

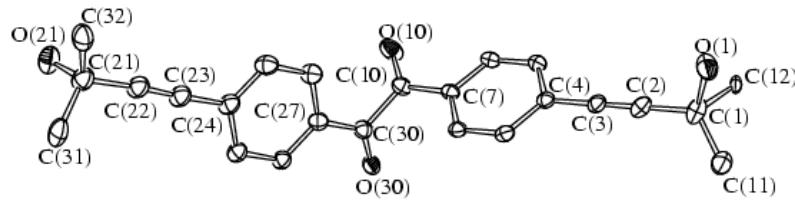
## Supporting Information

### A New Tetra-aryl-cyclopentadienone Based Low Molecular Weight Gelator: Synthesis, Self-Assembly Properties and Anion Recognition

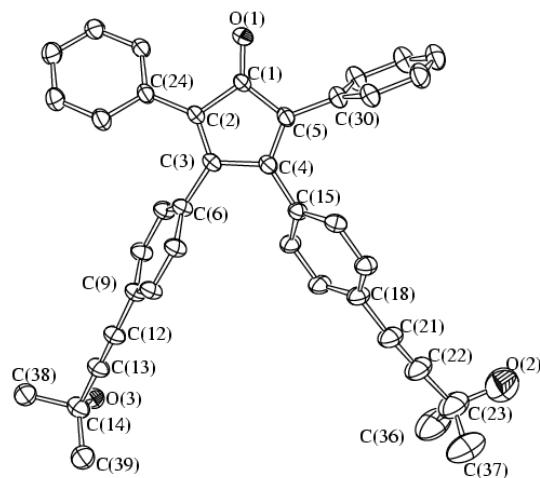
Mattia Boiani,<sup>a</sup> Andrea Baschieri,<sup>a</sup> Cristiana Cesari,<sup>b</sup> Rita Mazzoni,<sup>b</sup> Stefano Stagni,<sup>b</sup> Stefano Zacchini<sup>b</sup> and Letizia Sambri\*<sup>a</sup>

[a] M. Boiani, Dr. A. Baschieri, Dr. L. Sambri Dipartimento di Chimica Organica “A. Mangini”, University of Bologna, v.le Risorgimento 4, 40136 Bologna, Italy; Fax: (+) 39-051-2093654 ; e-mail: letizia.sambri@unibo.it

[b] Dr. R. Mazzoni, C. Cesari, Dr. S. Stagni, Prof. S. Zacchini Dipartimento di Chimica Fisica ed Inorganica, University of Bologna, v.le Risorgimento 4, 40136 Bologna, Italy.



**Figure 1S.** ORTEP drawing of **6a** with key atoms labelled. Displacement ellipsoids are at the 30% probability level. H-atoms have been omitted for clarity. Only the main images of the disordered groups are represented.



**Figure 2S.** ORTEP drawing of **1a** with key atoms labelled. Displacement ellipsoids are at the 30% probability level. H-atoms have been omitted for clarity. Only the main images of the disordered groups are represented.

**Table 1S.** Selected Bond Lengths ( $\text{\AA}$ ) and angles ( $^\circ$ ) for **6a**. Only the main images of disordered groups are considered.

C(10)-C(30)	1.538(6)		
C(10)-O(10)	1.211(5)	C(30)-O(30)	1.213(5)
C(7)-C(10)	1.466(6)	C(27)-C(30)	1.444(7)
C(3)-C(4)	1.434(6)	C(23)-C(24)	1.431(9)
C(2)-C(3)	1.193(6)	C(22)-C(23)	1.171(11)
C(1)-C(2)	1.471(6)	C(21)-C(22)	1.462(12)
O(10)-C(10)-C(7)	123.4(4)	O(30)-C(30)-C(27)	127.8(5)
O(10)-C(10)-C(30)	117.6(4)	O(30)-C(30)-C(10)	115.8(4)
C(7)-C(10)-C(30)	118.9(4)	C(27)-C(30)-C(10)	116.2(5)
C(2)-C(3)-C(4)	177.4(5)	C(22)-C(23)-C(24)	175.3(12)
C(3)-C(2)-C(1)	179.2(5)	C(23)-C(22)-C(21)	175.6(13)

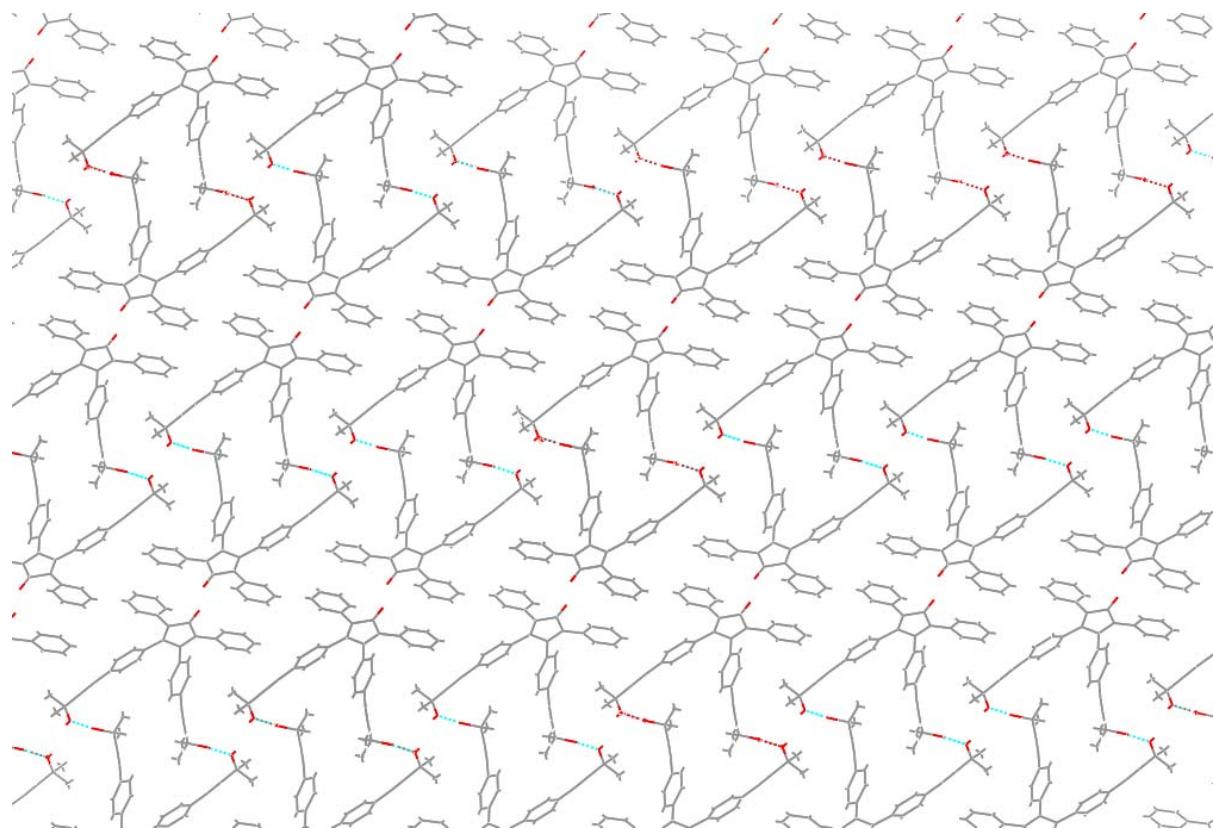
**Table 2S.** Selected Bond Lengths ( $\text{\AA}$ ) and angles ( $^\circ$ ) for **1a**. Only the main images of disordered groups are considered.

