

# Support Information

# **Amorphous manganese oxide-coated montmorillonite as an efficient catalyst for water oxidation**

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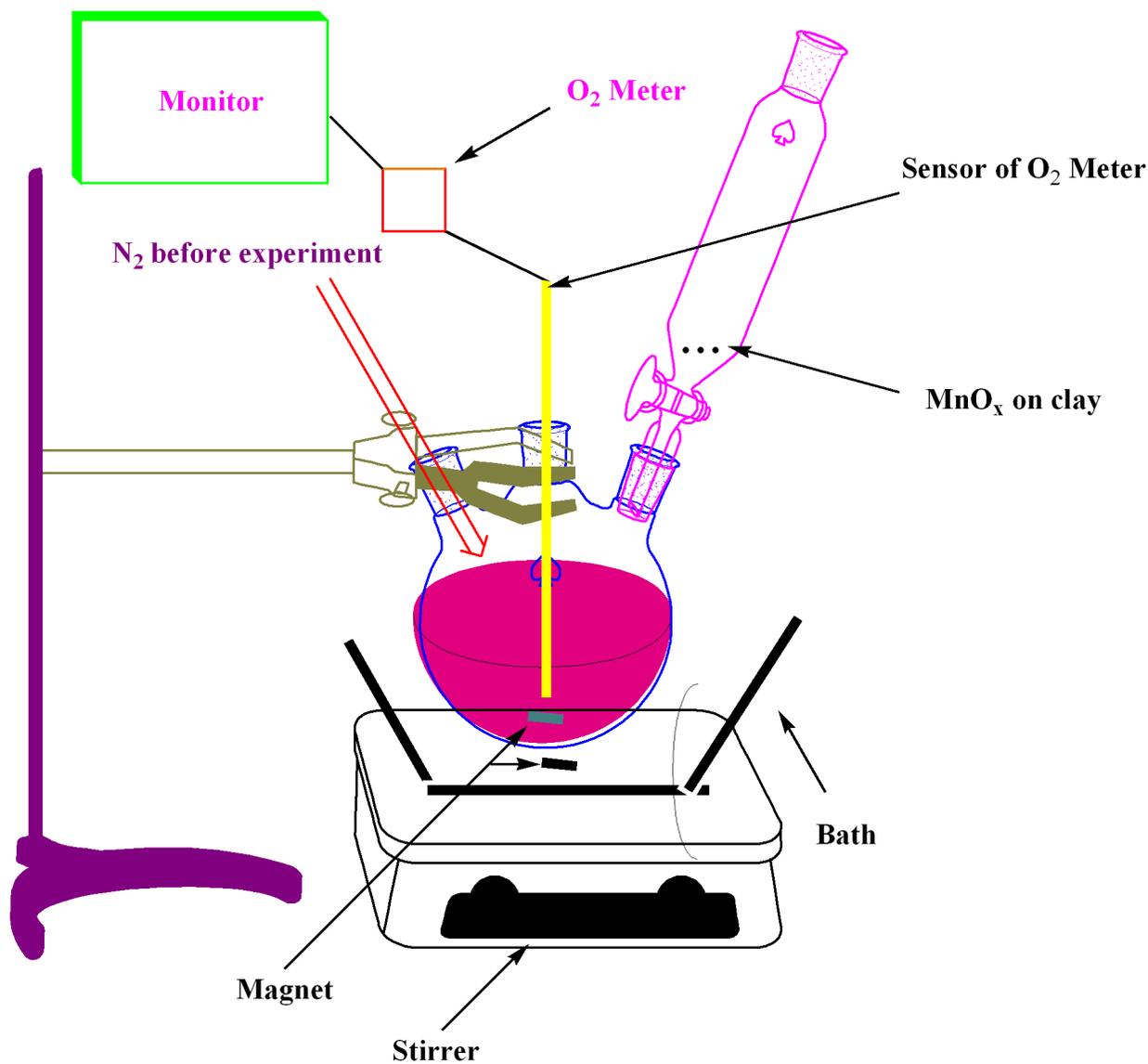
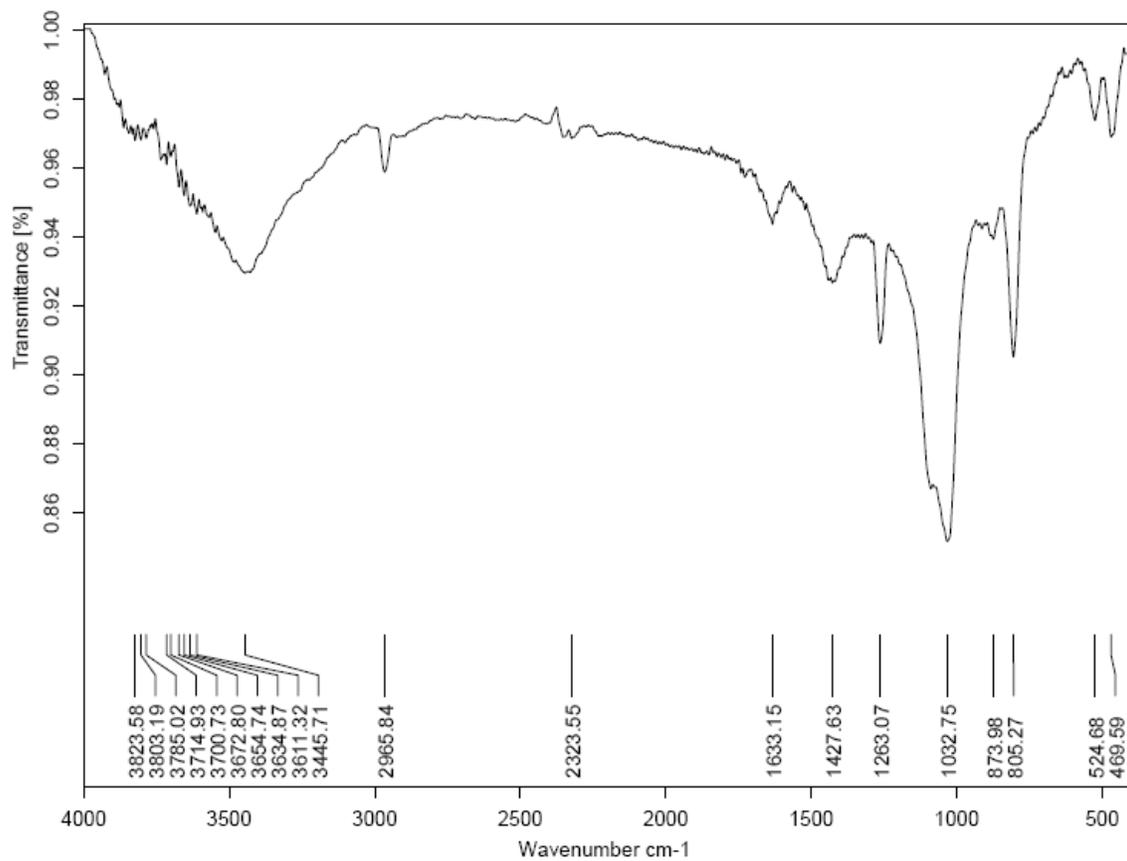
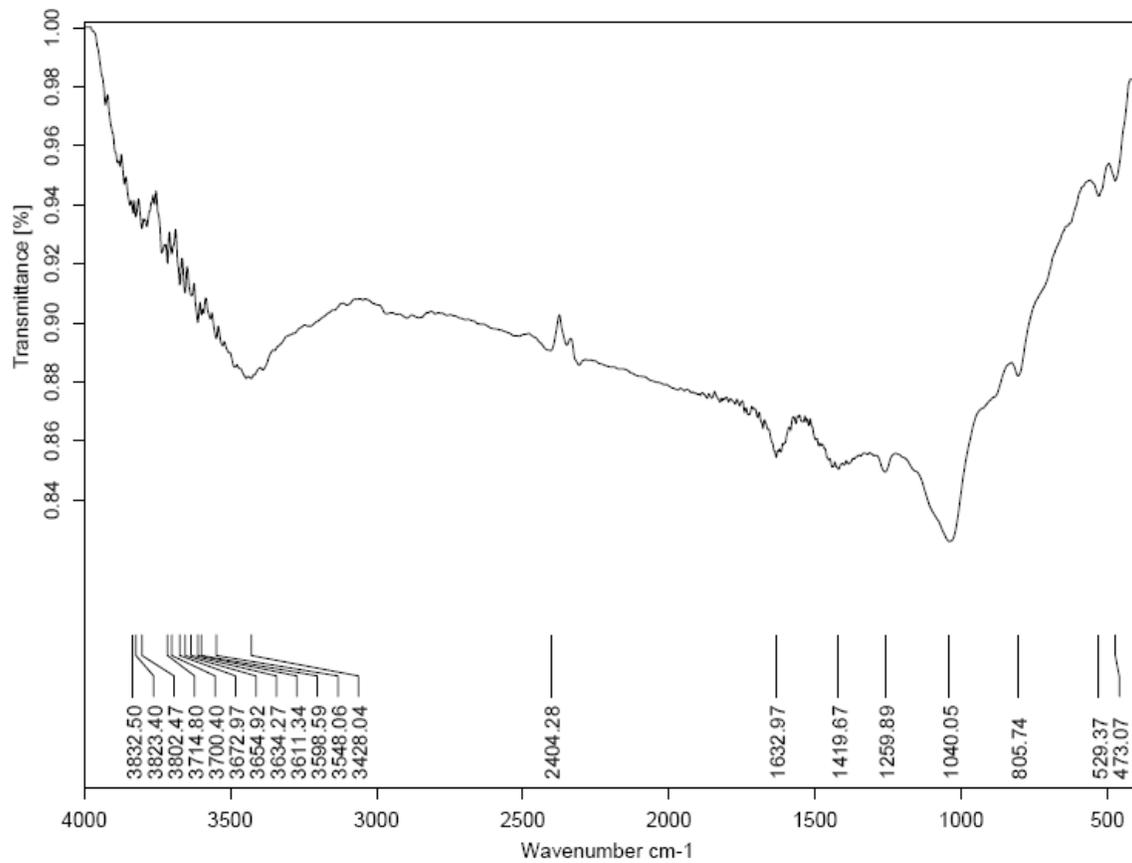


