

Supporting Information

Fluoride ligands exhibit marked departures from the hydrogen bond acceptor behavior of its heavier halogen congeners

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Table S1. Model geometries (Å, deg) used in ab initio calculations.

Figure S1. Spatially normalized plots⁹ of hydrogen bond distances (represented as R_{HX}^3) against angle at the hydrogen (represented as $1 - \cos(T)$, $T = 180 - (\text{D}-\text{H}\cdots\text{X})$) for hydrogen bonds $\text{D}-\text{H}\cdots\text{X}-\text{M}$; $\text{D} = \text{C}$ (top), $\text{D} = \text{N}$ (middle), $\text{D} = \text{O}$ (bottom), $\text{X} = \text{F}$ (left), $\text{X} = \text{Cl}$, $\text{X} = \text{Br}$, $\text{X} = \text{I}$ (right).

Figure S2. Spatially normalized plots⁹ of hydrogen bond distances (represented as R_{HX}^3) against angle at the halogen (represented as $1 - \cos(A)$, $A = 180 - (\text{H}\cdots\text{X}-\text{M})$) for hydrogen bonds $\text{D}-\text{H}\cdots\text{X}-\text{M}$; $\text{D} = \text{C}$ (top), $\text{D} = \text{N}$ (middle), $\text{D} = \text{O}$ (bottom), $\text{X} = \text{F}$ (left), $\text{X} = \text{Cl}$, $\text{X} = \text{Br}$, $\text{X} = \text{I}$ (right).

Figure S3. Calculated electrostatic potential in the plane perpendicular to the metal coordination plane for the model compounds *trans*-PdX(Me)(PH₃)₂; $\text{X} = \text{F}$, Cl , Br , I . Electrostatic potential represented with color gradation and contoured at 10 kcal/mol intervals. Only negative and zero contours shown; atoms that lie below the coordination plane are not shown.

