Supplementary Material (ESI) for Molecular BioSystems This journal is (c) The Royal Society of Chemistry, 2010 Stacking interaction in the middle and at the end of a DNA helix studied with non-natural nucleotides

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## **Supplementary information**

**Table S1** Thermodynamic parameters for DNA duplex formations measured in 1 MNaCl-phosphate buffer $^{a}$ 

Sequence $(5' \rightarrow 3')$	$-\Delta H^{\circ}$	$-\Delta S^{\circ}$	$-\Delta G^{\circ}_{37}$	$T_{ m m}$
	$(\text{kcal mol}^{-1})$	$(\operatorname{cal} \operatorname{mol}^{-1} \mathrm{K}^{-1})$	$(\text{kcal mol}^{-1})$	(°C)
GTGTAA <sup>phe</sup> ATGTC				
/ GACATFTACAC	76.5±1.9	216±6	9.53±0.22	49.6
GTGTTA <sup>phe</sup> TTGTC				
/ GACAAFAACAC	79.0±1.5	225±5	9.10±0.14	47.4
GTGTGA <sup>phe</sup> GTGTC				
/ GACACFCACAC	74.4±1.9	205±6	10.9±0.3	56.6
amama , phramama b				
GIGICA <sup>Phe</sup> CIGIC <sup>®</sup>	72.4	202	10.4	54.0
/ GACAGFGACAC	/3.4	203	10.4	54.9
GTGTCA <sup>naph</sup> CTGTC <sup>b</sup>				
/ GACAGTGACAC	76.5	209	11.5	60.2
	97 1+1 8	276+5	11.6+0.2	54.1
	<i>J</i> 1.1±1.0	210-5	11.0-0.2	JT.1
GTGTTATTGTC				
/ GACAATAACAC	86.2±1.5	242±5	11.0±0.1	54.1

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Sequence $(5' \rightarrow 3')$	$-\Delta H^{\circ}$	$-\Delta S^{\circ}$	$-\Delta G^{\circ}_{37}$	T <sub>m</sub>
	$(\text{kcal mol}^{-1})$	$(cal mol^{-1}K^{-1})$	$(\text{kcal mol}^{-1})$	(°C)
GTGTGAGTGTC				
/ GACACTCACAC	92.6±2.3	256±7	13.2±0.3	61.0
GTGTCACTGTC <sup>b</sup>				
/ GACAGTGACAC	89.0	248	12.6	57.4
GTGTCACTGTC <sup>b</sup>				
/ GACAGFGACAC	69.5	199	7.4	42.9
GTGTAC <sup>phe</sup> ATGTC				
/ GACATFTACAC	76.0±1.4	215±4	9.25±0.14	48.5
GTGTTC <sup>pne</sup> TTGTC				
/ GACAAFAACAC	75.1±1.9	214±6	8.91±0.23	47.1
nha				
GTGTGC <sup>pne</sup> GTGTC				
/ GACACFCACAC	82.7±1.3	231±4	$11.2 \pm 0.1$	55.6
am am a c <sup>ube</sup> am am a				
GTGTCC <sup>pm</sup> CTGTC				
/ GACAGFGACAC	67.5±5.9	185±18	$10.2\pm0.2$	59.0
ananaananhanana				
GIGICC	75 1 4 0	207+15	11 4 0 2	(0.0
/ GACAGFGACAC	/5.1±4.8	207±15	11.4±0.2	60.8
	<u>80 4+1 7</u>	250+5	11 0±0 <b>2</b>	56 9
UACAIUIACAC	09.4±1.7	230±3	11.9±0.2	30.8
GTGTTCTTGTC				
	82 6+1 7	229+5	11 5+0 2	56.8
	02.0±1.7		11.5±0.2	50.0
GTGTGCGTGTC				
/ GACACGCACAC	89 6±1 8	243±5	14 3±0 2	66 3
	09.0-1.0	213-0	11.5-0.2	00.5
GTGTCCCTGTC				
/ GACAGGGACAC	85.7±3.5	235±10	12.9±0.3	65.3
GTGTCCCTGTC				
/ GACAGFGACAC	64.3±3.9	180±12	7.84±0.16	45.2

<sup>*a*</sup>Data was obtained using the buffer containing 1 M NaCl, 10 mM Na<sub>2</sub>HPO<sub>4</sub> (pH 7.0), and 1 mM Na<sub>2</sub>EDTA. The error values were calculated from the difference between the parameters determined by the  $T_{\rm m}^{-1}$  versus log ( $C_{\rm t}/4$ ) plot and curve fitting to the two-state model.  $T_{\rm m}$  was calculated at 100  $\mu$ M.

