Influence of electron-withdrawing groups on the ionic liquid characteristics of 1-alkyl-*n*-cyanopyridinium and 1-alkyl-*n*-(trifluoromethyl)pyridinium salts

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Supplemental Data

IR spectra of selected Systems

 $\begin{bmatrix} C_2^4 CNPy \end{bmatrix} \begin{bmatrix} tSO_4 \end{bmatrix} (3) \\ \begin{bmatrix} C_2^4 CNPy \end{bmatrix} \begin{bmatrix} NTf_2 \end{bmatrix} (4) \\ \begin{bmatrix} C_4^4 CNPy \end{bmatrix} \begin{bmatrix} NTf_2 \end{bmatrix} (7) \\ \begin{bmatrix} C_6^4 CNPy \end{bmatrix} \begin{bmatrix} NTf_2 \end{bmatrix} (10) \\ \begin{bmatrix} C_8^4 CNPy \end{bmatrix} \begin{bmatrix} NTf_2 \end{bmatrix} (13) \\ \begin{bmatrix} C_1^3 CNPy \end{bmatrix} \begin{bmatrix} MeSO_4 \end{bmatrix} (21) \\ \begin{bmatrix} C_2^3 CNPy \end{bmatrix} \begin{bmatrix} tSO_4 \end{bmatrix} (23) \\ \begin{bmatrix} C_2^3 CNPy \end{bmatrix} \begin{bmatrix} tSO_4 \end{bmatrix} (23) \\ \begin{bmatrix} C_2^3 CNPy \end{bmatrix} \begin{bmatrix} NTf_2 \end{bmatrix} (24) \\ \begin{bmatrix} C_1^2 CNPy \end{bmatrix} \begin{bmatrix} NTf_2 \end{bmatrix} (30) \\ \begin{bmatrix} C_2^2 CNPy \end{bmatrix} \begin{bmatrix} tSO_4 \end{bmatrix} (31) \\ \begin{bmatrix} C_2^2 CNPy \end{bmatrix} \begin{bmatrix} NTf_2 \end{bmatrix} (32)$

Labelled plots of the crystallographic asymmetric units and principle cation-anion hydrogen-bonds in the crystal structures of 1, 2, 4, 22, 24, 32, (and closest contacts for 37)

1-Methyl-4-cyanopyridinium methylsulfate (1)
1-Methyl-4-cyanopyridinium bis{(trifluoromethyl)sulfonyl}imide (2)
1-Ethyl-4-cyanopyridinium bis{(trifluoromethyl)sulfonyl}imide (4)
1-Methyl-3-cyanopyridinium bis{(trifluoromethyl)sulfonyl}imide (22)
1-Ethyl-3-cyanopyridinium bis{(trifluoromethyl)sulfonyl}imide (24)
1-Ethyl-2-cyanopyridinium bis{(trifluoromethyl)sulfonyl}imide (32)
1-Ethyl-3-(trifluoromethyl)pyridinium bis{(trifluoromethyl)sulfonyl}imide (37)

IR spectra of selected Systems

IR data were recorded for twelve of the ionic liquid salts (in the liquid state between KBr disks, on a Perkin-Elmer Spectrum 100 FT-IR) for eleven ionic liquids; **3**, **4**, **7**, **10**, **13**, **21**, **23**, **24**, **30**, **31**, **32**. The spectra showed a nitrile vibrational band at ~2249.5 cm⁻¹ which had variable intensity between the different salts, but very little variation in position (shown in the table below).

Compound	Alkyl	Nitrile position	Anion	CN stretch /cm ⁻¹	Appearance
3	C_2H_5	4	EtSO ₄	2246	S
4	C_2H_5	4	NTf_2	2249	W
7	C_4H_9	4	NTf_2	2248.5	W
10	C_6H_{13}	4	NTf_2	2247.5	W
13	C_8H_{17} 4		NTf_2	2247	W
21	CH_3	3	MeSO ₄	2250.5	S
23	C_2H_5	3	$EtSO_4$	2250	S
24	C_2H_5	3	NTf_2	2255	S
30	CH_3	2	NTf_2	2250	W
31	C_2H_5	2	$EtSO_4$	2249	W
32	C_2H_5	2	NTf ₂	2252.5	W
			Average	2249.5	
			Stdev	2.5	













Labelled plots of the crystallographic asymmetric units and principle cation-anion hydrogen-bonds in the crystal structures of 1, 2, 4, 22, 24, 32, (and closest contacts for 37)

