

Electronic supplementary information

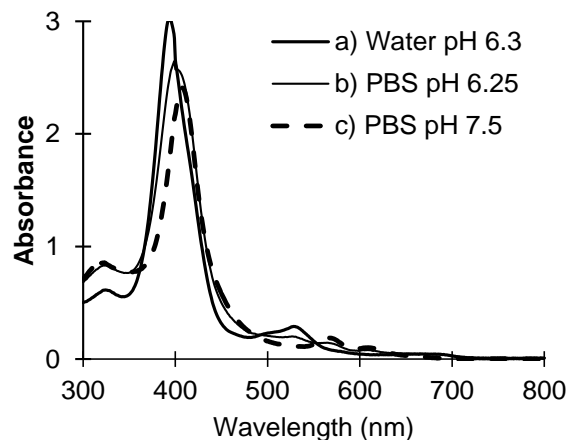


Fig.ESI1: UV-Vis spectra of 60 μM FeTSPP in water pH 6.3 (a), in PBS pH 6.25 (b), pH 7.5 (c).

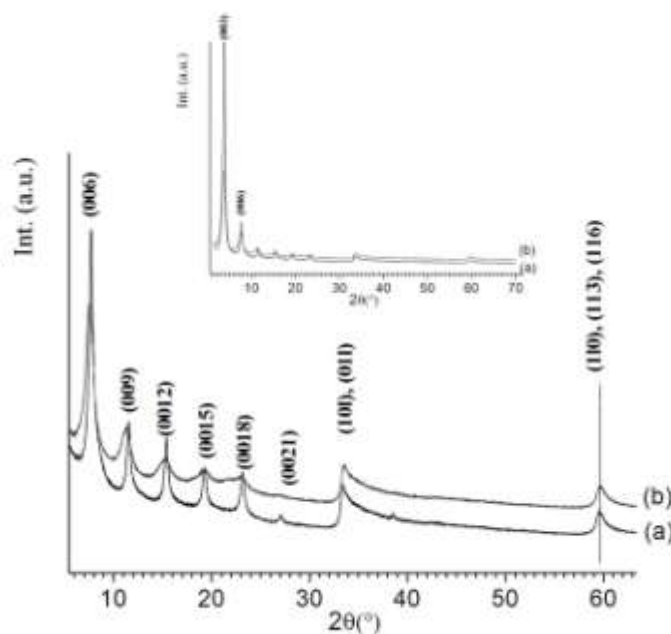


Fig.ESI2: X-ray powder diffraction patterns in the 2θ range 5-65° (inset 1-30°) of (a) $\text{Zn}_2\text{Cr-ZnTSPP}$ and (b) $\text{Zn}_2\text{Cr-FeTSPP}$. Labels give the Bragg reflections in the space group $R\bar{3}m$.

FTIR spectra (Fig.ESI3) show besides the vibration bands of the ZnCr LDH framework at 3500-3400 cm^{-1} (ν_{OH}) and at 576 and 507 cm^{-1} corresponding to the ν_{MO} lattice vibrations, the characteristic bands due to the TSPP porphyrins. IR spectra of TSPP porphyrins are characterized

by four main regions recorded between 2000 and 400 cm^{-1} . Vibration bands in the region 1600-1350 cm^{-1} are due to stretching vibrations of C=C and C=N in phenyl and pyridyl aromatic ring. Bands situated between 1200 and 1010 cm^{-1} are typical of phenyl-sulfonate groups and those between 1000 and 900 cm^{-1} of $\nu\text{C-H}$ rock and $\delta\text{N-H}$ pyrrole groups. Finally, the 700-800 cm^{-1} region is attributed to out of plane C-H vibrations¹. The fundamental stretching vibrations $\nu_{\text{as}}(\text{SO}_3)$ and $\nu_{\text{s}}(\text{SO}_3)$ are observed at 1176 and 1037 cm^{-1} for $\text{Zn}_2\text{Cr-ZnTSP}$ and at 1180 and 1038 cm^{-1} for $\text{Zn}_2\text{Cr-FeTSP}$. The position of these two bands are practically the same than that observed reported for the corresponding ZnAl-porphyrin hybrids, with bands at 1171 and 1036 cm^{-1} for ZnAl-ZnTSP² and 1184 and 1039 cm^{-1} for ZnAl-FeTSP³. These peak positions are similar to the free porphyrins, suggesting that SO_3^- groups of intercalated porphyrins interact with the LDH layers via weak noncovalent interactions.

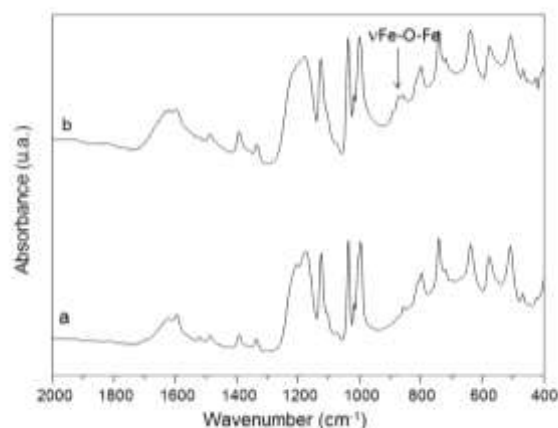


Fig.ESI3: FTIR spectra of $\text{Zn}_2\text{Cr-ZnTSP}$ (a) and $\text{Zn}_2\text{Cr-FeTSP}$ (b).

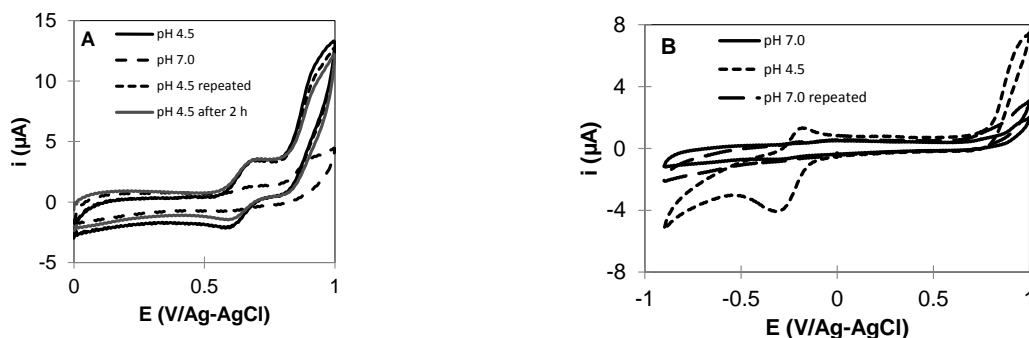


Fig.ESI4: Cyclic voltammograms recorded in different PBS using the same $\text{Zn}_2\text{Cr-ZnTSP/GCE}$ (20th cycles) (A) and $\text{Zn}_2\text{Cr-FeTSP/GCE}$ (10th cycles) (B) (0.1 M PBS under argon, $\nu = 50 \text{ mV s}^{-1}$).

References

1. K. Nakamoto, ed., *Infrared and Raman spectra of Inorganic and coordination compounds* Wiley, New York, 1997.
2. E. Káfuňková, C. Taviot-Gueho, P. Bezdika, M. Klementov, P. Kov, P. Kubt, J. Mosinger, M. Pospil and K. Lang, *Chem. Mater.*, 2010, **22**, 2481-2480.
3. M. Halma, A. K. Dias de Freitas Castro, C. Taviot-Gueho, V. Prevot, C. Forano, F. Wypych and S. Nakagaki, *J. Catal.*, 2008, **257**, 233-243.