

## Supplementary data

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### One-electron oxidation-induced dimerizing C-C coupling of a 2,5-diamino-1,4-benzoquinonediimine: a chemical and electrochemical investigation<sup>†</sup>

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**Extended Hückel calculations**

**Table S1:** Atomic parameters used for the extended Hückel calculations<sup>a</sup>

Atom	s shell		p shell	
	H <sub>ii</sub> (eV)	Exponent $\zeta$	H <sub>ii</sub> (eV)	Exponent $\zeta$
H	-13.4	1.3		
C	-21.4	1.625	-11.4	1.625
N	-26.0	1.95	-13.4	1.95

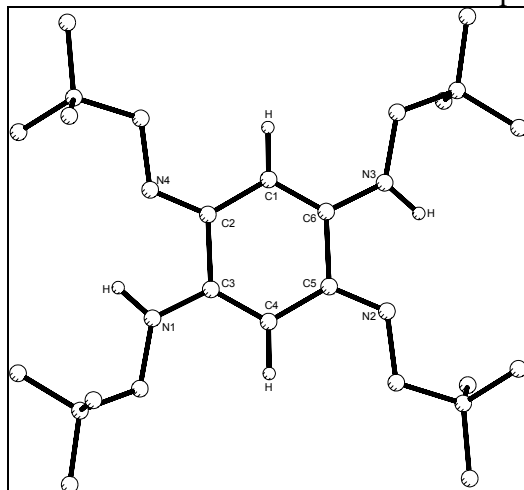
<sup>a</sup> from a) R. Hoffmann, W. N. Lipscomb, *J. Chem. Phys.* **1962**, *36*, 2179; b) R. Hoffmann, W. N. Lipscomb, *J. Chem. Phys.* **1962**, *37*, 2872; (c) R. Hoffmann, *J. Chem. Phys.* **1963**, *39*, 1397.

**Table S2:** Mulliken orbital populations obtained <sup>a</sup> for [3] and [3]<sup>+</sup>.

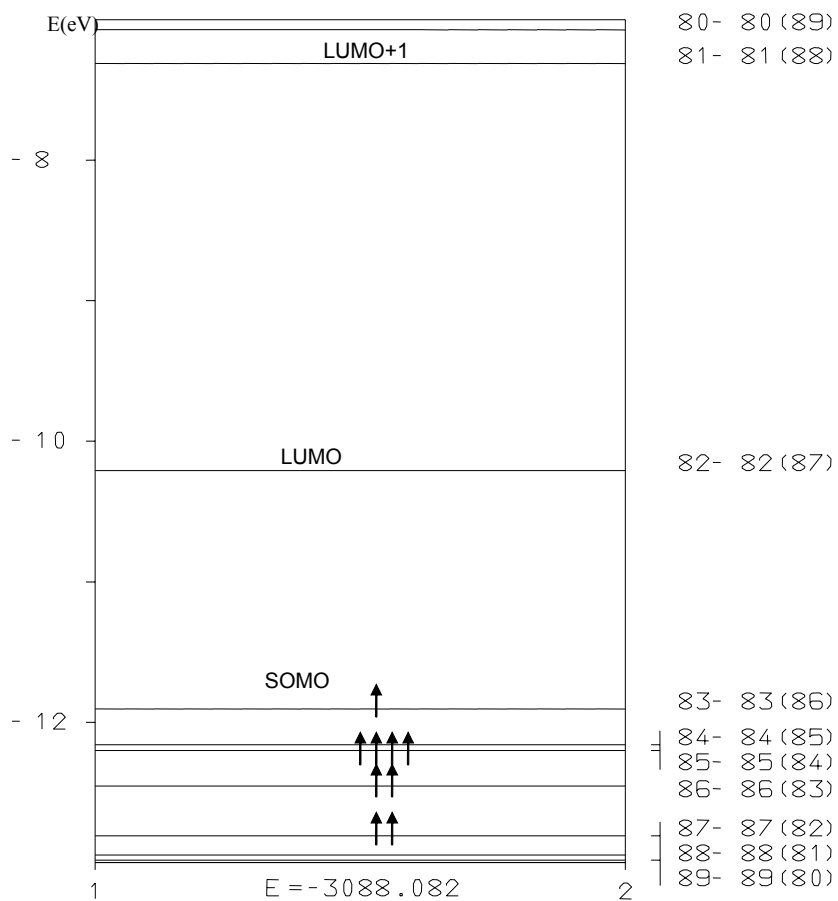
Atom/Orbitals	[3]	[3] <sup>+</sup>
<b>C(1)</b>		
s	1.163	1.163
p <sub>x</sub>	0.968	0.956
p <sub>y</sub>	1.220	0.986
p <sub>z</sub>	0.944	0.944
<b>net charge</b>	<b>-0.296</b>	<b>-0.049</b>
<b>C(2)</b>		
s	1.095	1.095
p <sub>x</sub>	0.781	0.781
p <sub>y</sub>	0.755	0.746
p <sub>z</sub>	0.902	0.902
<b>net charge</b>	<b>0.467</b>	<b>0.477</b>
<b>C(3)</b>		
s	1.106	1.106
p <sub>x</sub>	0.795	0.795
p <sub>y</sub>	0.802	0.781
p <sub>z</sub>	0.903	0.903
<b>net charge</b>	<b>0.394</b>	<b>0.415</b>
<b>C(4)</b>		
s	1.163	1.163
p <sub>x</sub>	0.968	0.956
p <sub>y</sub>	1.220	0.986
p <sub>z</sub>	0.944	0.944
<b>net charge</b>	<b>-0.296</b>	<b>-0.049</b>
<b>C(5)</b>		
s	1.095	1.095
p <sub>x</sub>	0.781	0.781
p <sub>y</sub>	0.755	0.746
p <sub>z</sub>	0.902	0.902
<b>net charge</b>	<b>0.467</b>	<b>0.477</b>
<b>C(6)</b>		
s	1.106	1.106
p <sub>x</sub>	0.795	0.795
p <sub>y</sub>	0.802	0.781
p <sub>z</sub>	0.903	0.903
<b>net charge</b>	<b>0.394</b>	<b>0.415</b>
<b>N(1)</b>		
s	1.362	1.362
p <sub>x</sub>	1.185	1.184
p <sub>y</sub>	1.670	1.546
p <sub>z</sub>	1.141	1.141
<b>net charge</b>	<b>-0.358</b>	<b>-0.232</b>
<b>N(2)</b>		
s	1.456	1.456
p <sub>x</sub>	1.591	1.591
p <sub>y</sub>	1.549	1.484

