

Experimental determination of the conformational free energies (A Values) of fluorinated substituents in cyclohexane by dynamic ¹⁹F NMR spectroscopy, Part 2: extension to fluoromethyl, difluoromethyl, pentafluoroethyl, trifluoromethylthio and trifluoromethoxy groups.

Y.CARCENAC, M. TORDEUX, C. WAKSELMAN, P. DITER

SIRCOB, UMR CNRS 8086, Université de Versailles-Saint-Quentin en Yvelines, 45 Avenue des Etats-Unis, 78035 Versailles, France. E-mail: patrick.diter@chimie.uvsq.fr

Chemical shifts must be read with one digit after the comma, but all the numbers are required and are given by the software of simulated spectra.

Table 1: ¹⁹F chemical shifts ^a of 4-methyl-1-pentafluoroethylcyclohexane (4a)

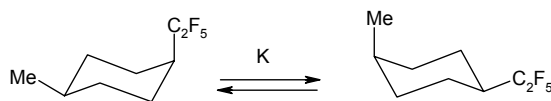
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		δ_{obs}^b	$\delta_{\text{ax}}^{b,c}$	$\delta_{\text{eq}}^{b,c}$
298	-124.000	-122.969	<i>-116.083</i>	<i>-124.551</i>
293	-124.035	-123.020	<i>-116.057</i>	<i>-124.568</i>
288	-124.070	-123.078	<i>-116.031</i>	<i>-124.585</i>
283	-124.105	-123.134	<i>-116.005</i>	<i>-124.601</i>
278	-124.139	-123.192	<i>-115.979</i>	<i>-124.618</i>
273	-124.174	-123.251	<i>-115.953</i>	<i>-124.635</i>
268	-124.208	-123.311	<i>-115.928</i>	<i>-124.651</i>
263	-124.240	-123.368	<i>-115.902</i>	<i>-124.668</i>
258	-124.274	-123.433	<i>-115.876</i>	<i>-124.685</i>
253	-124.307	-	-	-
248	-124.339	-	-	-
243	-124.372	-	-	-
238	-124.404	-	-	-
233	-124.435	-	-	-
228	-124.466	-	-	-
223	-124.496	-	-	-
218	-124.523	-	-	-
213	-124.552	-	-	-
208	-124.579	-	-	-
203	-124.605	-	-	-
198	-124.631	-	-	-
193	-124.656	-	(-124.896)	-115.537
188	-124.679	-	-124.918	-115.515
183	-124.702	-	-124.935	-115.492
178	-124.724	-	-124.953	-115.464
173	-124.744	-	-124.969	-115.438
168	-124.764	-	-124.985	-115.408

^a Chemical shifts are in ppm and were measured in THF-d⁸ relative to CFCl₃.

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression : $\delta_{\text{ax}} = -114.542 - 5.17 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -125.547 + 3.34 \cdot 10^{-3} T$

Table 2: Thermodynamic parameters of *cis*-4-methyl-1-pentafluoroethylcyclohexane



T / K	K ^a	ΔG° / J.mol ⁻¹
298	4.352	-3643.8
293	4.499	-3663.3
288	4.676	-3693.2
283	4.859	-3719.6
278	5.057	-3746.0
273	5.274	-3774.0
268	5.509	-3802.2
263	5.745	-3822.8
258	6.035	-3855.7

^a Equilibrium constants were calculated from the values of the chemical shifts :

$$K = (\delta_{ax} - \delta_{obs}) / (\delta_{obs} - \delta_{eq})$$

Table 3: ^{19}F chemical shifts ^a of 4-isopropyl-1-pentafluoroethylcyclohexane (4b)

