Supporting Information

Formation of i-motif structure at neutral and slightly alkaline pH

Jun Zhou, ^{a,b} Chunying Wei, ^c Guoqing Jia, ^{a,b} Xiuli Wang, ^{a,b} Zhaochi Feng, ^a and Can Li*,a

^aState Key Laboratory of Catalysis, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, Dalian 116023, China.

^bGraduate School of Chinese Academy of Sciences, Beijing 100039, China.

^cKey Laboratory of Chemical Biology and Molecular Engineering of Ministry of Education, Institute of Molecular Science, Shanxi University, Taiyuan 030006, China.

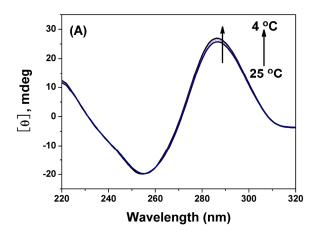
*Corresponding author. State Key Laboratory of Catalysis, Dalian Institute of

Chemical Physics, Chinese Academy of Sciences, Dalian 116023, China. Tel.:

86-411-84379070. Fax: 86-411-84694447. Web site: http://www.canli.dicp.ac.cn

Email: canli@dicp.ac.cn

Fig. S1



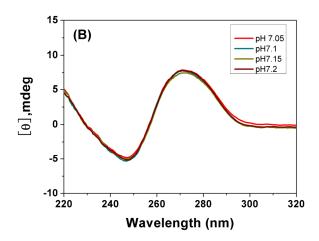


Fig. S1 Circular dichroism spectra of HT recorded at different temperatures at pH 5.0 (A), and at 25°C with different pH (B) in 10 mM cacodylate buffer.

Fig. S2

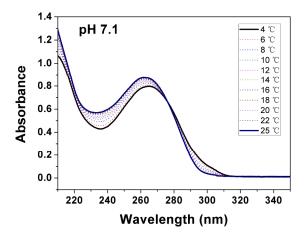


Fig. S2 UV absorption spectra of HT in 10 mM cacodylate buffer at pH 7.1 as a function of temperature.

Fig. S3

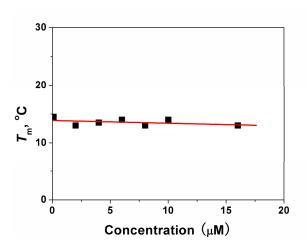


Fig. S3 $T_{\rm m}$ of HT at different strand concentration monitored by fluorescence (0.1 μ M) and CD (2-16 μ M) melting method (pH 7.0).

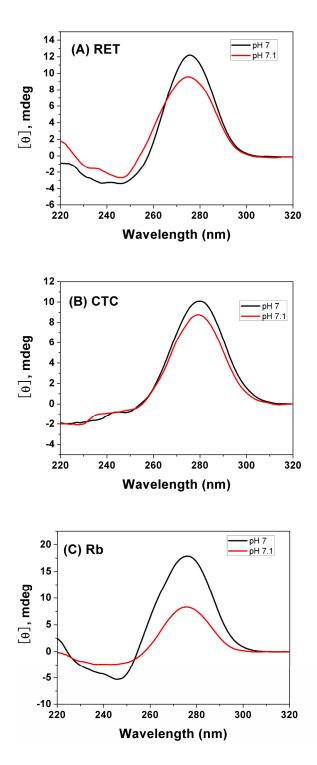


Fig. S4. Circular dichroism spectra of RET (A), CTC (B) and Rb (C) recorded at 25°C with pH 7.0 (black) and 7.1 (red) in 10 mM cacodylate buffer.