C(1)-O(1)	1.221(9)	C(3)-C(4)	1.516(11)
C(1)-C(2)	1.520(10)	C(1)-C(5)	1.471(11)
C(2)-C(3)	1.337(10)	C(4)-C(5)	1.354(10)
C(3)-C(6)	1.499(11)	C(4)-C(15)	1.474(11)
C(9)-C(12)	1.438(11)	C(18)-C(21)	1.443(13)
C(12)-C(13)	1.186(11)	C(21)-C(22)	1.189(12)
C(13)-C(14)	1.487(12)	C(22)-C(23)	1.474(15)
C(5)-C(1)-C(2)	108.0(7)	C(3)-C(2)-C(1)	105.6(7)
C(2)-C(3)-C(4)	110.4(7)	C(5)-C(4)-C(3)	108.9(7)
C(4)-C(5)-C(1)	107.1(7)		
C(13)-C(12)-C(9)	173.7(11)	C(22)-C(21)-C(18)	175.1(12)
C(12)-C(13)-C(14)	179.8(12)	C(21)-C(22)-C(23)	176.5(14)

Crystal data and collection details for **1a** and **6a** are listed in Table 6.

**Table 3S.** Crystal data and details of the structure refinement for **1a** and **6a**.

	<b>1a</b>	<b>6a</b>
Formula	$\text{C}_{39}\text{H}_{32}\text{O}_3$	$\text{C}_{24}\text{H}_{22}\text{O}_4$
$F_w$	548.65	374.42
$\lambda, \text{\AA}$	0.71073	0.71073
Temperature, K	293(2)	100(2)
Crystal system	Triclinic	Monoclinic
Space group	$P\bar{1}$	$C2/c$
$a, \text{\AA}$	6.0489(17)	23.681(8)
$b, \text{\AA}$	13.222(4)	4.9756(18)
$c, \text{\AA}$	20.272(6)	33.914(12)
$\alpha, {}^\circ$	73.883(4)	90
$\beta, {}^\circ$	86.839(4)	94.726(5)
$\gamma, {}^\circ$	77.762(4)	90
Cell volume, $\text{\AA}^3$	1522.3(7)	3983(2)
Z	2	8
$D_c, \text{g cm}^{-3}$	1.197	1.249
$\mu, \text{mm}^{-1}$	0.074	0.084
$F(000)$	580	1584
$\theta$ limits, ${}^\circ$	1.64-22.46	2.02-25.03
Reflections collected	11783	17291
Independent reflections	3958 ( $R_{\text{int}} = 0.1420$ )	3419 ( $R_{\text{int}} = 0.0479$ )
Data/restraints /parameters	3958/321/377	3419/622/366
Goodness on fit on $F^2$	1.008	1.020
$R1 (I > 2\sigma(I))$	0.1055	0.0986
$wR2$ (all data)	0.3554	0.2948
Largest diff. peak and hole, e. $\text{\AA}^{-3}$	0.614 and -0.577	0.779 and -0.678



**Figure 3S:** View along the *a* axis of the crystal packing of **1a**. Oxygen atoms are in red; H-bonds dotted blue lines.

**Table 4S:**  $T_g$  of **1a**-gel in toluene at various concentrations ( $c$ ) and the natural logarithm of the concentration and  $1/T_g$  values.

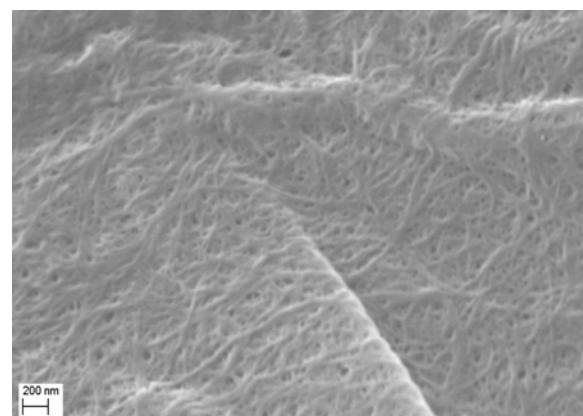
entry	$c$ (M)	$T_g$ (°K)	$\ln c$	$1/T$
1	0.01260	323	-4.37406	0.003096
2	0.01360	324	-4.29769	0.003086
3	0.01800	326	-4.01738	0.003067
4	0.02500	329	-3.68888	0.00304
5	0.03650	332	-3.31044	0.003012
6	0.04500	335	-3.10109	0.002985

**Table 5S:**  $T_g$  of **1a**-gel in nitromethane at various concentrations ( $c$ ) and the natural logarithm of the concentration and  $1/T_g$  values.

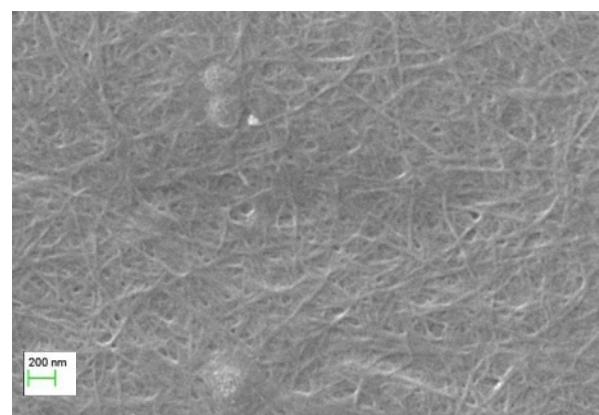
entry	$c$ (M)	$T_g$ (K)	$\ln c$	$1/T$
1	0.00869	321	-4.74558	0.003115
2	0.01330	328	-4.31999	0.003049
3	0.02000	333	-3.91202	0.003003
4	0.02600	336	-3.64966	0.002976
5	0.03320	341	-3.40521	0.002933

