Electronic Supplementary Information for:

Linearly π -extended squaraine dyes enable the spectral response of dye-sensitized solar cells in the NIR region over 800 nm

Takeshi Maeda,*^{*a*} Shigeki Arikawa, ^{*a*} Hidekazu Nakao,^{*a*} Shigeyuki Yagi, ^{*a*} Hiroyuki Nakazumi*^{*a*}

^{*a*} Department of Applied Chemistry, Graduate School of Engineering, Osaka Prefecture University, Naka-ku, Sakai 599-8531, Japan

E-mail: tmaeda@chem.osakafu-u.ac.jp, nakazumi@chem.osakafu-u.ac.jp

1. Synthesis

Preparation of compound 1

4.4 3,4-Dihydroxy-3-cyclobutene-1,2-dione (0.50 g, mmol) and 1-heptyl-5-iodo-2,3,3-trimethyl-3H-indolium iodide (4.72 g, 9.23 mmol) were dissolved in a mixture of 1-butanol (4 mL) and benzene (1 mL). Then a catalytic amount of quinoline was added and the solution was heated under reflux for 6 h. After cooling, the solvent was removed on a rotary evaporator, and the residue was purified by silica gel column chromatography (eluent; CHCl₃/hexane/EtOAc, 7/2/1, v/v/v). Compound 1 was precipitated from CHCl₃/hexane solution as a green crystal (3.0 g, 82%). ¹H NMR (CDCl₃, 400 MHz): δ 7.60–7.57 (m, 4H), 6.72 (d, J = 8.2 Hz, 2H), 5.92 (s, 2H), 3.98-3.81 (m, 4H), 1.74 (m, 16H), 1.37–1.24 (m, 16H), 0.85 (t, J = 6.9 Hz, 6H). MALDI TOF-MS: m/z calcd for $[M(C_{40}H_{50}I_2N_2O_2) + H]^+$, 845.20; found 845.46. IR (KBr, cm⁻¹): 2952, 2920, 1601, 1499, 1452, 1342, 1275, 1171, 1084, 1042. Anal.Calcd. for C₄₀H₅₀I₂N₂O₂; C, 56.88; H, 5.97; N, 3.32 %. Found: C, 56.86; H, 5.64; N, 3.18 %.

Preparation of compound 3

In a two-necked round-bottom flask equipped with a condenser, **1** (0.30 g, 0.36 mmol), Pd(PPh₃)₄ (12 mg, 0.010 mmol), and CuI (2.1 mg, 0.011 mmol) were dissolved in tetrahydrofuran (10 mL) under an N₂ atmosphere. To the solution was added 3-(1-methyl)ethyloxy-4-(tributylstannyl)cyclobut-3-ene-1,2-dione (**2**) (0.23 g, 0.53 mmol), and the mixture was stirred at 50 °C for 27 h. After cooling, the solvent was removed at reduced pressure, and the residue was purified by silica gel column chromatography (eluent; CHCl₃/EtOAc, 9/1, v/v). Compound **3** was precipitated from CHCl₃/hexane solution as a green crystal (0.12 g, 38%). ¹H NMR (CDCl₃, 400 MHz): δ 8.06 (d, *J* = 8.3 Hz, 1H), 7.97 (s, 1H), 7.66–7.63 (m, 2H), 7.04 (d, *J* = 8.4 Hz, 1H), 6.80 (d, *J* = 8.1 Hz, 1H), 6.80 (d, *J* = 8.4 Hz, 1H), 6.03 (s, 2H) 5.66–5.60 (m, 1H), 4.15–3.88 (m, 4H), 1.82–1.79 (m, 16H), 1.58 (d, *J* = 6.2 Hz, 6H), 1.38–1.28 (m, 16H), 0.88 (t, *J* = 6.6 Hz, 6H). MALDI TOF-MS: m/z calcd for [M(C₄₇H₅₇IN₂O₅]⁺, 856.33; found 855.99. IR (KBr, cm⁻¹): 2953, 2918, 1782, 1738, 1601, 1504, 1391, 1344, 1273, 1171, 1082, 1042. Anal.Calcd. for C₄₇H₅₇IN₂O₅: C, 65.88; H, 6.70; N, 3.27 %. Found: C, 65.70; H, 6.52; N, 3.15 %.

Preparation of compound 4

Compound **3** (0.40 g, 0.46 mmol) was dissolved in a mixture of THF–3.6% HCl aq. (THF/HCl aq., 10/1(v/v), 37 mL) and then stirred for 20 h at 60 °C. After removal of solvent, the residue was washed with water and a mixture of CHCl₃ and hexane (CHCl₃/hexane, 1/2, v/v), and dried *in vacuo* to give **4** as a purple solid (0.37 g, 97%), which was used in the next step without further purification. ¹H NMR (DMSO-*d*₆, 400 MHz): δ 8.02–7.98 (m, 2H), 7.87 (s, 1H), 7.64 (d, *J* = 9.8 Hz, 1H), 7.40 (d, *J* = 8.4 Hz, 1H), 7.14 (d, *J* = 8.5 Hz, 1H), 5.81 (s, 1H), 5.78 (s, 1H), 4.07–4.02 (m, 4H), 1.66–1.64 (m, 16H), 1.32–1.26 (m, 16H), 0.81 (t, *J* = 6.8 Hz, 6H). MALDI TOF-MS: m/z calcd for [M(C₄₄H₅₁IN₂O₅)]⁺, 814.28; found 814.98. IR (KBr, cm⁻¹): 2953, 2926, 1765, 1599, 1580, 1502, 1400, 1344, 1273, 1169, 1105, 1086, 1041.

