

## **Supporting Information**

### **Nanoporous material based on heteroleptic bilayers built up from bis- 575 phosphonium and *p*-sulfonato-calix[4]arene ions**

Mohamed Makha,\* Yatimah Alias, Colin L. Raston\* and Alexandre N. Sobolev

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**1-Results of the combinatorial approach based on molar ratios of components: identification and characterisation of complexes 3-5**

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Code	SO <sub>3</sub> Na-calix[4]arene	(Ph <sub>3</sub> P <sup>+</sup> -CH <sub>2</sub> PhCH <sub>2</sub> -P <sup>+</sup> Ph <sub>3</sub> ) 2Cl <sup>-</sup>	YbCl <sub>3</sub> .6H <sub>2</sub> O	pH	σ (mS/cm <sup>-1</sup> )	Notes
Z32	1	1	2	6.05	4.60	Structure complex 3 (2D porous)
Z42	1	1	3	6.08	4.70	Z42j – isostructural Z32

Code	SO <sub>3</sub> Na-calix[4]arene	(Ph <sub>3</sub> P <sup>+</sup> -CH <sub>2</sub> PhCH <sub>2</sub> -P <sup>+</sup> Ph <sub>3</sub> ) 2Cl <sup>-</sup>	Er(NO <sub>3</sub> ) <sub>3</sub> .5H <sub>2</sub> O	pH	σ (mS/cm <sup>-1</sup> )	Notes
Er42	1	1	3	6.18	4.06	Structure complex 4 (Expanded Bilayer)
Er32	1	1	2	6.20		

Code	SO <sub>3</sub> Na-calix[4]arene	(Ph <sub>3</sub> P <sup>+</sup> -CH <sub>2</sub> PhCH <sub>2</sub> -P <sup>+</sup> Ph <sub>3</sub> ) 2Cl <sup>-</sup>	Cs(NO <sub>3</sub> )	pH	σ (mS/cm <sup>-1</sup> )	Notes
Cs32	1	1	2	9.14	5.37	Structure complex 5 (compact structure) Isostructural to 4La3R
Cs42	1	1	3	9.57		

Code	SO <sub>3</sub> Na-calix[4]arene	(Ph <sub>3</sub> P <sup>+</sup> -CH <sub>2</sub> PhCH <sub>2</sub> -P <sup>+</sup> Ph <sub>3</sub> ) 2Cl <sup>-</sup>	LaCl <sub>3</sub> .7H <sub>2</sub> O	pH	σ (mS/cm <sup>-1</sup> )	Notes
4La1	1	1	1	6.95	3.32	
4La2	1	1	2	6.59	4.31	
4La3	1	1	3	6.57	3.89	isostructural to complex 5
4La4	1	2	1	6.72	4.20	

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W1	1	0	1	6.36	2.13	
W2	1	1	1	6.12	2.16	
W4	1	3	1	6.22	3.74	Too small unit cell
W6	1	5	1	6.37	4.45	Too small unit cell
X1	1	1	0	8.25	1.39	
X4	1	1	3	6.41	4.78	
X6	1	1	5	6.34	6.68	

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Y1	1	0	1	6.22	1.71	
Y2	1	1	1	6.14	2.05	
Y4	1	3	1	5.59	3.52	Too small unit cell
Y6	1	5	1	6.17	4.52	Too small unit cell
Z1	1	1	0	8.90	1.58	

Z4	1	1	3	6.10	4.87	
Z6	1	1	5	6.18	6.43	

Code	SO <sub>3</sub> Na-calix[4]arene	(Ph <sub>3</sub> P <sup>+</sup> -CH <sub>2</sub> PhCH <sub>2</sub> -P <sup>+</sup> Ph <sub>3</sub> ) 2Cl <sup>-</sup>	GdCl <sub>3</sub> .6H <sub>2</sub> O	pH	σ (mS/cm <sup>-1</sup> )	Notes
X32	1	1	2	6.14	2.25	
X42	1	1	3	6.21	4.88	