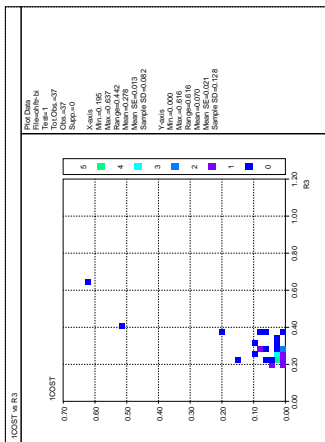
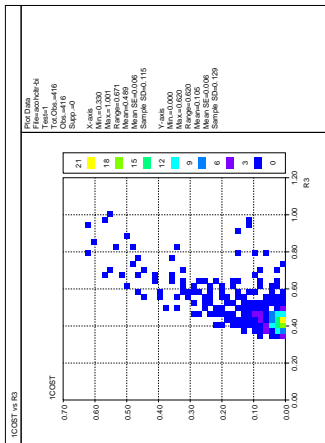
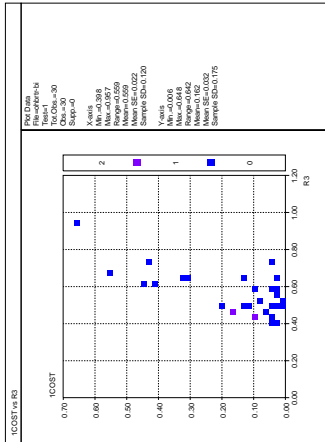
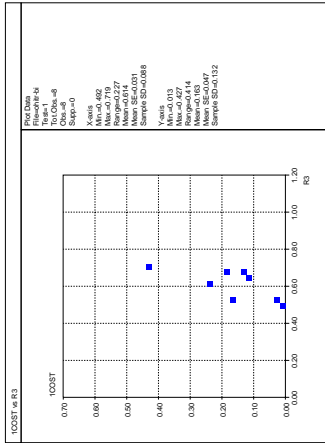
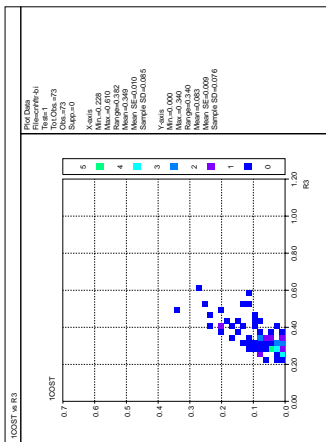
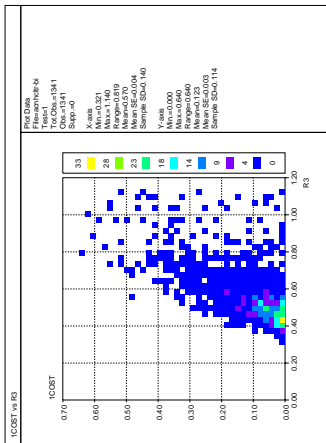
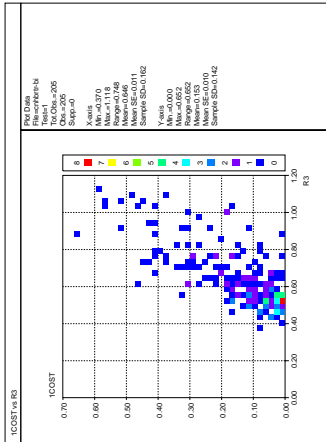
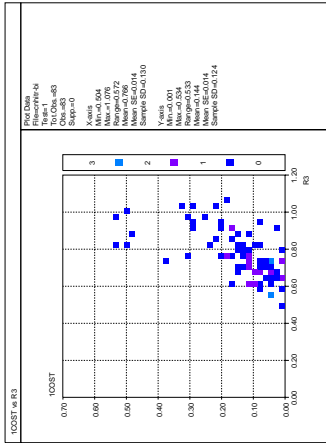
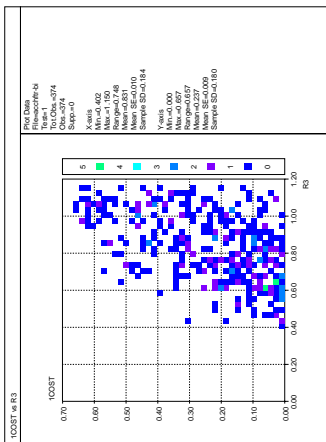
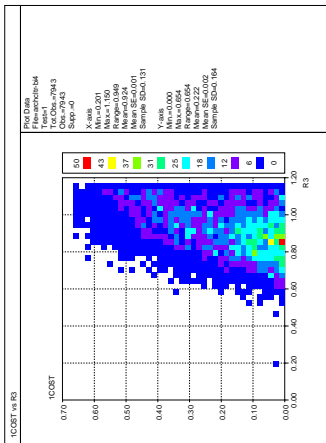
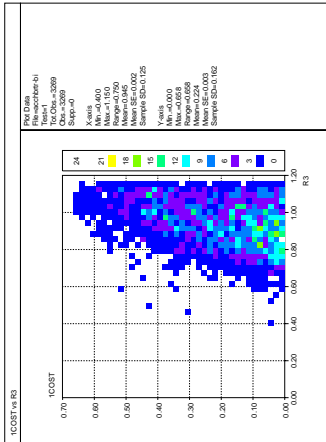
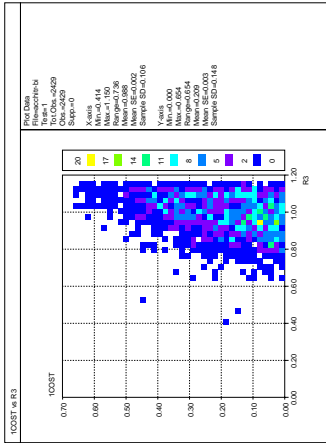
Table S1. Model geometries (Å, deg) used in ab initio calculations