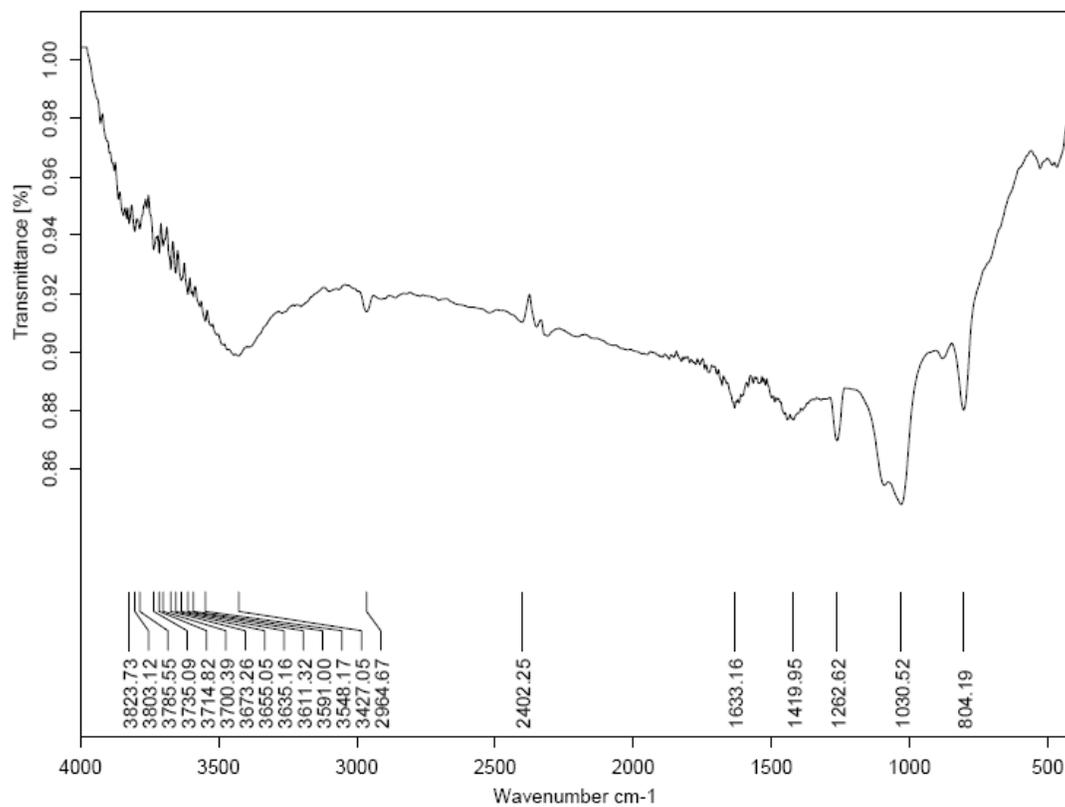
Fig. S1. The reactor set-up for oxygen evolution experiment from aqueous solution in the presence of  $(\text{NH}_4)_2\text{Ce}(\text{NO}_3)_6$  (Ce(IV)) and catalyst.