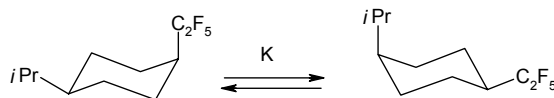
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
298	-121.307	-119.520	<i>-113.507</i>	<i>-122.028</i>
293	-121.344	-119.603	<i>-113.476</i>	<i>-122.043</i>
288	-121.378	-119.681	<i>-113.445</i>	<i>-122.059</i>
283	-121.415	-119.764	<i>-113.414</i>	<i>-122.074</i>
278	-121.453	-119.855	<i>-113.384</i>	<i>-122.089</i>
273	-121.489	-119.945	<i>-113.353</i>	<i>-122.105</i>
268	-121.526	-120.037	<i>-113.322</i>	<i>-122.120</i>
263	-121.563	-120.133	<i>-113.291</i>	<i>-122.135</i>
258	-121.598	-120.227	<i>-113.260</i>	<i>-122.151</i>
238	-121.735	-	-	-
218	-121.855	-	-	-
198	-121.957	-	-	-
193	-121.984	-	-	-
188	-122.008	-	-122.365	-112.830
183	-122.030	-	-122.383	-112.798
178	-122.052	-	-122.397	-112.765
173	-122.074	-	-122.412	-112.739

^a Chemical shifts are in ppm and were measured in THF- d^8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression : $\delta_{\text{ax}} = -111.671 - 6.16 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -122.923 + 3.07 \cdot 10^{-3} T$

Table 4: Thermodynamic parameters of *cis*-4-isopropyl-1-pentafluoroethylcyclohexane



T / K	K ^a	$\Delta G^\circ / \text{J.mol}^{-1}$
298	2.398	-2167.0
293	2.511	-2242.6
288	2.622	-2308.1
283	2.749	-2379.4
278	2.897	-2458.3
273	3.053	-2533.2
268	3.224	-2608.3
263	3.416	-2686.0
258	3.622	-2760.5

^a Equilibrium constants were calculated from the values of the chemical shifts :

$$K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$$

Table 5: ^{19}F chemical shifts ^a of 4-methyl-1-fluoromethylcyclohexane (9a)

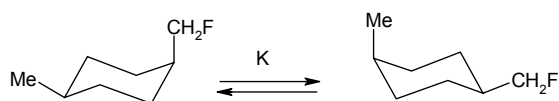
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		δ_{obs}^b	$\delta_{\text{ax}}^{b,c}$	$\delta_{\text{eq}}^{b,c}$
298	-223.819	-223.485	<i>-222.710</i>	<i>-224.539</i>
293	-223.782	-223.460	<i>-222.694</i>	<i>-224.482</i>
288	-223.744	-223.435	<i>-222.679</i>	<i>-224.425</i>
283	-223.706	-223.409	<i>-222.663</i>	<i>-224.368</i>
278	-223.668	-223.383	<i>-222.647</i>	<i>-224.311</i>
273	-223.628	-223.357	<i>-222.632</i>	<i>-224.254</i>
268	-223.589	-223.330	<i>-222.616</i>	<i>-224.197</i>
263	-223.549	-223.302	<i>-222.601</i>	<i>-224.140</i>
258	-223.508	-223.275	<i>-222.585</i>	<i>-224.083</i>
253	-223.466	-223.247	<i>-222.569</i>	<i>-224.026</i>
248	-223.425	-	-	-
243	-223.382	-	-	-
238	-223.339	-	-	-
233	-223.294	-	-	-
228	-223.249	-	-	-
223	-223.203	-	-	-
218	-223.158	-	-	-
213	-223.112	-	-	-
208	-223.066	-	-	-
203	-223.021	-	-	-
198	-222.975	-	-	-
193	-222.927	-	-222.382	-223.345
188	-222.877	-	-222.367	-223.289
183	-222.823	-	-222.349	-223.230
178	-222.771	-	-222.333	-223.172
173	-222.720	-	-222.317	-223.113
168	-222.667	-	-222.302	-223.052

^a Chemical shifts are in ppm and were measured in THF- d^8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression : $\delta_{\text{ax}} = -221.777 - 3.13 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -221.142 + 1.14 \cdot 10^{-2} T$

Table 6: Thermodynamic parameters of *cis*-4-methyl-1-fluoromethylcyclohexane



T / K	K^a	$\Delta G^\circ / \text{J} \cdot \text{mol}^{-1}$
298	0.735	762.7
293	0.749	704.4
288	0.764	645.6
283	0.778	592.0
278	0.792	537.9
273	0.808	483.3
268	0.823	433.9
263	0.837	389.6
258	0.854	339.0
253	0.870	293.5

