## **Supplementary Information**

## Non-invasive analysis in micro-reactors using Raman spectrometry with a specially designed probe

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Figure S-1. Schematic of the micro-reactor.



Figure S-2. a) Axial and b) transverse beam resolution measured at the glass-air interface by recording the intensity of the residual laser light. Spectral acquisition time, 0.5 s.



Figure S-3. First derivative Raman spectra of the pure reagents, butanol (red) and acetic anhydride (blue), and products, butyl acetate (black) and acetic acid (green), of the esterification reaction. Spectral acquisition time, 2 s.



Figure S-4. Example Raman spectrum of the reaction mixture in the micro-reactor. The peaks at 308 and 635 cm<sup>-1</sup> arise from butyl acetate while that at 670 cm<sup>-1</sup> arises from acetic anhydride. The broad peak at about 1300 cm<sup>-1</sup> arises from the glass micro-reactor. Spectral acquisition time, 1 s.



Figure S-5. Example 1<sup>st</sup> derivative Raman spectrum of the reaction mixture in the microreactor. The peaks at 300 and 675 cm<sup>-1</sup> arise from butyl actetate and acetic anhydride, respectively; the magnitude of these peaks can be used to monitor the progression of the esterification reaction. Spectral acquisition time, 1 s.



Figure S-6. a) Refraction of the laser beam by a bubble or thermal lens and b) ethanol spectra acquired under normal operation (black) and in the presence of a bubble or thermal lens (red). Spectral acquisition time, 2 s.