

Supplementary Material

Cationic side-chains control DNA/RNA binding properties and antiproliferative activity of dicationic dibenzotetraaza[14]annulene derivatives.

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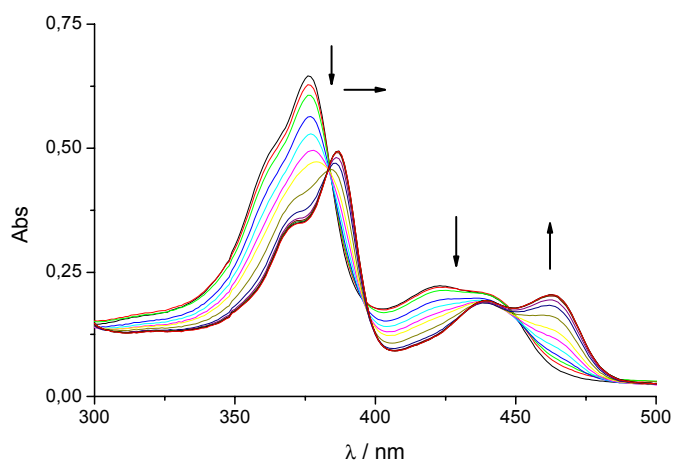
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Solubility

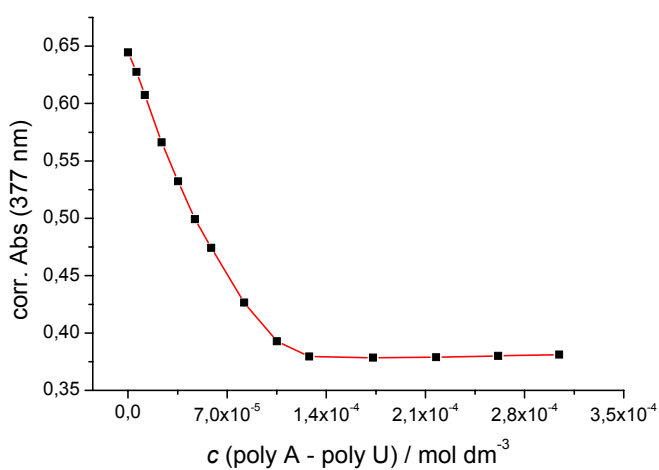
Studied compounds were dissolved in sodium cacodylate buffer, $I = 0.05 \text{ mol dm}^{-3}$ at pH = 7.0 (**4**, $c = 1.54 \times 10^{-3} \text{ mol dm}^{-3}$; **5**, $c = 1.44 \times 10^{-3} \text{ mol dm}^{-3}$; **6**, $c = 1.55 \times 10^{-3} \text{ mol dm}^{-3}$; **7**, $c = 1.67 \times 10^{-3} \text{ mol dm}^{-3}$).

The absorbencies of buffered aqueous solutions of studied compounds are proportional to their concentrations up to $c = 3\text{--}4 \times 10^{-5} \text{ mol dm}^{-3}$. Buffered aqueous solutions of studied compounds were stable for at least one week, when stored in a dark and cold place ($+ 8^{\circ}\text{C}$).

Spectrophotometric titrations



a)



b)

Figure S1. a) Changes in UV/Vis spectrum of **4** ($c = 1.53 \times 10^{-5} \text{ mol dm}^{-3}$) upon titration with poly A – poly U; b) Dependence of **4** absorbance at $\lambda_{\text{max}} = 377 \text{ nm}$ on $c(\text{poly A} - \text{poly U})$, at pH=7, sodium cacodylate buffer, $I = 0.05 \text{ mol dm}^{-3}$.

