Inclusion of anthraquinone derivatives by the cucurbit[7]uril host

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Electronic Supplementary Information



Figure S1. ¹H NMR spectra (DCl/D₂O, pH = 2.0) of guest **1**.



Figure S2. ¹³C NMR spectra (DCl/D₂O, pH = 2.0) of guest 1.



Figure S3. FAB MS spectra of guest 1.



Figure S4. ¹H NMR spectra (DCl/D₂O, pH = 2.0) of guest **2**.



Figure S5. ¹³C NMR spectra (DCl/D₂O, pH = 2.0) of guest 2.



Figure S6. FAB MS spectra of guest 2.



Figure S7. MALDI-TOF MS spectra of CB7•1 complex.



Figure S8. MALDI-TOF MS spectra of CB7•2 complex.



Figure S9. ¹H NMR spectra (DCl/D₂O, pH = 2.0) of guest **2** in the presence of 1.5 equiv of CB7 and 0 equiv(**A**), 62.7 equiv (0.3 M) (**B**), 104.5 equiv (0.5 M) (**C**) 146.4 equiv (0.7 M) (**D**) and 230.0 equiv (1.1 M) (**E**) of NaCl.



Figure S10. Cyclic voltammetric response on glassy carbon (0.071 cm^2) of guest **1** in 0.1 M HCl solution in the absence (red) and in the presence of 0.5 equiv (blue), 1.0 equiv (black), 1.5 equiv (green) and 2.0 equiv (brown) **CB7**. Scan rate: 0.1 V/sec.



Figure S11. Cyclic voltammetric response on glassy carbon (0.071 cm^2) of guest **2** in 0.1 M pH 2 phosphate buffer (also containing 0.1 M NaCl) solution in the absence (red) and in the presence of 0.5 equiv (blue), 1.0 equiv (brown), 1.5 equiv (green) and 2.0 equiv (black) **CB7**. Scan rate: 0.1 V/sec.



Figure S12. Electronic absorption spectra (25 °C, 0.01 M HCl) of (A) 10 μ M guest **1** and (B) 10 μ M guest **2** in the presence of increasing concentrations (0-80 μ M) of CB7.