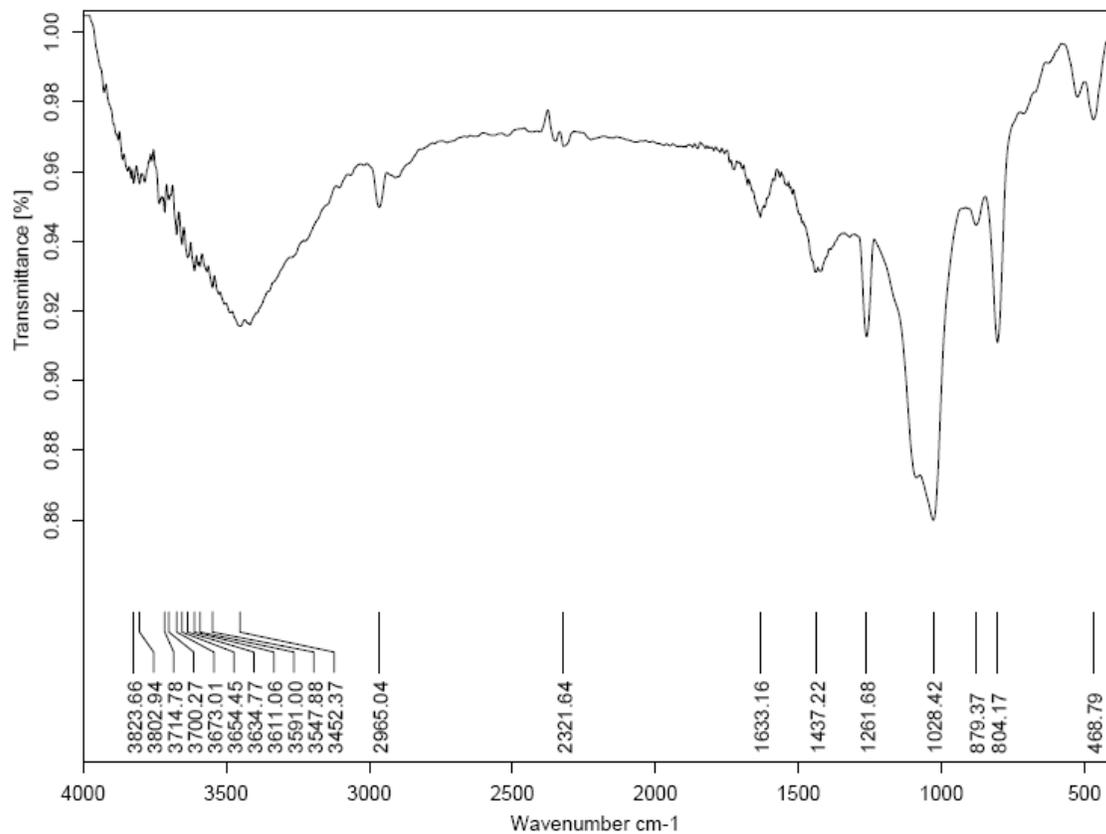
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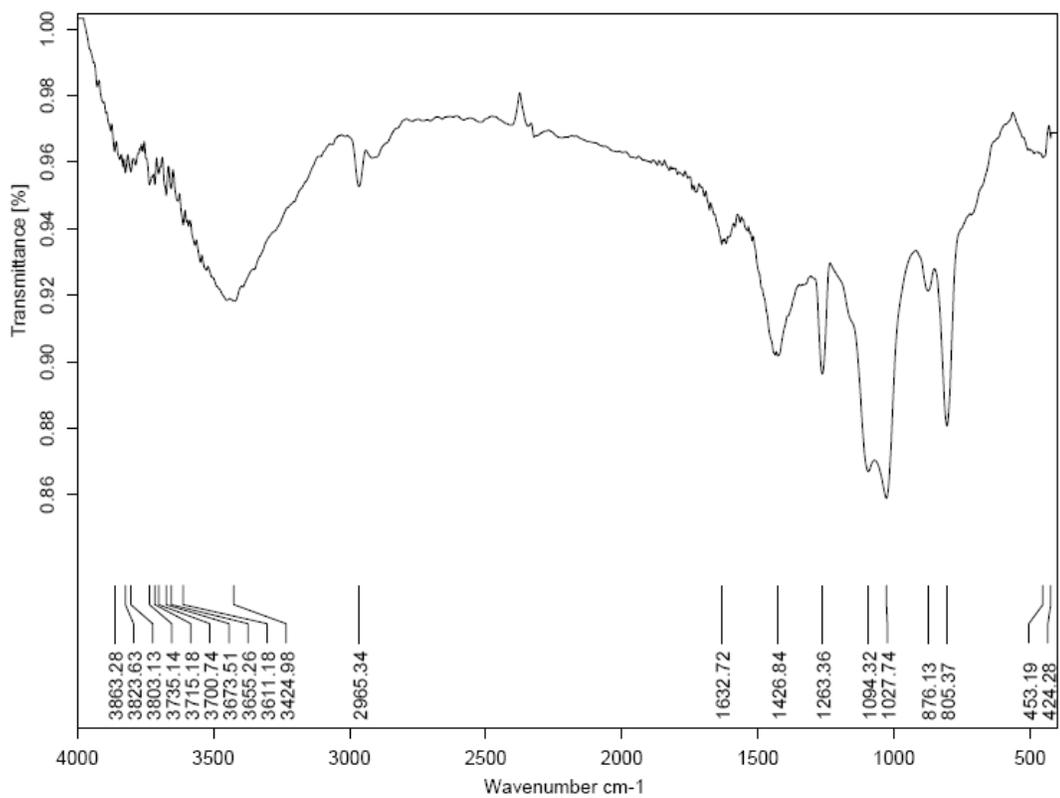