^a Equilibrium constants were calculated from the values of the chemical shifts :

$$K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$$

Table 7: ^{19}F chemical shifts ^a of 4-isopropyl-1-fluoromethylcyclohexane (9b)

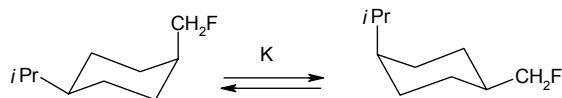
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
298	-222.867	-222.423	<i>-222.003</i>	<i>-223.569</i>
293	-222.824	-222.405	<i>-221.983</i>	<i>-223.521</i>
288	-222.780	-222.385	<i>-221.964</i>	<i>-223.474</i>
283	-222.735	-222.365	<i>-221.944</i>	<i>-223.426</i>
278	-222.690	-222.345	<i>-221.924</i>	<i>-223.378</i>
273	-222.644	-222.323	<i>-221.904</i>	<i>-223.330</i>
268	-222.597	-222.302	<i>-221.885</i>	<i>-223.283</i>
263	-222.550	-222.280	<i>-221.865</i>	<i>-223.235</i>
258	-222.503	-222.258	<i>-221.845</i>	<i>-223.187</i>
253	-222.455	-222.236	<i>-221.825</i>	<i>-223.139</i>
233	-222.274	-	-	-
213	-222.074	-	-	-
193	-221.890	-	-221.590	-222.565
188	-221.835	-	-221.569	-222.519
183	-221.777	-	-221.549	-222.471
178	-221.720	-	-221.529	-222.422
173	-221.661	-	-221.511	-222.375

^a Chemical shifts are in ppm and were measured in THF- d^8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression : $\delta_{\text{ax}} = -220.826 - 3.95 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -220.723 - 9.55 \cdot 10^{-3} T$

Table 8: Thermodynamic parameters of *cis*-4-isopropyl-1-fluoromethylcyclohexane



T / K	K^{a}	$\Delta G^{\circ} / \text{J.mol}^{-1}$
298	0.367	2485.2
293	0.377	2373.3
288	0.387	2272.4
283	0.396	2178.3
278	0.407	2078.4
273	0.416	1990.6
268	0.425	1906.1
263	0.435	1821.2
258	0.444	1740.3
253	0.454	1659.7

^a Equilibrium constants were calculated from the values of the chemical shifts :

$$K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$$

Table 9: ^{19}F chemical shifts ^a of 4-methyl-1-difluoromethylcyclohexane (10a)

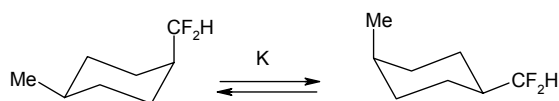
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		δ_{obs}^b	$\delta_{\text{ax}}^{b,c}$	$\delta_{\text{eq}}^{b,c}$
298	-125.879	-125.462	<i>-124.069</i>	<i>-126.684</i>
293	-125.876	-125.480	<i>-124.128</i>	<i>-126.652</i>
288	-125.873	-125.497	<i>-124.187</i>	<i>-126.621</i>
283	-125.868	-125.513	<i>-124.246</i>	<i>-126.590</i>
278	-125.864	-125.530	<i>-124.305</i>	<i>-126.559</i>
273	-125.859	-125.547	<i>-124.364</i>	<i>-126.528</i>
268	-125.853	-125.563	<i>-124.423</i>	<i>-126.496</i>
263	-125.845	-125.579	<i>-124.482</i>	<i>-126.465</i>
258	-125.840	-125.595	<i>-124.541</i>	<i>-126.434</i>
253	-125.832	-125.610	<i>-124.600</i>	<i>-126.403</i>
248	-125.824	-	-	-
243	-125.814	-	-	-
238	-125.804	-	-	-
233	-125.794	-	-	-
228	-125.782	-	-	-
223	-125.770	-	-	-
218	-125.755	-	-	-
213	-125.741	-	-	-
208	-125.720	-	-	-
203	-125.704	-	-125.186	-126.085
198	-125.685	-	-125.248	-126.059
193	-125.664	-	-125.307	-126.031
188	-125.641	-	-125.367	-126.003
183	-125.617	-	-125.426	-125.971
178	-125.590	-	-125.485	-125.938
173	-125.561	-	-125.561	-125.903
168	-125.534	-	-125.599	-125.866

