

Experimental determination of the conformational free energies (A Values) of fluorinated substituents in cyclohexane by dynamic ^{19}F NMR spectroscopy, Part 1: description of the method for the trifluoromethyl group.

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Chemical shifts must be read with one digit after the comma, but all the numbers are required and are given by the software of simulated spectra.

Table 1: ^{19}F NMR chemical shifts (ppm) of 4-methyl-1-trifluoromethylcyclohexane^a

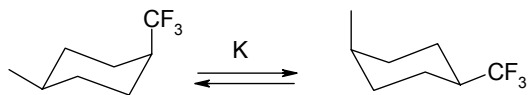
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		δ_{obs} ^b	δ_{ax} ^{b,c}	δ_{eq} ^{b,c}
298	-71.912	-70.341	<i>-64.402</i>	<i>-72.547</i>
293	-71.902	-70.355	<i>-64.366</i>	<i>-72.514</i>
288	-71.890	-70.370	<i>-64.329</i>	<i>-72.480</i>
283	-71.878	-70.385	<i>-64.293</i>	<i>-72.446</i>
278	-71.865	-70.399	<i>-64.256</i>	<i>-72.412</i>
273	-71.851	-70.416	<i>-64.220</i>	<i>-72.379</i>
268	-71.836	-70.432	<i>-64.183</i>	<i>-72.345</i>
263	-71.820	-70.449	<i>-64.146</i>	<i>-72.311</i>
258	-71.803	-70.471	<i>-64.110</i>	<i>-72.277</i>
253	-71.785	-70.493	<i>-64.073</i>	<i>-72.244</i>
243	-71.746	-	-	-
223	-71.657	-	-	-
213	-71.605	-	-	-
203	-71.548	-	-63.686	-71.901
193	-71.491	-	-63.635	-71.839
188	-71.460	-	-63.600	-71.807
183	-71.428	-	-63.562	-71.772
178	-71.393	-	-63.526	-71.738

^a Chemical shifts are in ppm and were measured in THF- d_8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression: $\delta_{\text{ax}} = -62.224 - 7.31 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -70.536 - 6.75 \cdot 10^{-3} T$

Table 2: Equilibrium parameters of *cis*-4-methyl-1-trifluoromethylcyclohexane



T / K	K ^a	$\Delta G^\circ / \text{J.mol}^{-1}$
298	2.690	-2452
293	2.775	-2486
288	2.863	-2519
283	2.956	-2551
278	3.052	-2579
273	3.157	-2610
268	3.267	-2638
263	3.383	-2665
258	3.520	-2700
253	3.666	-2733

^a Equilibrium constants were calculated from the values of the chemical shifts: $K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$

Table 3: ^{19}F NMR chemical shifts (ppm) of 4-ethyl-1-trifluoromethylcyclohexane^a

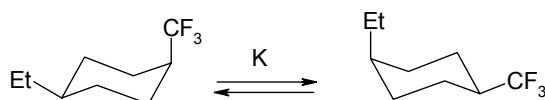
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
298	-73.746	-72.191	-66.227	-74.392
293	-73.735	-72.214	-66.189	-74.358
288	-73.724	-72.238	-66.151	-74.323
283	-73.712	-72.261	-66.114	-74.288
278	-73.699	-72.283	-66.076	-74.254
273	-73.686	-72.304	-66.038	-74.219
268	-73.671	-72.328	-66.000	-74.184
263	-73.656	-72.355	-65.962	-74.150
258	-73.639	-72.383	-65.925	-74.115
253	-73.622	-72.412	-65.887	-74.080
248	-73.603	-72.459	-65.849	-74.046
228	-73.515	-	-	-
208	-73.411	-	-	-
198	-73.353	-	-	-
188	-73.290	-	-65.396	-73.635
183	-73.257	-	-65.358	-73.602
178	-73.223	-	-65.321	-73.567
173	-73.187	-	-65.282	-73.531
168	-73.152	-	-65.245	-73.496

^a Chemical shifts are in ppm and were measured in THF- d_8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression: $\delta_{\text{ax}} = -63.974 - 7.56 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -72.327 - 6.93 \cdot 10^{-3} T$

Table 4: Equilibrium parameters of *cis*-4-ethyl-1-trifluoromethylcyclohexane



T / K	K^{a}	$\Delta G^{\circ} / \text{J} \cdot \text{mol}^{-1}$
298	2.710	-2470
293	2.811	-2518
288	2.918	-2564
283	3.031	-2609
278	3.149	-2651
273	3.272	-2691
268	3.409	-2732
263	3.562	-2778
258	3.730	-2823
253	3.911	-2869