Preparation of compound 6a

Compound **4** (0.13 g, 0.17 mmol) and 1-butyl-2,3,3-trimethyl-3*H*-indolium iodide (**5a**) (68 mg, 0.20 mmol) were dissolved in a mixture of 1-butanol (6.0 mL) and benzene (1.5 mL). Then a catalytic amount of quinoline was added and the solution was heated at 65 °C for 18 h. After cooling, the solvent was removed on a rotary evaporator, and the residue was purified by silica gel column chromatography (eluent; CHCl₃/acetone, 4/1, v/v). Compound **6a** was precipitated from CHCl₃/hexane solution as a dark purple crystal (0.13 g, 75%). ¹H NMR (DMSO-*d*₆, 400 MHz): δ 8.13 (d, *J* = 8.8 Hz, 1H), 8.11 (s, 1H), 7.94 (s, 1H), 7.78–7.73 (m, 2H), 7.71 (d, *J* = 8.4 Hz, 1H), 7.53 (t, *J* = 7.8 Hz, 1H), 7.49–7.46 (m, 2H), 7.23 (d, *J* = 8.4 Hz, 1H), 6.30 (s, 1H), 5.86 (s, 2H), 4.51–4.40 (m, 2H), 4.18–4.01 (m, 4H), 1.79–1.68 (m, 24H), 1.45–1.25 (m, 18H), 0.95 (t, *J* = 7.3 Hz, 3H), 0.85 (t, *J* = 6.7 Hz, 6H). MALDI TOF-MS: m/z calcd for [M(C₅₉H₇₀IN₃O₄)]⁺, 1011.44; found 1011.79. IR (KBr, cm⁻¹): 2953, 2926, 1742, 1612, 1558, 1493, 1394, 1342, 1319, 1265, 1167, 1082, 1038. Anal.Calcd. for C₅₉H₇₀IN₃O₄: C, 70.02; H, 6.97; N, 4.15 %. Found: C, 69.94; H, 7.01; N, 4.08 %.

Preparation of compound 6b

Compound 4 (0.15 g, 0.18 mmol) and 3-butyl-2-methyl-1,3-benzothiazol-3-ium iodide (**5b**) (74 mg, 0.22 mmol) were dissolved in a mixture of 1-butanol (3.7 mL) and benzene (0.9 mL). Then a catalytic amount of quinoline was added and the solution was heated at 90 °C for 4.5 h. After cooling, the solvent was removed on a rotary evaporator, and the residue was purified

by silica gel column chromatography (eluent; CHCl₃/EtOAc/CH₃OH, 40/10/1, v/v/v). Compound **6b** was precipitated from CHCl₃/hexane solution as a dark purple crystal (90 mg, 49%). ¹H NMR (CDCl₃, 400 MHz): δ 8.25–8.20 (m, 2H), 7.83 (d, J = 8.3 Hz, 1H), 7.64–7.58 (m, 3H), 7.53–7.50 (m, 2H), 7.02 (d, J = 8.3 Hz, 1H), 6.73 (d, J = 8.2 Hz, 1H), 6.44 (s, 1H), 6.01 (s, 1H), 5.96 (s, 1H), 4.41 (t, J = 7.9 Hz, 2H), 4.08–3.85 (m, 4H), 1.92–1.88 (m, 2H), 1.82–1.77 (m, 16H), 1.56–1.51 (m, 2H), 1.40–1.25 (m, 16H), 1.04 (t, J = 7.3 Hz, 3H), 0.88 (t, J = 6.7 Hz, 6H). MALDI TOF-MS: m/z calcd for [M(C₅₆H₆₄IN₃O₄S)]⁺, 1001.37; found 1001.76. IR (KBr, cm⁻¹): 2953, 2926, 1738, 1601, 1493, 1393, 1340, 1265, 1167, 1081, 1038. Anal.Calcd. for C₅₆H₆₄IN₃O₄S: C, 67.12; H, 6.44; N, 4.19 %. Found: C, 67.49; H, 6.77; N, 4.08 %.

2. DFT Calculations of TSQa-b

The ground-state geometries of **TSQa-b** were optimized in the gas phase by DFT calculations with the Gaussian 09 program (**Fig. 5**).¹ The calculations were performed on the models in which alkyl substituents have been replaced by methyl groups with the B3LYP exchange-correlation functional under a 6-31G basis set.

 Gaussian 09, Revision A.1, Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, Jr., J. A.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, N. J.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, Ö.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian, Inc., Wallingford CT, 2009.

3. ¹H-NMR Spectra