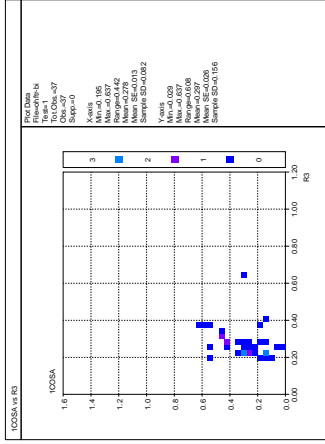
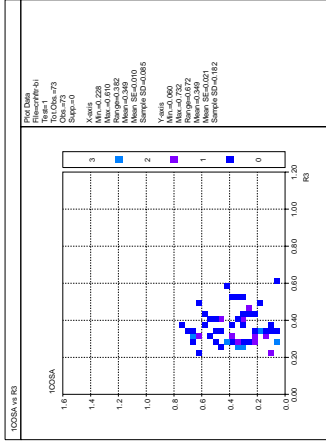
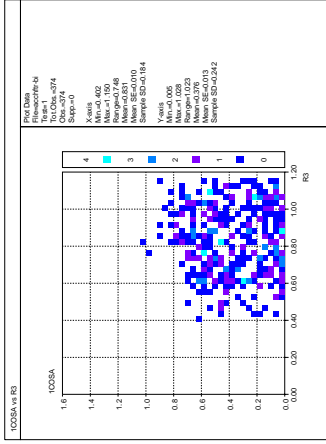
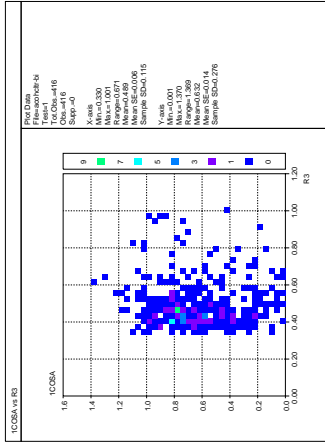
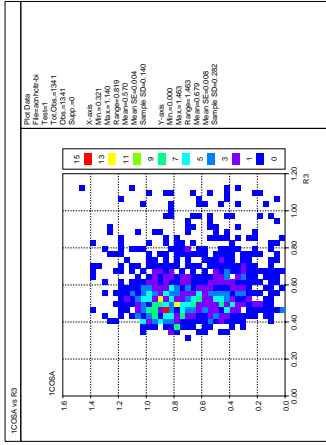
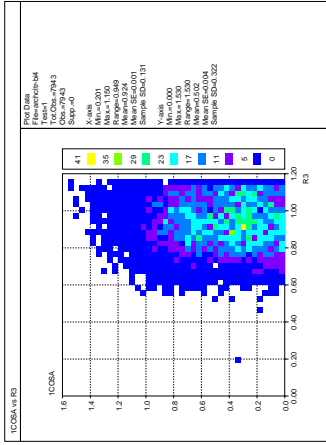
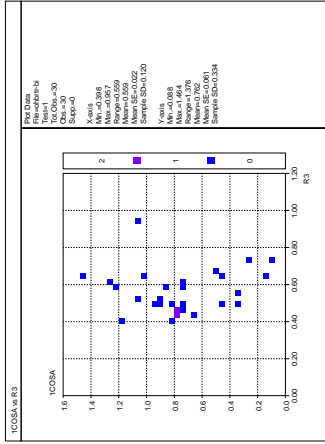
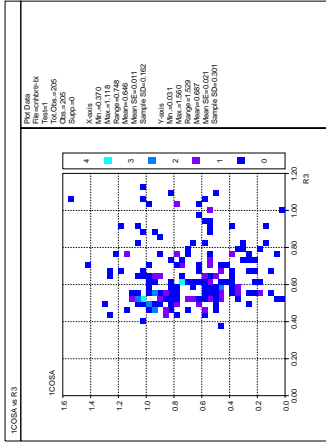
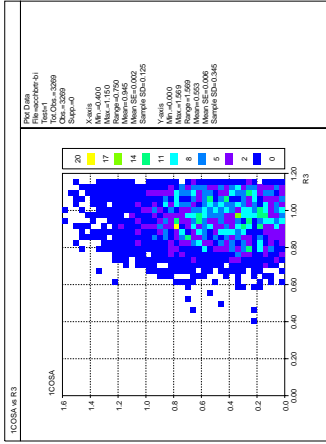
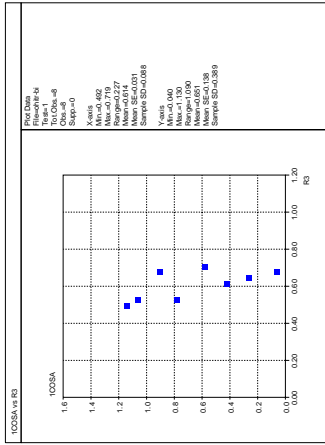
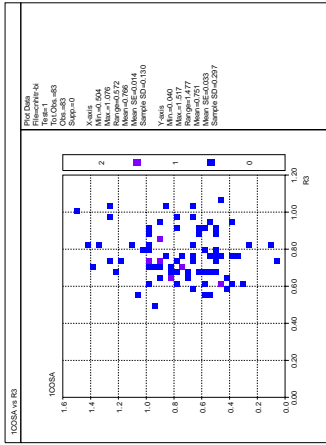
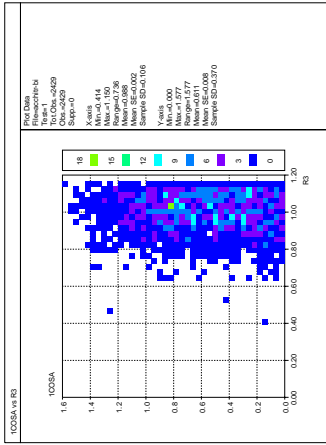
Pd-F	1.87 ^a	Pd-P	2.33 ^a
Pd-Cl	2.42 ^a	Pd-C	2.05 ^a
Pd-Br	2.55 ^a	P-H	1.409 ^c
Pd-I	2.75 ^a	C-H	1.095 ^c
P-Pd-X	90.0 ^b	Pd-P-H	118.12 ^c
P-Pd-C	90.0 ^b	Pd-C-H	112.59 ^c
C-Pd-X	180.0 ^b		