<sup>b</sup>Data are derived from the reference 19.

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 Table S2
 Thermodynamic parameters of self-complementary DNA duplexes measured in 1

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	A 7 70	4.00	1.00	7	
Sequence $(5' \rightarrow 3')$	$-\Delta H^{\circ}$	$-\Delta S^{\circ}$	$-\Delta G^{\circ}_{37}$	$I_{\rm m}$	
	(kcal mol <sup>-</sup> )	(cal mol <sup>-</sup> K <sup>-</sup> )	(kcal mol <sup>-1</sup> )	(°C)	
Core sequence 1					
ATGCGCAT <sup>b</sup>	62.0	171	9.3	54.3	
AATGCGCAT <sup>b</sup>	64.6	175	10.2	60.9	
A <sup>phe</sup> ATGCGCAT <sup>b</sup>	53.9	138	11.1	72.5	
A <sup>naph</sup> ATGCGCAT <sup>b</sup>	56.8	146	11.5	72.7	
<u>A</u> ATGCGCAT <u>T</u> <sup>b</sup>	75.8	208	11.2	61.2	
ATGCGCATA <sup>b</sup>	63.1	171	10.0	60.1	
ATGCGCAT <u>A<sup>phe b</sup></u>	56.2	148	10.1	64.1	
ATGCGCATA <sup>naph b</sup>	53.1	138	10.3	66.8	
<u>T</u> ATGCGCAT <u>A</u>	73.6±2.8	203±9	$10.7 \pm 0.2$	59.8	
CATGCGCAT	65.9±3.2	183±10	9.48±0.15	55.1	
C <sup>phe</sup> ATGCGCAT	$69.2 \pm 1.7$	170±5	11.3±0.1	65.5	
C <sup>naph</sup> ATGCGCAT	$65.4\pm2.0$	174±6	$11.3 \pm 0.2$	66.6	
<u>CATGCGCATG</u>	74.3±2.7	206±8	10.9±0.3	59.9	
ATGCGCATC	63.8±2.9	177±8	8.94±0.09	53.5	
ATGCGCATC <sup>phe</sup>	64.0±2.6	173±8	10.3±0.1	61.6	
ATGCGCATC <sup>naph</sup>	57.6±2.2	153±6	$10.3 \pm 0.2$	63.9	
GATGCGCATC	73.0±4.8	203±15	10.3±0.2	57.3	
Core sequence 2					
TGCGCA <sup>b</sup>	47.4	129	7.6	50.3	
ATGCGCA <sup>b</sup>	55.0	151	8.2	51.4	
A <sup>phe</sup> TGCGCA	55.1±2.3	145±7	$10.2 \pm 0.2$	64.1	
A <sup>naph</sup> TGCGCA	51.5±3.3	133±10	$10.2 \pm 0.2$	67.1	
<u>A</u> TGCGCA <u>T</u> <sup>b</sup>	62.0	171	9.3	54.3	
TGCGCA <u>A</u>	48.2±4.0	132±12	7.39±0.25	47.6	
TGCGCA <u>A<sup>phe</sup></u>	48.6±3.4	131±11	7.96±0.16	53.0	
TGCGCAA <sup>naph</sup>	46.7±2.6	126±8	7.77±0.10	50.9	
<u>T</u> TGCGCA <u>A</u>	57.4±4.0	158±13	8.64±0.18	52.9	
<u>C</u> TGCGCA	57.1±3.4	159±10	7.66±0.13	48.1	
C <sup>phe</sup> TGCGCA	57.3±6.0	154±9	9.49±0.09	59.0	
<u>C<sup>naph</sup>TGCGCA</u>	52.6±3.0	140±9	9.35±0.18	61.5	
<u>C</u> TGCGCA <u>G</u>	62.3±5.1	171±15	9.36±0.18	56.5	
TGCGCA <u>C</u>	37.8±2.8	98.6±9.0	6.92±0.19	49.5	
TGCGCA <u>C<sup>phe</sup></u>	55.7±2.0	151±6	8.49±0.16	53.1	
TGCGCA <u>C<sup>naph</sup></u>	54.4±2.9	148±9	8.78±0.15	54.1	
GTGCGCAC	58.7±1.9	156±6	9.94±0.05	62.3	