**Figure 4S:** SEM images of xerogels of **1a** in various solvents

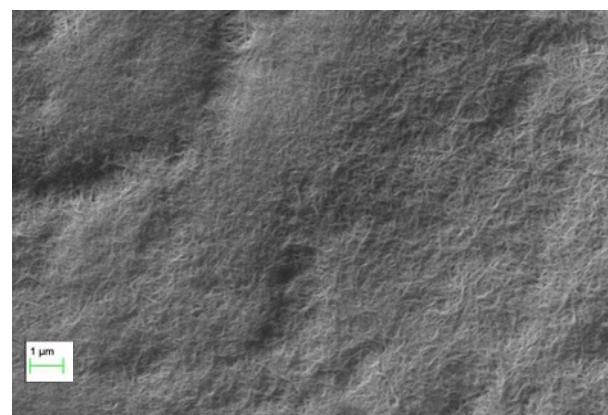
Acetonitrile



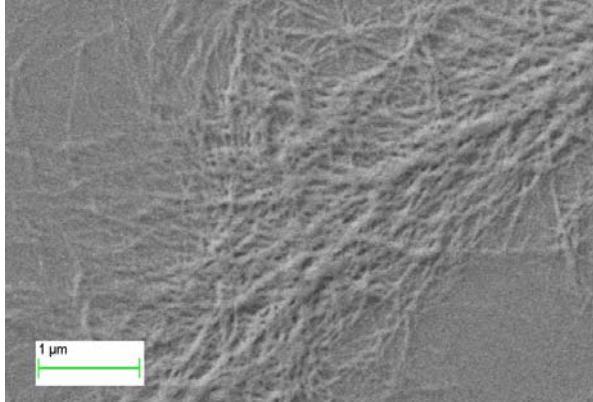
Nitromethane



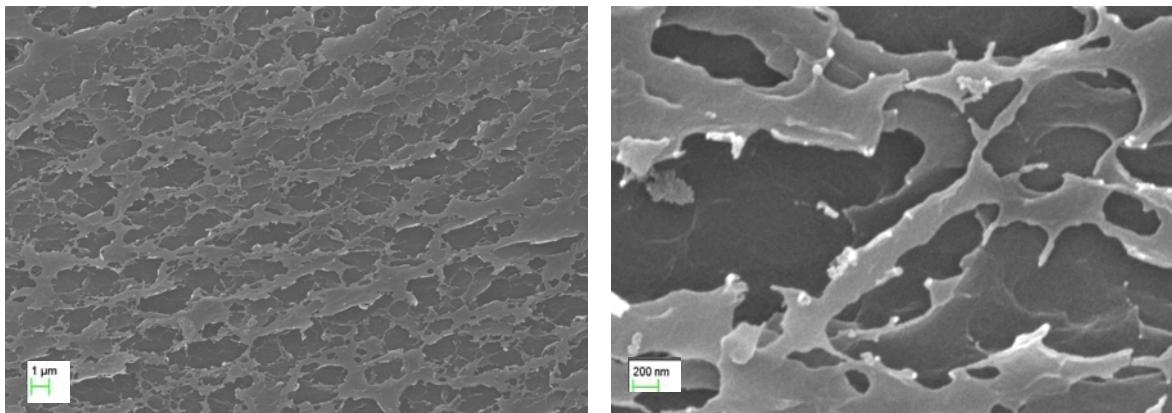
Benzene



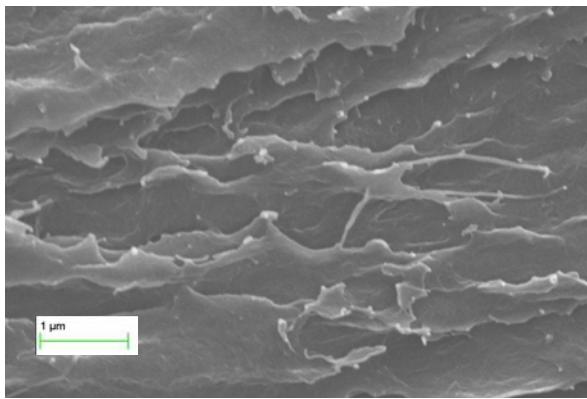
toluene



