

Optical and magnetic properties of the photo-induced state in the coordination network $\text{Na}_2\text{Co}_4[\text{Fe}(\text{CN})_6]_{3.3} \cdot 14\text{H}_2\text{O}$.

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Supporting information

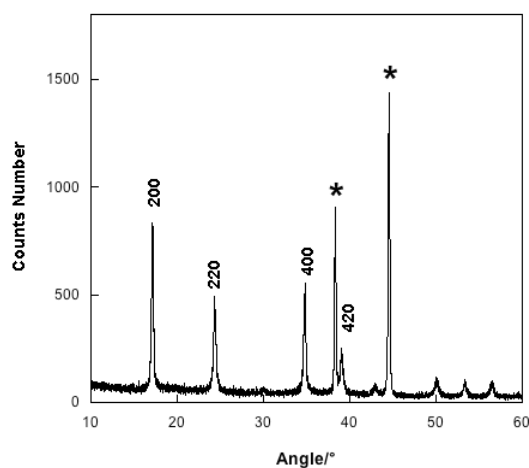


Figure S1: X-ray powder pattern at room temperature for $\text{Na}_2\text{Co}_4[\text{Fe}(\text{CN})_6]_{3.3} \cdot 14\text{H}_2\text{O}$.

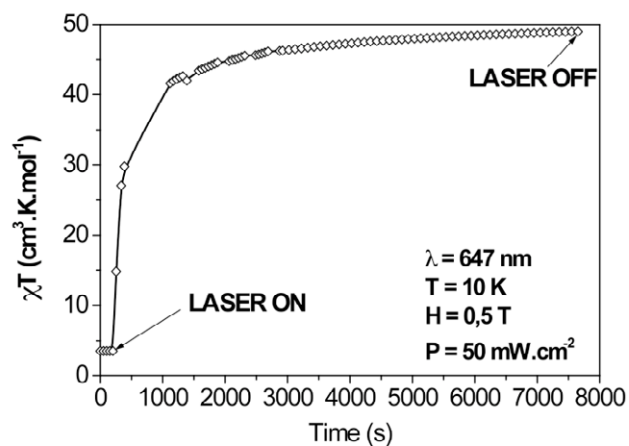


Figure S2: χT vs time at 10 K under a red light at 647 nm ($50 \text{ mW}/\text{cm}^2$) for $\text{Na}_2\text{Co}_4[\text{Fe}(\text{CN})_6]_{3.3} \cdot 14\text{H}_2\text{O}$

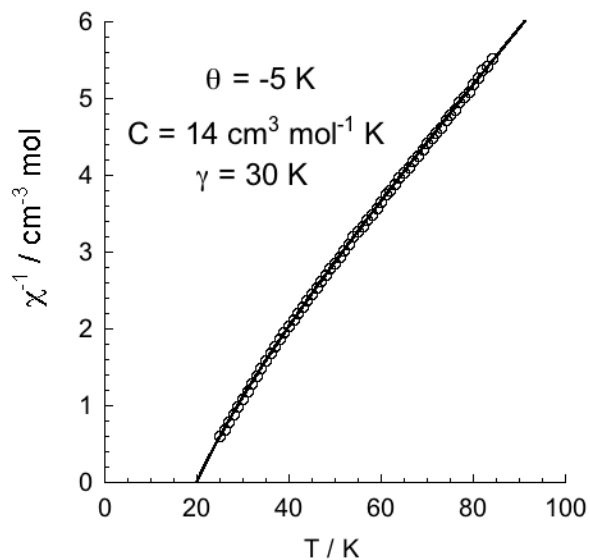


Figure S3: $1/\chi$ vs T at 10 K for $\text{Na}_2\text{Co}_4[\text{Fe}(\text{CN})_6]_{3.3} \cdot 14\text{H}_2\text{O}$ in the PI state. (•) experimental points. (—) fit using the theoretical law for a ferrimagnetic network AB:

The fit parameters are Θ , which is the Curie-Weiss constant, C the Curie constant ($=C_A+C_B=3.3 \cdot C_{\text{Fe}}+4C_{\text{Co}}$), γ being a parameter proportional to the molecular field coefficient W .¹

$$\frac{1}{\chi} = \frac{T+\theta}{C} + \frac{\gamma}{T-\theta}$$

The best parameters are given on the figure S3. The Θ value is in agreement with the ferrimagnetic nature of the compound.²

¹ L. Néel, *Ann. Phys.* 1948, t. 3, 10.

² O. Sato, Y. Einaga, A. Fujishita, K. Hashimoto, *Inorg. Chem.*, 1999, 38, 4405.