

Solvothermal syntheses, characterizations and properties of three transition-metal (Ni^{II} , Co^{II}) imino-carboxylate-diphosphonates

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Fig. SA1. One-dimensional O–H···O hydrogen-bonded chain of **1** down the *c* axis. State the Ni-centred octahedra are shaded in green and that the hydrogen atoms are in white. Hydrogen bonds are drawn as dotted blue lines.

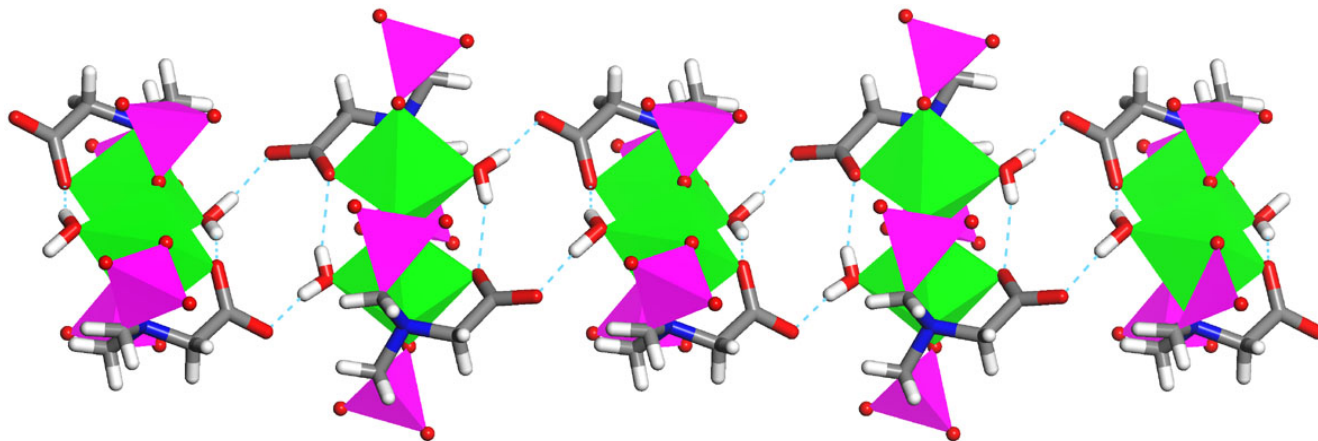


Fig. SA2. Three-dimensional O–H···N (NH₄⁺) hydrogen-bonded network of **1** down the *x* axis. Hydrogen bonds are drawn as dotted blue lines. 1,3-PDA cations are omitted for clarity.

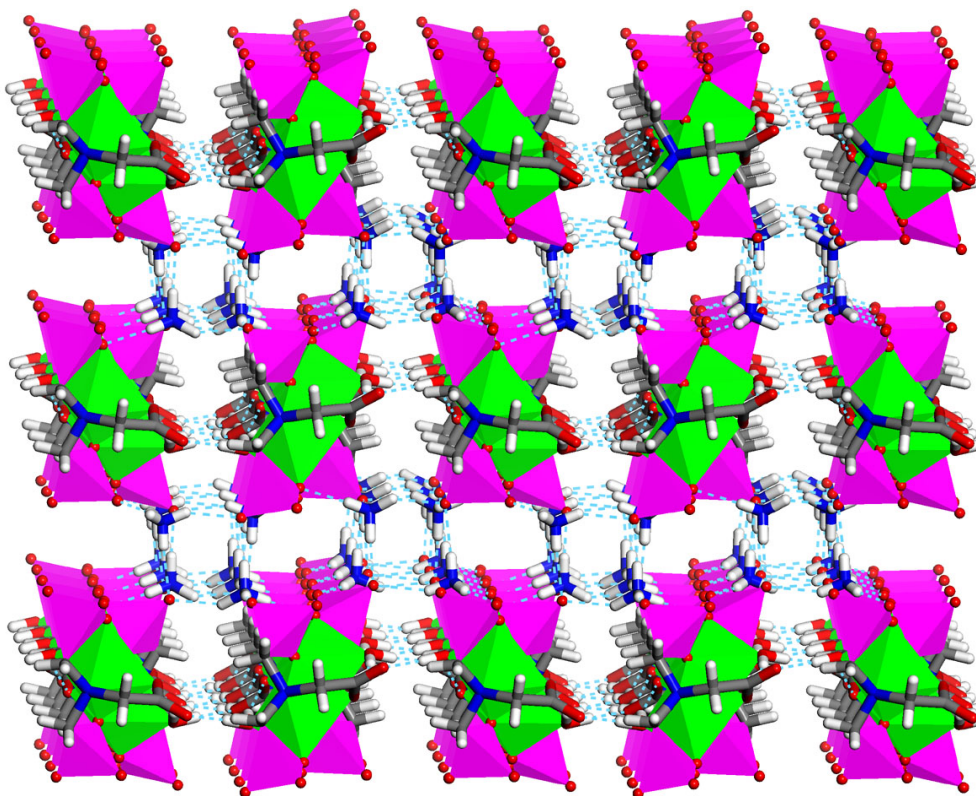


Fig. SA3 One-dimensional O–H···O hydrogen-bonded chain in the crystal structure of **2** down the *b* axis. Hydrogen bonds are drawn as dotted blue lines. State the Ni-centred octahedra are shaded in green and that the hydrogen atoms are in white.