**Molar ratios of tectons (Sulfonato-calix[4]arene: (Ph<sub>3</sub>P<sup>+</sup>-CH<sub>2</sub>PhCH<sub>2</sub>-P<sup>+</sup>Ph<sub>3</sub>: Ln<sup>3+</sup>) at different pH**

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Code	SO <sub>3</sub> Na-calix[4]arene	(Ph <sub>3</sub> P <sup>+</sup> -CH <sub>2</sub> PhCH <sub>2</sub> -P <sup>+</sup> Ph <sub>3</sub> ) 2Cl <sup>-</sup>	GdCl <sub>3</sub> .6H <sub>2</sub> O	pH (a)	pH	σ (mS/cm <sup>-1</sup> )	Notes
X32pH4	1	1	2	5.92	3.59	8.50	
X42pH4	1	1	3	5.83	4.01	7.14	
X32pH2	1	1	2	5.81	1.59	14.20	
X42pH2	1	1	3	5.78	2.33	10.05	

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Z32pH4	1	1	2	5.72	3.57	7.21	
Z42pH4	1	1	3	5.65	3.91	7.66	
Z32pH2	1	1	2	5.91	2.15	10.11	
Z42pH2	1	1	3	5.66	1.83	14.77	

Code	SO <sub>3</sub> Na-calix[4]arene	(Ph <sub>3</sub> P <sup>+</sup> -CH <sub>2</sub> PhCH <sub>2</sub> -P <sup>+</sup> Ph <sub>3</sub> ) 2Cl <sup>-</sup>	Er(NO <sub>3</sub> ) <sub>3</sub> .5H <sub>2</sub> O	pH (a)	pH	σ (mS/cm <sup>-1</sup> )	Notes
Er32pH4	1	1	2	6.06	3.89	8.23	
Er42pH4	1	1	3	5.87	3.74	7.39	
Er32pH2	1	1	2	6.11	2.04	11.84	
Er42pH2	1	1	3	5.98	2.28	9.79	

Code	SO <sub>3</sub> Na-calix[4]arene	(Ph <sub>3</sub> P <sup>+</sup> -CH <sub>2</sub> PhCH <sub>2</sub> -P <sup>+</sup> Ph <sub>3</sub> ) 2Cl <sup>-</sup>	Cs(NO <sub>3</sub> )	pH (a)	pH	σ (mS/cm <sup>-1</sup> )	Notes
Cs32pH4	1	1	2	9.60	3.94	7.46	
Cs42pH4	1	1	3	9.58	3.27	9.69	
Cs32pH2	1	1	2	9.54	2.22	11.52	
Cs42pH2	1	1	3	9.57	2.12	13.55	

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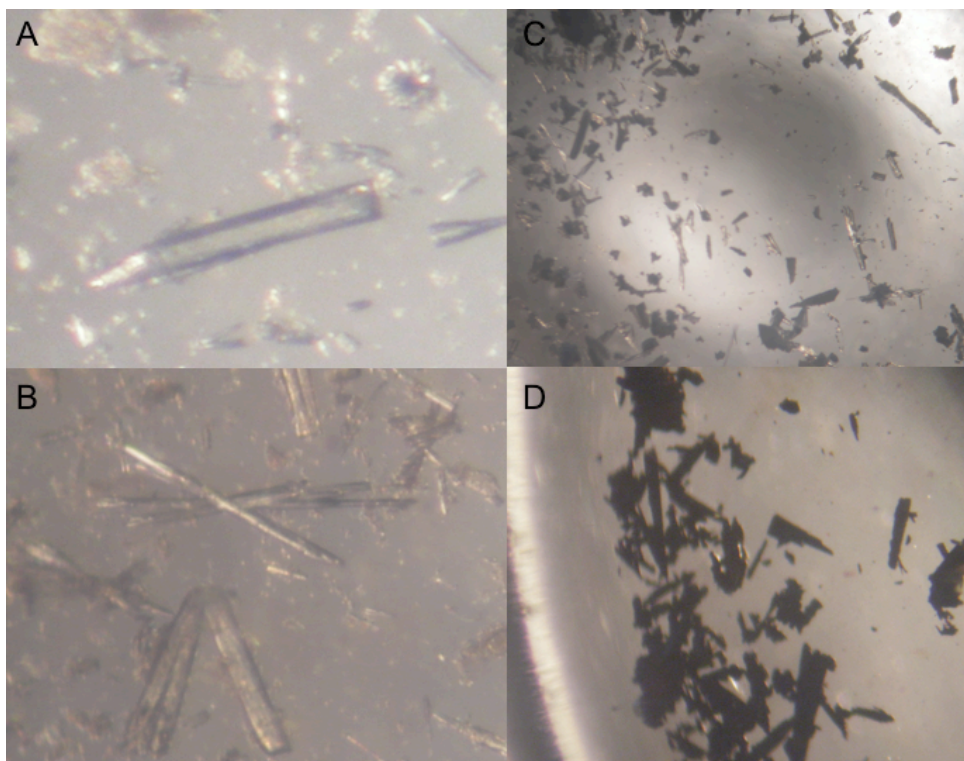
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655 **2- Assessment of porosity in complex 3**