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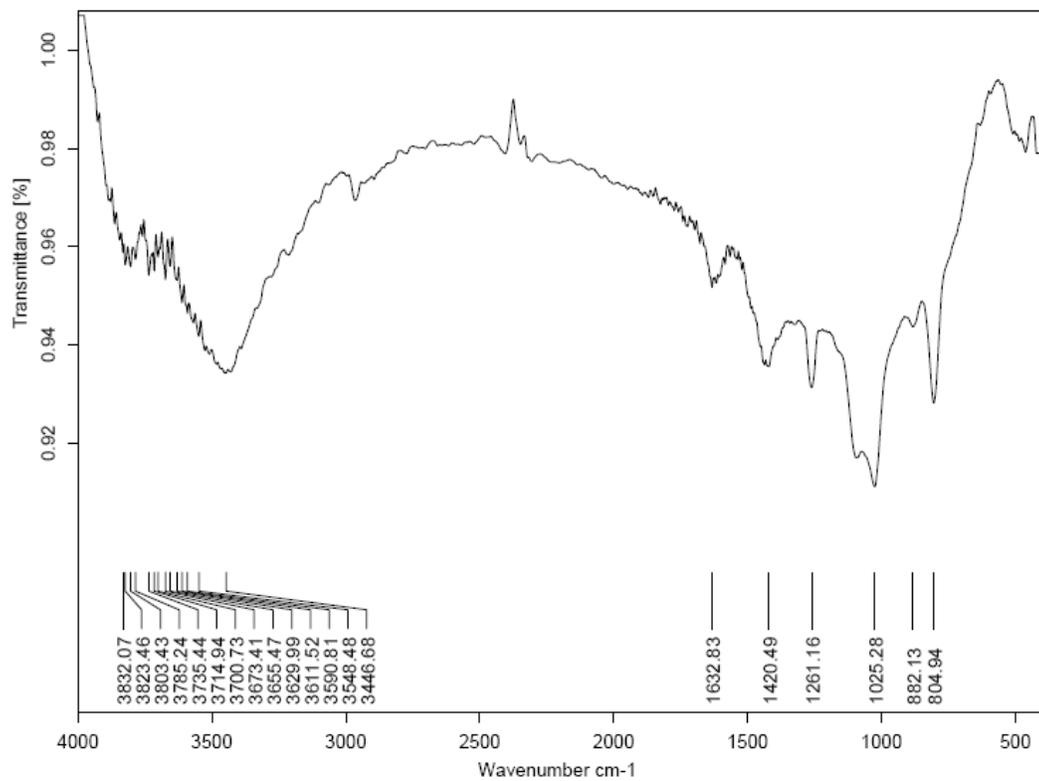


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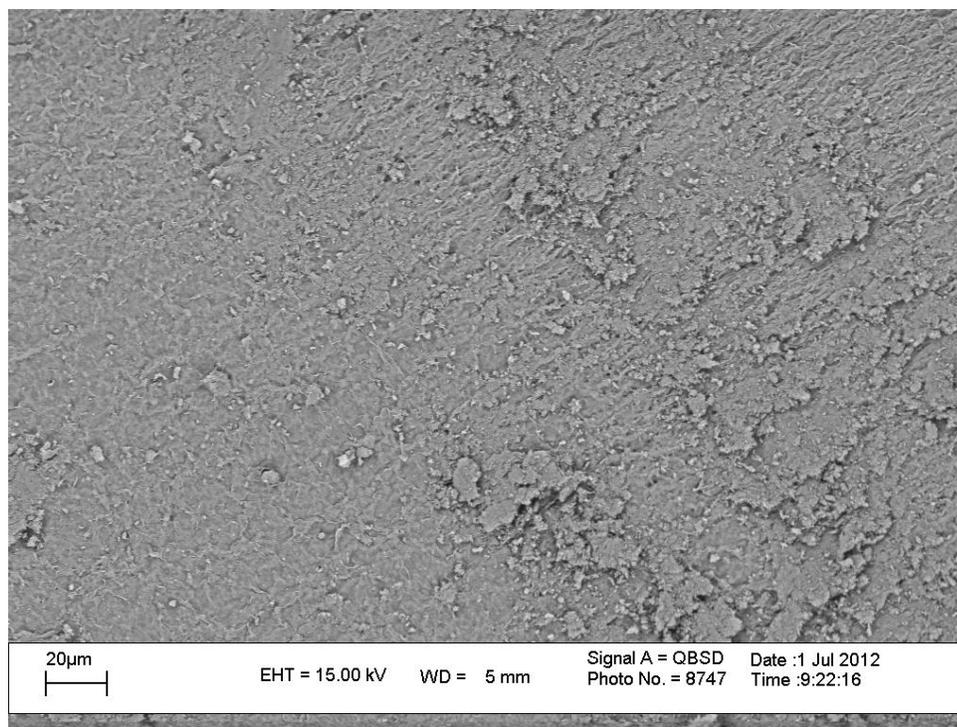