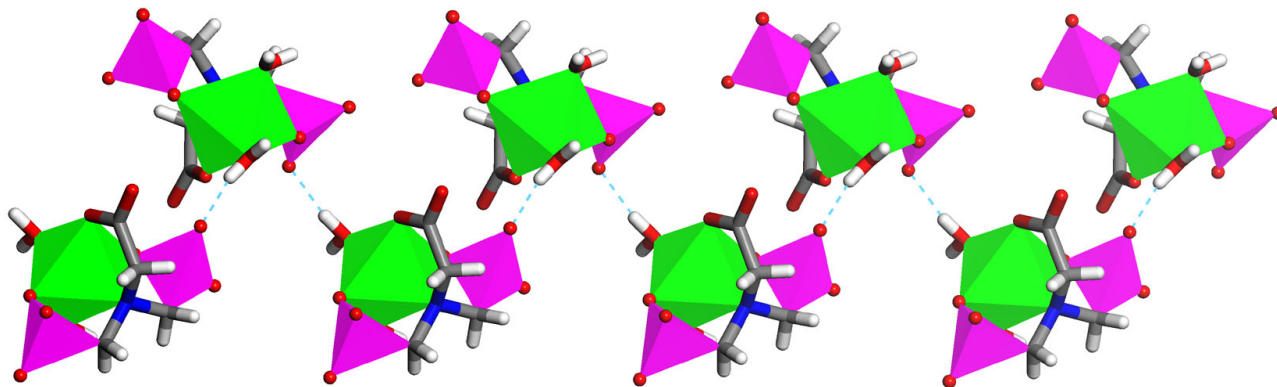


Fig. SA4 Two-dimensional O–H···O hydrogen-bonded network in the crystal structure of **2** in the *ac* plane. Hydrogen bonds are drawn as dotted blue lines.

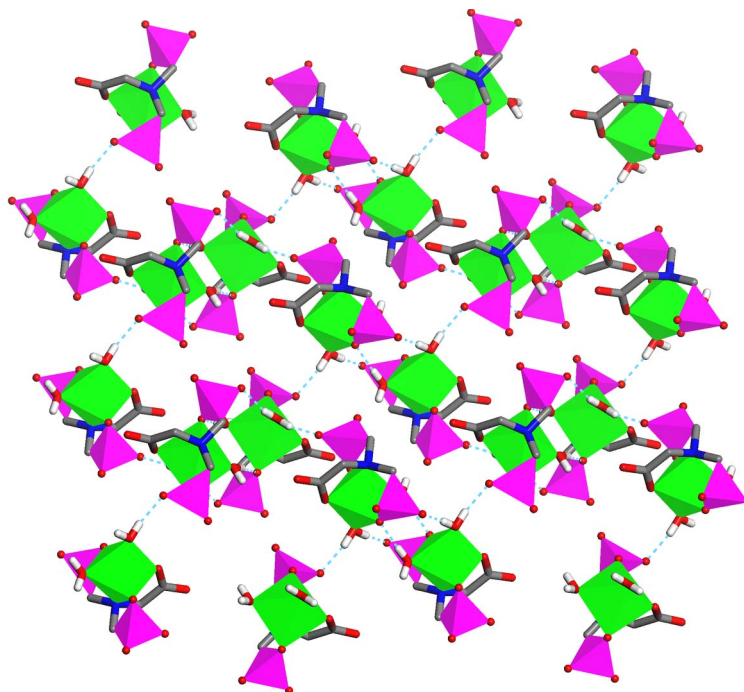


Fig. SA5. Two-dimensional network built from water chains of **2**. Hydrogen bonds are drawn as dotted blue lines.

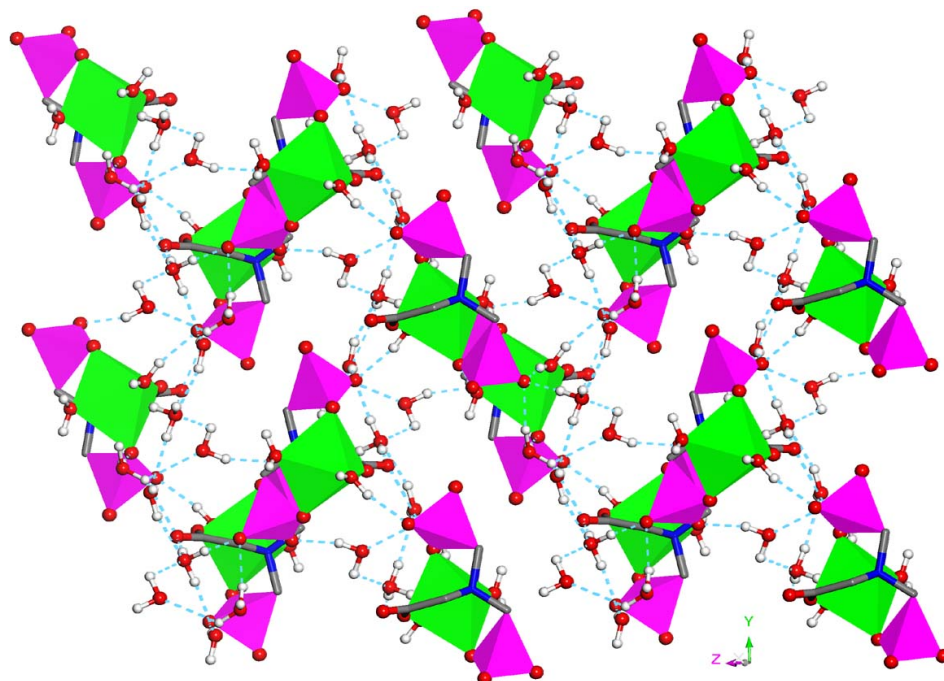


Fig. SA6 One-dimensional O–H···O hydrogen-bonded chain in the crystal structure of **3** down the *b* axis. Hydrogen bonds are drawn as dotted blue lines. State the Co-centred octahedra are shaded in green and that the hydrogen atoms are in white.