^a Chemical shifts are in ppm and were measured in THF- d^8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression : $\delta_{\text{ax}} = -127.582 + 1.18 \cdot 10^{-2} T$
 $\delta_{\text{eq}} = -124.824 - 6.24 \cdot 10^{-3} T$

Table 10: Thermodynamic parameters of *cis*-4-methyl-1-difluoromethylcyclohexane



T / K	K ^a	$\Delta G^\circ / \text{J} \cdot \text{mol}^{-1}$
298	1.141	-326.7
293	1.153	-346.3
288	1.164	-364.3
283	1.177	-383.2
278	1.191	-404.9
273	1.206	-424.9
268	1.220	-443.7
263	1.237	-465.3
258	1.257	-490.4
253	1.275	-511.5

^a Equilibrium constants were calculated from the values of the chemical shifts :

$$K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$$

Table 11: ^{19}F chemical shifts ^a of 4-isopropyl-1-difluoromethylcyclohexane (10b)

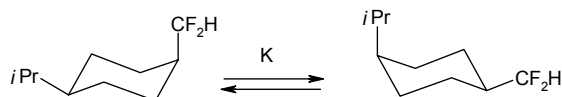
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
298	-123.097	-122.355	<i>-121.425</i>	<i>-123.961</i>
293	-123.092	-122.385	<i>-121.480</i>	<i>-123.927</i>
288	-123.087	-122.413	<i>-121.534</i>	<i>-123.894</i>
283	-123.081	-122.443	<i>-121.589</i>	<i>-123.860</i>
278	-123.075	-122.472	<i>-121.644</i>	<i>-123.826</i>
273	-123.068	-122.501	<i>-121.698</i>	<i>-123.793</i>
268	-123.061	-122.530	<i>-121.753</i>	<i>-123.759</i>
263	-123.053	-122.559	<i>-121.808</i>	<i>-123.725</i>
258	-123.043	-122.588	<i>-121.862</i>	<i>-123.692</i>
253	-123.034	-122.617	<i>-121.917</i>	<i>-123.658</i>
233	-122.984	-	-	-
213	-122.913	-	-	-
203	-122.867	-	-	-
193	-122.819	-	-	-
188	-122.785	-	-122.570	-123.250
183	-122.753	-	-122.627	-123.222
178	-122.719	-	-122.685	-123.190
173	-122.682	-	-122.79	-123.12
168	-122.641	-	-122.843	-123.083

^a Chemical shifts are in ppm and were measured in THF- d^8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression : $\delta_{\text{ax}} = -124.682 + 1.09 \cdot 10^{-2} T$
 $\delta_{\text{eq}} = -121.955 - 6.73 \cdot 10^{-3} T$

Table 12: Thermodynamic parameters of *cis*-4-isopropyl-1-difluoromethylcyclohexane



T / K	K^{a}	$\Delta G^{\circ} / \text{J.mol}^{-1}$
298	0.579	1352.0
293	0.586	1300.0
288	0.594	1247.9
283	0.602	1192.4
278	0.612	1136.4
273	0.621	1080.8
268	0.633	1020.2
263	0.645	960.1
258	0.658	897.9
253	0.673	832.7

^a Equilibrium constants were calculated from the values of the chemical shifts :

$$K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$$

Table 13: ^{19}F chemical shifts ^a of cyclohexyl trifluoromethyl sulfur (15a)