p <sub>z</sub>	1.394	1.394
<b>net charge</b>	<b>-0.990</b>	<b>-0.925</b>
<b>N(3)</b>		
s	1.362	1.362
p <sub>x</sub>	1.185	1.184
p <sub>y</sub>	1.670	1.546
p <sub>z</sub>	1.141	1.141
<b>net charge</b>	<b>-0.358</b>	<b>-0.232</b>
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p <sub>z</sub>	1.394	1.394
<b>net charge</b>	<b>-0.990</b>	<b>-0.925</b>

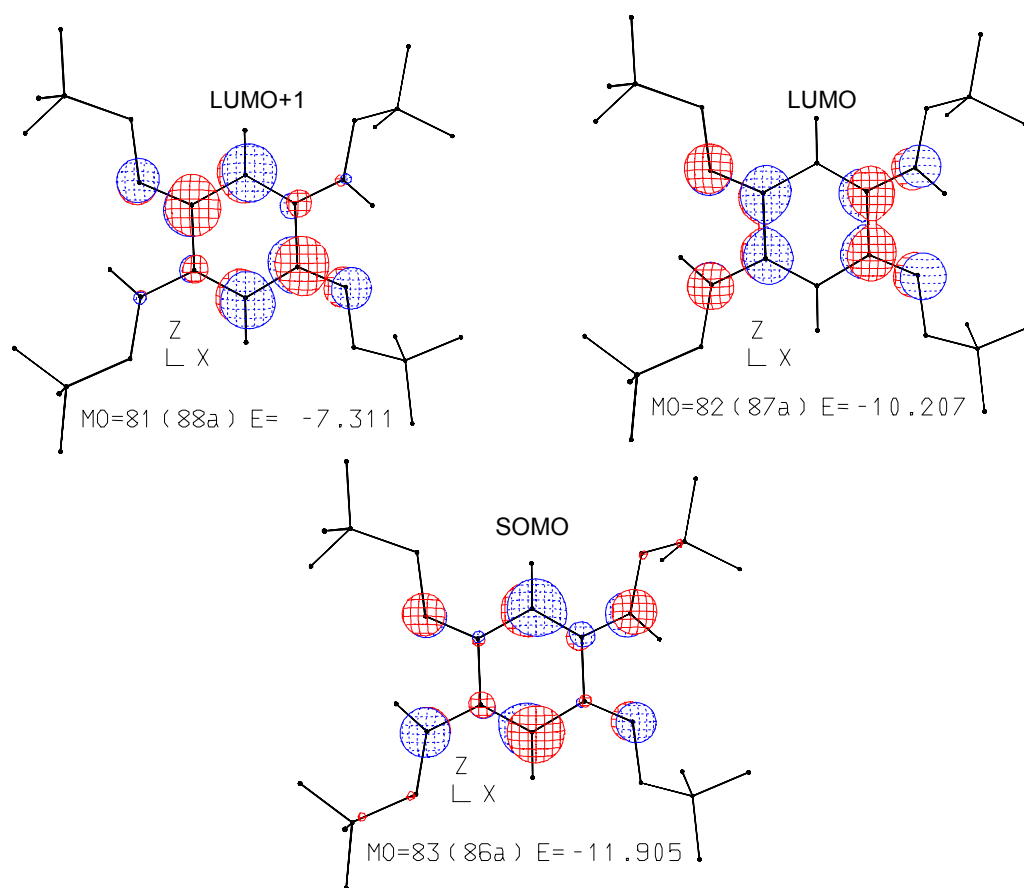
<sup>a</sup> in electron, from extended Hückel (EHT) calculations. The z axis is colinear with C(1)-C(4) and xz corresponds approximately to the plane of the C<sub>6</sub> ring.



**Figure S1:** Atom numbering in compound **3**



**Figure S2:** MO diagrams of the frontier orbitals for the  $[3]^+$



**Figure S3:** MO representation of the LUMO+x ( $x = 0-1$ ) and SOMO for  $[3]^+$