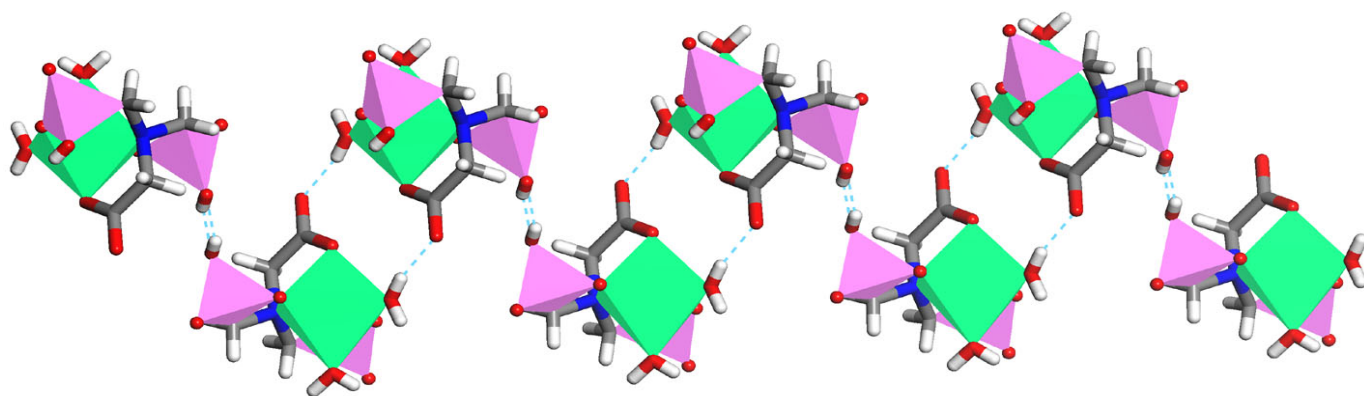


Fig. SA7 Two-dimensional O–H···O hydrogen-bonded network in the crystal structure of **3** in the *ab* plane. Lattice waters are omitted for clarity. Hydrogen bonds are drawn as dotted blue lines.

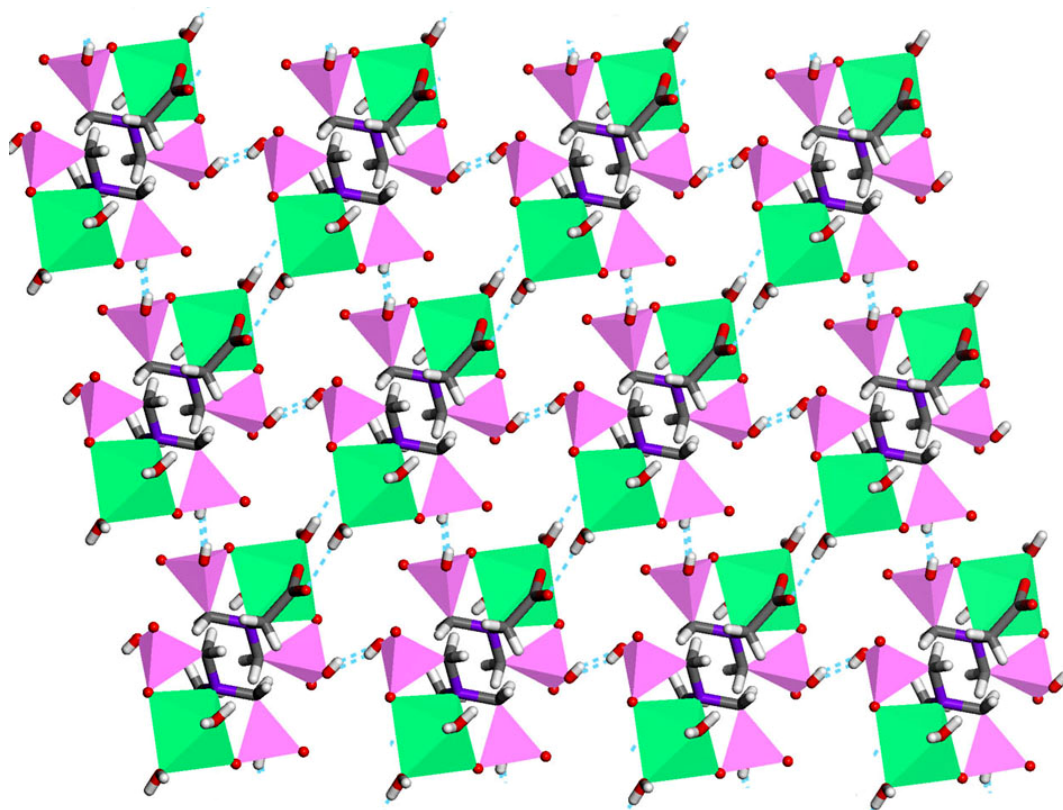
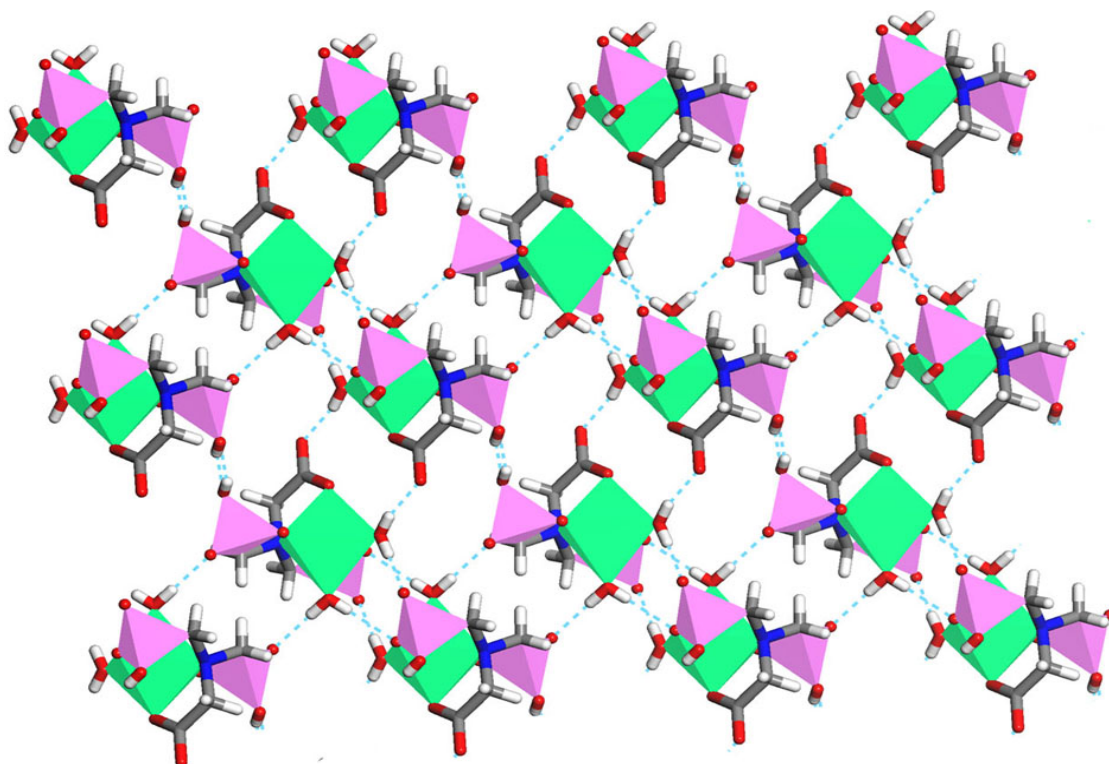