M NaCl-phosphate buffer<sup>a</sup>

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Sequence $(5' \rightarrow 3')$	$-\Delta H^{\circ}$	$-\Delta S^{\circ}$	$-\Delta G^{\circ}_{37}$	T <sub>m</sub>
	$(\text{kcal mol}^{-1})$	$(cal mol^{-1}K^{-1})$	$(\text{kcal mol}^{-1})$	(°C)
Core sequence 3				
GTGCGCAC	58.7±1.9	156±6	9.94±0.05	62.3
<u>A</u> GTGCGCAC	70.2±1.7	189±3	11.5±0.2	65.1
<u>A<sup>phe</sup>GTGCGCAC</u>	65.7±2.8	$171 \pm 8$	12.7±0.2	74.0
<u>A<sup>naph</sup>GTGCGCAC</u>	66.6±1.3	174±4	12.9±0.2	73.8
<u>A</u> GTGCGCAC <u>T</u>	75.7±1.6	204±4	12.5±0.1	67.3
GTGCGCAC <u>A</u>	71.9±1.7	194±5	11.9±0.1	66.3
GTGCGCAC <u>A<sup>phe</sup></u>	68.4±2.7	$184 \pm 8$	11.5±0.2	65.6
GTGCGCAC <u>A<sup>naph</sup></u>	67.0±3.1	178±9	11.9±0.3	68.5
<u>T</u> GTGCGCAC <u>A</u>	72.2±2.5	193±7	12.5±0.1	69.1
<u>C</u> GTGCGCAC	69.2±1.7	188±5	10.9±0.1	62.1
C <sup>phe</sup> GTGCGCAC	73.5±1.6	196±4	12.9±0.1	70.5
<u>C<sup>naph</sup>GTGCGCAC</u>	71.6±2.3	190±7	12.8±0.2	71.3
<u>C</u> GTGCGCAC <u>G</u>	79.8±2.1	214±6	13.5±0.2	70.7
GTGCGCAC <u>C</u>	67.5±1.8	182±5	10.9±0.1	63.2
GTGCGCAC <u>C<sup>phe</sup></u>	71.7±1.8	192±5	12.3±0.1	68.3
GTGCGCAC <u>C<sup>naph</sup></u>	67.6±2.2	179±6	12.1±0.2	69.5
<u>G</u> GTGCGCAC <u>C</u>	76.0±2.5	203±7	13.1±0.2	70.9
Core sequence A				
CTGCGCAG	62 3+5 1	171+15	9 36+0 18	56 5
	70 4+2 4	102+10	10.0+0.2	61.7
A <sup>phe</sup> CTCCCCAC	$70.4\pm 3.4$	$192\pm10$	10.9±0.2	01.7
<u>A<sup>n</sup></u> CIGCGCAG	$6/.9\pm2.9$	1/9±9	$12.4\pm0.2$	/1.3
<u>A '</u> CIGCGCAG	/6.0±2./	$203\pm7$	$13.1\pm0.1$	/0.5
<u>ACTGCGCAG1</u>	82.2±2.0	225±6	12.3±0.2	64.0
CTGCGCAG <u>A</u>	65.9±1.9	180±6	$10.3 \pm 0.1$	59.5
CTGCGCAG <u>Aphe</u>	$60.7 \pm 2.3$	161±7	$10.9\pm0.2$	65.4
CTGCGCAG <u>A<sup>napn</sup></u>	66.4±3.1	178±9	$11.2\pm0.2$	64.8
<u>T</u> CTGCGCAG <u>A</u>	78.1±3.2	214±10	11.8±0.2	62.6
<u>C</u> CTGCGCAG	61.7±2.8	167±9	10.0±0.2	59.3
<u>C<sup>phe</sup>CTGCGCAG</u>	76.6±2.2	207±6	12.5±0.2	66.6
<u>C<sup>naph</sup>CTGCGCAG</u>	73.0±3.1	195±9	12.6±0.2	69.1
<u>C</u> CTGCGCAG <u>G</u>	78.3±2.0	212±6	12.7±0.2	67.5
CTGCGCAG <u>C</u>	nd.	nd.	nd.	nd.
CTGCGCAG <u>C<sup>phe</sup></u>	58.5±3.6	155±10	10.4±0.3	64.0
CTGCGCAG <u>C<sup>naph</sup></u>	54.7±4.1	144±12	10.2±0.3	65.2
<u>G</u> CTGCGCAG <u>C</u> <sup>c</sup>	nd.	nd.	nd.	nd.