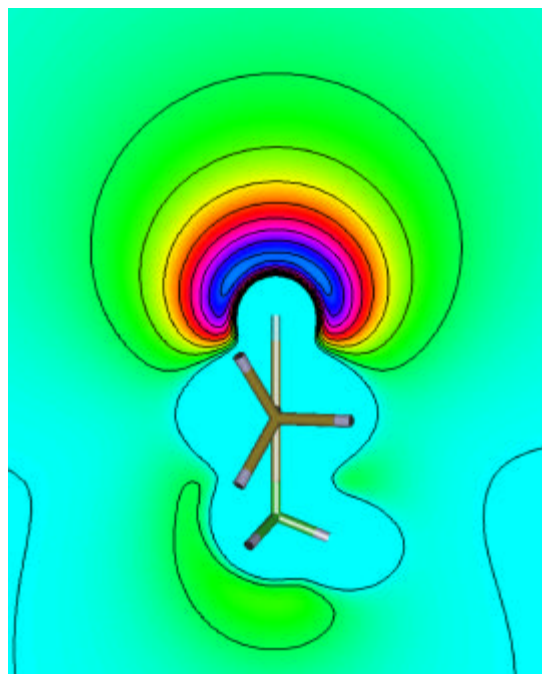
a) taken from representative literature crystal structures