Fig. SA8 Two-dimensional O–H···O hydrogen-bonded network in the crystal structure of **3** in the *bc* plane. Lattice waters are omitted for clarity. Hydrogen bonds are drawn as dotted blue lines.



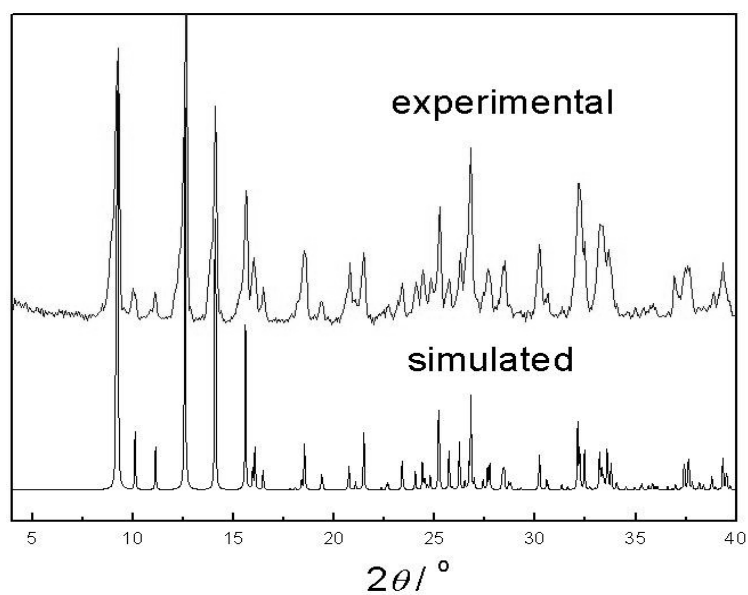


Fig. SB1 Experimental and simulated powder X-ray diffraction patterns of compound 1.

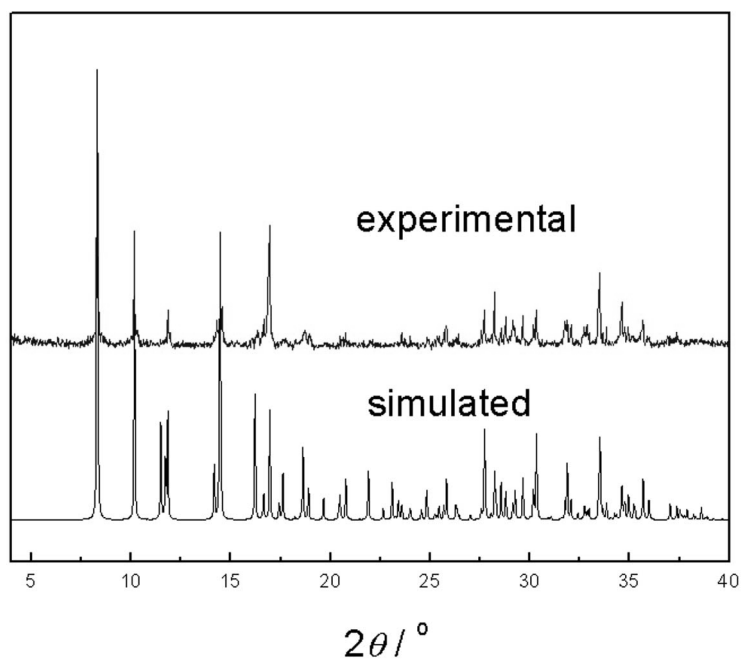


Fig. SB2 Experimental and simulated powder X-ray diffraction patterns of compound 2.

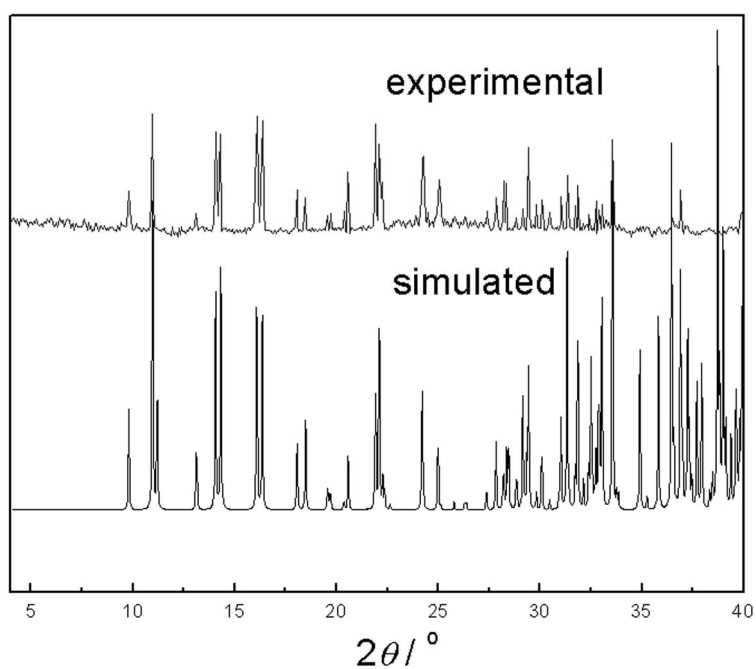


Fig. SB3 Experimental and simulated powder X-ray diffraction patterns of compound 3.