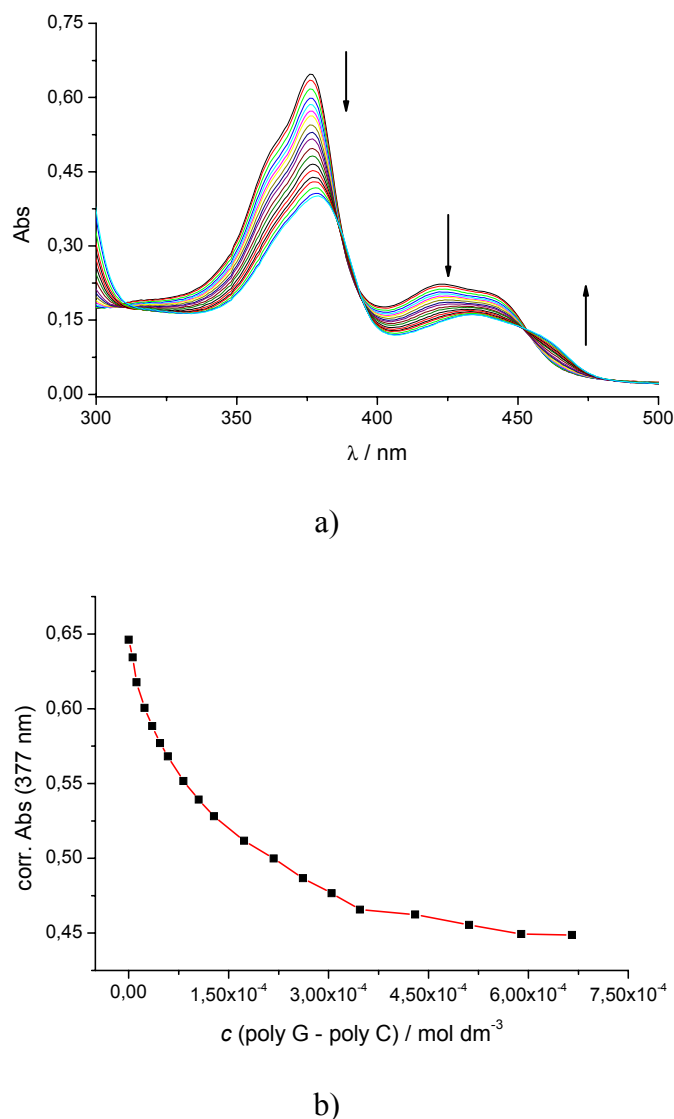


Figure S2. a) Changes in UV/Vis spectrum of **4** ($c = 1.53 \times 10^{-5} \text{ mol dm}^{-3}$) upon titration with poly G – poly C; b) Dependence of **4** absorbance at $\lambda_{\text{max}} = 377 \text{ nm}$ on $c(\text{poly G – poly C})$, at pH=7, sodium cacodylate buffer, $I = 0.05 \text{ mol dm}^{-3}$.

Viscometry measurements

Viscometry measurements were conducted with an Ubbelohde viscometer system AVS 350 (Schott). The temperature was maintained at $25 \pm 0.1 \text{ }^{\circ}\text{C}$. Aliquots of drug stock solutions were added to 5.5 ml of $5 \times 10^{-4} \text{ mol dm}^{-3}$ ct-DNA solution in sodium cacodylate buffer, $I = 0.05 \text{ mol dm}^{-3}$, pH=7, with a compound to DNA phosphate ratio r less than 0.2. Dilution never exceeded 4% and was corrected for in the calculations. The flow times were measured at least five times optically with a deviation of $\pm 0.2 \text{ s}$. The viscosity index α was obtained from the flow times at varying r according to the following equation:¹

$$L/L_0 = [(t_r - t_0) / (t_{\text{DNA}} - t_0)]^{1/3} = 1 + \alpha \cdot r$$

Where t_0 , t_{DNA} and t_r denote the flow times of buffer, free DNA and DNA complex at ratio $r_{[\text{compound}]} / [\text{ct-DNA}]$, respectively; L/L_0 is the relative DNA lengthening. The L/L_0 to $r_{[\text{compound}]} / [\text{ct-DNA}]$ -plot was fitted to a straight line that gave slope α . The error in α is ≤ 0.1 .

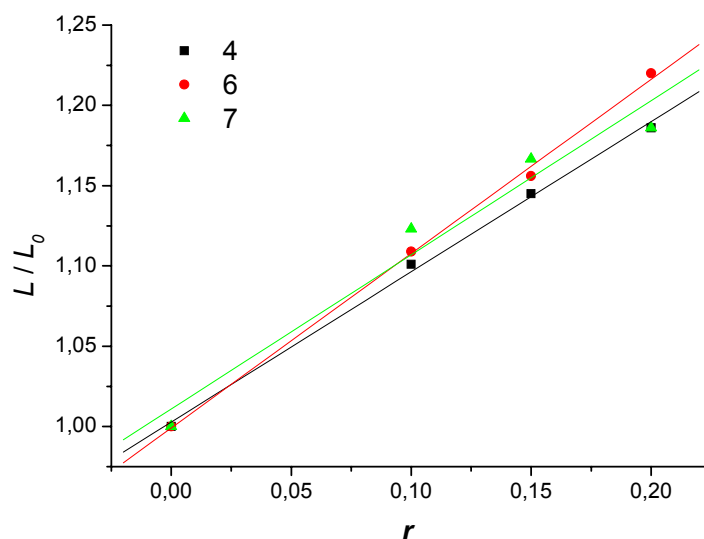


Figure S3. Relative DNA helix length extension (L/L_0) vs. ratio $r_{[\text{compound}]} / [\text{ct-DNA}]$ plot for **4**, **6** and **7** at pH=7, sodium cacodylate buffer, $I = 0.05 \text{ mol dm}^{-3}$. Results: $\alpha = 0.93 \pm 0.03$ (for **4**); 1.08 ± 0.04 (for **6**); 0.96 ± 0.1 (for **7**). It was not possible to determine α for compound **5** because of a precipitation.

¹ G. Cohen and H. Eisenberg, *Biopolymers* 1969, **8**, 45; M. Wirth, O. Buchardt, T. Koch, P. E. Nielsen and B. Nordén, *J. Am. Chem. Soc.* 1988, **110**, 932.