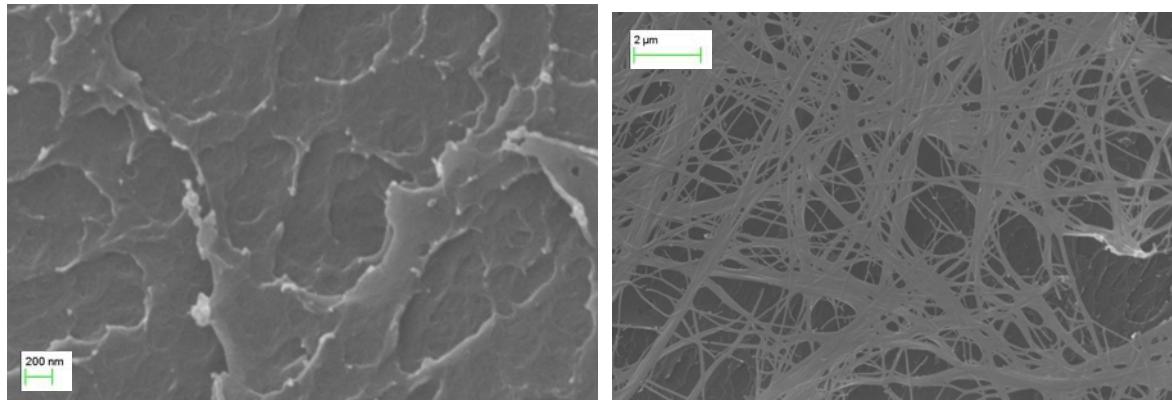
*o*-xylene



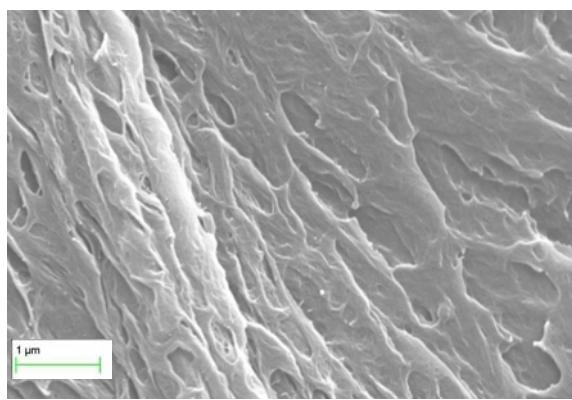
*m*-xylene



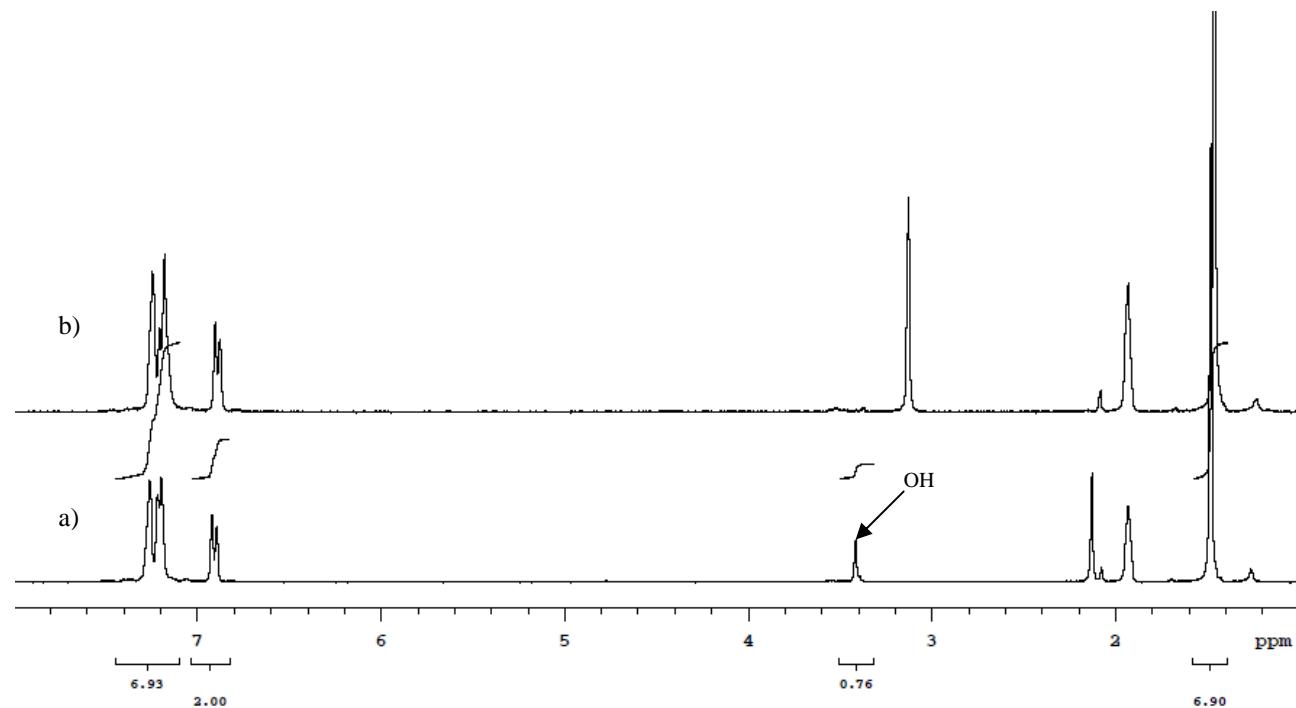
*p*-xylene



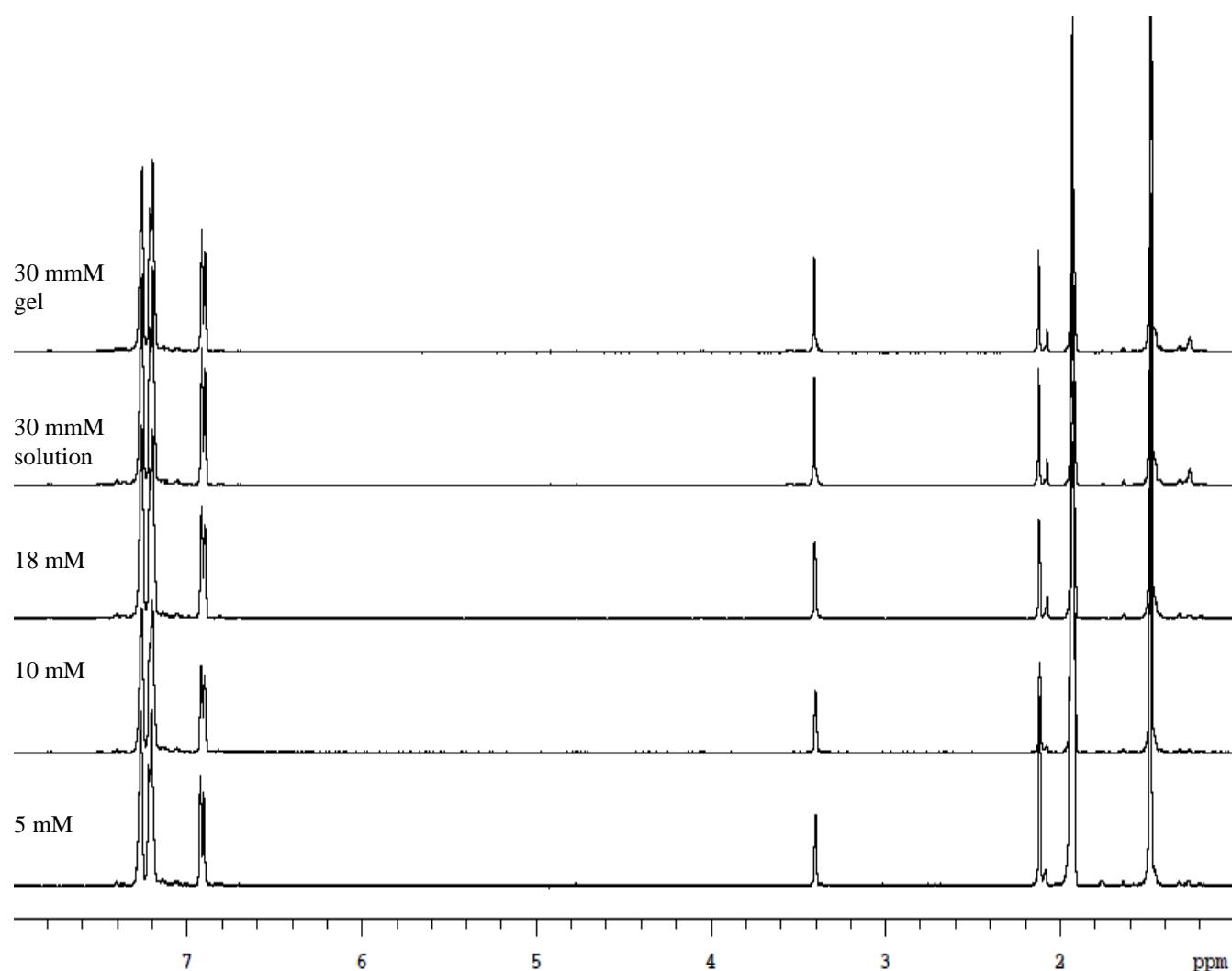
mesitilene



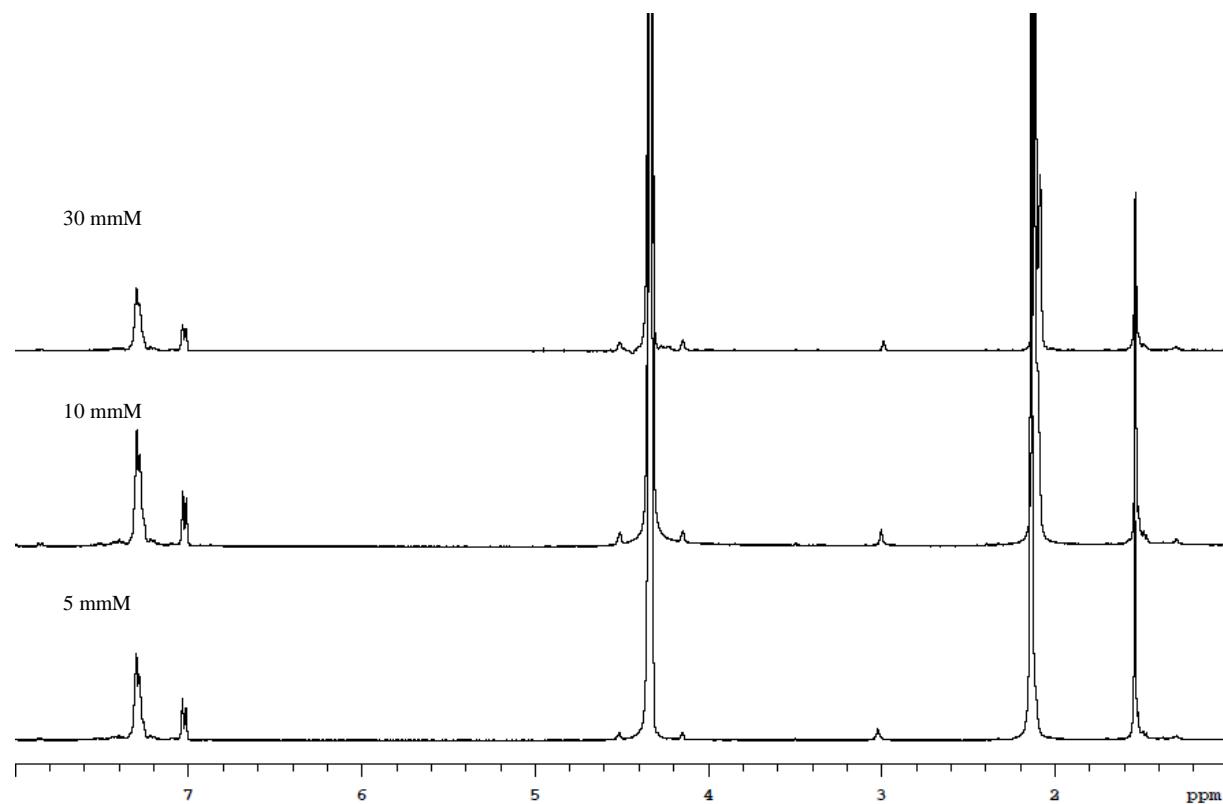