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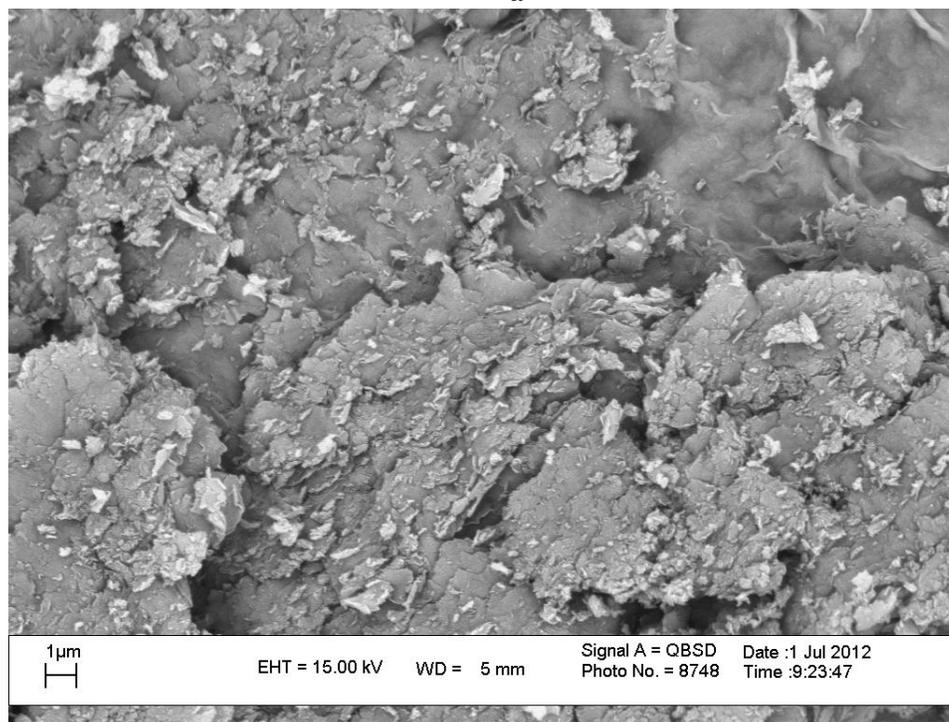


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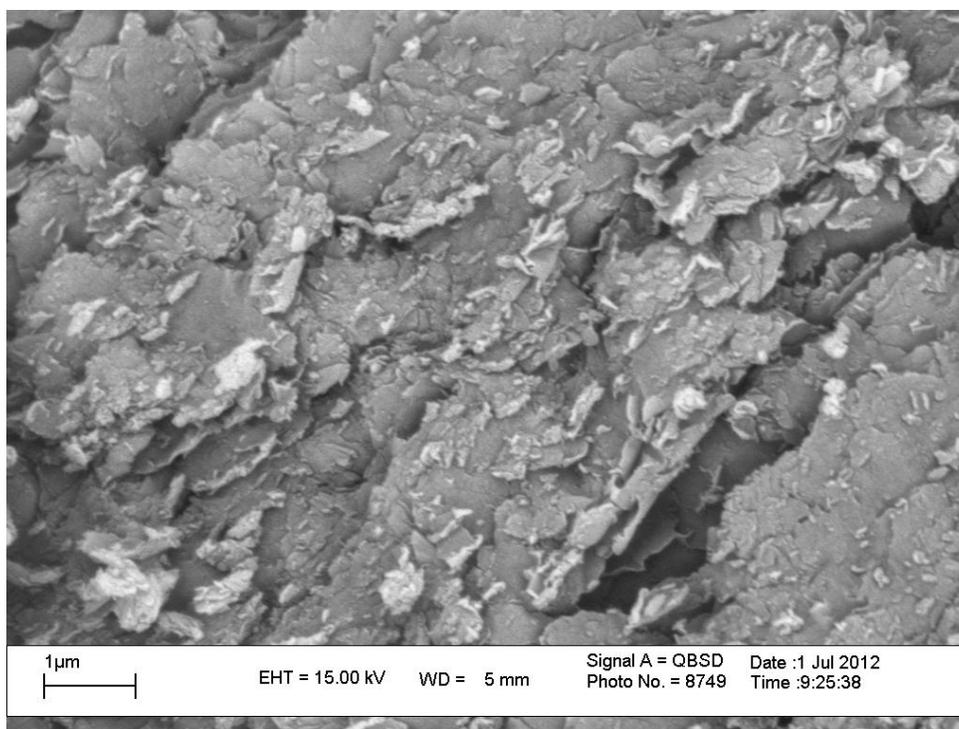
Fig. S2 IR spectra of the Amorphous manganese oxide-coated montmorillonite, clay (a), clay with manganese content of 3.0 wt % (b) and clay with manganese content of 5.0 % prepared at 100 (c), 300 (d), 400 (e) and 700 °C (f).



