Electronic Supplementary Material (ESI) for Metallomics

Metal Retention in Human Transferrin: Consequences of Solvent Composition in Analytical Sample Preparation Methods

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SUPPLEMENTARY INFORMATION

UV-VIS Spectra: UV-VIS spectroscopy was used to monitor the loss of iron from Tf under extreme pH conditions and due to urea addition (Figs. S1 and S2).

Circular Dichroism Spectra: CD spectroscopy was used to monitor the denaturing effects of guanidinium on the Fe-Tf complex (Fig. S3).

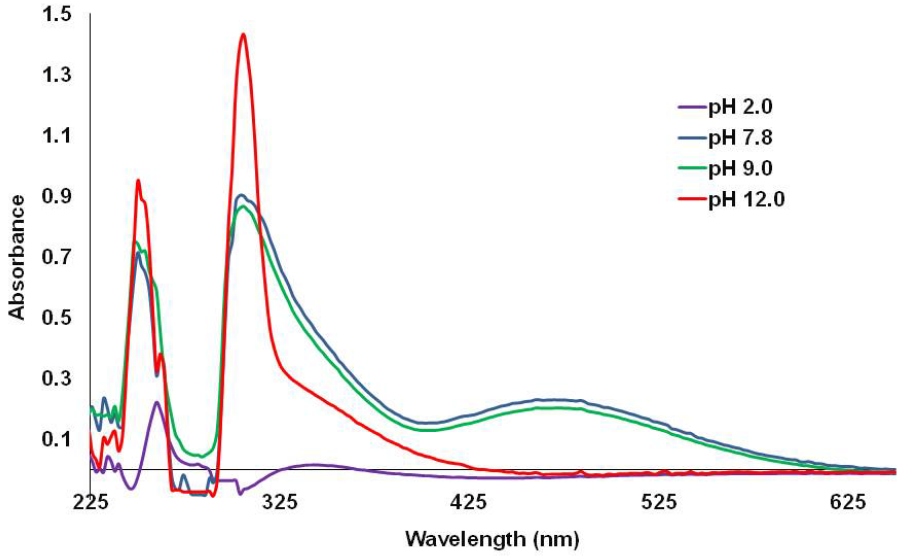

Fig. S1. UV-VIS spectra showing the effect of pH on Fe^{3+} retention in holo-Tf. Apo-Tf absorbances were subtracted from the Fe^{3+}-Tf absorbance spectra to better show absorbances resulting from iron binding.

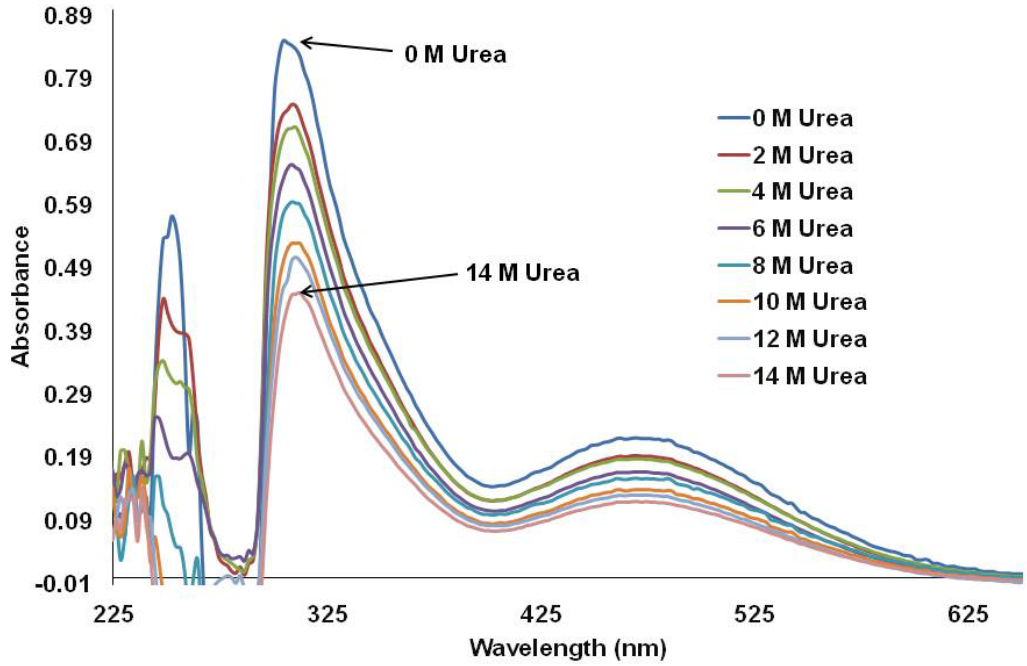

Fig. S2. UV-VIS absorbance spectra showing the effects of urea addition (0 – 14 M) on holo-Tf (50 µM). Apo-Tf absorbances were subtracted from the Fe^{3+}-Tf absorbance spectra to better show absorbances resulting from iron binding.

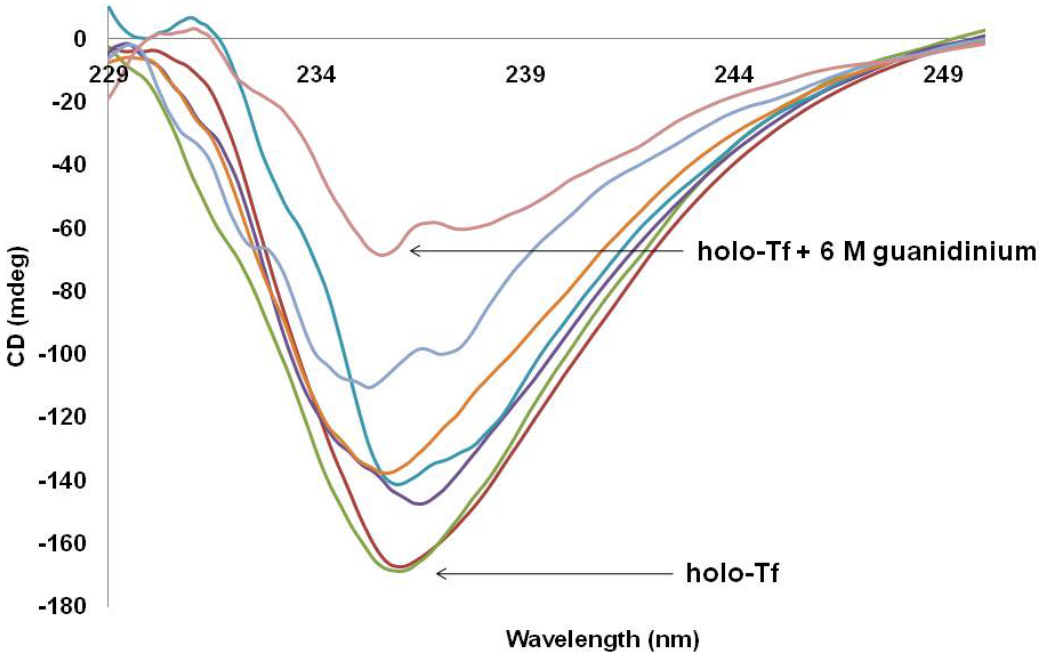

Fig. S3. CD spectra displaying the change in optical activity of holo-Tf (50 µM) upon addition of guanidinium (0 – 6 M).