<sup>a</sup> Data was obtained using the buffer containing 1 M NaCl, 10 mM Na<sub>2</sub>HPO<sub>4</sub> (pH 7.0), and 1 mM Na<sub>2</sub>EDTA.  $T_{\rm m}$  was calculated at 100  $\mu$ M. <sup>b</sup> Data are derived from the references 10 and 18.

1 1				
$\Delta H^{\circ}$	$\Delta S^{\circ}$	$\Delta G^{\circ}_{37}$ ,	$T_{\rm m}$	
$(\text{kcal mol}^{-1})$	$(cal mol^{-1}K^{-1})$	$(\text{kcal mol}^{-1})$	(°C)	
-65.0	-176	-10.3	61.3	
-65.5	-178	-10.4	60.4	
-63.5	-165	-12.2	72.8	
-73.1±2.5	-194±7	-12.9±0.1	71.5	
-64.9	-168	-12.9	76.7	
-78.6±2.2	-206±6	-14.6±0.3	76.7	
-68.0±3.1	-188±9	$-9.63 \pm 0.07$	56.3	
-72.2±3.5	$-202\pm10$	$-9.62 \pm 0.05$	54.9	
-70.7±3.9	-191±12	-11.5±0.2	64.7	
-64.8±2.9	-173±9	-11.1±0.2	65.6	
-66.3±2.6	-178±7	-11.3±0.3	65.2	
-62.2±1.7	-165±5	-11.3±0.3	67.0	
	$\frac{\Delta H^{\circ}}{(\text{kcal mol}^{-1})}$ -65.0 -65.5 -63.5 -73.1 $\pm$ 2.5 -64.9 -78.6 $\pm$ 2.2 -68.0 $\pm$ 3.1 -72.2 $\pm$ 3.5 -70.7 $\pm$ 3.9 -64.8 $\pm$ 2.9 -66.3 $\pm$ 2.6 -62.2 $\pm$ 1.7	$\begin{array}{c cccc} \Delta H^{\circ} & \Delta S^{\circ} \\ (\text{kcal mol}^{-1}) & (\text{cal mol}^{-1}\text{K}^{-1}) \\ \hline & -65.0 & -176 \\ & -65.5 & -178 \\ & -63.5 & -165 \\ & -73.1\pm2.5 & -194\pm7 \\ & -64.9 & -168 \\ & -78.6\pm2.2 & -206\pm6 \\ & -68.0\pm3.1 & -188\pm9 \\ & -72.2\pm3.5 & -202\pm10 \\ & -70.7\pm3.9 & -191\pm12 \\ & -64.8\pm2.9 & -173\pm9 \\ & -66.3\pm2.6 & -178\pm7 \\ & -62.2\pm1.7 & -165\pm5 \end{array}$	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $

**Table S3** Thermodynamic parameters of DNA duplexes forming 2 or 3 nucleotide danglingends measured in 1 M NaCl-phosphate buffer<sup>a</sup>

<sup>*a*</sup> Melting curve was obtained with the buffer containing 1 M NaCl, 10 mM Na<sub>2</sub>HPO<sub>4</sub> (pH 7.0), and 1 mM Na<sub>2</sub>EDTA.  $T_{\rm m}$  was calculated at 100  $\mu$ M.

<sup>b</sup> Data are derived from the references 10 and 18.

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**Fig. S1** Melting curve of the DNA duplexes forming a dangling end of C (red),  $C^{phe}$  (violet), or  $C^{naph}$  (blue) at the 5' end (A) or the 3' end (B) of the core sequence of ATGCGCAT, monitored at 260 nm. The curves were obtained with 10  $\mu$ M DNA in 1 M NaCl–phosphate buffer. The fits to a two–state model are indicated in a black line.