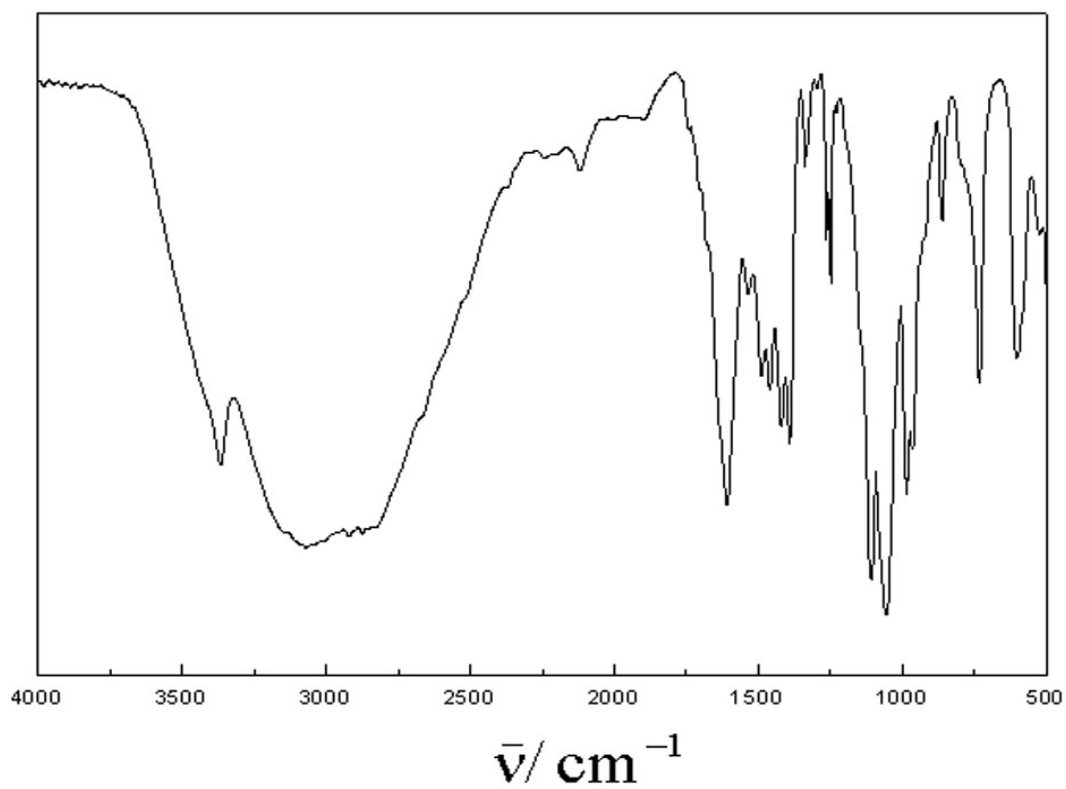


Fig. SC1 FTIR spectrum of the compound 1.

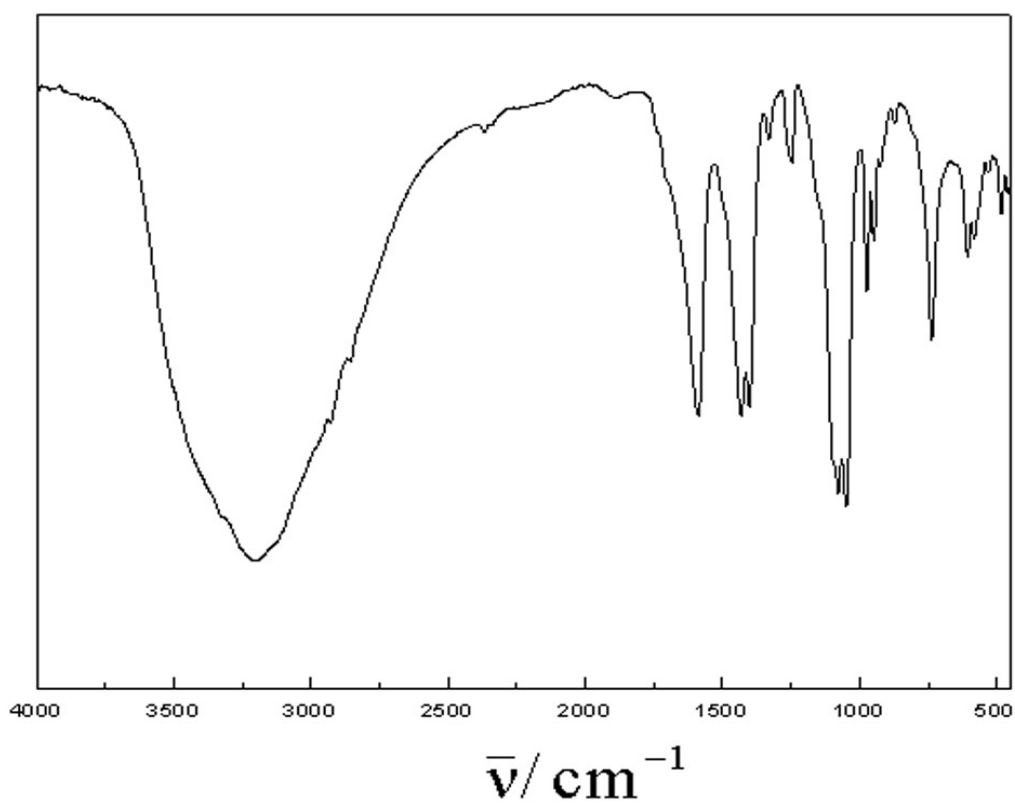


Fig. SC2 FTIR spectrum of the compound **2**.