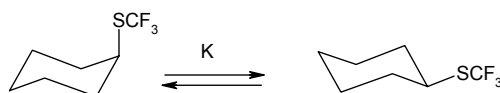
T / K	$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
298	-38.804	-39.980	-38.645
293	-38.798	-39.971	-38.631
288	-38.790	-39.961	-38.618
283	-38.782	-39.951	-38.605
278	-38.773	-39.940	-38.591
273	-38.763	-39.929	-38.578
268	-38.753	-39.917	-38.564
263	-38.741	-39.905	-38.551
258	-38.727	-39.892	-38.538
253	-	-	-
248	-	-	-
243	-	-	-
238	-	-	-
233	-	-	-
228	-	-	-38.455
223	-	-39.790	-38.441
218	-	-39.774	-38.431
213	-	-39.756	-38.419
208	-	-39.740	-38.408
203	-	-39.721	-38.393
198	-	-39.704	-38.381
193	-	-39.685	-38.366
188	-	-39.664	-38.350
183	-	-39.645	-38.335
178	-	-39.626	-38.321

^a Chemical shifts are in ppm and were measured in THF- d^8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression : $\delta_{\text{ax}} = -38.608 - 7.37 \cdot 10^{-3} T + 9.27 \cdot 10^{-6} T^2$
 $\delta_{\text{eq}} = -37.847 - 2.68 \cdot 10^{-3} T$

Table 14: Thermodynamic parameters of cyclohexyl trifluoromethyl sulfur



T / K	K^{a}	$\Delta G^{\circ} / \text{J.mol}^{-1}$
298	7.371	-4949.1
293	7.058	-4760.2
288	6.820	-4596.9
283	6.598	-4439.5
278	6.433	-4302.2
273	6.295	-4175.8
268	6.188	-4060.9
263	6.145	-3969.9
258	6.171	-3903.6

^a Equilibrium constants were calculated from the values of the chemical shifts :

$$K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$$

Table 15: ^{19}F chemical shifts ^a of 4-methylcyclohexyl trifluoromethyl sulfur (15b)

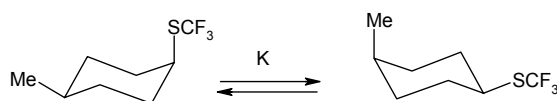
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
298	-41.600	-42.655	<i>-43.072</i>	<i>-41.559</i>
293	-41.595	-42.656	<i>-43.051</i>	<i>-41.551</i>
288	-41.589	-42.655	<i>-43.030</i>	<i>-41.544</i>
283	-41.584	-42.655	<i>-43.008</i>	<i>-41.536</i>
278	-41.578	-42.654	<i>-42.987</i>	<i>-41.528</i>
273	-41.571	-42.651	<i>-42.965</i>	<i>-41.5230</i>
268	-41.563	-42.649	<i>-42.944</i>	<i>-41.512</i>
263	-41.555	-42.647	<i>-42.923</i>	<i>-41.504</i>
258	-41.546	-42.645	<i>-42.901</i>	<i>-41.496</i>
253	-41.537	-42.645	<i>-42.880</i>	<i>-41.488</i>
248	-41.527	-	-	-
243	-41.516	-	-	-
238	-41.505	-	-	-
233	-41.492	-	-	-
228	-41.479	-	-	-
223	-41.465	-	-	-
218	-41.451	-	-	-
213	-41.437	-	-	-
208	-41.422	-	-42.684	-
203	-41.407	-	-42.665	-
198	-41.392	-	-42.646	-
193	-41.375	-	-42.626	-
188	-41.357	-	-42.604	-41.386
183	-41.339	-	-42.582	-41.378
178	-41.322	-	-42.559	-41.370
173	-41.303	-	-42.536	-41.362
168	-41.286	-	-42.514	(-41.355)

^a Chemical shifts are in ppm and were measured in THF- d_8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression : $\delta_{\text{ax}} = -41.797 - 4.28 \cdot 10^{-3} \text{ T}$
 $\delta_{\text{eq}} = -41.088 - 1.58 \cdot 10^{-3} \text{ T}$

Table 16: Thermodynamic parameters of 4-methylcyclohexyl trifluoromethyl sulfur



T / K	K^{a}	$\Delta G^{\circ} / \text{J.mol}^{-1}$
298	0.381	2393.9
293	0.358	2504.6
288	0.337	2607.1
283	0.316	2711.8
278	0.296	2812.8
273	0.278	2906.4
268	0.259	3008.6
263	0.241	3108.7
258	0.223	3218.1
253	0.203	3353.6