**Figure 5S:**  $^1\text{H}$  NMR spectrum of **1a** in  $\text{CD}_3\text{CN}$  (a) and  $\text{CD}_3\text{CN}+\text{D}_2\text{O}$  (b).



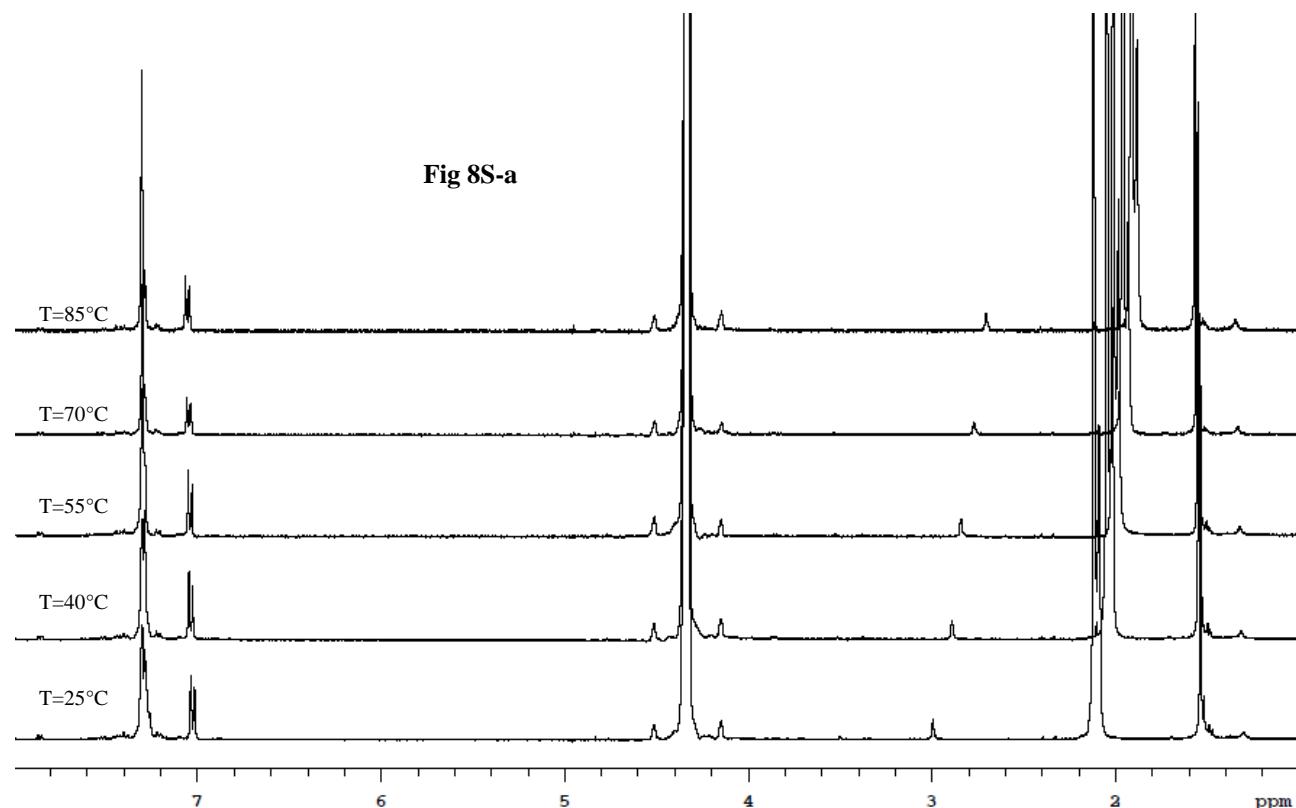
**Figure 6S:**  $^1\text{H}$  NMR spectrum of **1a** in  $\text{CD}_3\text{CN}$  at different concentrations at  $25^\circ\text{C}$