^a Equilibrium constants were calculated from the values of the chemical shifts: $K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$

Table 5: ^{19}F NMR chemical shifts (ppm) of 4-cyclohexyl-1-trifluoromethylcyclohexane^a

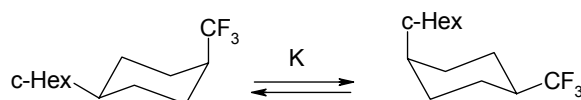
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
298	-73.788	-71.252	<i>-66.233</i>	<i>-74.304</i>
293	-73.778	-71.285	<i>-66.195</i>	<i>-74.273</i>
288	-73.768	-71.319	<i>-66.156</i>	<i>-74.242</i>
283	-73.756	-71.356	<i>-66.118</i>	<i>-74.211</i>
278	-73.744	-71.392	<i>-66.079</i>	<i>-74.180</i>
273	-73.731	-71.428	<i>-66.041</i>	<i>-74.149</i>
268	-73.717	-71.468	<i>-66.002</i>	<i>-74.118</i>
263	-73.702	-71.510	<i>-65.964</i>	<i>-74.087</i>
258	-73.686	-71.553	<i>-65.925</i>	<i>-74.056</i>
253	-73.669	-71.599	<i>-65.887</i>	<i>-74.025</i>
238	-73.613	-	-	-
223	-73.546	-	-	-
208	-73.470	-	-	-
198	-73.413	-	<i>-65.452</i>	<i>-73.683</i>
193	-73.383	-	<i>-65.426</i>	<i>-73.653</i>
188	-73.351	-	<i>-65.389</i>	<i>-73.624</i>
183	-73.321	-	<i>-65.349</i>	<i>-73.594</i>
178	-73.287	-	<i>-65.311</i>	<i>-73.561</i>
173	-73.251	-	<i>-65.273</i>	<i>-73.527</i>

^a Chemical shifts are in ppm and were measured in THF- d_8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression: $\delta_{\text{ax}} = -63.94 - 7.69 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -72.46 - 6.19 \cdot 10^{-3} T$

Table 6: Equilibrium parameters of *cis*-4-cyclohexyl-1-trifluoromethylcyclohexane



T / K	K^{a}	$\Delta G^{\circ} / \text{J} \cdot \text{mol}^{-1}$
298	1.645	-1233
293	1.704	-1298
288	1.766	-1362
283	1.835	-1428
278	1.906	-1491
273	1.980	-1550
268	2.063	-1613
263	2.152	-1675
258	2.248	-1737
253	2.354	-1801

^a Equilibrium constants were calculated from the values of the chemical shifts: $K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$

Table 7: ^{19}F NMR chemical shifts (ppm) of 4-isopropyl-1-trifluoromethylcyclohexane^a

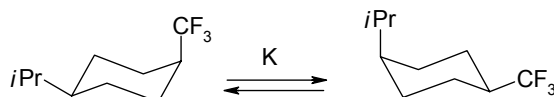
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
297	-73.767	-71.083	-66.371	-74.457
293	-73.760	-71.109	-66.336	-74.428
288	-73.748	-71.144	-66.293	-74.392
283	-73.737	-71.181	-66.249	-74.356
278	-73.723	-71.219	-66.206	-74.320
273	-73.710	-71.259	-66.162	-74.284
268	-73.696	-71.300	-66.119	-74.248
263	-73.680	-71.342	-66.075	-74.212
258	-73.664	-71.384	-66.032	-74.177
253	-73.647	-71.420	-65.988	-74.141
248	-73.629	-	-	-
228	-73.544	-	-	-
208	-73.443	-	-	-
198	-73.387	-	-	-
188	-73.324	-	-65.424	-73.674
183	-73.289	-	-65.381	-73.640
178	-73.251	-	-65.337	-73.602
173	-73.215	-	-65.294	-73.566

^a Chemical shifts are in ppm and were measured in THF- d^8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression: $\delta_{\text{ax}} = -63.79 - 8.69 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -72.32 - 7.18 \cdot 10^{-3} T$

Table 8: Equilibrium parameters of *cis*-4-isopropyl-1-trifluoromethylcyclohexane



T / K	K^{a}	$\Delta G^{\circ} / \text{J} \cdot \text{mol}^{-1}$
297	1.397	-8256
293	1.438	-885
288	1.493	-960
283	1.554	-1036
278	1.617	-1110
273	1.684	-1183
268	1.758	-1257
263	1.835	-1327
258	1.916	-1395
253	1.996	-1454

