

SUPPORTING INFORMATION

1D coordination polymer built on asymmetric $\mu_{1,1,3}$ -azide bridge: from unusual topology to magnetic properties and Cu(II)/Cu(I) redox reversibility

Jean-Bernard Tommasino, Guillaume Chastanet, Boris Le Guennic, Vincent Robert, and Guillaume Pilet

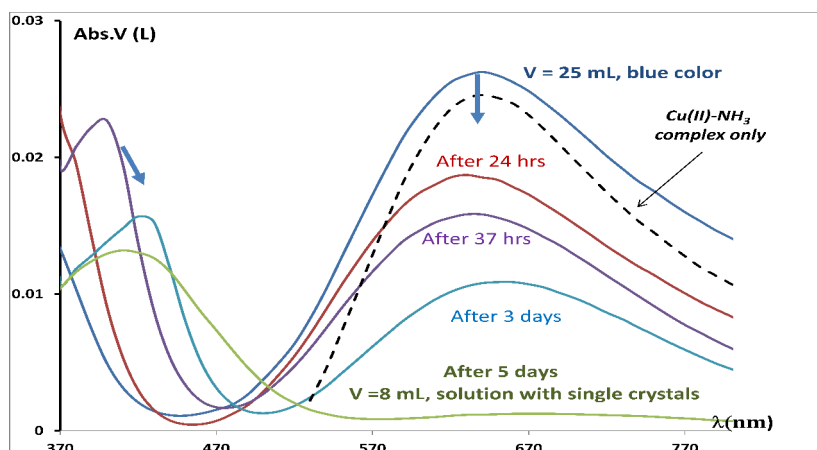


Figure S1 : Visible spectra during the formation of azide-Cu(II) complex (mixture of NaN_3 , Cu(II) salt, excess of NH_3 , H_2O 25% in methanol): solid line : spectra at different times. The green spectrum is identical at solid complex solubilized in solvent. Dashed line: Cu(II)- NH_3 complex in methanol.

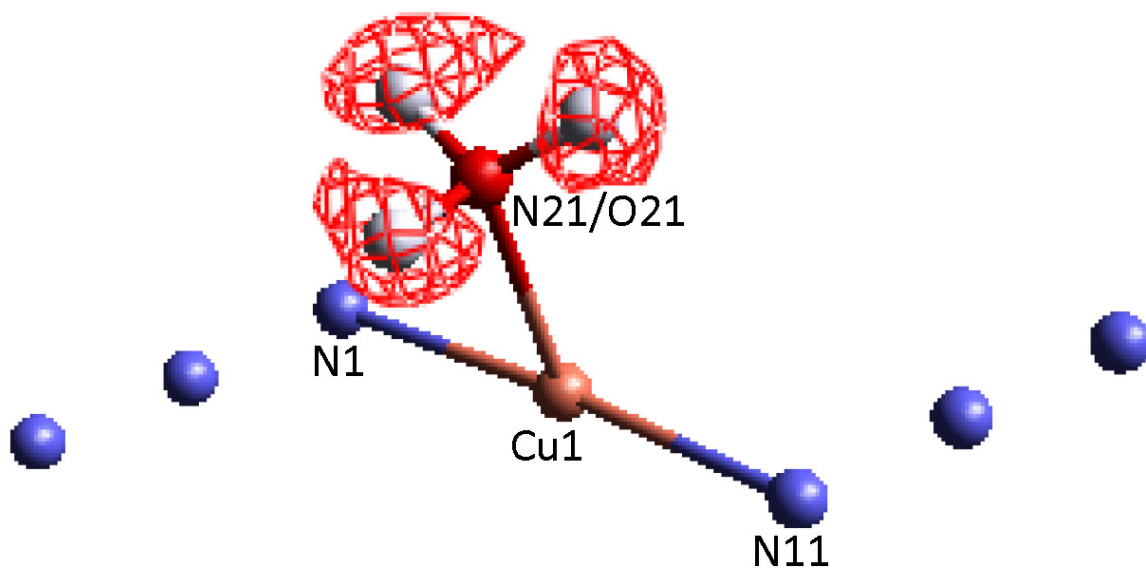


Figure S2: difference-map plot (occupancy of represented hydrogen atoms is equal to 0 in that case) showing (i) three potential positions for hydrogen atoms belonging to coordinated solvent molecules ($\text{H}_2\text{O}/\text{NH}_3$ statistical disorder) and (ii) that reported hydrogen atoms sites are reasonable. Red lines represent the positive electronic residual