**Figure 7S:**  $^1\text{H}$  NMR spectrum of **1a** in  $\text{CD}_3\text{NO}_2$  at different concentrations at  $25^\circ\text{C}$



**Figure 8S-a, b, c:** VT-<sup>1</sup>H NMR spectra of **1a** in CD<sub>3</sub>NO<sub>2</sub> (10mM) ranging from 25 to 85°C



**Fig 8S-b**

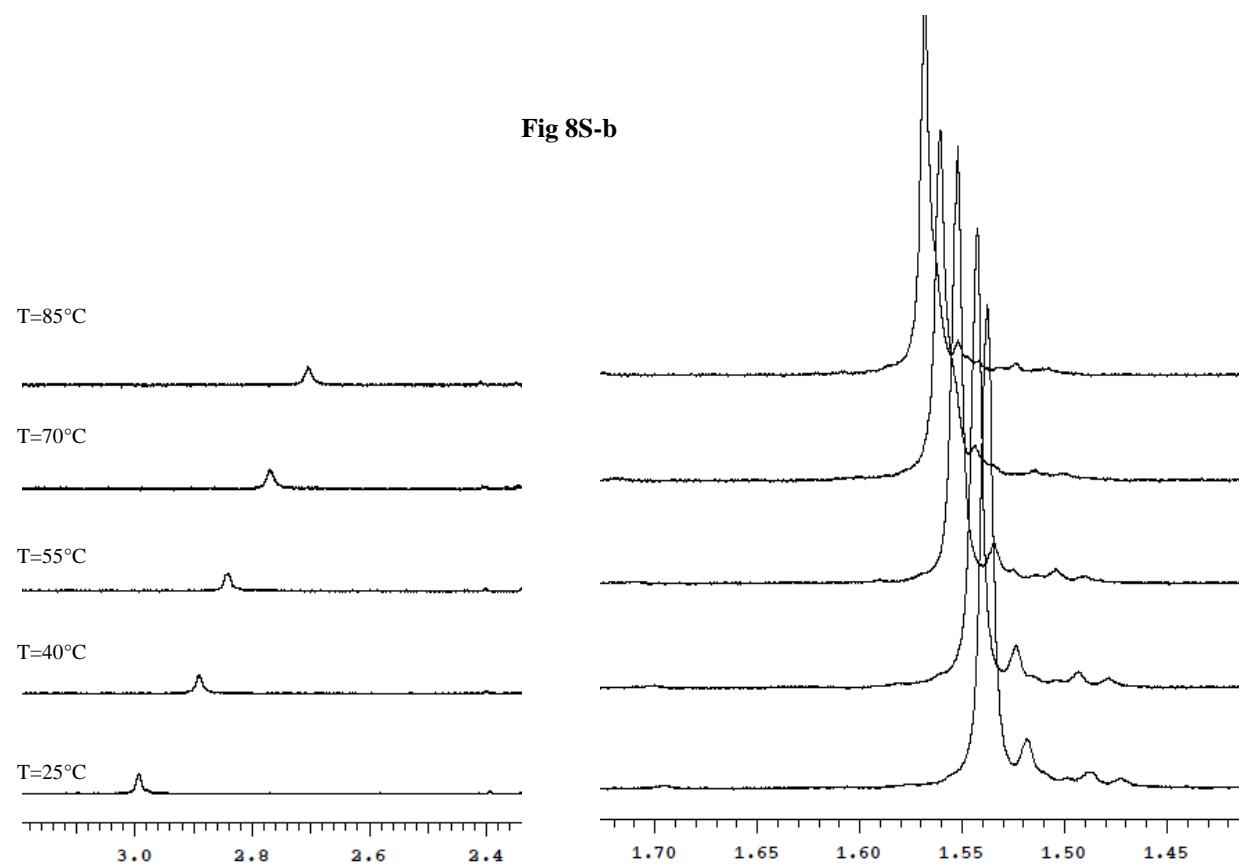


Fig 8S-c

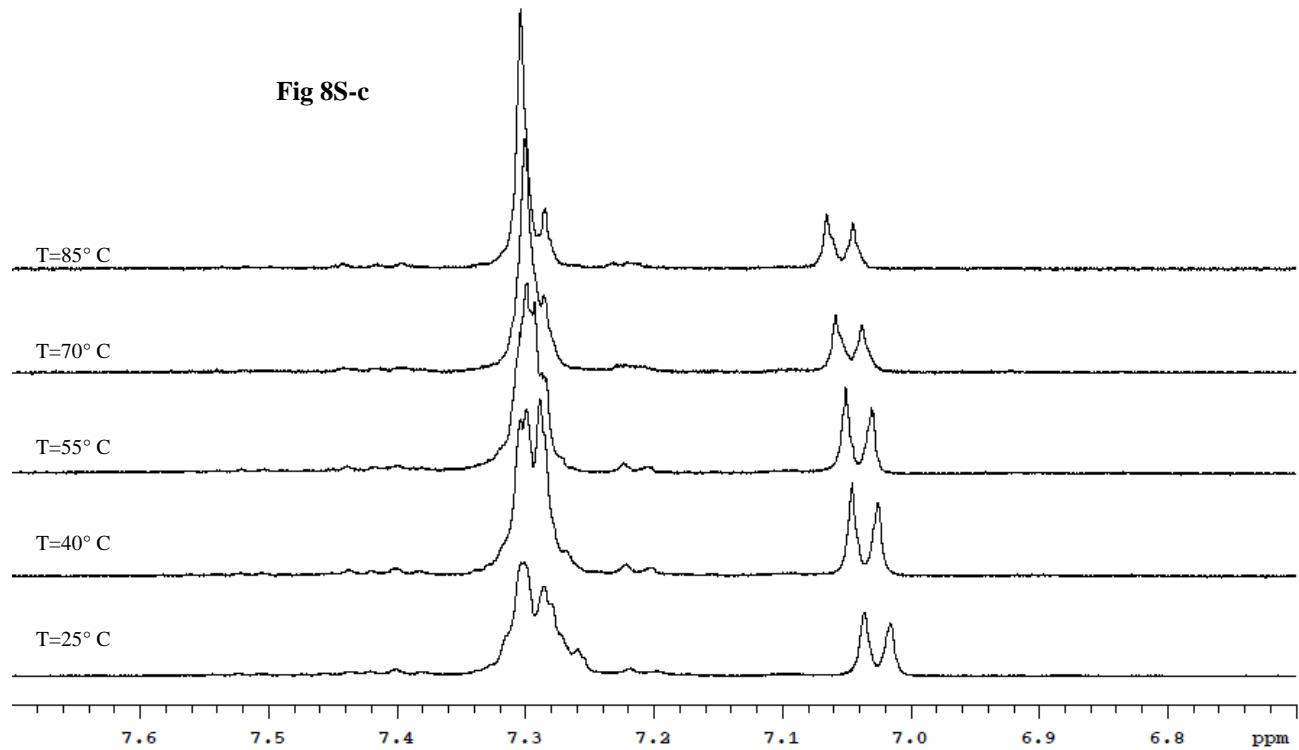
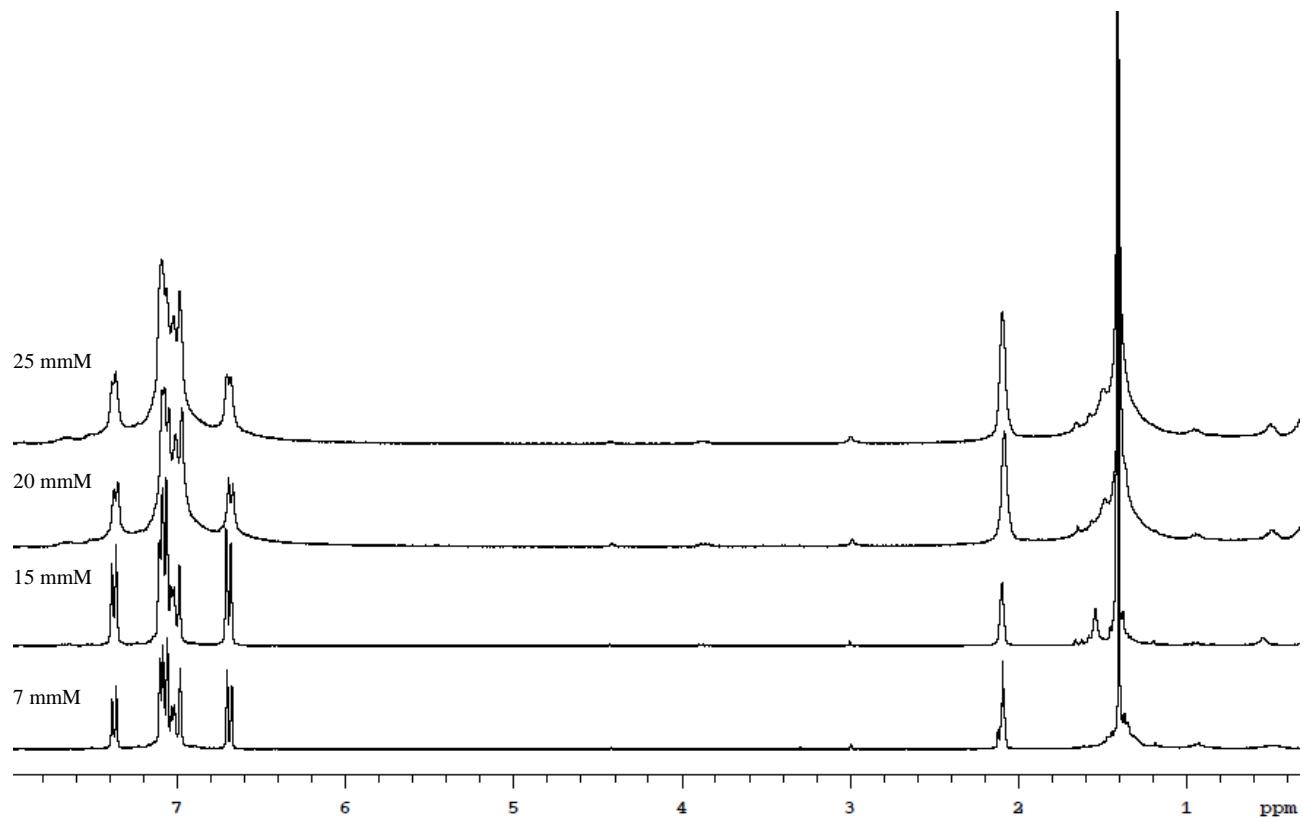
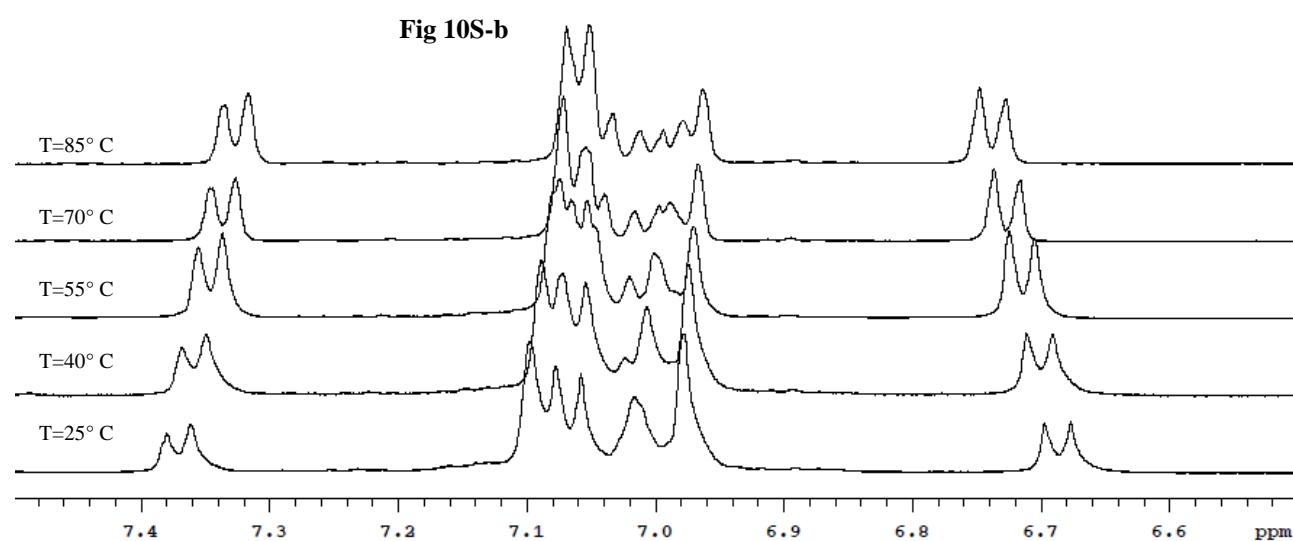
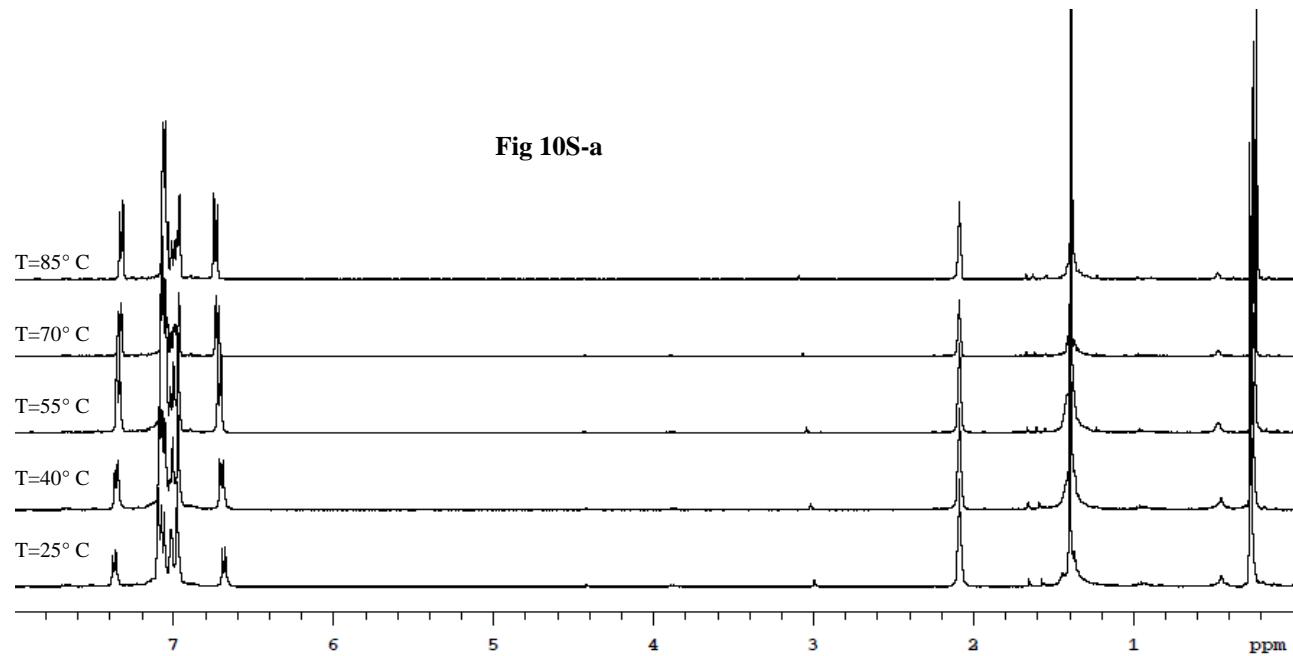
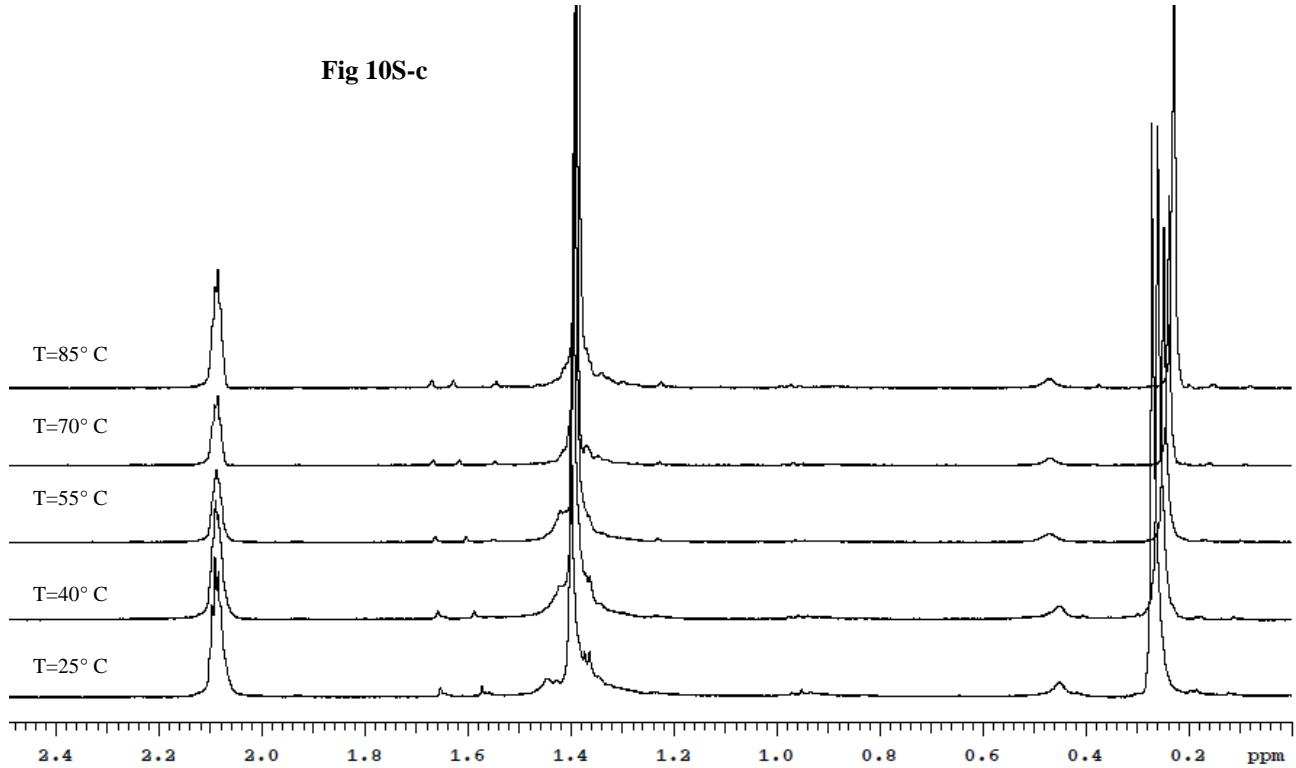


