

Electronic Supplementary Information For

A colorimetric and fluorescent chemosensor for copper ions in aqueous media and its application in living cells

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1 Experimental Sections

All titrations experiments were carried out in an aqueous buffer. UV-vis and fluorescence spectra were recorded on HITACHI 3010 UV-vis spectrometer and HITACHI F-4600 spectrometer, respectively. Excitation and emission slits are 5 nm. All proton and ^{13}C NMR were measured on Bruker AVANCE-400 NMR spectrometer. Confocal fluorescence microscopy was operated on Olympus FV-1000 laser scanning microscopy system equipped with a 488 nm laser head.

a Uv-vis absorption spectrum of copper complex upon addition of EDTA and TPEN

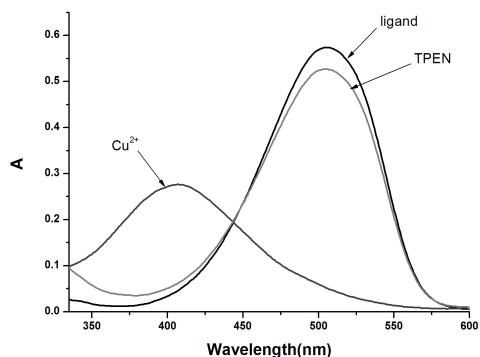
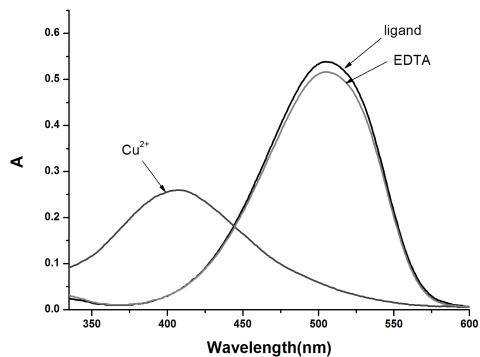


Figure S1. Uv-vis absorption of **1** (12.5 μM) upon addition of 4 equiv copper ion, then treated with 4 equiv EDTA (Top) or TPEN (Bottom) respectively.

b Job's plot

A series of solutions containing **1** and Cu(NO₃)₂ were prepared such that the sum of the total metal and **1** concentration remained constant at 25 μM . The molar fraction x of **1** was varied between 0.1 and 1.0. The corrected absorbance $A_{(\text{L/M})}-A_{(\text{L})}$ at 410 nm was plotted against the molar fraction of the sample solution.

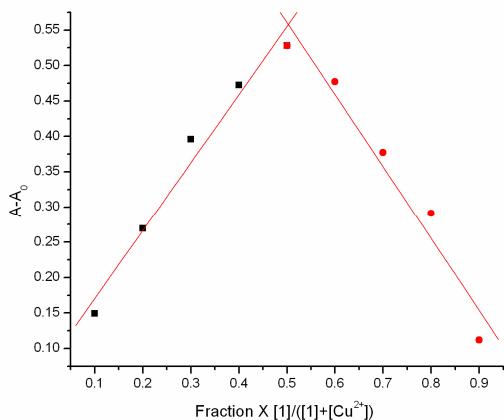


Figure S2. Job's plot of 1 in aqueous buffer.