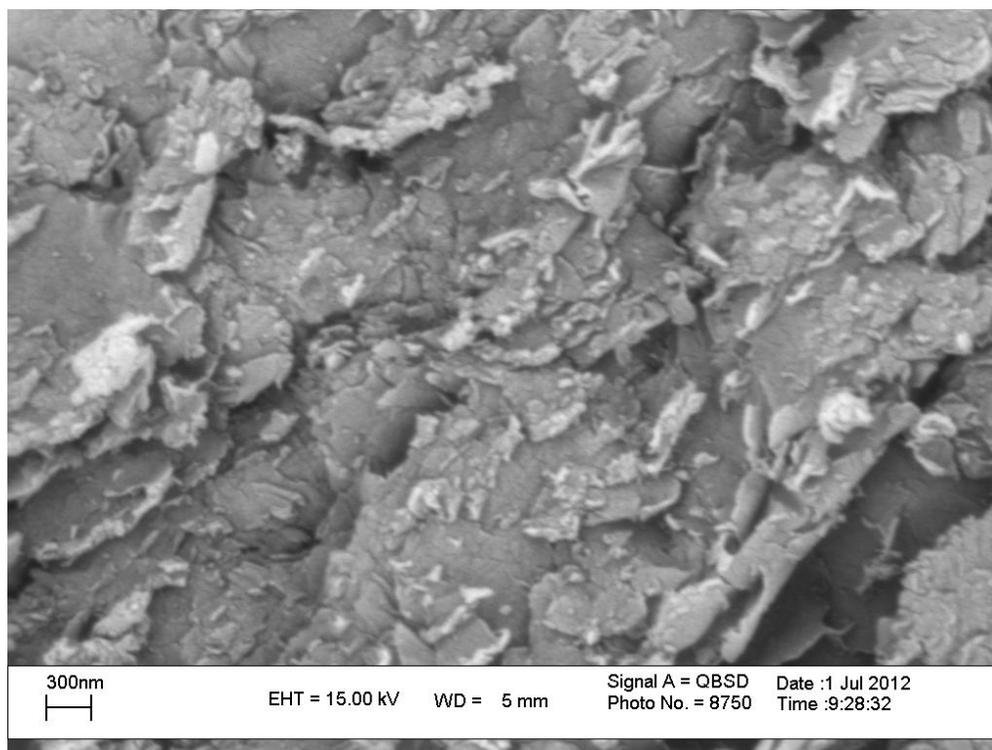
a



b

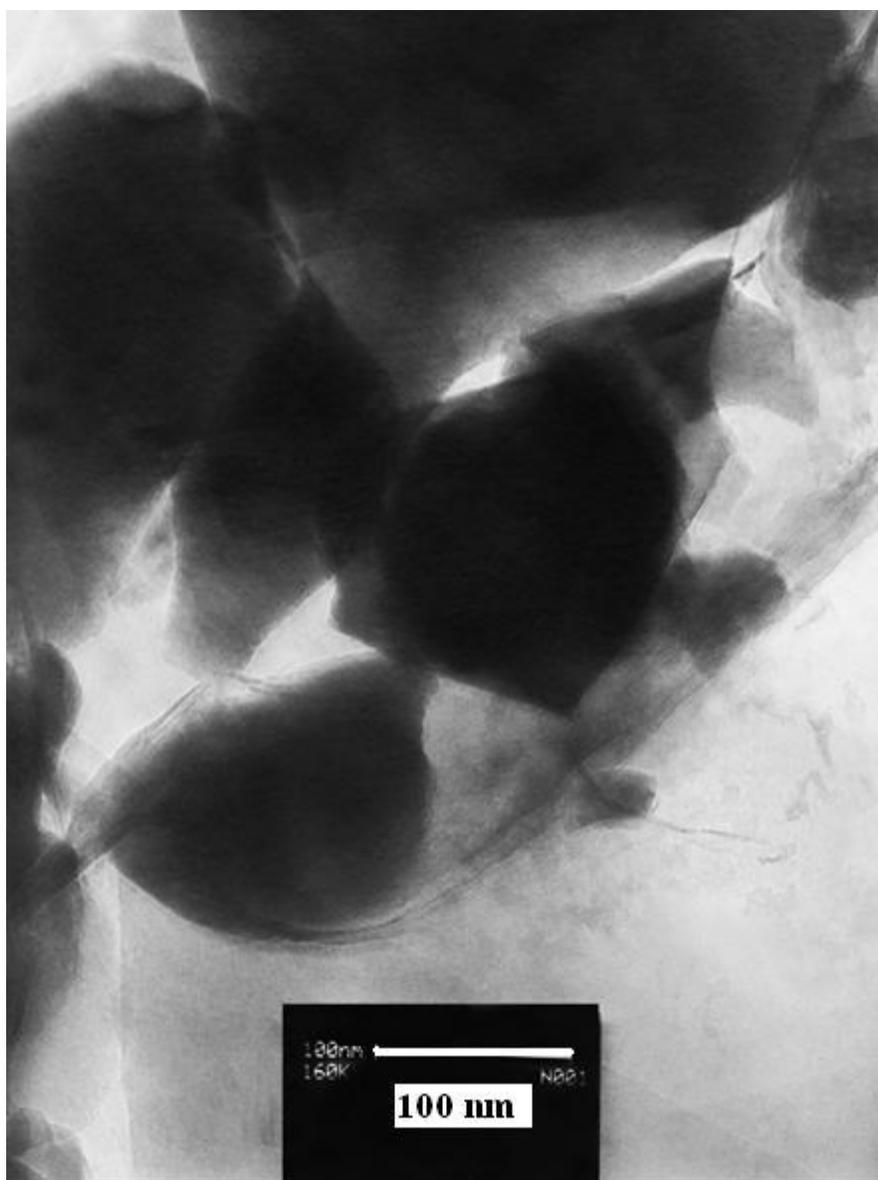


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