^a Equilibrium constants were calculated from the values of the chemical shifts: $K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$

Table 9: ^{19}F NMR chemical shifts (ppm) of 4-*tert*butyl-1-trifluoromethylcyclohexane^a

T / K	<i>trans</i> isomer	<i>cis</i> isomer
297	-73.785	-66.396
293	-73.777	-66.360
288	-73.766	-66.315
283	-73.754	-66.278
278	-73.741	-66.229
273	-73.728	-66.186
268	-73.714	-66.145
263	-73.697	-66.100
258	-73.681	-66.058
253	-73.664	-66.016
248	-73.646	-65.973
228	-73.560	-65.806
208	-73.458	-65.643
188	-73.337	-65.481
168	-73.198	-65.314

^a Chemical shifts are in ppm and were measured in THF- d^8 relative to CFCl_3 .

Table 10: ^{19}F NMR chemical shifts (ppm) of 4-phenyl-1-trifluoromethylcyclohexane^a

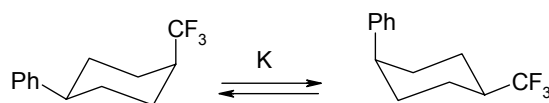
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		δ_{obs} ^b	δ_{ax} ^{b,c}	δ_{eq} ^{b,c}
298	-71.832	-66.772	-64.275	-72.489
293	-71.823	-66.741	-64.238	-72.459
288	-71.813	-66.710	-64.200	-72.429
283	-71.802	-66.686	-64.163	-72.399
278	-71.791	-66.656	-64.125	-72.369
273	-71.779	-66.622	-64.088	-72.339
268	-71.768	-66.586	-64.050	-72.309
263	-71.753	-66.543	-64.013	-72.279
258	-71.739	-	-	-
253	-71.723	-	-	-
233	-71.648	-	-	-
213	-71.556	-	-	-
203	-71.503	-	-	-
198	-71.479	-	-	-
193	-71.460	-	-63.488	-
188	-71.432	-	-63.452	-71.829
183	-71.402	-	-63.415	-71.799
178	-71.370	-	-63.376	-71.770
173	-71.338	-	-63.339	-71.739

^a Chemical shifts are in ppm and were measured in THF- d_8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression: $\delta_{\text{ax}} = -62.04 - 7.49 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -70.70 - 5.99 \cdot 10^{-3} T$

Table 11: Equilibrium parameters of *cis*-4-phenyl-1-trifluoromethylcyclohexane



T / K	K^a	$\Delta G^\circ / \text{J}\cdot\text{mol}^{-1}$
298	0.437	2053
293	0.438	2011
288	0.439	1972
283	0.442	1923
278	0.443	1883
273	0.443	1846
268	0.443	1814
263	0.441	1790

^a Equilibrium constants were calculated from the values of the chemical shifts: $K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$

Table 12: ^{19}F NMR chemical shifts (ppm) of trifluoromethylcyclohexane^a

T / K	δ_{obs}
298	-74.055
293	-74.052
288	-74.048
283	-74.043
278	-74.036
273	-74.029
268	-74.021
263	-74.012
258	-74.002
253	-73.988
248	-73.975
228	-73.911
208	-73.834
188	-73.714
168	-73.596

^aChemical shifts are in ppm and were measured in THF-d⁸ relative to CFCl₃.

Table 13: ^{19}F NMR chemical shifts (ppm) of 1,4-bis(trifluoromethyl)-cyclohexane^a

T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
297	-73.819	-70.207	-	-
288	-73.798	-70.174	-	-
278	-73.771	-70.134	-	-
268	-73.741	-70.092	-	-
258	-73.707	-70.047	-	-
248	-73.669	-70.001	-	-
228	-73.579	-	-	-
208	-73.485	-	-	-
188	-73.362	-	-65.655	-73.655
183	-73.330	-	-65.621	-73.617
178	-73.297	-	-65.590	-73.582
173	-73.260	-	-65.553	-73.545

^a Chemical shifts are in ppm and were measured in THF- d_8 relative to CFCl_3 .