Figure 9S: <sup>1</sup>H NMR spectrum of **1a** in toluene at different concentrations at 25°C

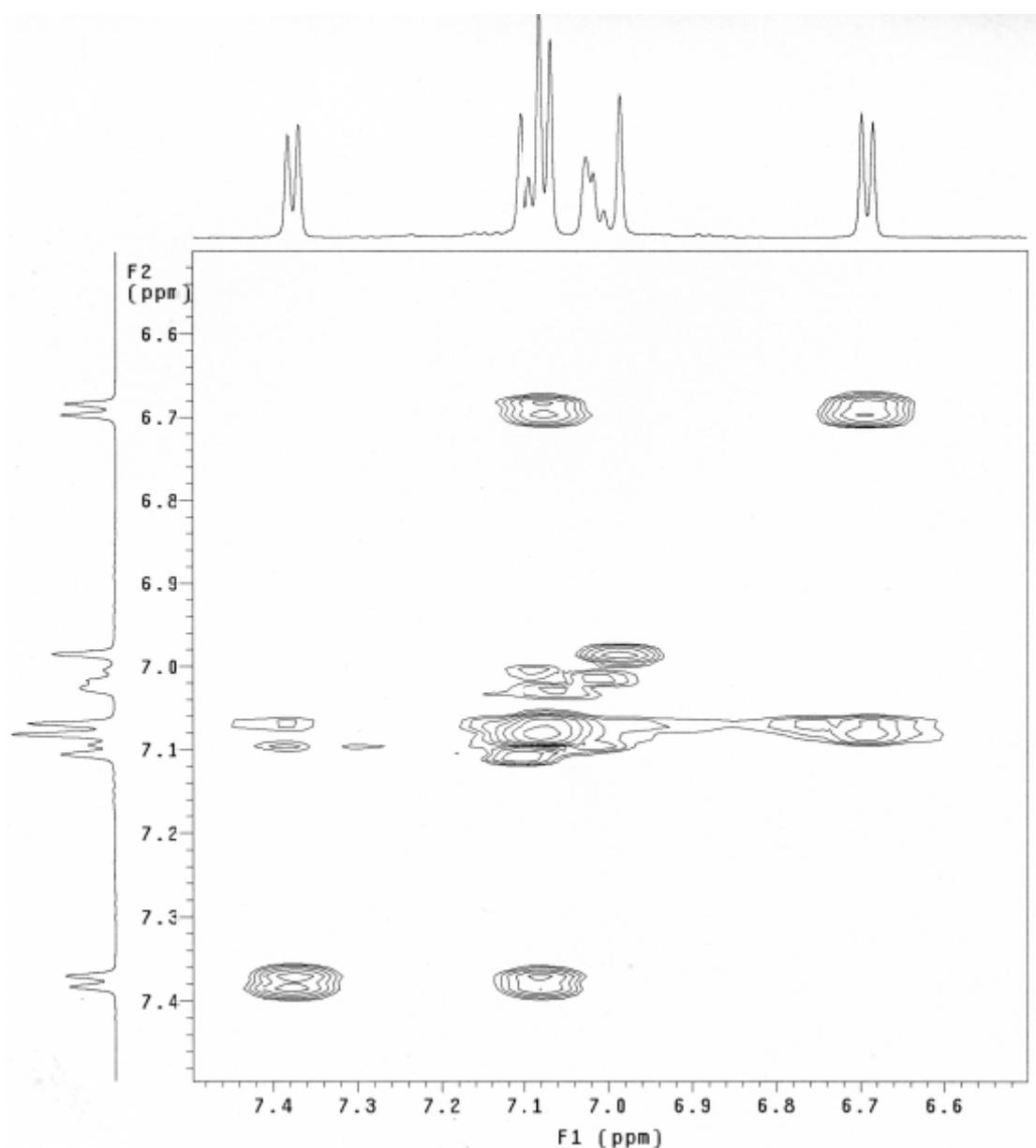


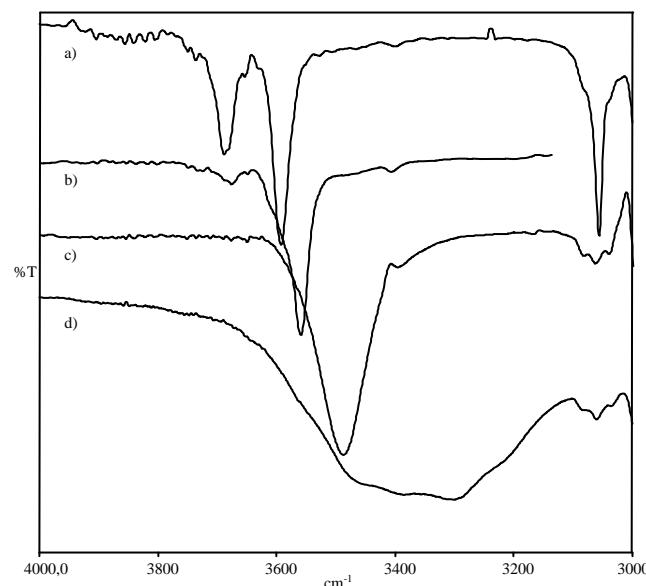
**Figure 10S-a, b, c:** VT-<sup>1</sup>H NMR spectra of **1a** in toluene (14.3 mM) ranging from 25 to 85°C





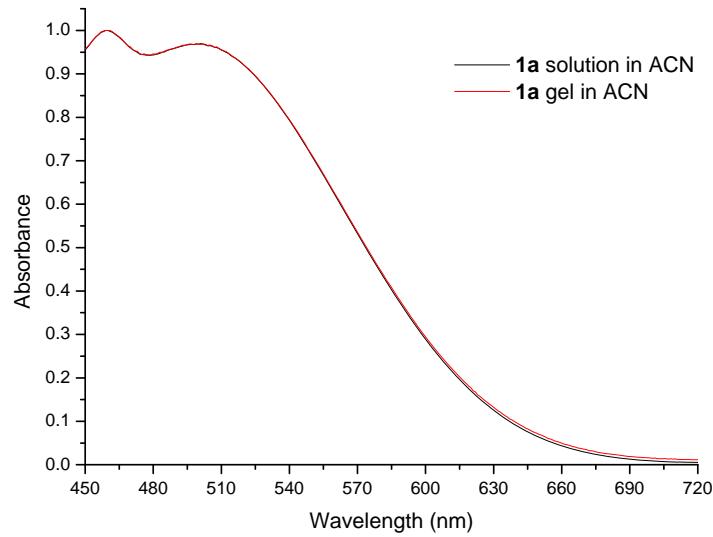
**Figure 11S:** Particular of COSY-NMR spectrum of **1a**-toluene gel (14.3 mM) at 25°C



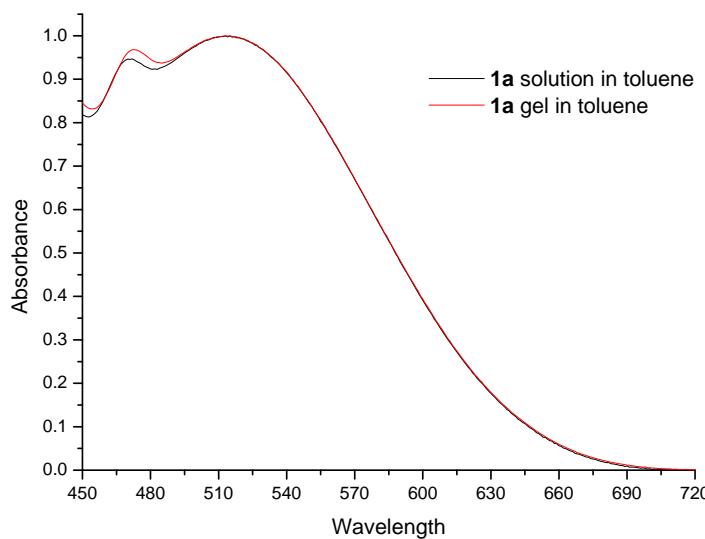


**Figure 12S.** FT-IR spectra of **1a**: a) in DCM solution, b) in *o*-xylene solution; c) in ACN solution and d) xerogel from *o*-xylene.

a)

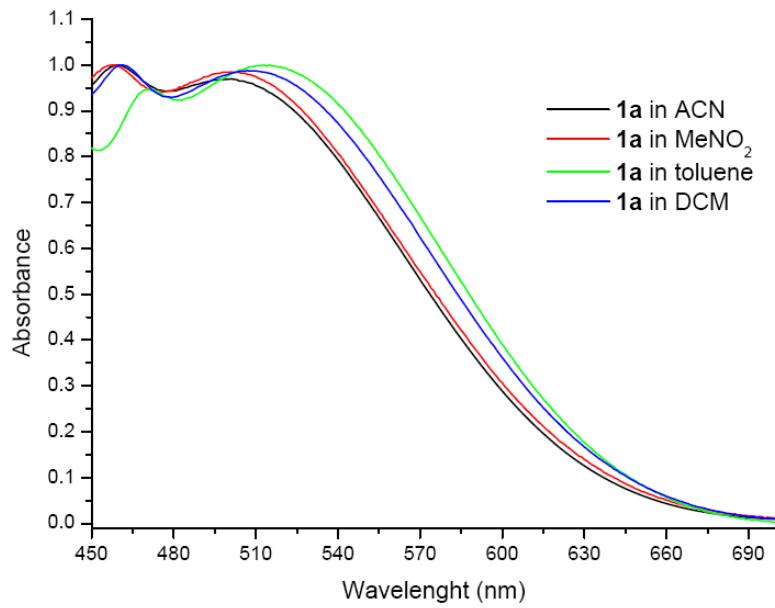


b)

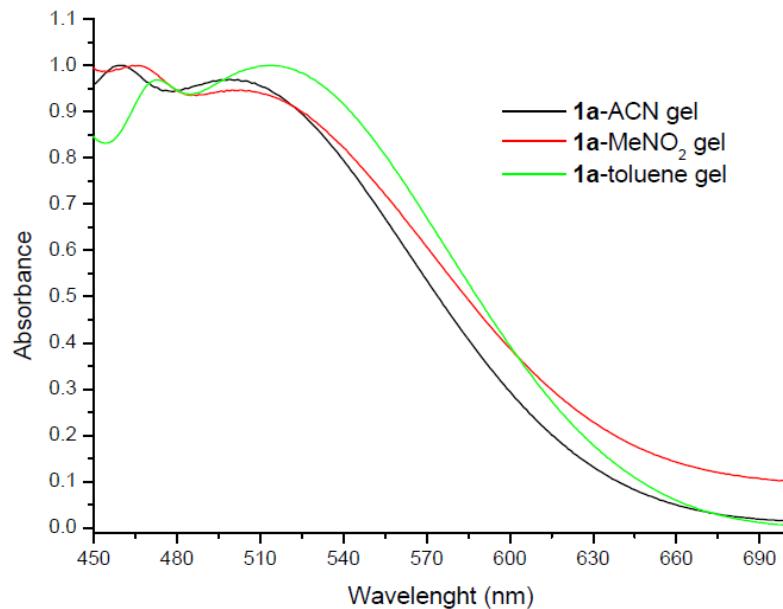


**Figure 13S:** Normalized UV/Vis absorption spectra at room temperature of **1a** in solution (1.0 mM) and in gel in different solvents. a) in ACN, **1a**-gel = 24.0 mM, b) in toluene, **1a**-gel = 12.6 mM.

a)



b)



**Figure 14S.** Normalized UV/Vis absorption spectra at room temperature. a) **1a**-solutions (1.0 mM) in ACN,  $\text{MeNO}_2$ , toluene and DCM; b) **1a**-gel in ACN (24.0 mM),  $\text{MeNO}_2$  (8.7 mM) and toluene (12.6 mM)