^a Equilibrium constants were calculated from the values of the chemical shifts :

$$K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$$

Table 17: ^{19}F chemical shifts ^a of trifluoromethoxycyclohexane (19a)

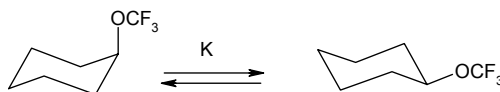
T / K	$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
298	-60.177	-60.500	-60.093
293	-60.175	-60.487	-60.093
288	-60.172	-60.474	-60.092
283	-60.169	-60.461	-60.091
278	-60.166	-60.448	-60.090
273	-60.163	-60.434	-60.089
268	-60.159	-60.421	-60.087
263	-60.155	-60.408	-60.085
258	-60.150	-60.395	-60.082
253	-60.145	-60.382	-60.079
248	-60.140	-60.368	-60.076
243	-	-	-
238	-	-	-
228	-	-	-
218	-	-60.290	-60.055
208	-	-60.262	-60.041
198	-	-60.238	-60.027
193	-	-60.225	-60.020
188	-	-60.211	-60.013
183	-	-60.198	-60.006
178	-	-60.183	-59.998
173	-	-60.176	-59.996

^a Chemical shifts are in ppm and were measured in THF- d^8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression : $\delta_{\text{ax}} = -59.713 - 2.64 \cdot 10^{-3} T$
 $\delta_{\text{eq}} = -59.514 - 3.88 \cdot 10^{-3} T + 6.50 \cdot 10^{-6} T^2$

Table 18: Thermodynamic parameters of trifluoromethoxycyclohexane



T / K	K^{a}	$\Delta G^{\circ} / \text{J.mol}^{-1}$
298	3.824	-3323.4
293	3.813	-3260.6
288	3.792	-3191.5
283	3.766	-3119.8
278	3.729	-3041.7
273	3.686	-2960.7
268	3.669	-2896.7
263	3.629	-2818.2
258	3.616	-2757.1
253	3.605	-2697.6
248	3.590	-2635.7

^a Equilibrium constants were calculated from the values of the chemical shifts :

$$K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$$

Table 19: ^{19}F chemical shifts ^a of 4-methoxy-1-trifluoromethoxycyclohexane (19d)

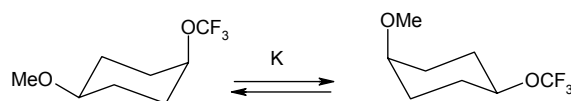
T / K	<i>trans</i> isomer ^b	<i>cis</i> isomer		
		$\delta_{\text{obs}}^{\text{b}}$	$\delta_{\text{ax}}^{\text{b,c}}$	$\delta_{\text{eq}}^{\text{b,c}}$
298	-60.392	-60.236	<i>-60.439</i>	<i>-60.112</i>
293	-60.388	-60.232	<i>-60.435</i>	<i>-60.109</i>
288	-60.384	-60.228	<i>-60.431</i>	<i>-60.105</i>
283	-60.379	-60.224	<i>-60.426</i>	<i>-60.101</i>
278	-60.375	-60.220	<i>-60.421</i>	<i>-60.097</i>
273	-60.369	-60.215	<i>-60.415</i>	<i>-60.093</i>
268	-60.364	-60.210	<i>-60.409</i>	<i>-60.088</i>
263	-60.358	-	-	-
258	-60.351	-	-	-
253	-60.343	-	-	-
248	-60.336	-	-	-
243	-60.329	-	-	-
238	-60.320	-	-	-
233	-60.312	-	-	-
228	-60.302	-	-	-
223	-	-	-	-
218	-	-	-	-
213	-	-	-	-
208	-	-	-	-
203	-	-	-	-
198	-	-	-	-
193	-	-	-	-
188	-	-	-60.007	-60.246
183	-	-	-60.001	-60.231
178	-	-	-59.995	-60.216
173	-	-	-59.990	-60.200
168	-	-	-59.984	-60.182

^a Chemical shifts are in ppm and were measured in THF- d_8 relative to CFCl_3 .