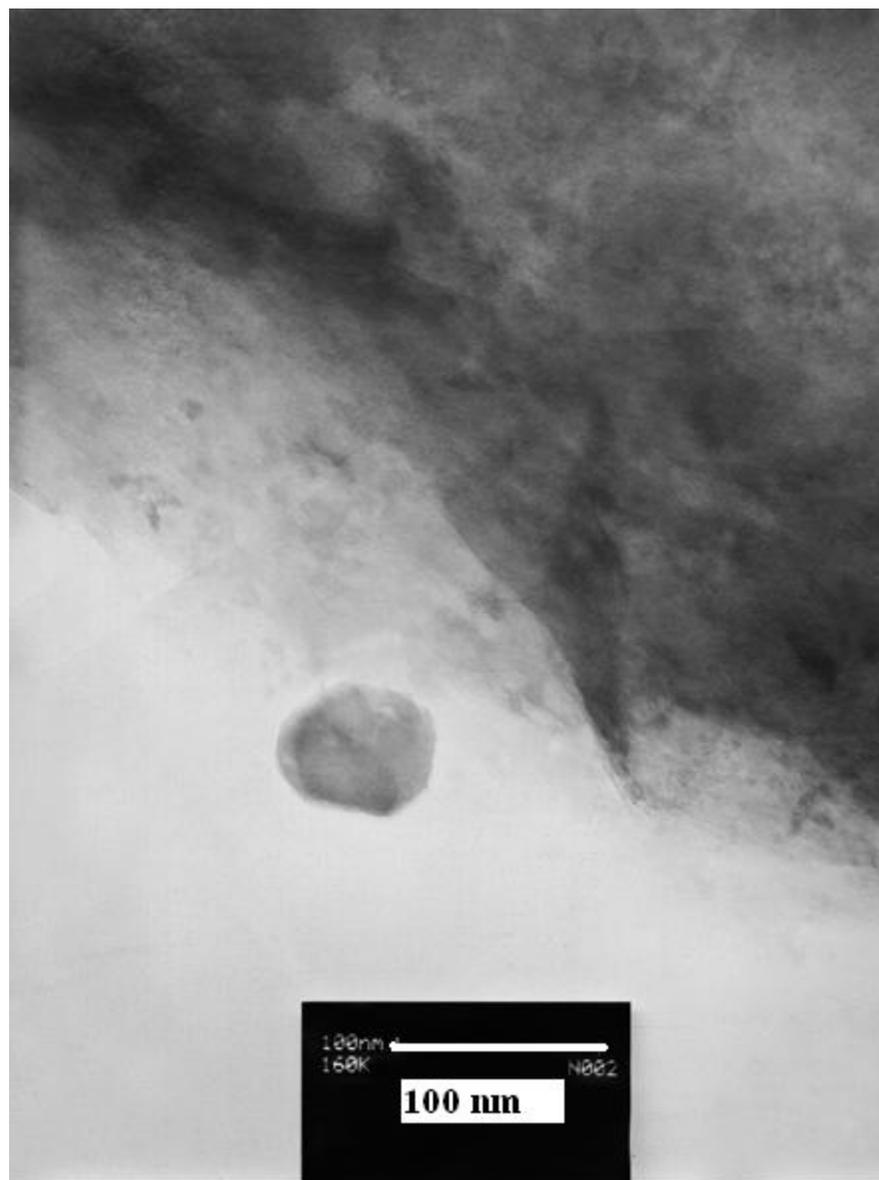


d

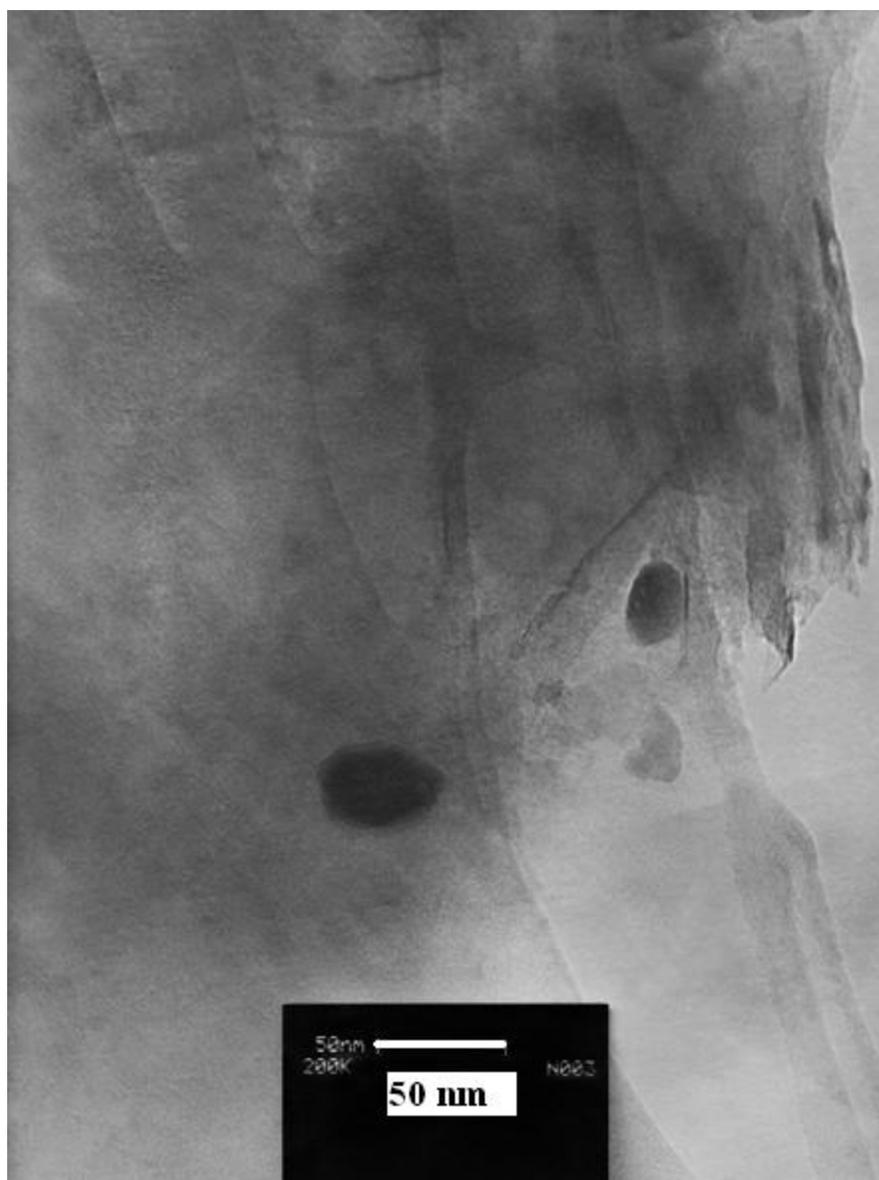
Fig. S3 SEM images of manganese oxides on clay with manganese content of 5.0 % (a-d).