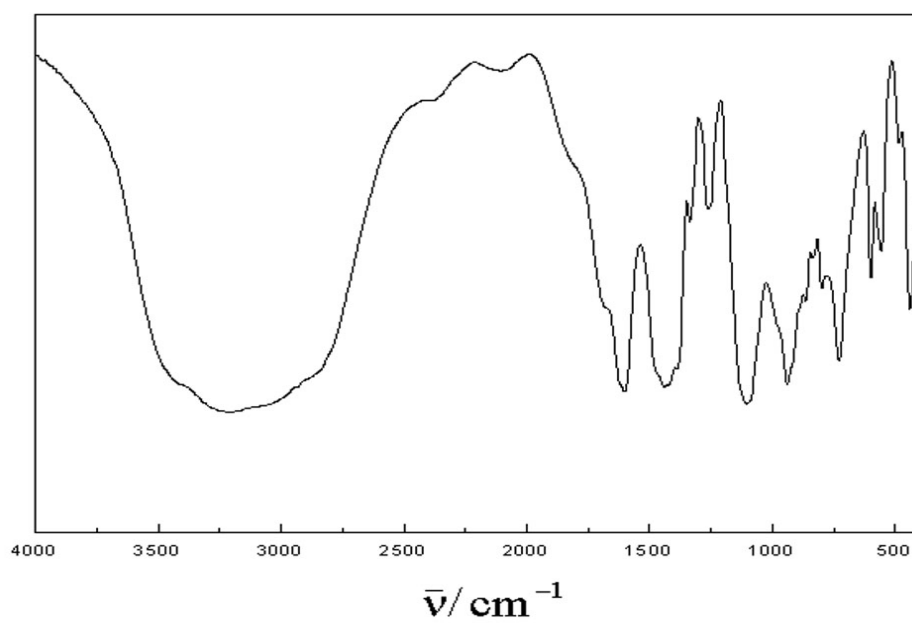


Fig. SC3 FTIR spectrum of the compound **3**.

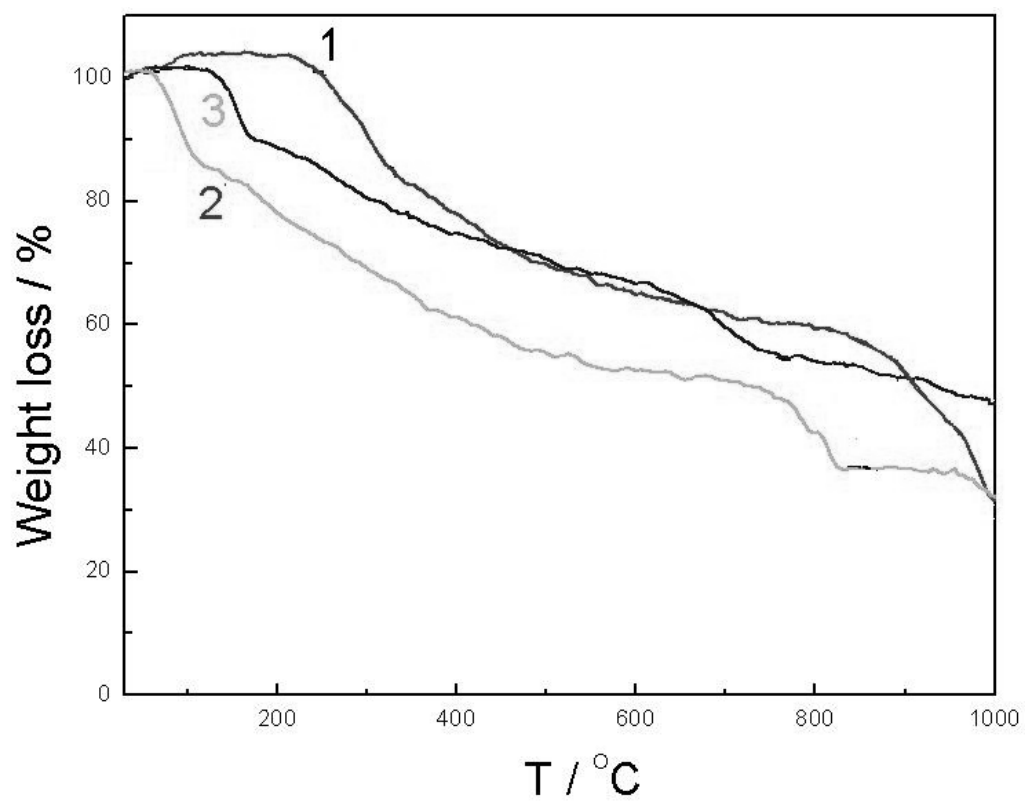


Fig. SD TGA curves of compounds 1–3.

Table S1. Selected Bond Lengths [\AA] and Angles [$^\circ$] of Compounds 1–3.