Table 14: ^{19}F NMR chemical shifts (ppm) of 4-isopropyl-1-trifluoromethylcyclohexane in toluene- d_8^a

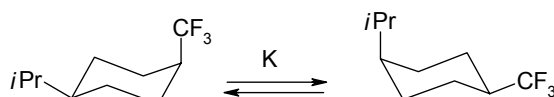
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		δ_{obs}^b	$\delta_{\text{ax}}^{b,c}$	$\delta_{\text{eq}}^{b,c}$
297	-78.903	-76.272	<i>-71.810</i>	<i>-79.418</i>
293	-78.888	-76.296	<i>-71.758</i>	<i>-79.383</i>
288	-78.868	-76.329	<i>-71.693</i>	<i>-79.340</i>
283	-78.847	-76.361	<i>-71.629</i>	<i>-79.296</i>
278	-78.826	-76.395	<i>-71.565</i>	<i>-79.252</i>
273	-78.803	-76.431	<i>-71.501</i>	<i>-79.209</i>
268	-78.779	-76.465	<i>-71.436</i>	<i>-79.165</i>
263	-78.753	-76.502	<i>-71.372</i>	<i>-79.121</i>
243	-78.633	-	-	-
223	-78.489	-	-	-
203	-78.325	-	-70.600	-78.597
193	-78.235	-	-70.462	-78.511
188	-78.190	-	-70.409	-78.465
183	-78.177	-	-70.387	-78.452
178	-78.174	-	-70.380	-78.448

^a Chemical shifts are in ppm and were measured in toluene- d_8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression: $\delta_{\text{ax}} = -67.982 - 1.289 \cdot 10^{-2} \cdot T$
 $\delta_{\text{eq}} = -76.823 - 8.73 \cdot 10^{-3} \cdot T$

Table 15: Equilibrium parameters of *cis*-4-isopropyl-1-trifluoromethylcyclohexane



T / K	K^a	$\Delta G^\circ / \text{J} \cdot \text{mol}^{-1}$
297	1.418	-863
293	1.470	-939
288	1.539	-1033
283	1.612	-1123
278	1.691	-1214
273	1.774	-1302
268	1.862	-1386
263	1.959	-1470

^a Equilibrium constants were calculated from the values of the chemical shifts: $K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$

Table 16: ^{19}F NMR chemical shifts (ppm) of 4-isopropyl-1-trifluoromethylcyclohexane in methanol- d_4^a

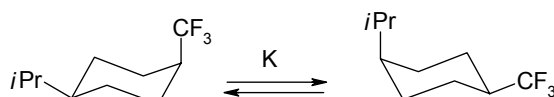
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		δ_{obs}^b	$\delta_{\text{ax}}^{b,c}$	$\delta_{\text{eq}}^{b,c}$
297	-71.972	-69.172	-64.518	-72.707
293	-71.968	-69.225	-64.485	-72.678
288	-71.962	-69.277	-64.444	-72.642
283	-71.954	-69.329	-64.403	-72.606
278	-71.944	-69.383	-64.361	-72.570
273	-71.934	-69.434	-64.319	-72.534
268	-71.922	-69.487	-64.278	-72.497
263	-71.909	-69.539	-64.237	-72.461
243	-71.850	-	-	-
223	-71.764	-	-	-
203	-71.647	-	-	-72.025
193	-71.580	-	-63.655	-71.956
188	-71.546	-	-63.619	-71.920
183	-71.509	-	-63.573	-71.882
178	-71.470	-	-63.532	-71.843

^a Chemical shifts are in ppm and were measured in CD_3OD relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression: $\delta_{\text{ax}} = -62.056 - 8.29 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -70.557 - 7.24 \cdot 10^{-3} T$

Table 17: Equilibrium parameters of *cis*-4-isopropyl-1-trifluoromethylcyclohexane



T / K	K^a	$\Delta G^\circ / \text{J}\cdot\text{mol}^{-1}$
297	1.316	-679
293	1.372	-771
288	1.436	-867
283	1.504	-960
278	1.576	-1051
273	1.650	-1137
268	1.730	-1221
263	1.814	-1302

^a Equilibrium constants were calculated from the values of the chemical shifts: $K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$

Table 18: ^{19}F NMR chemical shifts (ppm) of 4-methyl-1-trifluoromethylcyclohexanol^a