^b Shifts in bold are experimental values.

^c Shifts in italic were calculated by linear regression : $\delta_{\text{ax}} = -59.371 - 6.46 \cdot 10^{-3} T + 9.65 \cdot 10^{-6} T^2$
 $\delta_{\text{eq}} = -59.731 - 1.81 \cdot 10^{-3} T + 1.77 \cdot 10^{-6} T^2$

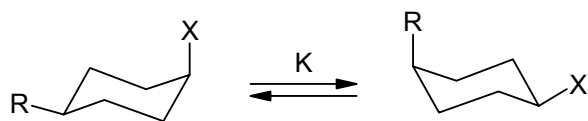
Table 20: Thermodynamic parameters of 4-methoxy-1-trifluoromethoxycyclohexane



T / K	K^a	$\Delta G^\circ / \text{J} \cdot \text{mol}^{-1}$
298	1.652	-1243.0
293	1.649	-1219.1
288	1.640	-1184.7
283	1.634	-1155.8
278	1.628	-1126.0
273	1.631	-1110.2
268	1.635	-1096.1
263	1.652	-1243.0

^a Equilibrium constants were calculated from the values of the chemical shifts : $K = (\delta_{\text{ax}} - \delta_{\text{obs}}) / (\delta_{\text{obs}} - \delta_{\text{eq}})$

Table 21. Thermodynamic parameters of 4-substituted-1-fluorinated cyclohexanes.



	R	X	$\Delta H^\circ / \text{kJ.mol}^{-1}$ ($\Delta H^\circ / \text{kcal.mol}^{-1}$) ^a	$\Delta S^\circ / \text{J. mol}^{-1}.\text{K}^{-1}$ ($\Delta S^\circ / \text{cal.mol}^{-1}.\text{K}^{-1}$) ^b	$\Delta G^\circ_{298\text{K}} / \text{kJ.mol}^{-1}$ ($\Delta G^\circ_{298\text{K}} / \text{kcal.mol}^{-1}$)
1a	Me	C ₂ F ₅	-5.23 ± 0.02 (-1.25 ± 0.01)	-5.33 ± 0.07 (-1.27 ± 0.02)	-3.64 (-0.87)
1b	<i>i</i> Pr	C ₂ F ₅	-6.59 ± 0.03 (-1.58 ± 0.01)	-14.86 ± 0.11 (-3.55 ± 0.04)	-2.17 (-0.52)
2a	Me	CH ₂ F	2.35 ± 0.04 (-0.56 ± 0.01)	-10.41 ± 0.15 (-2.49 ± 0.05)	0.76 (0.18)
2b	<i>i</i> Pr	CH ₂ F	2.97 ± 0.09 (-0.71 ± 0.02)	-18.21 ± 0.33 (-4.35 ± 0.08)	2.49 (0.59)
3a	Me	CF ₂ H	-1.54 ± 0.02 (-0.37 ± 0.01)	-4.10 ± 0.06 (-0.98 ± 0.02)	-0.33 (-0.08)
3b	<i>i</i> Pr	CF ₂ H	-2.07 ± 0.04 (-0.50 ± 0.01)	-11.51 ± 0.14 (-2.75 ± 0.04)	1.35 (0.32)
4a	H	SCF ₃	2.96 ± 0.37 (0.71 ± 0.09)	26.30 ± 1.34 (6.28 ± 0.32)	-4.95 (-1.18)
4b	Me	SCF ₃	8.47 ± 0.07 (2.02 ± 0.02)	20.36 ± 0.25 (4.86 ± 0.05)	2.39 (0.57)
5a	H	OCF ₃	0.87 ± 0.05 (0.21 ± 0.01)	14.08 ± 0.20 (3.36 ± 0.05)	-3.32 (-0.79)
5d	OMe	OCF ₃	0.23 ± 0.07 (0.05 ± 0.02)	4.92 ± 0.26 (1.18 ± 0.06)	-1.24 (-0.30)