c Binding constant

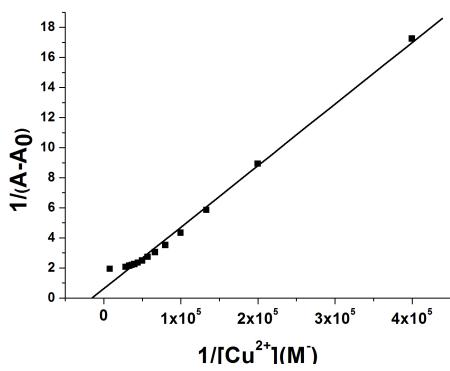
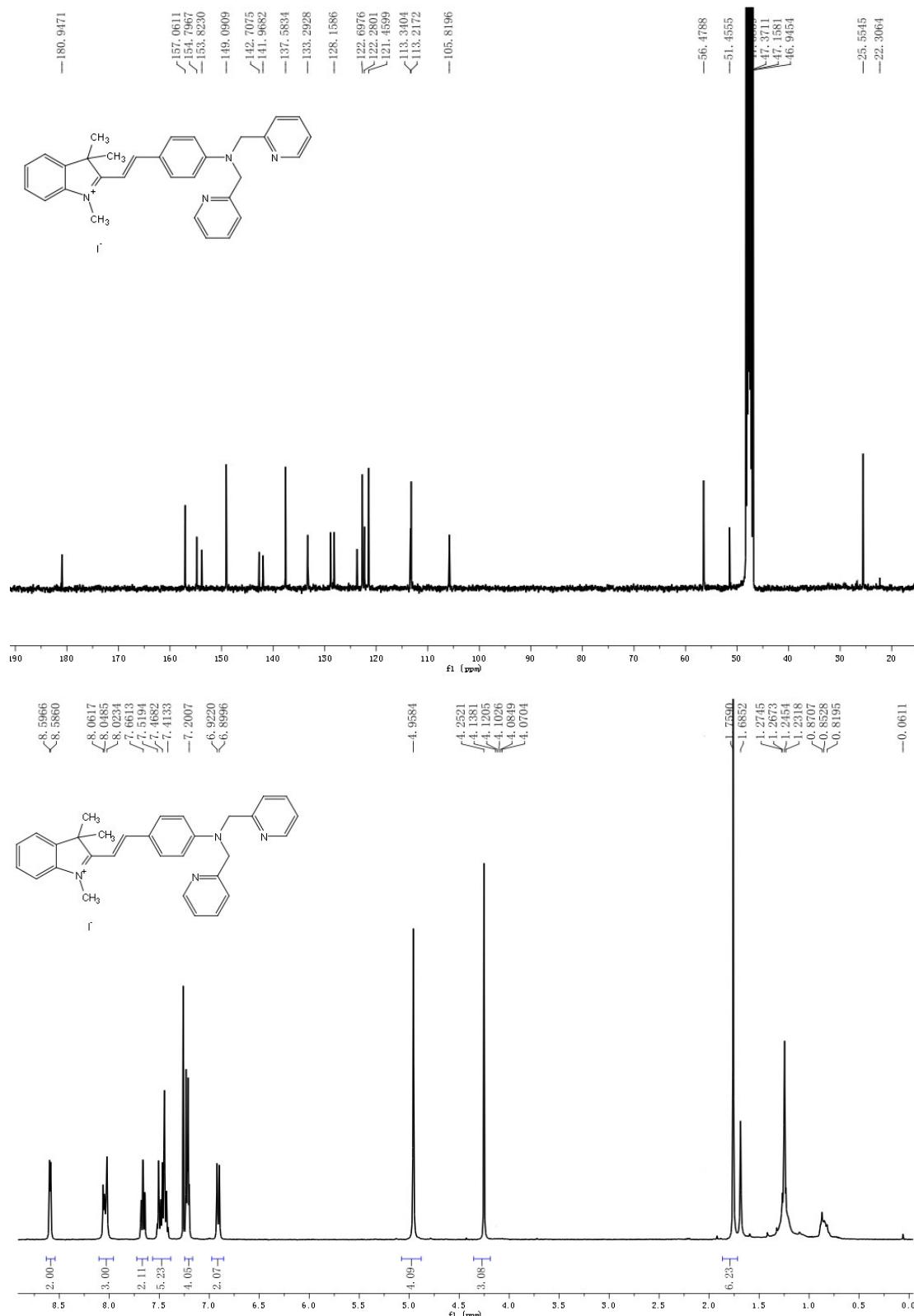


Figure S3. Benesi-hilderbrand plot according to Uv-vis absorbance at 410 nm.

2 NMR spectra and HRMS analysis report of 1



Peking University Mass Spectrometry Sample Analysis Report

Analysis Info

Analysis Name 0912204_20091214_000001.d
Sample F-DPA
Comment ESI Positive

Acquisition Date 12/14/2009 4:49:23 PM
Instrument Bruker Apex IV FTMS
Operator Peking University