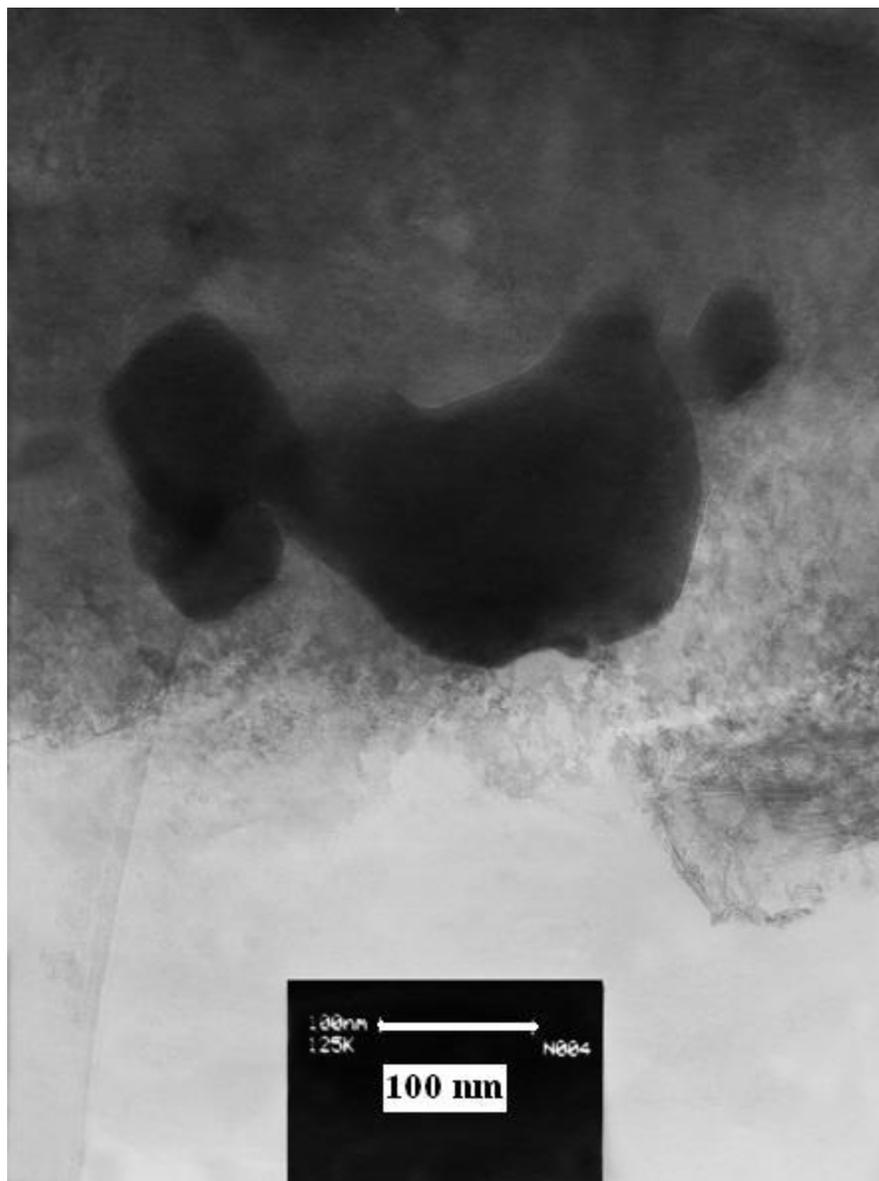
a



b



c



d

Fig. S4. TEM images of manganese oxides on clay with manganese content of manganese: 5% (a-d).

Table S1 The rate of water oxidation by the various manganese oxides as catalysts for water oxidation.

Compound	Oxidant	TOF <sup>a</sup>	References
Nano scale manganese oxide within NaY zeolite	Ce(IV)	2.62	1
Layered manganese-calcium oxide	Ce(IV)	2.2	2
Layered manganese-aluminium or zinc oxide	Ce(IV)	1.1	3
CaMn <sub>2</sub> O <sub>4</sub> ·H <sub>2</sub> O	Ce(IV)	0.54	4
Amorphous Manganese Oxides	Ru(bpy) <sub>3</sub> <sup>3+</sup>	0.06	5
	Ce(IV)	0.52	
CaMn <sub>2</sub> O <sub>4</sub> ·4H <sub>2</sub> O	Ce(IV)	0.32	4
Mn oxide nanoclusters	Ru(bpy) <sub>3</sub> <sup>3+</sup>	0.28	6
Manganese oxide-coated montmorillonite	Ce(IV)	0.22	This Work
Nano-sized α-Mn <sub>2</sub> O <sub>3</sub>	Ce(IV)	0.15	7
Octahedral Molecular Sieves	Ru(bpy) <sub>3</sub> <sup>3+</sup>	0.11	5
	Ce(IV)	0.05	
MnO <sub>2</sub> (colloid)	Ce(IV)	0.09	8
α-MnO <sub>2</sub> nanowires	Ru(bpy) <sub>3</sub> <sup>3+</sup>	0.059	9
CaMn <sub>3</sub> O <sub>6</sub>	Ce(IV)	0.046	10
CaMn <sub>4</sub> O <sub>8</sub>	Ce(IV)	0.035	10
α-MnO <sub>2</sub> nanotubes	Ru(bpy) <sub>3</sub> <sup>3+</sup>	0.035	9
Mn <sub>2</sub> O <sub>3</sub>	Ce(IV)	0.027	4
β-MnO <sub>2</sub> nanowires	Ru(bpy) <sub>3</sub> <sup>3+</sup>	0.02	9
Ca <sub>2</sub> Mn <sub>3</sub> O <sub>8</sub>	Ce(IV)	0.016	11
CaMnO <sub>3</sub>	Ce(IV)	0.012	11
Nano-sized λ-MnO <sub>2</sub>	Ru(bpy) <sub>3</sub> <sup>3+</sup>	0.03	12
Bulk α-MnO <sub>2</sub>	Ru(bpy) <sub>3</sub> <sup>3+</sup>	0.01	9
Mn Complexes	Ce(IV)	0.01-0.6	12-14
PSII	Sunlight	100-400 × 10 <sup>3</sup>	15

a: mmol O<sub>2</sub>/ mol Mn

References:

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