Compound 1							
Ni(1)-O(1) ^{#1}	2.037(2)	Ni(1)-O(9)	2.102(2)	P(1)-O(1)	1.534(2)	P(2)-O(4)	1.537(2)
Ni(1)-O(1)	2.094(2)	Ni(1)-N(1)	2.098(2)	P(1)-O(2)	1.521(2)	P(2)-O(5)	1.518(2)
Ni(1)-O(4)	2.043(2)	P(1)-C(1)	1.831(2)	P(1)-O(3)	1.510(2)	P(2)-O(6)	1.518(2)
Ni(1)-O(7)	2.051(2)	P(2)-C(2)	1.837(2)	O(4)-Ni(1)-N(1)	84.27(6)	O(1)-P(1)-C(1)	102.33(9)
O(1) ^{#1} -Ni(1)-O(4)	103.81(6)	O(7)-Ni(1)-N(1)	84.44(7)	O(3)-P(1)-O(1)	112.48(9)	O(3)-P(1)-O(2)	112.61(9)
O(1) ^{#1} -Ni(1)-O(7)	86.65(7)	O(1)-Ni(1)-N(1)	87.70(6)	O(2)-P(1)-O(1)	111.98(9)	O(6)-P(2)-O(4)	111.61(1)
O(4)-Ni(1)-O(7)	94.05(7)	O(1) ^{#1} -Ni(1)-O(9)	88.75(7)	O(3)-P(1)-C(1)	111.21(1)	C(3)-N(1)-Ni(1)	105.25(1)
O(1) ^{#1} -Ni(1)-O(1)	84.55(6)	O(4)-Ni(1)-O(9)	90.45(7)	O(2)-P(1)-C(1)	105.53(1)	C(1)-N(1)-Ni(1)	107.68(1)
O(4)-Ni(1)-O(1)	171.48(6)	O(7)-Ni(1)-O(9)	174.22(7)	P(2)-O(4)-Ni(1)	117.53(9)	C(2)-N(1)-Ni(1)	108.61(1)
O(7)-Ni(1)-O(1)	88.00(6)	O(1)-Ni(1)-O(9)	88.07(7)	Ni(1) ^{#1} -O(1)-Ni(1)	95.45(6)	P(1)-O(1)-Ni(1) ^{#1}	146.66(9)
O(1) ^{#1} -Ni(1)-N(1)	168.39(6)	N(1)-Ni(1)-O(9)	99.64(7)	C(4)-O(7)-Ni(1)	113.85(1)	P(1)-O(1)-Ni(1)	115.71(8)
						O(6)-P(2)-O(5)	113.66(1)
Compound 2							
Ni(1)-O(1)	2.038(2)	Ni(1)-O(10)	2.115(2)	P(1)-O(1)	1.544(2)	P(2)-O(4)	1.539(2)
Ni(1)-O(4)	2.042(2)	Ni(1)-N(1)	2.125(2)	P(1)-O(2)	1.517(2)	P(2)-O(5)	1.522(2)
Ni(1)-O(7)	2.055(2)	P(1)-C(1)	1.828(3)	P(1)-O(3)	1.524(2)	P(2)-O(6)	1.517(2)
Ni(1)-O(9)	2.086(2)	P(2)-C(2)	1.828(3)				
O(7)-Ni(1)-O(10)	178.23(8)	O(9)-Ni(1)-N(1)	173.76(8)	O(2)-P(1)-O(1)	111.84(1)	O(5)-P(2)-O(4)	113.67(1)
O(7)-Ni(1)-O(9)	90.43(8)	O(1)-Ni(1)-N(1)	85.82(7)	O(3)-P(1)-O(1)	111.59(1)	P(1)-O(1)-Ni(1)	116.17(1)
O(10)-Ni(1)-O(9)	91.33(8)	O(7)-Ni(1)-O(4)	92.34(8)	O(2)-P(1)-C(1)	109.33(1)	P(2)-O(4)-Ni(1)	113.92(1)
O(7)-Ni(1)-O(1)	92.69(8)	O(10)-Ni(1)-O(4)	87.46(8)	O(3)-P(1)-C(1)	106.75(1)	C(4)-O(7)-Ni(1)	115.03(2)
O(10)-Ni(1)-O(1)	87.28(8)	O(9)-Ni(1)-O(4)	92.04(8)	O(1)-P(1)-C(1)	103.72(1)	C(3)-N(1)-Ni(1)	107.03(2)
O(9)-Ni(1)-O(1)	95.59(8)	O(1)-Ni(1)-O(4)	170.83(7)	O(6)-P(2)-O(5)	112.71(1)	C(1)-N(1)-Ni(1)	106.53(1)
O(7)-Ni(1)-N(1)	83.42(8)	N(1)-Ni(1)-O(4)	87.15(7)	O(6)-P(2)-O(4)	111.47(1)	C(2)-N(1)-Ni(1)	107.89(2)
O(10)-Ni(1)-N(1)	94.81(8)	O(2)-P(1)-O(3)	113.04(1)				
Compound 3							
Co(1)-O(1)	2.084(2)	Co(1)-O(6)	2.128(2)	P(1)-O(3)	1.521(2)	P(2)-O(4)	1.508(2)
Co(1)-O(2)	2.051(2)	Co(1)-N(1)	2.183(2)	P(1)-O(8)	1.539(2)	P(2)-O(6)	1.528(2)
Co(1)-O(3)	2.115(2)	P(2)-C(1)	1.823(2)	P(1)-O(9)	1.496(2)	P(2)-O(7)	1.543(2)
Co(1)-O(5)	2.082(2)	P(1)-C(2)	1.824(2)				
O(2)-Co(1)-O(5)	91.50(8)	O(5)-Co(1)-O(6)	93.04(8)	O(6)-Co(1)-N(1)	84.21(7)	C(3)-O(2)-Co(1)	116.33(1)
O(2)-Co(1)-O(1)	179.01(8)	O(1)-Co(1)-O(6)	92.16(8)	O(9)-P(1)-O(3)	113.96(1)	P(1)-O(3)-Co(1)	116.10(9)
O(5)-Co(1)-O(1)	88.46(9)	O(3)-Co(1)-O(6)	169.57(6)	O(9)-P(1)-O(8)	112.42(1)	C(2)-N(1)-Co(1)	106.95(1)
O(2)-Co(1)-O(3)	89.66(8)	O(2)-Co(1)-N(1)	81.97(7)	O(3)-P(1)-O(8)	111.38(1)	C(1)-N(1)-Co(1)	106.01(1)
O(5)-Co(1)-O(3)	97.32(8)	O(5)-Co(1)-N(1)	172.95(8)	O(4)-P(2)-O(6)	113.27(1)	C(4)-N(1)-Co(1)	106.95(1)
O(1)-Co(1)-O(3)	89.37(8)	O(1)-Co(1)-N(1)	98.12(8)	O(4)-P(2)-O(7)	111.77(1)	P(2)-O(6)-Co(1)	116.18(9)
O(2)-Co(1)-O(6)	88.83(7)	O(3)-Co(1)-N(1)	85.36(7)	O(6)-P(2)-O(7)	112.15(1)		

Symmetry transformations used to generate equivalent atoms:

^{#1} -x+1, -y+1, -z ^{#2} -x+1/2, y-1/2, z

Table S2. Hydrogen Bonds [\AA] and Angles [$^\circ$] for Compounds 1–3.

