

Supporting Information

Proton-transfer Supramolecular Salts Resulted from 3,5-Dinitrobenzoic Acid and Aminomethyl Pyridine

Xue-Hua Ding, Lin-Fang Cui, Yong-Hua Li,* Shi Wang and Wei Huang*

Key Laboratory for Organic Electronics & Information Displays (KLOEID),
Institute of Advanced Materials (IAM), Nanjing University of Posts &
Telecommunications,
Nanjing 210046, China.

Fax: +86 25 85866396, +86 25 85866008

Tel: +86 25 85866396, +86 25 85866008

E-mail: iamyhli@njupt.edu.cn, wei-huang@njupt.edu.cn

Table 1. Crystallographic data for the complex of **1–3**

Compounds	1	2	3
Formula	C ₁₃ H ₁₂ N ₄ O ₆	C ₁₃ H ₁₂ N ₄ O ₆	C ₂₆ H ₂₄ N ₈ O ₁₂
Formula weight	320.27	320.27	640.53
Crystal system	monoclinic	orthorhombic	triclinic
Space group	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ 2 ₁ 2 ₁	<i>P</i> -1
<i>a</i> (Å)	6.5494 (13)	6.8770 (14)	7.4234 (15)
<i>b</i> (Å)	9.1960 (18)	9.2116 (18)	11.738 (2)
<i>c</i> (Å)	23.512 (5)	22.056 (4)	17.698 (4)

α (°)	90°	90°	102.99 (3)°
β (°)	90.39 (3)°	90°	100.52 (3)°
γ (°)	90°	90°	101.26 (3)°
V (Å ³)	1416.0 (5)	1397.2 (5)	1431.8 (5)
D_x (Mg m ⁻³)	1.502	1.522	1.486
μ (mm ⁻¹)	0.12	0.12	0.12
Z	4	4	2
T (K)	293	293	293
$F(000)$	664	664	664
θ range for data collection (°)	3.2–27.5	3.1–27.5	3.0–27.5
Index ranges	$-8 \leq h \leq 8,$ $-11 \leq k \leq 11,$ $-30 \leq l \leq 30$	$-8 \leq h \leq 8,$ $-11 \leq k \leq 11,$ $-28 \leq l \leq 28$	$-9 \leq h \leq 9,$ $-15 \leq k \leq 15,$ $-22 \leq l \leq 22$
Measured reflections	14201	14505	14831
Independent reflections	3246	3190	6511
Data/restraints/parameters	3246/0/208	3190/0/208	6511/0/416
R_{int}	0.045	0.041	0.063
Reflections with $I > 2\sigma(I)$	2348	2831	3674
$R_1 [I > 2\sigma(I)]$	0.055	0.039	0.062
wR_2 (all data)	0.148	0.092	0.167
S	1.09	1.08	1.03

Table 2. Hydrogen Bond Distances and Parameters for the complex of **1–3** (Å, °)

D–H...A	D–H	H...A	D...A	D–H...A
Compound 1				
N4–H4A...O2 ⁱ	0.89	1.93	2.810(2)	172
N4–H4B...O1 ⁱⁱ	0.89	2.52	3.374(2)	162
N4–H4B...O2 ⁱⁱ	0.89	2.15	2.908(2)	142

N4–H4C···O1 ⁱⁱⁱ	0.89	1.93	2.776(2)	157
C9–H9A···O3 ^{iv}	0.93	2.57	3.149(3)	121
Compound 2				
N4–H4A···O2 ^{iv}	0.89	1.98	2.7635(19)	146
N4–H4B···O1 ^v	0.89	1.93	2.8028(19)	167
N4–H4C···O1	0.89	1.95	2.8323(19)	170
C13–H13B···O4 ^{vi}	0.97	2.57	3.236(2)	126
Compound 3				
N6–H6A···N7 ^{iv}	0.89	2.02	2.900	171
N6–H6B···O1	0.89	1.80	2.668	166
N6–H6C···O5 ^{vii}	0.89	2.59	3.080	116
N6–H6C···N5 ^{viii}	0.89	2.51	3.139	129
N8–H8C···O7 ^{ix}	0.89	2.48	2.973	115
N8–H8C···O8 ^{ix}	0.89	2.04	2.921	169
N8–H8A···O7	0.89	1.88	2.750	167
N8–H8B···O2	0.89	1.95	2.757	151
C15–H15A···O1 ^{viii}	0.93	2.58	3.419	150
C22–H22A···O11 ^x	0.93	2.55	3.282	136
C26–H26B···O7 ^{ix}	0.97	2.57	3.168	120

Symmetry codes: (i) $x-1, y+1, z$; (ii) $-x+1, -y+1, -z$; (iii) $x, y+1, z$; (iv) $x-1, y, z$; (v) $x-1/2, -y+3/2, -z+1$; (vi) $-x, y-1/2, -z+3/2$; (vii) $x-1, y-1, z$; (viii) $-x+1, -y, -z+1$; (ix) $-x, -y, -z$; (x) $x, y-1, z$.

Table 3. Summary of Mass (m/g) and Moles (n/mmol) Used of Acid and Amine for the Synthesis and Elemental analyses results of Compounds **1-3**

Compound	Amine (m, n)	Acid (m, n)	Elemental analyses results
1	2-aminomethyl pyridine (0.029, 0.24)	3,5-dinitrobenzoic acid (0.047, 0.22)	Anal. Calc. For C ₁₃ H ₁₂ N ₄ O ₆ : C, 48.75; H, 3.78; N, 17.49. Found: C, 48.90; H, 4.00; N, 17.27 %.
2	3-aminomethyl pyridine (0.013, 0.11)	3,5-dinitrobenzoic acid (0.025, 0.12)	Anal. Calc. For C ₁₃ H ₁₂ N ₄ O ₆ : C, 48.75; H, 3.78; N, 17.49. Found: C, 48.92; H, 3.88; N, 17.37 %.
3	4-aminomethyl pyridine (0.038, 0.32)	3,5-dinitrobenzoic acid (0.064, 0.30)	Anal. Calc. For C ₁₃ H ₁₂ N ₄ O ₆ : C, 48.75; H, 3.78; N, 17.49. Found: C, 48.60; H, 4.01; N, 17.40 %.