### **Electronic Supplementary Information**

# Synthesis of Mesoporous LaPO<sub>4</sub> Nanostructures with Controllable Morphologies

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#### **Preparation of La-TOPO**

10.0 mmol of La(NO<sub>3</sub>)<sub>3</sub>·nH<sub>2</sub>O was dissolved in 10.0 ml of deioned water. After dissolution, lanthanum ions were extracted from the La(NO<sub>3</sub>)<sub>3</sub>·nH<sub>2</sub>O aqueous solution by 3.89 g of TOPO (~10 mmol). To reduce the viscosity of the organic phase containing TOPO and extracted lanthanum species, 7.0 ml of cyclohexane was added to dilute the organic phase, facilitating the separation of two phases and later transference by syringe. The extraction yield of lanthanum ions was about 60.0% in TOPO, and the final concentration of lanthanum was around 0.35 mol L<sup>-1</sup> in the TOPO-cyclohexane (denoted La-TOPO).

#### Preparation of H<sub>3</sub>PO<sub>4</sub>-EG

10.0 ml of 85%  $H_3PO_4$  solution was mixed with 10.0 ml of EG, followed by stirring and heating at 100 °C for 10 h to remove water. As a result, a clear solution with a high concentration of around 7.0 mol L<sup>-1</sup> was obtained (denoted  $H_3PO_4$ -EG).

## Synthesis of Eu<sup>3+</sup> doped and Ce<sup>3+</sup>/Tb<sup>3+</sup> codoped LaPO<sub>4</sub> nanostructures

*LaPO*<sub>4</sub>:*Ce*,*Tb*: The synthetic procedure for the LaPO<sub>4</sub>:Ce,Tb nanostructures was the same as that used for the synthesis of undoped LaPO<sub>4</sub> nanostructures, except that 1.14 ml of La-TOPO (0.35 M, ~0.4 mmol), 1.67 ml of Ce-TOPO (0.27 M, ~0.45 mmol), and 0.40 ml of Tb-TOPO (0.38 M, ~0.15 mmol) were used as the precursors in EG.

*LaPO*<sub>4</sub>:*Eu*: The synthetic procedure for the LaPO<sub>4</sub>:*Eu* nanostructures was also same as that used for the synthesis of undoped LaPO<sub>4</sub> nanostructures, except that 1.14 ml of La-TOPO (0.35 M, ~0.4 mmol) and 0.08 ml of Eu-TOPO (0.25 M, ~0.02 mmol) were used as the precursors in EG.



Fig. S1 SEM images of the obtained  $LaPO_4$  nanostructures in a typical synthesis with different reaction time. a) 10 min b) 3 h



Fig. S2 SEM images of the obtained LaPO<sub>4</sub> nanostructures in a typical synthesis with different reaction temperature. a) 120  $^{\circ}$ C c) 180  $^{\circ}$ C



Fig. S3 XRD patterns of LaPO<sub>4</sub> nanostructures obtained in a typical synthesis with different molar ratio of  $La^{3+}/H_3PO_4$ . (a) 1/2 (b) 1/20 (c) 1/100 (d) 1/200.



Fig. S4 SEM images obtained LaPO<sub>4</sub> nanostructures in a typical synthesis with different molar ratio of  $La^{3+}/H_3PO_4$ : (a) 5/1 (b) 1/4 (c) 1/10 (d) 1/30 (e) 1/40 (f) 1/60.



Fig. S5 Fourier Transform Infrared spectra (FT-IR) of (a) as-obtained LaPO<sub>4</sub> nanostructures in a typical synthesis and (b) LaPO<sub>4</sub> nanoparticles prepared by adding phosphate acid (85%) into La(NO<sub>3</sub>)<sub>3</sub> aqueous solution at room temperature.



**Fig. S6** XRD patterns of doped LaPO<sub>4</sub> nanostructures obtained in a typical synthesis. a) LaPO<sub>4</sub>:Eu, b) LaPO<sub>4</sub>:Ce,Tb.



Fig. S7 SEM images of LaPO<sub>4</sub>:Ce,Tb nanostructures synthesized in a typical synthesis with different molar ratio of  $La^{3+}/H_3PO_4$ : a) 1/2 b) 1/20 c) 1/100 d) 1/200



Fig. S8 Room-temperature excitation and emission spectra of LaPO<sub>4</sub>:Ce,Tb nanostructures synthesized in a typical synthesis with different molar ratio of La<sup>3+</sup>/H<sub>3</sub>PO<sub>4</sub> a) 1/2 b) 1/20 c) 1/100 d) 1/200 ( $\lambda_{ex} = 273$  nm and  $\lambda_{em} = 546$  nm).