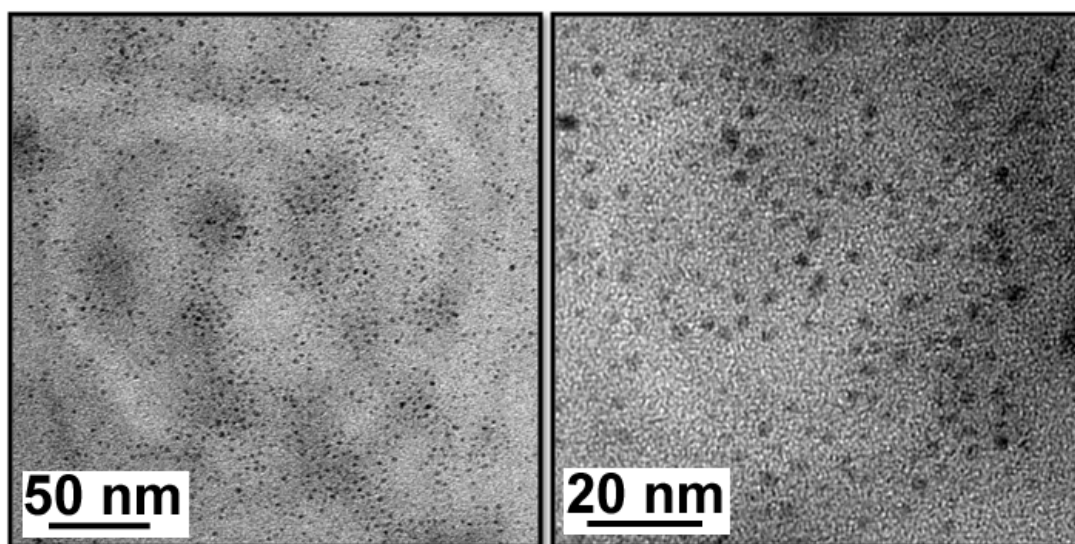


Fig. S5: TEM image of PVP stabilized copper nanoparticles

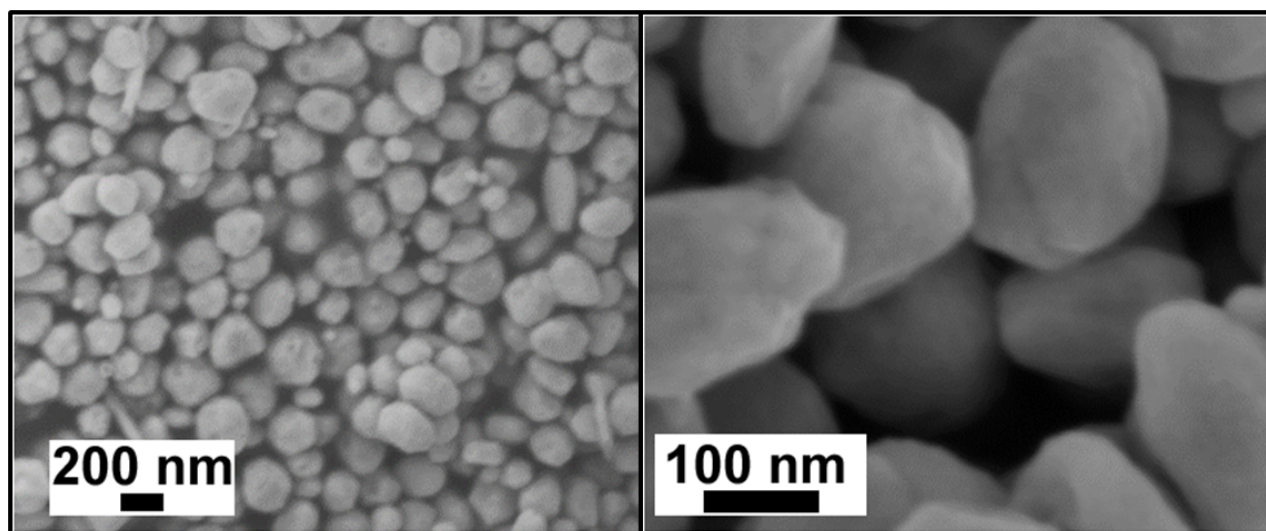


Fig. S6: FESEM (Low and High magnification) images of Cu(0) nanostructures

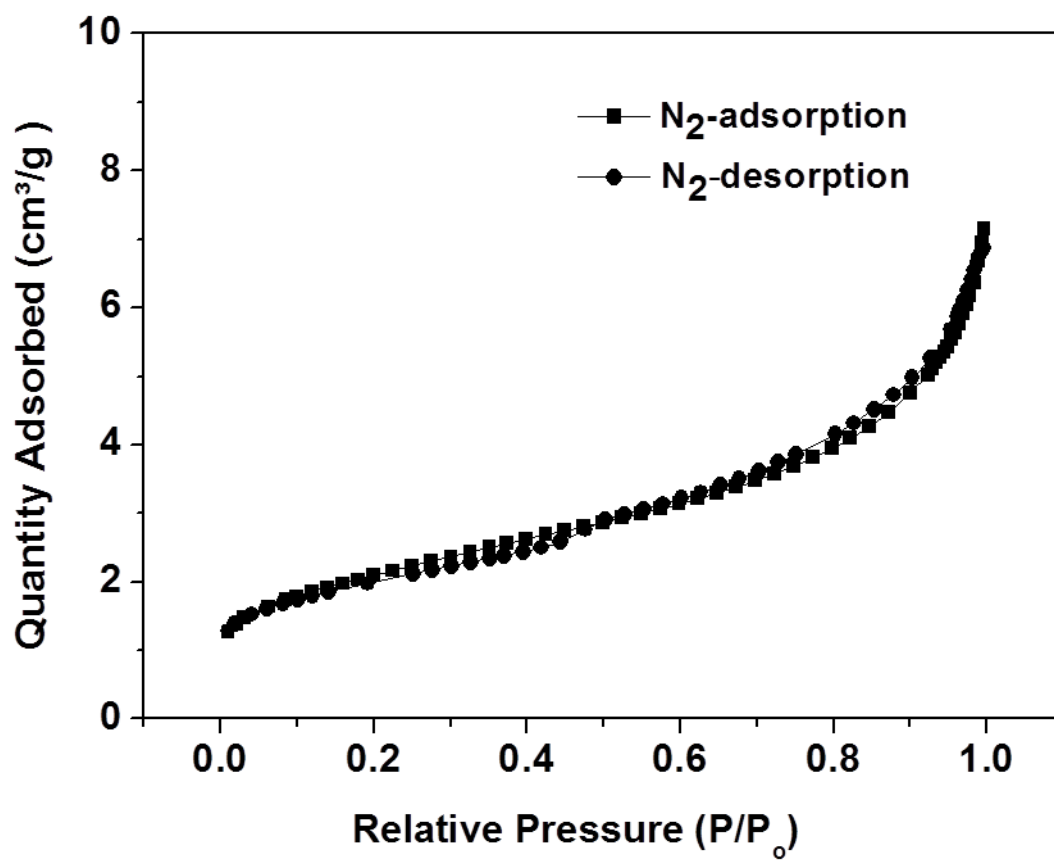


Fig. S7 N₂-physorption isotherm for Cu nanostructures (powder)