Extended π conjugation in 2*H*-1,4,2-diazaphosphole complexes

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Fig. S.1 Comparison of the experimental UV/Vis spectrum of 9 (CH₂Cl₂; solid line) with calculated vertical singlet excitations for cation 9^+ (A) and contact ion pair 9 (B) (TD-B3LYP/SV(P)/ECP-60-MWB(W) // RI-BLYP/SV(P)/ECP-60-MWB(W)).



Table S.1 Selected vertical singlet excitations calculated for the cation 9^+ (only the major orbital contributions given).

Fig. S.2 Visualization of selected molecular orbitals calculated for the cation **9**⁺ (B3LYP/SV(P)/ECP-60-MWB(W); isovalue 0.04 au).



Fig. S.3 Visualization of selected molecular orbitals calculated for the contact ion pair **9** (B3LYP/SV(P)/ECP-60-MWB(W); isovalue 0.04 au).

Table S.	2 Selected	vertical	singlet	excitations	calculated	for 3	3b (or	ily the	major	orbital	contribution
given).			U					5	5		

No.	ν [cm ⁻¹]	Oscillator strength	Orbital contributions	$ c ^2$ [%]
1	20596	0.002	HOMO $(d_2) \rightarrow \text{LUMO}(\pi^*)$	99
2	21326	0.017	HOMO-1 $(d_1) \rightarrow$ LUMO (π^*)	99
4	24498	0.051	HOMO-3 $(\pi_3) \rightarrow$ LUMO (π^*)	96
13	30564	0.147	HOMO-4 $(\pi_2) \rightarrow$ LUMO (π^*)	48
19	33960	0.051	HOMO-6 $(\pi_i) \rightarrow \text{LUMO}(\pi^*)$	87



Fig. S.4 Visualization of selected molecular orbitals calculated for **3b** (B3LYP/SV(P)/ECP-60-MWB(W); isovalue 0.04 au).



Fig. S.5 Visualization of selected molecular orbitals calculated for **4** (B3LYP/SV(P)/ECP-60-MWB(W); isovalue 0.04 au).