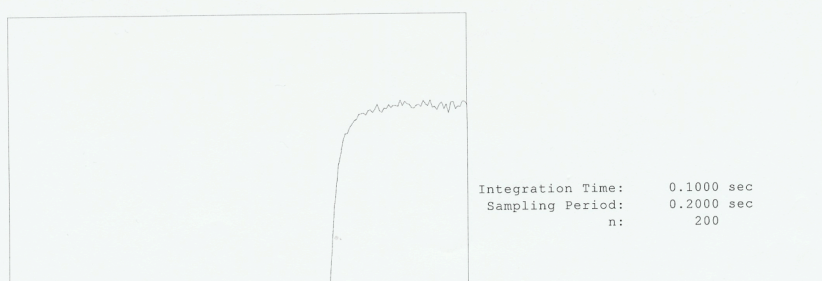
(A) Crystal images of complex 3, (B) crystals of complex 3 soaked in 2M HCl; (C) Isolated and dried crystals of complex 3 after HCl treatment and (D) crystals of complex 3 turned dark brown instantly after exposure to iodine (iodine is recovered from the crystals upon exposure to acetone).

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**3- ICP records on HCl leaching of Ytterbium**

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m/z	Range	Count	Mean	RSD%
174	50,000,000	30068120.0	8496536.0	153.03
171	20,000,000	13227383.0	3744787.3	153.81

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Tuning Parameters
===Plasma Condition===
RF Power : 1500 W
RF Matching : 1.8 V
Smpl Depth : 6 mm
Torch-H : -1.3 mm
Torch-V : 1.1 mm
Carrier Gas : 0.91 L/min
Makeup Gas : 0.2 L/min
Optional Gas : 0 %
Nebulizer Pump : 0.15 rps
Sample Pump : -- rps
S/C Temp : 2 degC

===Ion Lenses===
Extract 1 : 4 V
Extract 2 : -100.5 V
Omega Bias-cs : -32 V
Omega Lens-cs : 8 V
Cell Entrance : -26 V
QP Focus : 2 V
Cell Exit : -42 V