D-H...A	d(D-H)	d(H...A)	d(D...A)	$\angle(\text{DHA})$
Compound 1				
N(2)-H(4)...O(3)#1	0.89	1.94	2.821(2)	171.7
N(2)-H(5)...O(6)#3	0.89	1.88	2.752(2)	165.6
N(2)-H(6)...O(4)	0.89	2.25	2.894(2)	129.1
N(2)-H(6)...O(5)	0.89	2.39	3.190(3)	149.1
O(9)-H(1)...O(7)#1	0.74(3)	2.14(3)	2.849(2)	161(3)
O(9)-H(2)...O(8)#4	0.93(4)	1.81(4)	2.732(3)	171(3)
N(3)-H(7)...O(2)#2	0.85(3)	2.03(3)	2.879(2)	177(3)
N(3)-H(8)...O(2)	0.99(3)	1.95(3)	2.833(2)	147(3)
N(3)-H(9)...O(4)#4	0.88(3)	1.84(3)	2.696(2)	164(3)
N(3)-H(10)...O(5)#5	1.00(3)	1.74(3)	2.740(2)	175(3)
N(4)-H(11)...O(3)#1	0.87(3)	1.97(3)	2.832(2)	169(3)
N(4)-H(12)...O(2)#6	0.86(3)	1.96(3)	2.809(2)	169(3)
N(4)-H(13)...O(6)#7	0.95(4)	1.86(4)	2.799(2)	170(3)
N(4)-H(14)...O(5)#2	0.90(4)	2.05(4)	2.878(2)	152(3)
Compound 2				
O(9)-H(9A)...O(6)#1	0.880(2)	1.86(2)	2.730(3)	168(4)
O(9)-H(9B)...O(3)#2	0.861(2)	1.92(2)	2.770(3)	168(4)
O(10)-H(10A)...O(7W)#3	0.884(2)	1.761(2)	2.642(3)	174(4)
O(10)-H(10B)...O(1)#2	0.880(2)	1.84(2)	2.714(3)	175(4)
O(1W)-H(1A)...O(6)#4	0.92(2)	1.80(2)	2.709(4)	171(5)
O(1W)-H(1B)...O(3W)#5	0.91(2)	2.20(3)	3.035(6)	152(5)
O(2W)-H(2A)...O(2)	0.911(2)	1.85(2)	2.735(4)	162(4)
O(3W)-H(3A)...O(8)	0.940(2)	1.79(2)	2.727(4)	177(5)
O(3W)-H(3B)...O(1W)#4	0.932(2)	2.35(3)	3.276(8)	165(5)
O(3W)-H(3C)...O(3)#4	0.922(2)	1.82(2)	2.720(4)	163(4)
O(4W)-H(4A)...O(2)	0.880(2)	1.81(2)	2.685(3)	177(5)
O(4W)-H(4B)...O(5)#6	0.89(2)	1.92(2)	2.801(4)	179(5)
O(5W)-H(5B)...O(8)#7	0.969(2)	2.39(3)	3.206(4)	142(4)
O(5W)-H(5B)...O(2W)#7	0.969(2)	2.38(4)	3.055(4)	126(3)
O(5W)-H(5C)...O(3)#2	0.954(2)	1.94(2)	2.882(3)	170(4)
O(6W)-H(6A)...O(1)	0.958(2)	1.88(2)	2.824(3)	170(4)
O(6W)-H(6B)...O(5)#6	0.96(2)	2.10(3)	3.007(3)	156(4)
O(6W)-H(6C)...O(4)#1	0.949(2)	1.91(2)	2.848(3)	172(4)
O(7W)-H(7A)...O(8)#7	0.891(2)	1.86(2)	2.736(3)	166(3)
O(7W)-H(7A)...O(7)#7	0.891(2)	2.53(3)	3.221(3)	135(3)
O(7W)-H(7B)...O(5)	0.910(2)	1.80(2)	2.705(3)	171(4)
Compound 3				
O(5)-H(10)...O(11)#5	0.76(4)	1.98(4)	2.718(3)	166(4)
O(5)-H(11)...O(4)#3	0.81(4)	1.93(4)	2.725(3)	168(4)
O(1)-H(7)...O(9)#4	0.877(2)	1.80(2)	2.666(3)	167(3)
O(1)-H(8)...O(6)#3	0.864(2)	1.89(2)	2.725(3)	164(3)
O(12W)-H(16)...O(7)#2	0.99(4)	1.88(4)	2.855(3)	169(3)
O(13W)-H(19)...O(13W)#1	0.81(3)	2.01(3)	2.818(7)	178(4)

Symmetry transformations used to generate equivalent atoms:

Compound 1. #1 $-x+1, -y+1, -z$ #2 $-x+1, y, -z+1/2$ #3 $-x+3/2, -y+1/2, -z$ #4 $x, -y+1, z+1/2$ #5 $-x+3/2, y+1/2, -z+1/2$ #6 $x-1/2, y-1/2, z$ #7 $x-1, y, z$

Compound 2. #1 $-x+3/2, y-1/2, -z+1/2$ #2 $-x+1, -y, -z$ #3 $-x+1, -y+1, -z$ #4 $-x+2, -y, -z$ #5 $x-1/2, -y-1/2, z-1/2$ #6 $x, y-1, z$ #7 $-x+3/2, y+1/2, -z+1/2$

Compound 3. #1 $-x+2, -y+1, -z$ #2 $x+1, y, z$ #3 $-x+1, -y, -z$ #4 $-x+1, -y+1, -z$ #5 $-x+1, -y, -z+1$