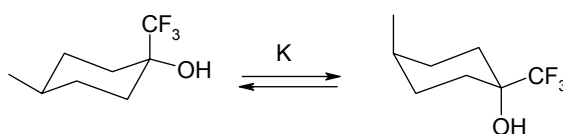
T / K	<i>cis</i> isomer ^b	<i>trans</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
298	-84.560	-83.111	-77.228	-85.300
293	-84.555	-83.140	-77.206	-85.275
288	-84.550	-83.169	-77.184	-85.250
283	-84.545	-83.199	-77.162	-85.225
278	-84.540	-83.228	-77.140	-85.201
273	-84.533	-83.257	-77.118	-85.176
268	-84.526	-83.287	-77.096	-85.151
263	-84.519	-83.319	-77.074	-85.126
258	-84.510	-83.355	-77.052	-85.102
253	-84.500	-83.403	-77.029	-85.077
248	-84.491	-	-	-
243	-84.480	-	-	-
238	-84.467	-	-	-
233	-84.454	-	-	-
228	-84.439	-	-	-
218	-84.408	-	-	-
208	-84.373	-	-	-
203	-84.354	-	-	-
198	-84.335	-	-	-
193	-84.314	-	-	-84.780
188	-84.292	-	-76.742	-84.756
183	-84.269	-	-76.718	-84.733
178	-84.245	-	-76.698	-84.707
173	-84.220	-	-76.675	-84.681
168	-84.208	-	-76.653	-84.657

^a Chemical shifts are in ppm and were measured in THF- d_8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression: $\delta_{\text{ax}} = -75.911 - 4.42 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -83.825 - 4.95 \cdot 10^{-3} T$

Table 19: Equilibrium parameters of *trans*-4-methyl-1-trifluoromethylcyclohexanol^a



T / K	K^{a}	$\Delta G^{\circ} / \text{J} \cdot \text{mol}^{-1}$
298	2.688	-2450
293	2.779	-2490
288	2.875	-2529
283	2.978	-2568
278	3.0859	-2604
273	3.198	-2639
268	3.321	-2674
263	3.456	-2711
258	3.607	-2752
253	3.806	-2812

^a K were calculated from the values of the chemical shifts: $K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$

Table 20: ^{19}F NMR chemical shifts (ppm) of 4-isopropyl-1-trifluoromethylcyclohexanol^a

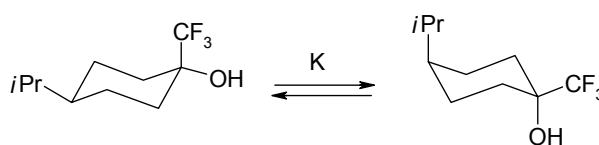
T / K	<i>cis</i> isomer ^b	<i>trans</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
293	-86.413	-83.984	<i>-79.089</i>	<i>-87.123</i>
288	-86.408	-84.038	<i>-79.064</i>	<i>-87.100</i>
283	-86.402	-84.094	<i>-79.038</i>	<i>-87.076</i>
278	-86.396	-84.151	<i>-79.013</i>	<i>-87.052</i>
273	-86.388	-84.208	<i>-78.987</i>	<i>-87.0277</i>
268	-86.380	-84.268	<i>-78.962</i>	<i>-87.004</i>
263	-86.371	-84.330	<i>-78.936</i>	<i>-86.980</i>
258	-86.362	-84.391	<i>-78.911</i>	<i>-86.956</i>
253	-86.352	-84.458	<i>-78.885</i>	<i>-86.932</i>
248	-86.340	-84.523	<i>-78.859</i>	<i>-86.908</i>
243	-86.328	-	-	-
238	-86.315	-	-	-
233	-86.301	-	-	-
228	-86.286	-	-	-
218	-86.251	-	-	-
208	-86.214	-	-	-
198	-86.172	-	-	-
193	-86.150	-	-	-86.620
188	-86.125	-	-78.5281	-86.598
183	-86.100	-	-78.5014	-86.574
178	-86.074	-	-78.4746	-86.550
173	-86.047	-	-78.4515	-86.526
168	-86.022	-	-78.4253	-86.500

^a Chemical shifts are in ppm and were measured in THF- d_8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression: $\delta_{\text{ax}} = -77.567 - 5.11 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -85.696 - 4.79 \cdot 10^{-3} T$

Table 21: Equilibrium parameters of *trans*-4-isopropyl-1-trifluoromethylcyclo-hexanol^a



T / K	K^{a}	$\Delta G^{\circ} / \text{J} \cdot \text{mol}^{-1}$
293	1.579	-1114
288	1.646	-1194
283	1.718	-1273
278	1.795	-1352
273	1.877	-1429
268	1.9667	-1507
263	2.064	-1584
258	2.167	-1658
253	2.285	-1738
248	2.409	-1813

^a Equilibrium constants were calculated from the values of the chemical shifts: $K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$