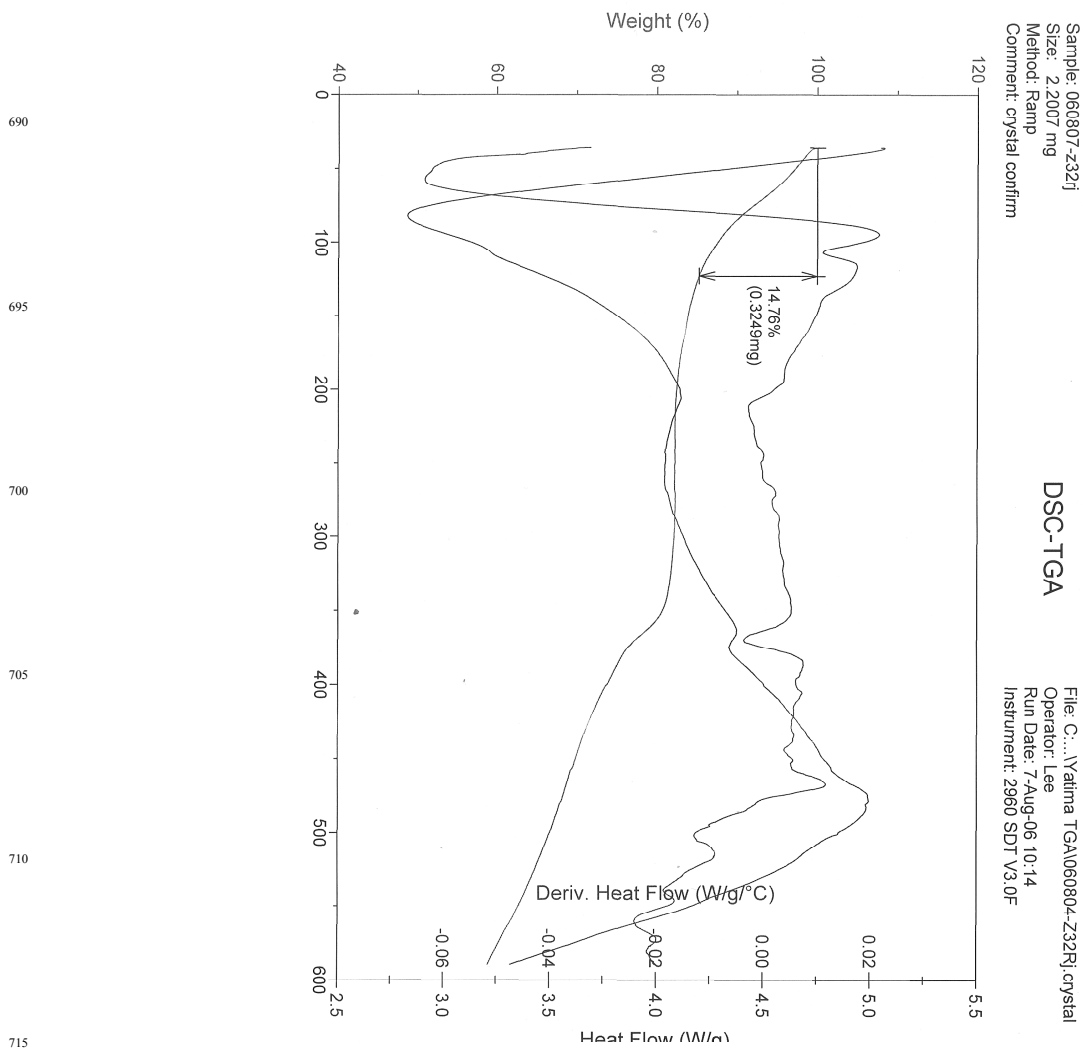
===Q-Pole Parameters===
AMU Gain : 130
AMU Offset : 124
Axis Gain : 1.0004
Axis Offset : -0.02
QP Bias : -3 V

===Detector Parameters===
Discriminator : 8 mV
Analog HV : 1940 V
Pulse HV : 1520 V

===Reaction Cell===
Reaction Mode : OFF
H2 Gas : 0 mL/min
He Gas : 0 mL/min
Optional Gas : --- %
    
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685 **4- TGA analysis on complex 3 with crystals retained morphology at 600 °C**