<u>1-Methyl-4-cyanopyridinium methylsulfate (1)</u>



D	Н	Α	D-H	HA	DA	D-HA	symm(A)
C6	H5	02	0.9500	2.4200	3.2412	144.00	1/2-x, $1/2+y$, $1/2+z$
C6	H5	04	0.9500	2.3900	3.1953	142.00	1/2-x, $1/2$ +y, $1/2$ +z
C8	H7	02	0.9800	2.5300	3.3104	137.00	, , , , , , , , , , , , , , , , , , ,
C8	H8	03	0.9800	2.5700	3.2631	128.00	1/2-x, $-1/2+y$, $1/2+z$
C5	Н9	03	0.9500	2.4200	3.1780	136.00	1/2-x, $-1/2+y$, $1/2+z$
C5	Н9	O4	0.9500	2.5600	3.2586	131.00	1/2-x, -1/2+y, 1/2+z

<u>1-Methyl-4-cyanopyridinium bis{(trifluoromethyl)sulfonyl}imide (2)</u>



D	Н	А	D-H	HA	DA	D-HA	symm(A)
C8	H1	02	0.9500	2.3400	3.2721	167.00	1-x,-y, 1-z
C7	H2	01	0.9500	2.3200	3.2689	172.00	1/2+x, 1/2-y, 1/2+z
C4	H3	04	0.9500	2.4600	3.3929	167.00	-1/2+x, 1/2-y, -1/2+z

<u>1-Ethyl-4-cyanopyridinium bis{(trifluoromethyl)sulfonyl}imide (4)</u>



D H A D-H HA DA D-HA symm(A)	.)
C10 H1 N1 0.9500 2.3100 3.2559 171.00 -1/2+x, -1/2	+y, z
C6 H3 O1 0.9500 2.4900 3.2082 132.00 -1/2+x, 1/2	+y,z
C6 H3 O4 0.9500 2.4400 3.2583 144.00 -1/2+x, 1/2	+y,z
C11 H9 N4 0.9900 2.5000 3.3400 143.00 1/2+x, 1/2-y,	1/2+z

<u>1-Methyl-3-cyanopyridinium bis{(trifluoromethyl)sulfonyl}imide (22)</u>



D	Η	А	D-H	HA	DA	D-HA	symm(A)
C7	H2	N2	0.9500	2.3800	3.1648	140.00	x, y, -1+z
C6	H3	O3	0.9500	2.4200	3.1379	132.00	x, y, -1+z
C9	H6	F3	0.9800	2.5300	3.3471	141.00	
C5	H7	O2	0.9500	2.5700	3.2681	130.00	1+x, y, z

<u>1-Ethyl-3-cyanopyridinium bis{(trifluoromethyl)sulfonyl}imide (24)</u>



D	Н	А	D-H	HA	DA	D-HA	symm(A)
C9	H7	01	0.9700	2.5700	3.4577	152.00	1/2-x, -1/2+y, 1/2-z
C5	H9	03	0.9300	2.5000	3.3742	157.00	

<u>1-Ethyl-2-cyanopyridinium bis{(trifluoromethyl)sulfonyl}imide (32)</u>



D	Η	А	D-H	HA	DA	D-HA	symm(A)
C9	H2	01	0.9300	2.4800	3.3007	147.00	1-x, -1/2+y, 1/2-z
C8	H3	04	0.9300	2.6000	3.4451	152.00	1-x, -1/2+y, 1/2-z
C7	H4	02	0.9300	2.3100	3.2270	167.00	-1/2+x, 1/2-y, 1-z
C18	H10	O3	0.9300	2.3300	3.1910	154.00	
C17	H11	06	0.9300	2.5000	3.4098	166.00	1/2+x, 1/2-y, -z
C15	H13	04	0.9300	2.5200	3.1608	126.00	-1/2+x, 1/2-y, -z

1-Ethyl-3-(trifluoromethyl)pyridinium bis{(trifluoromethyl)sulfonyl}imide (37)



D	Н	А	D-H	HA	DA	D-HA	symm(A)
C9	H2	08	0.9500	2.5900	3.2263	125.00	
C11	H7	O7	0.9900	2.4100	3.3058	150.00	1+x, y, z
C11	H8	O3	0.9900	2.4200	3.2892	146.00	
C7	H9	06	0.9500	2.5700	3.1855	123.00	1+x, y, z
C17	H11	01	0.9500	2.4900	3.1641	128.00	
C16	H12	O5	0.9500	2.5700	3.4251	149.00	x, y, -1+z
C19	H16	05	0.9900	2.5600	3.3631	139.00	x, y, -1+z
C19	H17	O2	0.9900	2.5400	3.3360	137.00	-1+x, y, z
C15	H18	O2	0.9500	2.5600	3.4295	152.00	-1+x, y, z
C15	H18	O4	0.9500	2.5400	3.1294	121.00	-1+x, y, z