b) idealized

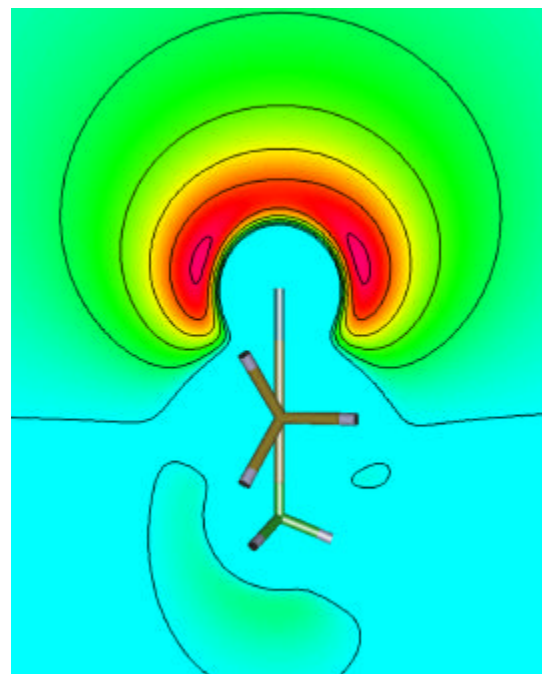
c) derived by geometry optimization on the fluoride complex with heavy atom positions fixed.



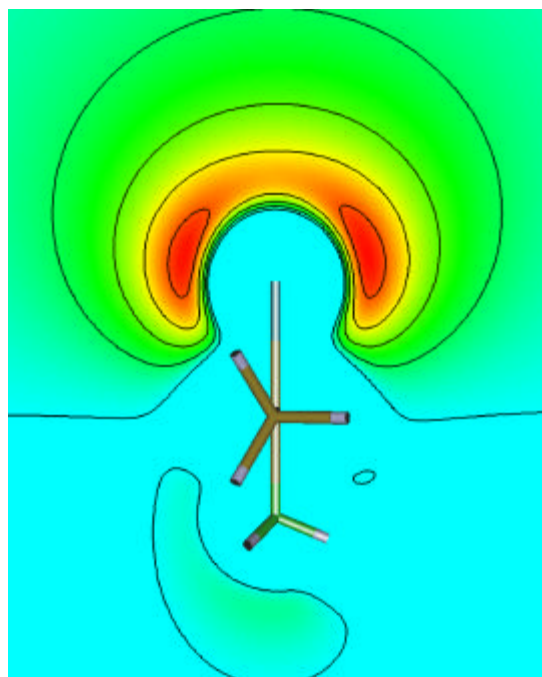




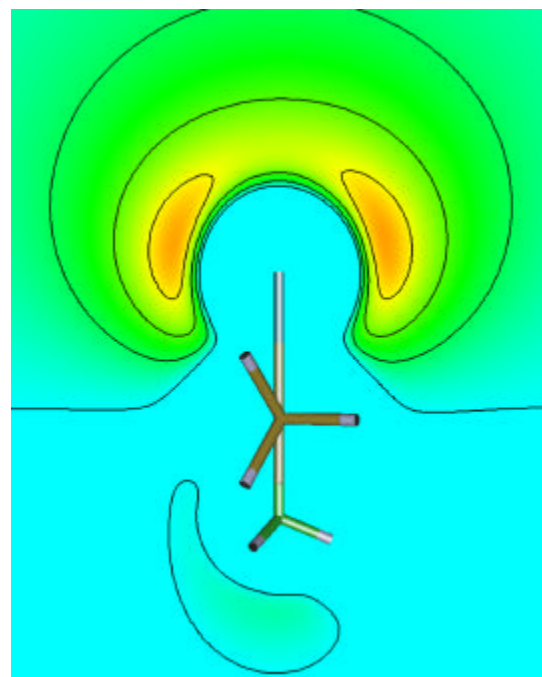
PdF(Me)(PH₃)₂



PdCl(Me)(PH₃)₂



PdBr(Me)(PH₃)₂



PdI(Me)(PH₃)₂

Figure S3.