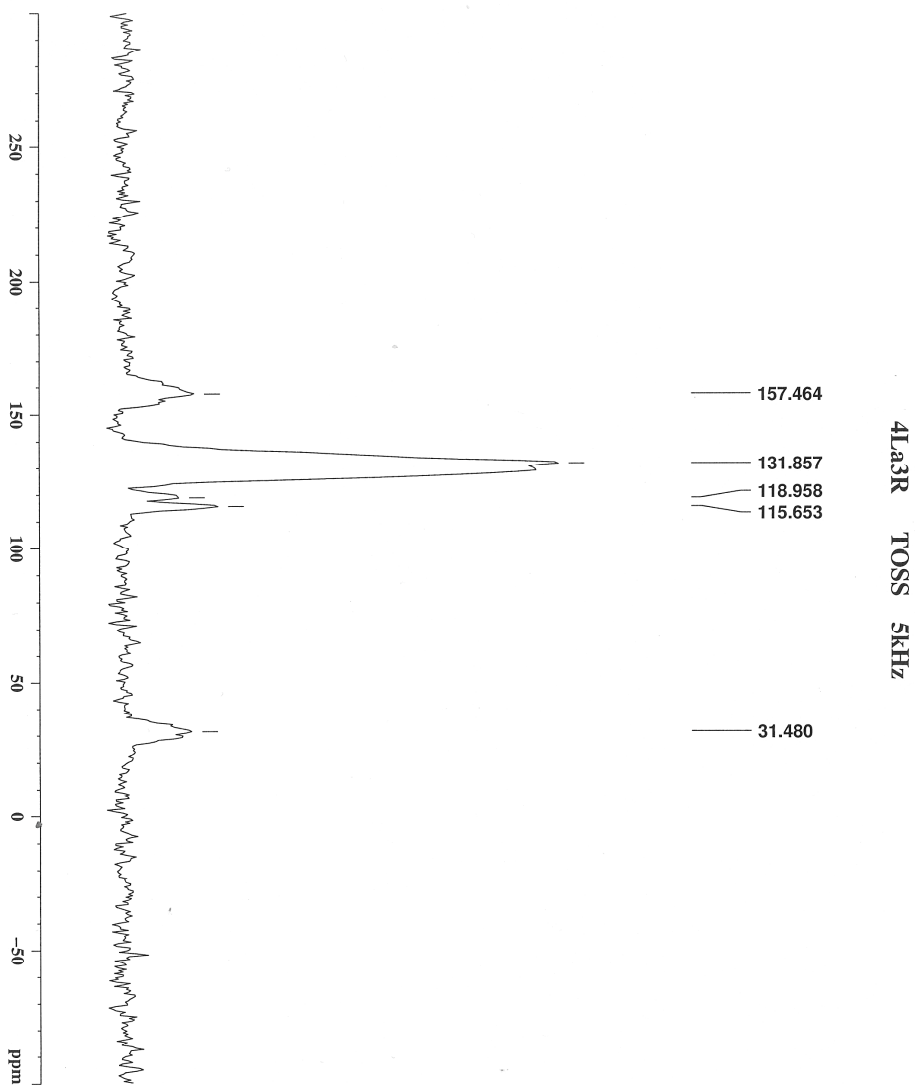


5- <sup>13</sup>C MASS, <sup>31</sup>P MASS NMR and TGA data on complex 5

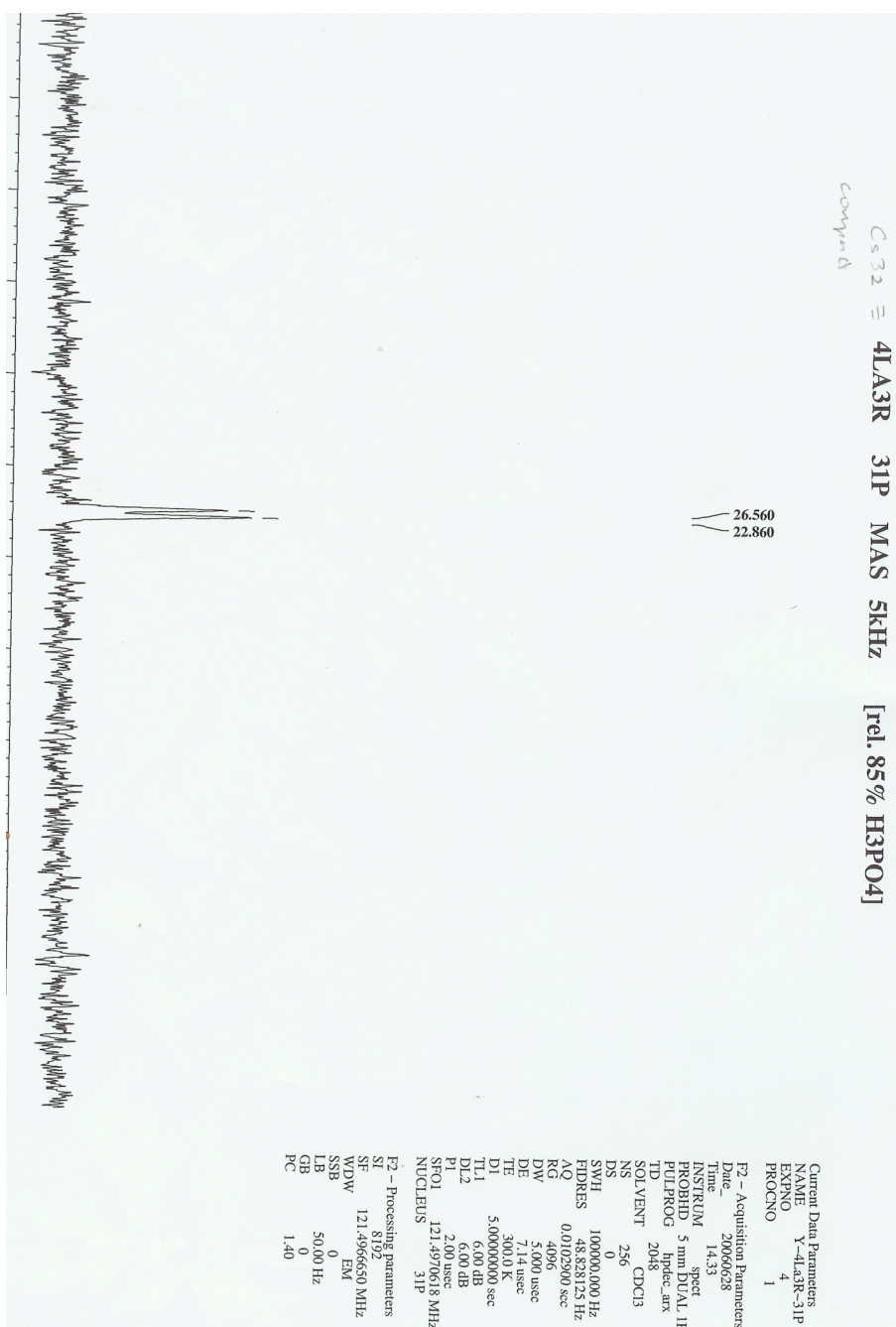
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Current Data Parameters  
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 EXPNO 3  
 PROCNO 1  
 F2 - Acquisition Parameters  
 Date\_ 20060621  
 Time 0:10  
 INSTRUM spect  
 PULPROG 5 mm DUAL H1  
 PL1PROG zgpg30  
 TD 1024  
 SOLVENT CDCl3  
 NS 640  
 DS 0  
 SWH 41666.668 Hz  
 FIDRES 40.690105 Hz  
 AQ 0.0123380 sec  
 RG 8192  
 DW 12.000 usec  
 DE 17.14 usec  
 TE 300.2 K  
 L1 5000  
 L2 8.00 usec  
 P3 4.50 usec  
 D5 0.00000857 sec  
 D1 2.00000000 sec  
 TL1 3.00 dB  
 DL2 5.37 dB  
 SFO2 300.1355000 MHz  
 DECNUC 1H  
 P15 2000.000 usec  
 D11 0.0000357 sec  
 D12 0.0000002 sec  
 D13 0.0001084 sec  
 D14 0.0001838 sec  
 D15 0.0000329 sec  
 SFO1 75.4764457 MHz  
 NUCLEUS 13C  
 F2 - Processing parameters  
 SI 1024



740

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750

Sample: 060807-4La3R.001  
Size: 8.2633 mg  
Method: Ramp  
Comment: crystal confirm

### DSC-TGA

File: C:\Yatima TGA\060804.4La3R.001  
Operator: Lee  
Run Date: 7-Aug-06 11:37  
Instrument: 2960 SDT V3.0F

